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ABSTRACT

A FIELD STUDY COUPLING SOIL FRACTIONATION AND SONIC ENERGY FOR ENHANCING THE *IN SITU* REMOVAL OF VOLATILE ORGANIC COMPOUNDS IN THE VADOSE ZONE

by Hassan Kaleem

Remediation of sites contaminated with hazardous wastes could be an expensive endeavor. There is, therefore, the need to explore techniques, which can reduce the remediation time and achieve regulatory specifications, thus reducing the cost involved in a site remediation exercise.

In this work, we investigated the use of sonic energy to enhance the *in situ* removal rate of trichloroethylene and dichloroethylene from a site in Hillsborough Township, New Jersey. The experiments were performed with and without sonic energy and each time the concentration of the trichloroethylene swept out from the site and the flowrate of the effluent gas were measured. The results obtained indicate that when sonic energy is used as an enhancement technique the removal rate of trichloroethylene increases by an average value of about 37.9 % and the concentration of trichloroethylene in the effluent stream increases by an average value of about 20.8 %. These results mean that sonic energy, when used as an enhancement technique, will reduce the remediation time and can help achieve regulatory specifications in a site clean-up exercise after coventional Vapor Extraction methods have reached assymptotic values.

It is recommended that further work be done to find the attenuation coefficients of the sonic field and also to determine the decay rate of the sonic intensity at this site.

A FIELD STUDY COUPLING SOIL FRACTIONATION AND SONIC ENERGY FOR ENHANCING THE *IN SITU* REMOVAL OF VOLATILE ORGANIC COMPOUNDS IN THE VADOSE ZONE

by Hassan Kaleem

A Masters Thesis Submitted to the Faculty of New Jersey Institute of Technology in Partial Fulfillment of the Requirements for the Degree of Master of Science in Chemical Engineering

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A FIELD STUDY COUPLING SOIL FRACTIONATION AND SONIC ENERGY FOR ENHANCING THE *IN SITU* REMOVAL OF VOLATILE ORGANIC COMPOUNDS IN THE VADOSE ZONE

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To my beloved family

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LIST OF SYMBOLS

- a = Constant
- A = Area for $Drying(ft^2 \text{ or } m^2)$
- b = Constant
- k = Removal Rate Constant (Hour⁻¹, Minute⁻¹ or Second⁻¹)
- M = Mass of Solids (lb or kg)
- R = Rate of Drying (lb/ft^2 -h or kg/m²-h)
- R_c = Rate at the First Critical Point in lb/ft²-h or kg/m²-h
- t = Drying Time (Hours, Minutes or Seconds)
- X = Free Moisture Content of Soil in Mass of Water per Unit Mass of Dry Soild (lb H₂O per lb Dry Solid or kg H₂O per kg Dry Soild)
- X_c = Free Moisture Content at the First Critical Point in Mass of Water per Unit Mass of Dry Soild (lb H₂O per lb Dry Solid or kg H₂O per kg Dry Soild)
- X_0 = Free Moisture Content of Soil in Mass of Water per Unit Mass of Dry Solid at Time t = 0 (lb H₂O per lb Dry Solid or kg H₂O per kg Dry Soild)

CHAPTER 1

INTRODUCTION

1.1 Overview

Soil contamination due to the dumping of hazardous waste is a major problem in the United States where an estimated 33000 to 50000 abandoned hazardous waste sites exist (Bloom, 1986). "The term hazardous wastes means waste or combination of wastes which because of the quantity, concentration, or physical, chemical, or infectious characteristics may;

• Cause, or significantly contribute to an increase in mortality or an increase in serious irreversible, or incapacitating reversible illness; or

• Pose a substantial present or potential hazard to human health or the environment when improperly treated stored, transported, or disposed of, or otherwise managed", (U.S Environmental Protection Agency, State Decision Makers' Guide for Hazardous Waste Management, U.S EPA OSW SW 412, Washington D.C 1977).

Today, there are sites around the world where unknown quantities of unknown wastes pose a potential environmental as well as a health problem. Table 1.1 gives some statistics of the quantities of hazardous wastes produced in some industrialized regions and Table 1.2 gives sources of hazardous wastes in the United States.

Location	Hazardous wastes (10 ⁶ tons / yr.)	Total population (millions)
United states	40.0	220
California	4.6	22
Ohio	5.0	12
Canada	4.0	25
Ontario	1.5	8
West Germany	3.5	62
Bavaria	0.5	11
Netherlands	1.2	13

Table 1.1 Quantities of Hazardous Wastes Produced in Industrialized Regions *.

*source: Adapted from Henry J.G., Leachate from Hazardous Waste Landfills, Toronto: University of Toronto, Solid and Hazardous waste Publication No.WM82-02,1982.

Table 1.2 Hazardous waste sources in the United States*

Industry	Basis	
	Dry	Wet
Primary Metals	40%	29%
Inorganic Chemicals	20%	12%
Organic Chemicals	20%	24%
Electroplating	10%	18%
other industries	10%	17%
Total	100%	100%

*source: U.S EPA, State Decision-Makers' Guide for Hazardous Waste Management, US EPA OSW SW 412, Washington DC, 1977. The complete EPA treaty on hazardous waste can be found in the May 19, 1980, issue of the Federal Register, vol. 45, No. 98 (Book 2 of 3), pages 33119 to 33137. The following is a listing of the substances considered as hazardous in that document.

- "Spent halogenated solvents used for degreasing, such as trichloroethylene, methylene chloride, and others.
- Spent nonhalogenated solvents such as xylene, acetone, ethyl benzene, ethyl ether, and others.
- Wastewater treatment sludge from electroplating operations.
- Dewatered air pollution control scrubber sludge from coke ovens and blast furnaces.
- Sludge generated during the production of various chromium compounds.
- API separator sludge from petroleum refineries.

Examples of wastes exempted from the hazardous waste lists are domestic sewage; Irrigation return flows; mine tailings; animal manure; fly ash and bottom ash; drilling fluids; and wastes from crude oil and natural gas. Also excluded from this list are radioactive and nuclear wastes, which, because of their special requirements, are controlled separately under the Atomic Energy Act of 1954 and amendments thereto", (U.S Environmental Protection Agency, State Decision Makers' Guide for Hazardous Waste Management, U.S EPA OSW SW 412, Washington D.C 1977).

Chapter 2 discusses some of the approaches to hazardous waste management and the options available for treatment and disposal of hazardous industrial wastes. Some of the existing methods for cleaning up sites contaminated with hazardous wastes are time consuming and expensive. The technology that is presented in this thesis is aimed at reducing the remediation time and achieving the regulatory specifications and thus reducing the cumulative cost involved in a hazardous waste site cleanup exercise. Pneumatic Fracturing and air injection have been successfully used to clean up a site in Highland Park, New Jersey. This site was contaminated with trichoroethylene and the remediation time was reduced from 10 to 2 years, (Fernandez, 1997). This improvement in remediation time justifies further investigation of processes or techniques that will enhance the currently used Pneumatic Fracturing followed by Vapor Extraction method. The research presented in this document discusses an enhancement technique for the conventional Pneumatic Fracturing coupled with sonic energy and Vapor Extraction to decontaminate a soil containing volatile organic compounds in the vadose zone.

1.2 Research Objective

The objective of the research presented in this document is to test the benefits of coupling sonic energy enhanced mass transfer with soil fracturing to remove trichloroethylene and dichloroethylene at the Derelco industrial site in Hillsborough, New Jersey. This entire field project is based on earlier laboratory work which showed that, overall, sonic energy enhances the removal of volatile organic contaminants, specifically ethanol, as well as water from a tank simulating the fractured vadose zone, (Fernandez, 1997).



CHAPTER 2

BACKGROUND OF THE STUDY

2.1 Overview of Current Remediation Technologies

Remediation technologies could be grouped into two major categories.

- 1. ex situ (involving excavation and transportation of the contaminated soil) and
- 2. *in situ* (involving treating the contaminated soil where it is found)

2.1.1 Excavation and Transportation of Soil

These methods involve *ex situ* treatment of the contaminated soil. The soil is dug up, transported to a remediation facility, treated to remove the volatile organic compounds and returned to its original site. Due to the excavation and transportation of soil involved for treatment *ex situ* methods tend to have a higher cost compared to the *in situ* methods. Some of the *ex situ* methods include;

2.1.1.1 Thermal Treatment: This method involves *ex situ* high-temperature (1200°C or more) treatment to dispose of organic wastes. A common example where this technique is used is in incinerators. In this method, the wastes to be incinerated are stored in secured containers. The usual vessels of storage for the waste depends on the type of waste being stored. Solid waste and thick sludge are stored in pits and liquid and semisolid wastes are stored in mixed or heated tanks. Drummed hazardous organics and organic solids and sludge are incinerated together with the drums when necessary. The heat content of the gases produced during the incineration can be utilized by passing the gases through a heat boiler to recover the heat and to use the recovered heat as required.

The gases are also neutralized and desulphurized by removing the sulphur in the form of heavy metal precipitates. Scrubbing and electrostatic precipitation are used to further purify the incinerator exhaust gases. The clean non corrosive gases can then used to turn a turbine in order to generate electricity, (Henry G.J and Heinke, G.W, 1989). A major disadvantage in the use of incinerators is the cost involved in building an industrial scale incinerator.

2.1.1.2 Secure Landfill: This method is an *ex situ* treatment technique in which it is intended that a secure landfill holds the organic and inorganic hazardous wastes in as concentrated a form as possible for an indefinite period. If it is necessary, Leachate is removed periodically for treatment and disposal.

2.1.1.3 Codisposal: This method is an *ex situ* treatment technique which involves the codisposal of hazardous waste with municipal refuse, it is aimed at using large quantities of refuse to absorb relatively small quantities of hazardous inorganic liquid wastes (and some organics) so that the contaminants can be decontaminated by the refuse and the surrounding soil, (Henry J.G, Heinke G. W, 1989).

2.1.1.4 Hazardous Waste Treatment Facility: This method is an *ex situ* treatment technique. In this method, the hazardous substances are transported to a hazardous waste treatment facility. In a hazardous waste treatment facility, organic wastes may be either incinerated or treated to yield a liquid effluent that is non-toxic and a concentrated sludge which is then sent to a secure landfill. Inorganic wastes are treated to make them non-

toxic, neutralized, and concentrated to produce a sludge which upon further treatment can be sent to a landfill. Chemical and Physical treatment methods, incineration and landfillng of the residues generated from these methods are the most common types of hazardous waste treatment processes, (Henry J.G, Heinke G. W, 1989). The high cost involved and the high demand for land makes *ex situ* techniques unattractive. Other methods used in conjunction with digging up are chemical treatment of the soil and washing of the soil.

2.1.2 In-Situ Removal of Contaminants

These methods involve the *in situ* treatment of the contaminated soil. Due to the *in situ* (local) treatment of soil involved, *in situ* methods tend to have a lower cost compared to the *ex situ* methods. Some of the *in situ* methods include;

2.1.2.1 Washing: This technique is a common *in situ* treatment method. Washing is most efficient in permeable soils contaminated with solid and liquid waste. This process involves treating the site with a solution that is able to dissolve the contaminants in the soil. The solvent travels through the pores of the permeable soil, thereby reaching the contaminants and dissolving them. The solvent finally reaches the ground water where it is pumped to the surface through extraction wells, treated and redistributed on the site again. The major advantages of this technology are the ability to remediate the site permanently and the moderate cost involved in implementing the technology. The major disadvantage of this technique is the fact that it can only be utilized in highly permeable soils.

2.1.2.2 Chemical Treatment: Most hazardous waste treatment processes include chemical treatment as an essential component. The commonly used chemical treatment processes are oxidation and reduction reactions, pH adjustments and ionic exchange. Ionic exchange involves separating all dissolved inorganics from the waste. Chemical fixation is also a common chemical treatment method for hazardous waste. It involves solidifying the wastes (both organic and inorganic) with cement. This Method tends to stabilize the inorganics into inert silicates or hydroxides. The application of this method to organic wastes is still under development. If the end products of this method are not confirmed to be completely stabilzed, a secure land fill is recommended, (Henry J.G, Heinke G. W, 1989).

2.1.2.3 Bioremediation: For low concentrations of toxic substances, biological methods of treatment can be employed. These methods involve using microorganisms to decompose the waste. The process can be carried out using aerobic or anaerobic microorganisms. In general, due to the fact that aerobic processes lead to higher growth of microorganisms they are less sensitive to the toxicity of the waste being decomposed. Typical examples of wastes which can be decomposed by biological processes include; phenols, oils and other refinery wastes, (Henry J.G, Heinke G. W, 1989).

2.1.2.4 Ultrasonics: Remediation of subsurface soils using ultrasound is currently under development as an *in situ* technology and has been discussed. Some of the applications of Ultrasound include:

- The enhancement of chemical treatment processes during the decontamination of soils.
- The elimination of microbes using ultrasonic energy. These microbes could decrease the permeability of the soil by clogging the pore spaces of the soil.
- The improvement in the permeability of tightly packed clays. This can be achieved by using the ultrasound to disperse the tightly packed clay formation.

Ultrasound is used as an enhancement technique for other methods of treatment. The ultrasound is used to increase the permeability of the soil so that other remediation agents (such as chemicals and microorganisms) can have easy passage through the soil being remediated. This leads to a faster rate of remediation because there is more contact between the remediation agent and the hazardous waste in the soil, (Fernandez, 1997).

There are other *in situ* techniques being used to treat soil contaminated with hazardous waste. Pneumatic Fracturing followed by Vapor Extraction has proven to be an effective technique. This thesis looks at an enhancement technique for the conventional Pneumatic Fracturing-Vapor Extraction method. It will involve coupling Sonic Energy, Pneumatic Fracturing and Vapor Extraction to remediate a site contaminated with trichloroethylene and dichloroethylene, in Hillsborough Township, New Jersey.

2.2 Pneumatic Fracturing

The Hazardous Substance Management Research Center (HSMRC, NJIT) has developed a new method for the remediation of tightly packed soils. This method known as Pneumatic Fracturing is aimed at increasing the permeability of tightly packed soils. The method was tested on a site in Somerville Township, New Jersey, and was proven to be effective in increasing the permeability of the soil, (EPA/540/AR-93/509, July 1993). This study showed that the method of Pneumatic Fracturing increases the permeability of tight soils. This is believed to be achieved by creating more fractures or widening the existing fractures. Another finding from the study was that the removal rate of the major contaminant in the soil (trichloroethylene) was also observed to increase. This result is believed to be a result of the releasing of the organics which were trapped within the formation prior to the Pneumatic Fracturing.

Furthermore, both the observed air flowrates and the removal rate of the organic contaminant were observed to increase significantly. The concentration of the organic contaminant removed was also observed to increase. This is believed to have been caused by the release of some trapped organics which were held within the fractures prior to the Pneumatic Fracturing. Finally, the study found that the nature of the site geology and the presence of built structures (e.g pipe lines) could have a profound impact on the nature of the fracturing, (Accutech Pneumatic Fracturing Extraction and Hot Gas Injection Phase1, Applications Report by EPA/540/AR-93/509, July 1993).

2.3 Vapor Extraction

Vapor Extraction is a very effective method during *in situ* site remediation. A major limitation to this technique however, is the permeability of the vadose zone. If the vadose zone is not sufficiently permeable, the organic contaminant will not be swept away with the air stream due to low air flow rates. Currently, there are different ways used to implement Vapor Extraction. One way is to draw the vacuum from the central well and inject air into the neighboring wells or leave the neighboring wells open to the

atmosphere. The neighboring wells could also be sealed. Another method is to inject air into the central well and draw the vacuum from the surrounding wells, (Applications Report by EPA/540/AR-93/509, July 1993).

2.4 Coupling Pneumatic Fracturing and Vapor Extraction

Vapor Extraction is an efficient approach for organic contaminant removal from soils that are sufficiently permeable and allow for easy passage of air through the pores of the formation. For soils that are not sufficiently permeable, Pneumatic Fracturing can be used to open up the fractures within the soil and thus increase the permeability of the soil. This will then lower the resistance within the fractures in the soil and lead to higher airflow through the soil. Examples of such soils include shale, silts and clays. When Vapor Extraction is then performed on such soils, there is a higher rate of removal of organic contaminant because the higher airflow through the fractures causes more of the organic contaminants to be swept away by the air passing through the pores of the soil. Pneumatic Fracturing usually involves injecting air intermittently (air at about 500 psig < 1 minute) into the soil, (Accutech Pneumatic Fracturing Extraction and Hot Gas Injection Phase1, Applications Report by EPA/540/AR-93/509, July 1993).

2.4.1 Theoretical Considerations

When air is injected into a formation it will travel through the path that offers the minimum resistance, which in this case is the fracture. As air passes through the fracture, the moisture in the fracture will evaporate into the air stream. Due to concentration gradients and pressure differences, and depending on the velocity of the air stream, the

moisture will be carried along with the air stream. Thus the air flowing through the fracture is drying the ground.

This real situation can be modeled after a tray dryer, in which a continuous stream of air is allowed to pass over a series of wet solids, in trays, thereby causing the moisture in the solids to be swept away by the air stream. Some limitations to applying this model to the actual fracture are summarized below:

- 1. Some of the air going into the fracture may be dispersed into the formation whereas in the tray dryer all the inlet air comes out.
- 2. The real fracture will not have a uniform dimension throughout its length and this could lead to variations in the airflow rate in the fracture (some parts of the fracture could be clogged by debris).

2.4.1.1 Drying of Porous Solids and Capillarity: Porous substances are made up of a network of pores which are connected to each other in a complicated manner. As a porous substance is being dried, moisture flows through the pores of the material by capillarity to the surface of the material. Figure 2.1 gives the moisture distribution in a porous material as it is being dried, (W.L McCabe, J.C Smith and P. Harriott, 1993).

Figure 2.1 shows that as the distance from the surface of the porous material being dried increases the moisture content of the material also increases and then slightly levels off for large distances from the surface of the material. This graph shows that in a typical Vapor Extraction process, in a porous soil, the surface moisture or organic contaminant will be the first to be swept away. The Figure also shows that the further the

moisture or organic contaminant is located from the surface, the less likely it is to be removed by the conventional Vapor Extraction techniques.



Figure 2.1 Moisture Distribution in Porous Slab being dried* *Adapted from (W.L McCabe, J.C Smith and P. Harriott, 1993).

Figure 2.2 shows the drying rate curve for a porous ceramic plate. The rate of drying of porous materials can be divided into three sections, (W.L McCabe, J.C Smith and P. Harriott, 1993). The three sections are shown as section AB, BC and CD in Figure 2.2.

2.4.1.1.1 Section AB: This is the initial rate of drying period. During this period the entire surface of the material is covered with moisture. The rate of removal is constant for this period. For this period the following equations are used to find the moisture content present in the soil at any time.



Figure 2.2 Drying Rate Curve for Porous Ceramic Plate Adapted from (W.L McCabe, J.C Smith and P. Harriott, 1993).

$$R = -(M/A)(dX/dt)$$
 2.1

For the constant rate period, segment AB, Figure 2.2, the solution of the above equation is given by;

$$X = X_0 - (R A/M) * t$$
 2.2

(adapted from W.L McCabe, J.C Smith and P. Harriott, 1993).

Equation 2.2 gives the free moisture content of the soil over time. These equations are used in calculating the moisture content of the soil.

The equations can be generalized to account for volatile organic compounds during the constant rate of removal period. For the purposes of this study the volatile organic contaminant content of the air stream above the organic contaminant in the soil (X) will be measured. The moisture lost from the soil is equal to the moisture gained by the air stream, thus moisture content swept out by the air stream until time t is proportional to $X_o - X$.

$$X_0 - X = (R A/M) * t$$
 2.3

For the falling rate period, segment BC and CD, Figure 2.2, the rate of drying of the porous soil is no longer constant. This falling rate period begins when the surface of the soil starts to dry up. The first part of this falling rate period, segment BC, Figure 2.2, corresponds to the case where the surface is partially covered by moisture and there still some areas of moisture on the surface. The mechanism of evaporation remains unchanged but the rate of removal of the moisture decreases linearly with time. This trend is followed until the surface no longer contains free moisture and the moisture content layer of the soil begins to receed to the interior of the soil. During this period the rate of removal of the moisture from the soil follows a parabolic shape (CD).

For the first falling rate period, segment BC, Figure 2.2, the rate is linear with X for most porous solids. Thus

$$\mathbf{R} = \mathbf{aX} + \mathbf{b} \tag{2.4}$$

In some situations, a straight line representing the falling rate period passes through the origin. For these cases

$$\mathbf{R} = \mathbf{a}\mathbf{X}$$
 2.5

For simplification purposes if Equation 2.5 is assumed, Equation 2.1 can be integrated.

or

$$\mathbf{X} = \mathbf{X}_{c} \, \mathbf{e}^{-\mathrm{MAt/a}} \qquad 2.6$$

$$\mathbf{X} = \mathbf{X}_{c} \, \mathbf{e}^{-\mathbf{k}\mathbf{t}} \tag{2.7}$$

Equation 2.7 indicates an exponential decay of the moisture content with time in the falling Rate region. The solution of Equation 2.1 with R = aX + b is given by

-Aat/Mb=
$$(Ln R/R_c) * X$$
 2.8

$$a = (R_c - R)/(X_c - X)$$
 2.9
From Equation 2.8 and 2.9

$$Xc-X = -A[R_c-R]t/M Ln(R/R_c)$$
 2.10

Equation 2.10 has the same form as Equation 2.3 with the constant rate term in Equation 2.3 replaced by the falling rate term.

2.5 Overview of Acoustics

2.5.1 Acoustic Properties of Soil

Properties of sound in soil depend very much on the nature of the soil. Properties like the speed of sound, amplitude of sound waves and attenuation of sound in soil depend on the porosity of the soil, moisture content of the soil, type of minerals present in the soil and the state of consolidation of the soil. It has been shown that in clay sound waves travel faster and with little change in velocity and also with some attenuation, (Fernandez, 1997).

2.5.2 Sonic Drying

Sonic energy increases the drying rate of solids exposed to a sound field but the extent of the improvement varies considerably with the material. Further research has led to the conclusion that sonic drying is most efficient at sound frequencies between 7kHz and 20kHz and sound intensities greater than 145Db, (Fernandez, 1997).

2.5.3 Sonic Generators

Fernandez, 1997, investigated the different types of sound generators and gives the main classes of sonic generators as;

- 1. Electrostatic Generators
- 2. Electrodynamic Generators
- 3. Magnetostrictive Generators
- 4. Piezoelectric Generators

5. Pneumatic Generators (Dynamic Generators and Static Generators)

The principle governing the operation of Electrostatic Generators is that electrostatic forces between the plates of a condenser can change the spacing between the plates .The attraction and repulsion between these plates creates sound vibrations in the air that is near the plates, (Fernandez, 1997).

The principle that governs the operation of Electrodynamic Generators is that an electric current when passed through a coil will generate a magnetic field which causes vibrations in a magnetic plate. These vibrations produce resonance and hence enhance the sound being generated, (Fernandez, 1997).

Magnetostrictive Generators use a voltage applied to a metallic material. The material expands when the voltage is applied to it and contracts to its normal shape when the voltage is removed. Thus, when an alternating voltage is applied to the material a series of vibrations are produced. These vibrations produce an acoustic field, (Fernandez, 1997).

The principle that governs the operation of Piezoelectric Generators is based on the fact that a crystal with piezoelectric properties will build a charge when brought into contact with a voltage. This charged crystal will be attracted to other oppositely charged crystals. When the voltage is reversed the charge on the crystal is also reversed. A series of alternating charges causes the crystal to vibrate. If the vibration of the crystal coincides with its resonance frequency a sound field is generated, (Fernandez, 1997).

Pneumatic Generators produce sound waves using air. Pneumatic Generators can be divided into two subclasses, Static Generators and Dynamic Generators.

Dynamic Generators operate on the principle that when a rotating device is allowed to interrupt a jet of air intermittently, a sound wave is generated. An example of a Dynamic Generator is a Siren, (Fernandez, 1997). A major disadvantage of the siren is its susceptibility to mechanical damage due to its moving parts at high revolutions per second.

The principle governing Static Generators is that a jet of air emerging from a converging nozzle, close to the speed of sound, causes waves to form at the tip of the nozzle. When a resonant cavity is placed in the path of the air jet, a sound wave is produced, (Fernandez, 1997). Whistles are examples of Static Generators. Whistles have a major advantage over sirens in that whistles have no moving parts.

For this research, a whistle will be used as the sonic generator (refer to Chapter 4 for description and specification). This selection is based on previous laboratory work (Fernandez, 1997) and the fact that the whistle is more mechanically resistant compared to the siren.

2.6 Coupling Pneumatic Fracturing, Sonic Energy Enhanced Mass Transfer and Vapor Extraction

This method, the subject of this investigation, will be an enhancement of the technique presented in Section 2.4. Previous laboratory studies have revealed that this technique could lead to promising results (Fernandez, 1997). These studies determined that

focusing sonic energy into a fracture can lead to an enhancement in the removal rate of liquids that are trapped inside tightly packed soils. Laboratory studies performed on drying of solids in the presence of a sonic field show that several effects are obtained when a sonic intensity of about 160Db is allowed to pass through a fracture, (Fernandez, 1997).

The first effect is the lowering of the net total pressure within the fracture by the compression and dilation of the air within the fracture caused by the sonic energy. These compressions and dilations tend to lower the net total pressure within the fractures because it is known that dilation regions dominate over compression regions. Thus the lower net total pressure causes some of the liquid in the fractures to vaporize and be carried away by the air stream. In addition the lower net total pressure within the fracture causes a pressure gradient to develop which acts as a driving force for liquid to move towards the fractures, (Fernandez, 1997).

Another effect when a sound field of 160 Db is focussed into a fracture is the lowering of the gas – liquid interface film. This is achieved due to the higher turbulence that is built up in the region because of the presence of the sonic field. This decrease in the interface film increases both mass and heat transfer and causes more liquid to be evaporated, (Fernandez, 1997).

Furthermore, moisture that is trapped within the fractures and separated by air bubbles will be released because the sonic energy causes the bubbles to heat and expand. The pressure gradient is in the direction of the fracture and hence, the capillary contents move toward the fracture zone, (Fernandez, 1997). A last effect that can be obtained when an intense sonic field of 160Db is focussed into a fracture is Cavitaion. This is caused by the intense sound field. It is believed that the effect of Cavitaion may have a positive influence on the removal rate of the contaminant, (Fernandez, 1997).

2.7 Trichloroethylene (TCE) and Dichloroethylene (DCE)

2.7.1 Dichloroethylene (DCE)

Dichloroethylene (CHClCHCl) has two stereo isomeric forms. It is not highly flammable, but, the hot gases containing dichloroethylene can be ignited and continue to burn so long as heat is supplied. It is colorless and boils at a low temperature. It is a slightly toxic liquid with a pleasant odor. Some of its uses are as a solvent for oils, gums, resins, waxes, rubber and cellulose acetate. It is also used in the extraction of dyes, perfumes and organic material. The percentages of the different isomers present determines the physical properties of this mixture. The lethal dose of DCE (based on a 30 minutes test on cats) is 44000 PPM, (Ibert Mellan, 1939).

2.7.2 Trichloroethylene (TCE)

Trichloroethylene is a non-flammable and non-explosive organic compound. It is not combustible and has a very low affinity for common metals. Trichloroethylene is considered to be a toxic liquid with a low boiling point. It is also a colorless, mobile, heavy liquid with a smell similar to that of chloroform. Trichloroethylene (CHClCCl₂) belongs to a group of compounds known as chlorinated solvents. Members of this group include (Carbon tetrachloride, pentachloroethane, tetrachloroethane, perchloroethylene, dichloroethylene and methyl chloroform). It is the second most toxic member of the group after trichloromethane (Table 2.5). During World War II, it was used extensively in the degreasing of metal parts. Presently, it is widely used as a solvent for gum, crude rubber, dyes, bitumen, fats, waxes and grease. It is miscible with acetone, benzene, toluene, ethyl and methyl acetates and alcohol. Trichloroethylene is also miscible with most organic solvents. Most of the trichloroethylene produced is used for degreasing metal parts (87 %). Major factors that limit its production are air pollution legislation and OSHA restrictions.

Table 2.1 - 2.5 gives some physical properties of Trichloroethylene, (Ibert Mellan, 1939).

Temperature/°C	Specific Gravity (referred to water at 4°C)
0	1.4996
15	1.4762
30	1.4514
45	1.4262
59.5	1.3997

 Table 2.1 Specific Gravity of Trichloroethylene at Different Temperatures

source: Industrial Solvents, Ibert Mellan, 1939.

Temperature/°C	Solubility/
	g. water per 100g. TCE
-2	0.01
10	0.017
18	0.025
28	0.035
70	0.09

 Table 2.2 Solubility of Water in Trichloroethylene, at Different Temperatures

source: Industrial Solvents, Ibert Mellan, 1939.

Table 2.3 Viscosity of Trichloroethylene at Different Temperatu	ires
---	------

Temperature/°C	Viscosity/ centipoise
-10	0.78
25	0.550
50	0.446
55	0.440
75	0.371

source: Industrial Solvents, Ibert Mellan, 1939.

Temperature/°C	Vapor Pressure/mm Hg
-10	10
10	33
20	60
30	90
50	210
70	450
80	630

 Table 2.4 Vapor Pressure of Trichloroethylene at Different Temperatures

source: Industrial Solvents, Ibert Mellan, 1939.

 Table 2.5 Relative Toxicity*

Carbon Tetrachloride (CCl ₄)	1.0
Perchloroethylene (C_2Cl_4)	1.6
Dichloroethylene (C ₂ H ₂ Cl ₂)	1.7
Trichloroethylene (C_2HCl_3)	1.7
Trichloromethane (CHCl ₃)	2.2

source:Industrial Solvents, Ibert Mellan, 1939.* CCl₄ arbitrarily taken as1

Sec. 1

Other physical properties: (Source: Ibert Mellan, 1939).

Molecular weight		131.40
Melting Point		73°C
Boiling Point		87.2°C
Weight per gallon		12.20 lb. (20°C)
Color		Water-white
Threshold limit valu	e	100 PPM
Vapor density		4.45 (air = 1)
Refractive index		1.4735 (27°C)
Surface Tension		32.0 dynes/cm (25°C)
Specific Heat		0.229 cal./gm (23°C)
Solubility in water		0.1% by wt. (25°C)
Latent heat of vapor	ization	57.3 cal./g. at BP)
Coefficient of expan	sion (per °C)	0.00115 to 20°C
Dielectric constant		3.42
Lethal dose (in 30 m	inutes for cats)	37000 PPM

CHAPTER 3

SITE DESCRIPTION

3.1 Background

The actual field studies will be carried out at the Derelco site, at Hillsborough, N.J, contaminated with TCE and DCE. The study will involve sending a sound wave (generated by a whistle) through an air injection well, at points where a fracture connects the air injection well with the extraction well. The whistle has a major advantage because it has no moving parts. The aim of this field study is to test the benefits of coupling Pneumatic Fracturing, Sonic Energy Enhanced Mass Transfer and Vapor Extraction. McLaren/Hart teamed up with our research group at the New Jersey Institute of Technology to perform a field test using sonic energy enhancement coupled with soil fracturing for soil remediation at the Hillsborough site, NJ.

The site had been previously investigated by McLaren/Hart Environmental Engineering. On October 26 1985, a fire at the site destroyed a National Diagnostics Inc. building, located on the site, causing the release of organic contaminants into the site. The major contaminants that were involved are trichloroethylene and dichloroethylene.

The site is located near the Sommerville traffic circle, on route 206 south. It is a flat partly paved site and generally slopes slightly from the Northwest towards the Southwestern direction. The surrounding area is a light industrial area and a tributary of the Royce Brook River runs eastward by the northern border of the site. Also located close to the site, about 100 feet, are a few medium sized building structures, (Boland etal, Ultrasonic Field Demonstration Work Plan, NJIT, 1998).

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3.2 Site Geology

The geology of the Derelco Site is described by Mclaren/Hart as "underlain by a thin veneer of unconsolidated sediments overlaying shale and siltstone bedrock." The deposits range in thickness from one to three feet. These deposits are believed to derive from the local bedrock and consist primarily of a heterogeneous mixture of silt and clay.

Rock samples by McLaren/Hart in September of 1990 revealed "a reddish brown siltstone with interbedded shale layers" and bedrock of "fair quality" with moderate fracture spacing: 30 centimeters to 1 meter apart. This study by McLaren/Hart also revealed three highly fractured zones within the bedrock. These zones appeared at a depth of 18 feet and at the intervals of 33 to 35 and 64 to 66 feet. Smaller fractures were also encounted at 29 feet, 40 feet, 55 feet, and 75 feet. Fractures are oriented both vertically and horizontally. Horizontal fractures occurred along bedding planes that dip five to ten degrees to the west. The vertical fractures are planar and parallel to the strike of the formation which, run Northwest to Southwest, (Boland etal, Ultrasonic Field Demonstration Work Plan, NJIT, 1998).

3.3 Site Hydrology

Previous investigations of the Derelco Site by McLaren/Hart Environmental Engineering, revealed that the ground water circulation at the site occurred at a depth of about thirty feet below the ground level and the circulation is limited to the fractures located in this region. The studies also showed that the ground water circulated towards the Northeastern direction, (Boland etal, Ultrasonic Field Demonstration Work Plan, NJIT, 1998).

3.4 Well Layout

The locations of the wells within the site are depicted in Figure 3.1. Measured data for the wells at the Derelco site, Hillsborough Township, N.J, prior to the field studies, are given in Table 3.1 and Table 3.2.

Well Number	X(ft)	Y(ft)	Depth(ft)	Conc. of TCE (PPMv)	Bore-hole size (inch)	Well Depth (ft)
1	53.3	25.6	26.80	1.8		85.00
2	47.9	19.7	7.00	30.0		19.10
3	58.1	20.0	6.90	13.2		17.55
4	36.5	19.2	7.90	59.5	4.0	21.65
5	30.5	31.1	8.42	13.0		21.22
6	23.9	30.1	8.49	6.8		24.27
7	24.7	22.3	8.32	7.6	4.0	22.16
8 fw	29.5	21.0	8.30	11.5	3.5	20.50
9	10.6	26.5	9.04	14.3		22.90
10	19.9	23.6	8.90	3.6		22.02
11	23.9	16.3	8.40	7.5	4.0	22.04
12	28.2	15.3	8.32	7.9		15.10
13	33.2	14.2	8.32	13.8	4.0	22.08
14	27.9	10.4	7.90	8.7		24.86
15	25.9	0.0	7.60	8.5		20.76

Table 3.1 Measured Data, Derelco Site Hillsborough Township, New Jersey

Source: (Boland etal, Ultrasonic Field Demonstration Work Plan, NJIT, 1998).

Well Number	TCE, PPMv	DCE, PPMv
1		
2		
3		
4	23.14	4.66
5	19.98	2.32
6		
7	0.02	0.01
8	4.60	2.90
9	1.11	0.70
10	0.02	0.011
11	0.0	0.52
12	0.29	0.17
13	2.14	1.38
14	0.59	0.61
15		
15		

Table 3.2 Well Water Sample Analysis, Derelco Site Hillsborough Township, NJ.

Source: (Boland etal, Ultrasonic Field Demonstration Work Plan, NJIT, 1998).



Figure 3.1 Location of Wells at the Derelco Site, Hillsborough Township, New Jersey Source: (Boland etal, Ultrasonic Field Demonstration Work Plan, NJIT, 1998)

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CHAPTER 4

UTILITIES AND EQUIPMENT SPECIFICATION

The utilities and equipment specification for the ultrasonic field test are divided into two main categories.

- The injection system utilities and equipment specification
- The extraction system utilities and equipment specification

4.1 Injection System - Equipment Specification

The major components of the injection system are shown in Figures C.2, C.3, C.4a and C.4b (Appendix C). Other parts of the injection system are described below:

4.1.1 Compressor

The compressor serves the purpose of providing the compressed air needed for the sonic experiments. The compressor used for the experiments was a rented, medium sized, Smith brand, 100 cubic feet per minute compressor capable of providing the desired flowrates (20- 40 SCFM). During operation, the compressor runs at 120- 160 °F range and 95-115 Psi range. There are gauges to monitor the operating Temperature, Pressure, Battery Charge and Fuel Level. There is easy accessibility to the parts of the compressor and adjustments can be made to the compressor during varied process conditions. During cold morning operation the choke is used and the compressor is allowed to warm up for about twenty minutes until the steady operating temperature (160 °F) is attained. During warm up, the discharge valve is closed and during operation it is opened towards the outward direction to an angle of 45°. By means of bleed valves, located underneath the

compressor, all condensation can be bled out of the compressor at specific intervals of time. A valve which can be regulated to attain the desired outlet airflow rate controls the discharge of the compressor.

4.1.2 Flow Manifold

The flow manifold arrangement, for the sonic field-test, is shown in Figure C.1. The flow manifold is mounted on a rectangular board made from wood. The board is supported vertically by four stands that stand out at an angle of 30° from the surface of the board.

The flow manifold consists of four, 3/4 inch ID flowmeters (2.4 - 24.0) SCFM. Each flowmeter is equipped with a 1/2 inch Sch. 40 Bronze Globe valve at the inlet and a 1/2 inch Sch. 40-IPS Forged-Brass Ball valve at the outlet. By means of these valves, air can be made to travel through the desired flowmeters. Each flowmeter is also equipped with a 1/8 inch ID, 0-100 PSIG, pressure gauge at its outlet. The pressure gauges indicate the pressure drop through the injection system and the pressure drop through the whistles and their feed lines. All four flow meters are connected on a single line, at the inlet, by means of four 1 inch Sch.40 Steel Tees, seven nipples and three tubing unions (1 inch Sch. 40). This single inlet line to the four flowmeters is connected to a compressor. There is also a 0-100PSIG pressure gauge placed on this line. By means of 25 feet of 1/2 inch OD hoses, the outlet of each flowmeter is connected to the well-head assembly (Figure C.2).

4.1.3 Well-head Assembly for Ultrasonic Field Test

The Well-head assembly, for the Ultrasonic Field Test is shown in Figure C.2. The wellhead assembly together with the flow manifold arrangement will be the parts of the Ultrasonic Bore-hole Test Assembly above ground level. The rest of the assembly will be submerged into the fracture well (Number 4).

The well-head assembly consists of a 1.5 inch Sch. 40 Tee made of steel. To the lower end of this Tee are connected three steel pipes (3 feet 1 1/4inch Sch 40) in series by means of a 1 1/4inch Sch. 40 Galvanized nipple 1 inch long. To the upper end of the Tee a pipe adapter is connected which makes it possible to draw in air (from two of the four flowmeters) through a Twin self-extracting Nylon air Hose (25 feet of 1/2 inch OD plastic Hose). To the third end of the Tee, a 1 inch Sch. 40, 45 degree lateral steel Tee is connected by means of a bushing nipple. This 45 degree lateral Tee makes it possible to draw in two extra lines of air (from the two other flowmeters) by means of 25 feet of 1/2 inch OD plastic Hose.

4.1.4 Extension Pipes and Fittings

The extension pipes and fittings are shown in Figure C.3. The extension pipes are 3 feet, 1 1/4inch Sch. 40 made of steel. Within this extension pipe, two lines made of 1/2 inch OD Teflon tubing run from the well head assembly and extend all the way down to the end of the extension pipe where the ultrasonic device (Whistles) is located. The 1/2 inch OD Teflon tubings are connected to the air feed tubes of the two whistles. This makes it possible to send air from the well-head assembly through the extension pipes and fittings and then through the whistles. It is similarly possible to bypass the whistles by sending the air through the 45 degree lateral Tee.

4.1.5 Ultrasonic Transducer (Whistle)

Two unidirectional whistles are the main parts of the ultrasonic system. They were purchased from Applied Ultrasonics, Bethel, Connecticut. Each of the whistles is about 3 inches long and are protected by two aluminum plates (Figure C.4b). These aluminum plates protect the whistle from scraping along the wall of the bore-hole.

The whistles all face a single direction to focus and maximize sonic energy into the fractures. Each of the two whistles is equipped with a 1/2 inch air feed tube. Figure C.4a is a schematic of the ultrasonic device showing the two whistles and the aluminium plates. Figure C.4b is a detailed schematic showing the two whistles and the aluminium plates . Table C.1 shows the Bill of materials needed for the entire injection assembly.

4.1.6 Packers

The sonic system will be coupled with two, four foot packers to isolate and prevent air leakage from the sonic zone. These packers will be placed before and after the sonic generators in the bore-hole to isolate the treatment zone. Figure C.4 depicts the relative positions of the packers and the sonic system. The packers have an external tubing through which an external nitrogen or air source can be used to inflate them as required.

4.2 Extraction System - Equipment Specification

4.2.1 Packers

The extraction unit is basically made up of a packer arrangement (separated by a perforated pipe fitting). This packer arrangement is coupled to three steel Sch. 40 pipes each 4 feet long and 1 1/4 inch in nominal diameter and in series (Figure C.5 and C.6).

4.2.2 Photoionization Detector and Field-GC

The analytical system used for this field study can be divided into two parts. A portable GC was used to determine the type of organic contaminant present in the site and a Photoionization Detector was used for measuring the concentrations over time of the contaminants. The GC was used during the initial stages of the study to find out which were the major contaminants present in the site.

4.2.2.1 Photoionization Detector: The Photoionization Detector (PID) was used to measure the concentration of the contaminant over time. The PID was a Thermo Brand which can be run in battery or in direct current mode. For the purposes of this field study the PID was run on battery mode. This required overnight charging of the battery after each day's run. For the purposes of this field study the PID was calibrated using a standard sample of TCE and it was connected to the extraction line. The PID is equipped with an internal vacuum pump and during operation it continuously draws in a sample from the extraction line, through an inlet port located at its front end, into an analyzing chamber and gives the instantaneous concentrations of the contaminant in parts per million by volume, PPMv. It is capable of displaying up to two analyses every second.

The exhaust gases are continuously emitted via an exhaust port located at the rear end of the PID, to the atmosphere.

4.2.2.2 Field – GC: The GC used for this field study is a Scentograph Plus II, designed by Sentex Systems Inc.. It is a portable gas chromatrograph which is capable of performing the following functions;

- Calibrating a known Sample
- Analysing Samples
- Collecting and Injecting samples into the analyzing chamber
- Separating Compounds
- Detecting Compounds
- Identifying and Integrating Peaks.
- Displaying and storing data of concentrations, retention times and operating conditions.
- Continuos Operation
- Automatic recalibration

The Scentograph Plus II could operate in one of the following two modes.

- 1. Calibration Mode
- 2. Sample Analysis Mode

Calibration Mode:

During the calibration mode, a sample of known concentration is introduced into the analyzing chamber of the GC. A retention time is associated with each sample. An area is then associated with each sample depending on its concentration. The data obtained from

the calibration run are then stored in the GC and the graph of the calibration run is displayed on the screen. This graph serves as the basis for calculating the concentrations of all samples to be analyzed until another calibration run is performed.

Sample Analysis Mode:

When the Scentograph is operating in sample analysis mode, the sample to be analyzed is introduced through the sample analysis ports, which are located behind the GC. The concentration of the sample is calculated by comparing the area under the graph displayed to the area under the graph of the calibration run, at the same retention time. Compounds that are detected and which do not match any retention time obtained during calibration are listed as "unknown". The Scentograph gives very accurate results due to the fact that calibration can be performed as frequently as desired, (Scentograph "plus II" Operating Manual, 1993).

For the purpose of this field study, the calibration gas used is a standard mixture of TCE (12 PPM), cis-DCE (12 PPM) and trans-DCE (15 PPM).

4.2.3 Flow Manifold

The flow manifold for the extraction system is an electronic flowmeter. The electronic flow meter is connected to the extraction line, after the PID. It is capable of displaying instantaneous flowrates in SCFM, electronically. Also connected to the extraction line is a piezometer tube. The piezometer tube is connected after the electronic flow meter. Based on the area of the pipe to which the piezometer is connected and the reading on the piezometer, the flow rate through the extraction line can be determined. With this installation in place, the accuracy of the electronic flow meter was confirmed.

CHAPTER 5

EXPERIMENTAL APPROACH AND RESULTS

The field study that was performed at the Derelco Site encumbers the following;

- 1. Site Dewatering
- 2. Permeability Studies
- 3. Experimental studies

5.1 Site Dewatering

The first activity carried out on the Site was dewatering. The purpose of dewatering was to get the water level below the fracture zone, about 13 feet below ground surface (BGS). This is necessary to keep the fracture zone free of water and to allow free passage of injected air into the fractures.

5.1.1 Method

The dewatering process at the Derelco Site is an ongoing process. The process involves first pumping water from the wells into a drum and from the drum into a larger reservoir (Figure 5.1). The water is then treated and discharged. Pump 1 (Figure 5.1) is a positive displacement pump. Pump 2 (Figure 5.1) is operated with a float level control system. Pump 2 stops when the float is in the horizontal position and starts when the float is above the horizontal position. Depending on the water level in the drum, the float will be either above the horizontal position (drum more than a third filled with water) or below the horizontal position (drum almost empty).





Pump 1 pumps water out of the wells into the drum and pump 2 pumps from the drum into the reservoir. With this system in place, the water level in the wells is about 13 feet BGS. When there is the need for aggressive pumping, the pumps can be switched from automatic operation to manual operation. Thus with this system in place the ground water can be kept at the desired level, which is greater than 13 feet BGS.

At the beginning of a pumping cycle the water level in the borehole rapidly rises as soon as pump 1 is switched off. It usually takes about 5 minutes to rise five feet. However, when aggressive pumping has gone on for about 60 minutes, the water level tends to rise slowly, 2 feet every 15 to 30 minutes. Pump 2 automatically starts when there is enough water in the drum to keep the float, controlling pump 2, above the horizontal inclination. Pump 1 can either be manually turned on to pump water out of the well or can be set to pump water out of the well at specified intervals of time. All the water recovered from the wells is pumped to an on-site water treatment facility from which the water is recirculated into the local drainage system. Some typical values of the water level 8 minutes after a pumping cycle are given below.

Well Number	Water level, feet BGS	
1	23.8	
2	12.8	
3	10.7	
4	17.5	
5	14.5	
6	13.4	
7	10.4	
8	14.6	
9	12.8	
10	13.2	

 Table 5.1 Water Level in Selected Wells after a Pumping Cycle

Well Number	Water level, feet BGS	
11	10.6	
12	Not accessible	
13	13.0	
14	13.2	
15	12.0	

 Table 5.1 Water Level in Selected Wells after a Pumping Cycle (Continued)

The data obtained from the Site dewatering experiments suggest that a period of about one hour of aggressive dewatering is required prior to the start of the experimental runs.

5.2 Field Permeability Studies

The aim of the permeability study was to find out if the resistance in the existing fractures was still low enough to permit easy passage of air. Low permeability will suggest that the soil fracturing process will have to be repeated at the Derelco Site. The permeability study was also designed to find out if the resistance in the piping and tubing were low enough to attain the desired air flowrates. The desired flowrate through the fractures, for the purposes of this study, is 10 - 20 SCFM. The field permeability study was carried out on both wells number 8 and number 7. The method involved connecting the extraction system to each of the wells in turn (Figure 5.2). Air was then sucked out of the wells at different levels of vacuum and the flowrate was measured for each different level of vacuum. First, the vacuum pump was set to maximum vacuum level and then

gradually decreased for well 8 (Table 5.2). The flowrate was measured for each point until the minimum vacuum level was reached. The procedure was repeated for well 7 by increasing the vacuum and the flowrates were checked until the maximum vacuum level was attained (Table 5.3).



Figure 5.2 Field Permeability Studies

The field permeability data are shown in Tables 5.2 and 5.3.

Flowrate/SCFM-Air	Vacuum/ Inches of water	Well Number
15.50	54.5	8
13.75	50.0	8
12.50	45.0	8
11.25	40.0	8
9.50	35.0	8
8.25	30.0	8
5.50	25.0	8
1.75	20.0	8

Table 5.2 Field Permeability Studies

Table 5.3	Field Perm	eability	Studies
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Flowrate/SCFM-Air	Vacuum/inches of water	Well number
1.75	20	7
5.75	25	7
7.75	30	7
9.00	35	7
10.25	40	7
11.50	45	7
13.00	50	7
14.00	55	7
14.25	56	7

The results obtained show that the maximum flowrate attainable, using the existing system is 15.50 SCFM. The major factor limiting the flow was the resistance within the copper lines. In order to reach the desired flowrates (10 - 20 SCFM) the flow manifold was redesigned to include an electronic flow meter and a 1 ½ inch copper piping fitting. The ³/₄ inch copper piping was by-passed and this new design increased the flow-rates almost 3 fold. The maximum flow rate was about 50 SCFM, reached at the maximum vacuum of 55 inches of water.

The permeability study showed that the fractures were still open and that the resistance within them was low enough to allow for the desired amount of air (10 - 20 SCFM) to pass through the formation.

5.3 Preliminary Field Experimental Studies

Prior to the experimental study, a preliminary study was conducted. The aim of this study was to get familiarized with the operation of the equipment and also to have an idea of the major contaminants present at the site. The first part of this study involved drawing samples of air from the intended extraction well (using the vacuum pump) and analyzing the concentrations of the contaminants with both the GC and a PID. Two runs were conducted with a time lapse of about 11 minutes between the two runs. The vacuum pump was left running for the period between the two runs. The results obtained from the GC were checked with a PID to verify the reliability of the data obtained. The results are summarized in Table 5.4.

Contaminant	Concentration, PPMv	Concentration, PPMv
	GC	PID
TCE	5.58	6.6
DCE	0	0

 Table 5.4 Preliminary Field Study

For each case, the detector was placed at the exhaust of the vacuum pump and the sample was taken at the same time that the GC started to sample. Other results obtained from the first part of the preliminary study are summarized in Table A.34 and Table A.35 (Appendix A).

The first part of the preliminary field study led to the following conclusion:

- Table A.34 shows that the pressure drop through the whistles and the feed lines increases substantially with flowrate.
- 2. The major contaminant present appears to be TCE (Table A.35).
- 3. There may be traces of DCE and other unknowns (Table A.35).

The second part of the preliminary field study involved performing the first sonic experiments. For this purpose two preliminary experiments were performed. The process flow diagram for the preliminary experiments is shown in Figure 5.3.



Figure 5.3 Field Layout

5.3.1 First Preliminary Run (Saturday, September 26, 1998)

The first preliminary run was carried out on September 26, 1998. The run started around 10.00 AM. The weather conditions for this day were mild with a cool morning and perfectly dry conditions. There were no strong winds on this day.

5.3.1.1 Experimental Procedure for Preliminary Experiment 1: The experimental procedure involved the following;

Well number 4 was used as the injection well and well number 8 was used as the extraction well. This arrangement was chosen because at a distance of about 9 feet down well number 8 there was a constriction which prevented the whistle array from going lower into the well. Also the well video tapes showed that well number 4 and well number 8 had very good fractures. All the other wells were sealed with rubber caps which tightened by a bolt and expanded against the inside wall of the borehole pipe. The distance between the centers of the two wells is 7 feet and 8 inches. The injection system was lowered into well number 4 so that the center of the two whistles was at 13.3 feet BGS. The extraction system was lowered into well number 8 so that the center of the perforated pipe was at 14.9 feet BGS. The equipment was laid out as shown in Figure 5.3. The packers were then inflated with an external source of Nitrogen until there was a tight seal in each of the two wells. The compressor was turned on for about five minutes to allow all the water vapor to evaporate. The vacuum pump was then turned on and the compressed air from the compressor was then connected to the flow manifold. The flow meters were then adjusted so that no flow went through flowmeters 1 and 2 (whistles shut) and 5 SCFM was allowed to go through each of the flowmeters 3 and 4 (total flow = 10 SCFM, whistle by pass open).

The PID was then used to measure the concentrations of the gases going through the extraction line, as a function of time. The PID was used to measure concentrations for a total of 3772 seconds (about one hour). After this period the flow was switched from the by pass to the whistles. This was done in order to find the impact of the sonic energy on the amount of organic removed. The flow was switched so that no flow went through flow meters 3 and 4(by pass shut) and then flowmeters 1 and 2 were each turned on to 5 SCFM (total flow = 10 SCFM, whistles open). The PID was then allowed to take samples for a total duration of 2090 seconds (35 minutes).

The flow was then switched back to flowmeters 3 and 4 so that no flow went through 1 and 2 (whistles shut). A flow of 5 SCFM was allowed to go through each of the flow meters 3 and 4 (by pass open). The PID was again used to sample for 1055 seconds (18 minutes). This entire cycle completed the first preliminary experiment. The total duration of this experiment was about 1 hour and 55minutes.

5.3.2 Second Preliminary Run (Saturday, September 26, 1998)

The second preliminary run was carried out on September 26, 1998. The run started around 1.30 PM. The weather conditions for this day were mild with a warm afternoon and perfectly dry conditions. There were no strong winds on this day.

5.3.2.1 Experimental Procedure for Preliminary Experiment 2: The procedure for preliminary Experiment 2 was as follows;

The flow meters were adjusted so that no flow went through flowmeters 1 and 2 (whistles shut) and 5 SCFM was allowed to go through each of the flowmeters 3 and 4 (total flow = 10 SCFM, whistle by pass open). The PID was then used to measure the concentrations of the gases going through the extraction line, as a function of time. The PID was used to measure the concentration of TCE for a total of 3085 seconds (about 52 minutes). After this period the flow was switched from the by pass to the whistle. This was done in order to find the impact of the sonic energy on the amount of organic removed. The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10SCFM, whistles open). The PID was then used to measure concentrations for a total duration of 2930 seconds (about 49 minutes). This entire cycle completed the second preliminary experiment. The entire experiment was for a duration of about 1 hour and 40 minutes.

5.3.3 Results of the Preliminary Studies

The results obtained for the preliminary field sonic experimental studies are presented in Figure 5.4 - Figure 5.7 and the measured data are shown in Tables A.1 - A.7.



Figure 5.4 Removal Rate of Organic (First Preliminary Study)

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Figure 5.5 Removal Rate of Organic (Second Preliminary Study)



Figure 5.6 Average Rate of Removal of Organic at Assymptotic Value (First Preliminary Study)


Figure 5.7 Average Rate of Removal of Organic at Assymptotic Value (Second Preliminary Study)

In Figure 5.4, initially the experiments are performed using air alone. During this period the rate of removal of the organic contaminant decreases from a peak value of about 0.00025 cubic feet per minute until it stabilizes at a constant rate of removal of about 0.00017 cubic feet per minute. The rate of removal remains at this constant value for about 20 minutes. When the Sonic energy is turned on, the sonic energy causes the rate of removal to increase from the original constant value of 0.00017 cubic feet per minute to a higher constant rate of removal of about 0.00023 cubic feet per minute. The rate of removal set this higher constant rate of removal of about 0.00023 cubic feet per minute. The rate of removal remains at this higher constant value for about 30 minutes. When the operating conditions are turned back to air, the rate of removal again drops from about 0.00023 cubic feet per minute to an average value of about 0.00017 cubic feet per minute and stays at this value until the end of the first preliminary study.

Figure 5.5 Shows the results obtained for the second preliminary study. Initially, the experiments are performed using air alone. During this period the rate of removal of the organic contaminant decreases from a peak value of about 0.0003 cubic feet per minute until it stabilizes at a constant rate of removal of about 0.00017 cubic feet per minute. The rate of removal remains at this constant value for about 40 minutes. When the sonic energy is turned on, the sonic energy causes the rate of removal to increase from the original constant value of 0.00017 cubic feet per minute to a higher constant rate of removal of about 0.00023 cubic feet per minute. The rate of about 0.00023 cubic feet per minute. The rate of removal to increase from the original constant value of 0.00017 cubic feet per minute to a higher constant rate of removal of about 0.00023 cubic feet per minute. The rate of removal remains at this higher constant value for about 43 minutes, end of the second preliminary study. Figure 5.6 and Figure 5.7 give the average values of the constant rate periods over the total duration of the experiments. The percentage improvement, calculated from Figure 5.6 is

about 37.5% for the first part and 29.4% for the second part. The percentage improvement, calculated from Figure 5.7 is about 35%.

The preliminary field study shows that sonic energy can lead to a significant improvement in both the rate of removal and the amount of organic contaminant removed. The results of the preliminary studies also show that the existing experimental system, experimental approach and the site under investigation were well suited to perform the sonic field experiments.

5.4 Detailed Field Experimental Studies

The experimental study involved four runs. The dates of the experimental runs were :

- Experimental Run 1 : Sunday September 27, 1998.
- Experimental Run 2 : Saturday November 7, 1998.
- Experimental Run 3 : Saturday November 14, 1998.
- Experimental Run 4 : Sunday November 15, 1998.
- Additional Field Experiment: Sunday November 8, 1998.

The experiments were carried out each time switching between the use of sonic energy and without the use of sonic energy. A summary of the experiments is given below. Experimental Run 1 : Sonic energy – no sonic energy –Sonic energy - no sonic energy

Experimental Run 2 : Sonic energy- no sonic energy-Sonic energy- no sonic energy

Experimental Run 3 : no sonic energy-Sonic energy- no sonic energy-Sonic energy

Experimental Run 4 : Sonic energy- no sonic energy-Sonic energy- no sonic energy

Additional Field Experiment: no sonic energy-Sonic energy- no sonic energy-Sonic energy- no sonic energy

5.4.1 Setup of Experimental Apparatus

The entire setup for the sonic field study is shown in Figure 5.3.

5.4.2 Experimental Runs

Each of the experimental runs is described in detail in the following sections. The results obtained are also presented here and the discussion of the results, for each of the runs, is given in Chapter 6.

5.4.2.1 Experimental Run 1, Sunday September 27, 1998: The first experimental run was carried out on September 27, 1998. The run started around 9.15 AM. It was a nice day with some sunshine and dry conditions. There were no strong winds on this day.

5.4.2.1.1 Experimental Procedure: The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10SCFM, whistles open). The PID was then used to measure the concentration of TCE for a total duration of 2130 seconds (about 36 minutes). The flow meters were then adjusted so that no flow went through flowmeters 1 and 2 (whistles shut) and 5 SCFM was allowed to go through each of the flowmeters 3 and 4 (total flow = 10 SCFM, whistle by pass open). The PID was then used to measure the

concentrations of the gases going through the extraction line, as a function of time. The PID was used to sample for a total of 2975 seconds (about 50 minutes).

After this period the flow was switched from the by pass to the whistle. This was done in order to find the impact of the sonic energy on the amount of organic removed. The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10 SCFM, whistles open). The PID was then used to measure the concentration of TCE for a total duration of 2905 seconds (48 minutes). The flow was again switched back to flowmeters 3 and 4 so that no flow went through 1 and 2 (whistles shut). A flow of 5 SCFM was allowed to go through each of the flow meters 3 and 4 (whistle by pass open). The PID was again used to measure the concentration of TCE for 1840 seconds (about 30 minutes). This entire cycle completed the first experimental run. The total duration of this experiment was about 2 hours and 45 minutes.

5.4.2.1.2 Results of Experimental Run 1, Sunday September 27, 1998: The results obtained for the first experimental run are presented in Figure 5.8 – Figure 5.13 and Tables A.8 –A.12.



Figure 5.8 Concentration of Trichloroethylene in Effluent Stream (Experiment 1)



Figure 5.9 Removal Rate of Organic (Experiment 1)



Figure 5.10 Assymptotic Values of Concentration of Trichloroethylene in Effluent Stream (Experiment 1)



Figure 5.11 Assymptotic Values of Removal Rate of Organic (Experiment 1)



Figure 5.12 Average Value of Concentration of Trichloroethylene in Effluent Stream at Assymptotic Value (Experiment 1)



Figure 5.13 Average Rate of Removal of Organic at Assymptotic Value (Experiment 1)

5.4.2.2 Experimental Run 2, Saturday November 7, 1998: The Second experimental run was carried out on November 7, 1998. The run started around 12.44 PM. It was a sunny day with some gentle wind and dry conditions.

5.4.2.2.1 Experimental Procedure: The procedure for Experimental Run 2 was as follows;

The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10 SCFM, whistles open). The PID was then used to measure the concentration for a total duration of 3355 seconds (about 60 minutes). The flow meters were then adjusted so that no flow went through flowmeters 1 and 2 (whistles shut) and 5 SCFM was allowed to go through each of the flowmeters 3 and 4 (total flow = 10 SCFM, whistle by pass open). The PID was then used to measure the concentrations of the gases going through the extraction line, as a function of time. The PID was used to sample for a total of 3630 seconds (about 60 minutes).

After this period the flow was switched from the by pass to the whistle. This was done in order to find the impact of the sonic energy on the amount of organic removed. The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10 SCFM, whistles open). The PID was then used to take samples for a total duration of 3000 seconds (50 minutes). The flow was again switched back to flowmeters 3 and 4 so that no flow went through 1 and 2 (whistles shut). A flow of 5 SCFM was allowed to go through each of the flow meters 3 and 4 (whistle by pass open). The PID was again used

to measure the concentration of TCE for 2460 seconds (about 41 minutes). This entire cycle completed the second experimental run. The total duration of this experiment was about 3 hours and 30 minutes.

5.4.2.2.2 Results of Experimental Run 2, Saturday November 7, 1998: The results obtained for the second experimental run are presented in Figure 5.14 – Figure 5.19 and Tables A.13 – A.17.



Figure 5.14 Concentration of Trichloroethylene in Effluent Stream (Experiment 2)



Figure 5.15 Removal Rate of Organic (Experiment 2)



Figure 5.16 Assymptotic Values of Concentration of Trichloroethylene in Effluent Stream (Experiment 2)



Figure 5.17 Assymptotic Values of Removal Rate of Organic (Experiment 2)







Figure 5.19 Average Rate of Removal of Organic at Assymptotic Value (Experiment 2)

5.4.2.3.1 Experimental Procedure: The procedure for Experimental Run 3 was as follows;

The flow rate of air was switched so that no flow went through flowmeters 1 and 2 (whistles shut) and then flowmeters 3 and 4 were each turned on to 5 SCFM (total flow = 10SCFM, whistle by pass open). The PID was then used to measure the concentration of TCE for a total duration of 3000 seconds (about 50 minutes). The flow meters were then adjusted so that no flow went through flowmeters 3 and 4 (whistles open) and 5 SCFM was allowed to go through each of the flowmeters 1 and 2 (total flow = 10 SCFM, whistle by pass shut). The PID was then used to measure the concentrations of the gases going through the extraction line, as a function of time. The PID was used to measure the concentration of TCE for a total duration of 3000 seconds (about 50 minutes).

After this period the flow was switched from the whistle to the by pass. This was done in order to find the impact of the sonic energy on the amount of organic removed. The flow was switched so that no flow went through flow meters 1 and 2 (whistles shut) and then flow meters 3 and 4 were each turned on to 5 SCFM (total flow = 10 SCFM, whistle by pass open). The PID was then used to take samples for a total duration of 2580 seconds (43 minutes). The flow was again switched back to flowmeters 1 and 2 so that no flow went through 3 and 4 (whistles open). A flow of 5 SCFM was allowed to go through each of the flow meters 1 and 2 (whistles open). The PID was again allowed to sample

for 3600 seconds (about 60 minutes). This entire cycle completed the third experimental run. The total duration of this experiment was about 3 hours and 23 minutes.

5.4.2.3.2 Results of Experimental Run 3, Saturday November 14, 1998: The results obtained for the third experimental run are presented in Figure 5.20 – Figure 5.25 and Tables A.19 - A.22.



Figure 5.20 Concentration of Trichloroethylene in Effluent Stream (Experiment 3)



Figure 5.21 Removal Rate of Organic (Experiment 3)

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Figure 5.22 Assymptotic Values of Concentration of Trichloroethylene in Effluent Stream (Experiment 3)



Figure 5.23 Assymptotic Values of Removal Rate of Organic (Experiment 3)



Figure 5.24 Average Value of Concentration of Trichloroethylene in Effluent Stream at Assymptotic Value (Experiment 3)



Figure 5.25 Average Rate of Removal of Organic at Assymptotic Value (Experiment 3)

5.4.2.4 Experimental Run 4, Sunday November 15, 1998: The fourth experimental run was carried out on November 15, 1998. The run started around 12.15 PM. It was a cold day with some sunshine and conditions were generally dry. There were some strong winds on this day.

5.4.2.4.1 Experimental Procedure: The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10 SCFM, whistles open). The PID was then used to take samples for a total duration of 1945 seconds (about 32 minutes). The flow meters were then adjusted so that no flow went through flowmeters 1 and 2 (whistles shut) and 5 SCFM was allowed to go through each of the flowmeters 3 and 4 (total flow = 10 SCFM, whistle by pass open). The PID was then used to measure the concentrations of the gases going through the extraction line, as a function of time. The PID was used to sample for a total of 1895 seconds (about 32 minutes).

After this period the flow was switched from the by pass to the whistle. This was done in order to find the impact of the sonic energy on the amount of organic removed. The flow was switched so that no flow went through flow meters 3 and 4 (whistle by pass shut) and then flow meters 1 and 2 were each turned on to 5 SCFM (total flow = 10 SCFM, whistles open). The PID was then used to measure conentrations for a total duration of 2030 seconds (34 minutes). The flow was again switched back to flowmeters 3 and 4 so that no flow went through 1 and 2 (whistles shut). A flow of 5 SCFM was allowed to go through each of the flow meters 3 and 4 (whistle by pass open). The PID was again allowed to sample for 2260 seconds (about 38 minutes). This entire cycle

completed the fourth experimental run. The total duration of this experiment was about 2 hours and 15 minutes.

5.4.2.4.2 Results of Experimental Run 4, Sunday November 15, 1998: The results obtained for the fourth experimental run are presented in Figure 5.26 – Figure 5.31 and Tables A.23 – A.27.



Figure 5.26 Concentration of Trichloroethylene in Effluent Stream (Experiment 4)



Figure 5.27 Removal Rate of Organic (Experiment 4)



Figure 5.28 Assymptotic Values of Concentration of Trichloroethylene in Effluent Stream (Experiment 4)



Figure 5.29 Assymptotic Values of Removal Rate of Organic (Experiment 4)



Figure 5.30 Average Value of Concentration of Trichloroethylene in Effluent Stream at Assymptotic Value (Experiment 4)



Figure 5.31 Average Rate of Removal of Organic at Assymptotic Value (Experiment 4)

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5.4.2.5 Additional Experiment, Saturday November 8, 1998: The additional experiment was carried out on November 8, 1998. The run started around 10.00 AM. It was a cold day with some light showers. There were some strong winds on this day.

5.4.2.5.1 Experimental Procedure: The additional experiment was performed to find the influence of some of the operating parameters on the amount and rate of removal of the organic contaminant. The particular operating parameter investigated was the vacuum drawn from the extraction well.

The flow was switched so that no flow went through flow meters 1 and 2 (whistles shut) and then flow meters 3 and 4 were each turned on to 5 SCFM (total flow = 10 SCFM, whistle by pass open). The vacuum pump was then adjusted to a reading of 27 inches of water vacuum. The PID was then used to measure the concentration of TCE for a total duration of 2380 seconds (about 40 minutes).

The flow meters were then adjusted so that no flow went through flow meters 3 and 4 (whistles open) and 5 SCFM was allowed to go through each of the flow meters 1 and 2 (total flow = 10 SCFM, whistle by pass shut).

The vacuum pump was then adjusted to a lower reading of 20 inches of water vacuum. This adjustment was made in order to determine the influence of operating at different levels of vacuum on the sonic experiments. The objective was to determine by how much the vacuum could be reduced when sonic energy was used. The PID was then used to measure the concentrations of the gases going through the extraction line, as a function of time. The PID was used to measure the concentration of TCE for a total of 3030 seconds (about 50 minutes).
After this period the flow was switched from the whistle to the by pass. In order to determine the influence of operating at different levels of vacuum, the vacuum pump was again adjusted to a higher reading of 27 inches of water vacuum. The flow was then switched so that no flow went through flow meters 1 and 2 (whistles shut) and then flow meters 3 and 4 were each turned on to 5 SCFM (total flow = 10 SCFM, whistle by pass open). The PID was then used to measure the concentration of TCE for a total duration of 2140 seconds (40 minutes).

The flow was again switched back to flowmeters 1 and 2 so that no flow went through 3 and 4 (whistles open). The vacuum pump reading was again set to 20 inches of water vacuum. A flow of 5 SCFM was allowed to go through each of the flow meters 1 and 2 (whistles open). The PID was again allowed measure the concentration of TCE for 2160 seconds (about 36 minutes).

After this period, the whistles were shut and the by pass was opened. The vacuum pump was then adjusted to a lower reading of 20 inches of water vacuum. The PID was then allowed to sample for another 30 minutes. This entire cycle completed the additional experimental run. The total duration of this additional experiment was about 3 hours and 15 minutes.

5.4.2.5.2 Results of the Additional Experiment, Sunday November 8, 1998: The results obtained for the additional experimental run are presented in Figure 5.32 – Figure 5.37 and Tables A.28 – A.33.



Figure 5.33 Removal Rate of Organic (Additional Experiment, *Inches of Water Vacuum)



Figure 5.32 Concentration of Trichloroethylene in Effluent Stream (Additional Experiment, *Inches of Water Vacuum)



Figure 5.34 Assymptotic Values of Removal Rate of Organic (Additional Experiment, *Inches of Water Vacuum)



Figure 5.35 Assymptotic Concentration of Trichloroethylene in Effluent Stream (Additional Experiment, *Inches of Water Vacuum)



Figure 5.36 Average Concentration of Trichloroethylene in Effluent Stream at Assymptotic Value (Additional Experiment, *Inches of Water Vacuum)



Figure 5.37 Average Rate of Removal of Organic at Assymptotic Value (Additional Experiment, *Inches of Water Vacuum)

CHAPTER 6

DISCUSSION OF EXPERIMENTAL RESULTS

6.1 Experimental Run 1 (Sunday, September 27, 1998)

The results of experimental run one are presented in Figure 5.8 to Figure 5.13.In Figure 5.8, initially the experiments are performed using sonic energy (whistles turned on). At the start of the experiment, the concentration of trichloroethylene in the effluent stream decreases rapidly from a value of about 40 PPMv until it stabilizes at a constant value of about 22 PPMv. This initial decline in the concentration can be explained by the fact that initially only the contaminants on the surface of the fractures are being removed. The amount of these surface contaminants removed is directly proportional to their concentration on the surface. Thus the concentration of the contaminant on the surface is the limiting condition that controls the amount of contaminant removed. As the concentration of the contaminant on the surface decreases the concentration of trichloroethylene in the effluent stream also decreases. This trend is followed until the contaminants on the surface of the fractures are depleted.

After this period, Figure 5.8, the concentration of trichloroethylene in the effluent stream stays at this constant value of about 22 PPMv. This observation can be explained by the fact that during this period the contaminants trapped within the fractures are being removed. As these contaminants are being removed from a tight formation, the limiting condition for the concentration of trichloroethylene in the effluent stream during this period is the penetration ability of the sonic energy into the fractures. Since the amplitude and frequency and the intensity of the sound at the source from the sonic device remain

constant during the experiment the penetration ability of the sonic energy into the formation also remains constant and hence the concentration of the effluent stream is constant for this period.

When the method of operation is switched from sound to air, with no sound, Figure 5.8, the concentration of trichloroethylene in the effluent stream drops to a lower constant value of about 16.5 PPMv. This observation can be explained by the fact that during this period the contaminants that are being removed are still those that are trapped within the formation. As the air carries less energy and produces less vibrations within the fractures, compared to the sound, the ability of the air to penetrate the tight geological formation is considerably decreased. This explains the observed drop in concentration of trichloroethylene in the effluent stream (from about 22 PPMv with sound to about 16.5 PPMv with air).

The observed initial increase in the concentration of trichloroethylene in the effluent stream when the air is first turned on can be explained by the fact that the air may be sweeping out some contaminants that were left loose by the vibrations of the sonic energy. This explanation is justified by the fact that the duration of this increase lasts for only about 5 to 10 seconds compared to the total duration of the experimental run (about 11000 seconds). The subsequent increase in the concentration of trichloroethylene in the effluent stream (to about 22 PPMv) when the sound is again switched on and the drop in the concentration of trichloroethylene in the effluent stream (to about 16.5 PPMv) when the sound is again switched off (Figure 5.8) follow the same explanation as given above.

Figure 5.9 is obtained from Figure 5.8 by converting the concentrations of TCE in the effluent stream to removal rates (Appendix B). In Figure 5.9, initially the experiments are performed using sonic energy (whistles turned on). At the start of the experiment, the removal rate of trichloroethylene, which is the product of the air flow rate and concentration of TCE, decreases rapidly from a value of about 0.0004 cubic feet per minute until it stabilizes at a constant value of about 0.000255 cubic feet per minute. When the mode of operation is switched from sound to no sound the effluent flowrate remains unchanged but the removal rate of TCE drops from 0.000255 cubic feet per minute to 0.000189 cubic feet per minute. When the sound is again turned on and then switched off, the effluent flowrate still remains the same but the removal rate of the TCE increases and then decreases respectively (Table 6.1). The trend follows the same explanation as given for Figure 5.8. The results indicate that the sonic energy is effective in removing the TCE from the soil.

Experiment	Mode of	Concentration of	Effluent	Removal Rate of
-	Operation	TCE at the	Flowrate,	TCE at
		Assymptotic Value,	scfm	Assymptotic
		PPMv		Value, Cubic Feet
				Per Minute
1	sound	22.0	11.3	0.000255
1	no sound	16.5	11.3	0.000189
1	sound	22.0	11.3	0.000240
1	no sound	16.5	11.3	0.000180
2	sound	33.0	12.1	0.000393

Table 6.1 Average Concentration	of TCE and Removal Rat	tes at Assym	ptotic Values
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Experiment	Mode of	Concentration of	Effluent	Removal Rate of
	Operation	TCE at the	Flowrate,	TCE at
		Assymptotic Value,	scfm	Assymptotic
		PPMv		Value, Cubic Feet
				Per Minute
2	no sound	30.0	10.1	0.000300
2	sound	33.0	12.5	0.000413
2	no sound	30.0	10.7	0.000330
3	nosound	18.0	9.5	0.000121
3	sound	20.0	12.8	0.000256
3	no sound	18.0	10.4	0.000186
3	sound	20.0	12.5	0.000260
4	sound	21.0	11.0	0.000233
4	no sound	16.0	10.0	0.000159
4	sound	19.9	11.3	0.000222
4	no sound	16.0	9.6	0.000155

 Table 6.1 Average Concentration of TCE and Removal Rates at Assymptotic Values.

 (Continued)

Figure 5.10 shows the assymptotic values for Figure 5.8 and Figure 5.11 shows the corresponding constant rate plot for Figure 5.9. These are the periods during which the penetration ability of the method employed limit both the amount and rate of removal of the organic contaminant. From Figure 5.10 and 5.11, it is evident that both the amount and the removal rate of the organic are increased when sonic energy is focused into the fractures. The percentage increase when sonic energy is focussed into the fractures is calculated by taking the averages of the assymptotic values of Figure 5.10 and 5.11. This

is shown in Figure 5. 12 (average concentration of trichloroethylene in the effluent stream at assymptotic value, from Figure 5.10) and Figure 5.13 (average rate of removal of organic at assymptotic value, from Figure 5.11). The percentage improvement in concentrations for the first sound-air pair, as calculated from Figure 5.12 is 34.8% and from the second-air pair is 33.08%. Also, the percentage improvement in removal rate for the first sound-air pair, as calculated from Figure 5.13 is 34.92% and from the second-air pair, as calculated from Figure 5.13 is 34.92% and from the second-air pair is 33.33%.

These results show that there is a remarkable improvement when sonic energy is used as an enhancement technique to remove volatile organic compounds.

6.2 Experimental Run 2 (Saturday, November 7, 1998)

The results of experimental run 2 are presented in Figure 5.14 to Figure 5.19. In Figure 5.14, initially the experiments are performed using sonic energy (whistles turned on). At the start of the experiment, the concentration of trichloroethylene in the effluent stream builds up from a low value of about 23 PPMv to a high value of about 33 PPMv. This observation can be explained by the fact that at the time that this run was started there was not a considerable amount of organic contaminant left on the surface of the fractures. This is a result of the fact that the vacuum pump had been left running prior to the experiment, while the experimental conditions were being prepared, setting the compressor and checking the water levels. The initial build up period also justifies the fact that the sonic energy needs a finite time to penetrate the fractures and manifest its effect. This time is measured as about 30 minutes, from Figure 5.14. After this period,

Figure 5.14, the concentration of trichloroethylene in the effluent stream stays at a constant value of about 33 PPMv (Table 6.1).

When the method of operation is switched from sound to no sound, Figure 5.14, the concentration of trichloroethylene in the effluent stream drops to a lower constant value of about 30 PPMv. The subsequent increase in the concentration of trichloroethylene in the effluent stream (to about 33PPMv) when the sound is again switched on and the drop in the concentration of trichloroethylene in the effluent stream (to about 30 PPMv) when the sound is switched off (Table 6.1) follow the same explanation as given for Figure 5.8. Table 6.1 shows that these changes in concentration are not as large as in experiment 1 because the effluent flow rate also changed. Had the flow rate stayed constant at 12.1 SCFM the concentration would have been lower than 30 PPMv.

Figure 5.15 is similar to Figure 5.14 with the concentrations converted to removal rates (refer to Appendix B for sample calculation). Figure 5.15 unlike Figure 5.14 takes into account the variation of the effluent flowrate. In Figure 5.15, initially the experiments are performed using sonic energy (whistles turned on). At the start of the experiment, the removal rate of trichloroethylene builds up from a low value of about 0.00027 cubic feet per minute to a high value of about 0.000393 cubic feet per minute (Table 6.1). After this period, Figure 5.15, the removal rate of trichloroethylene stays at a constant value of about 0.000393 cubic feet per minute.

When the method of operation is switched from sound to no sound the removal rate of trichloroethylene drops to a lower constant value of about 0.0003 cubic feet per minute but the effluent flowrate also drops from 12.1 scfm to 10.1 scfm (Table 6.1). This

drop in effluent flowrate means that the drop in the rate of removal would be higher if the effluent flowrates were maintained at the same value. There is no single explanation for the observed drop in the effluent flowrate. A possible explanation however is the possibility of the air being lost within the formation and hence resulting in an unsteady state balance.

The subsequent increase in the removal rate of trichloroethylene (to about 0.000413 cubic feet per minute) when the sound is again switched on and the drop in the removal rate of trichloroethylene (to about 0.00033 cubic feet per minute) when the sound is again switched off (Table 6.1) follow the same explanation as given for Figure 5.8.

Figure 5.16 shows the assymptotic values for Figure 5.14 and Figure 5.17 shows the corresponding constant rate plot for Figure 5.15. These are the periods during which the penetration ability of the method employed limits both the amount and rate of removal of the organic contaminant. From Figure 5.16 and 5.17, it is evident that both the amount and the removal rate of the organic are increased when sonic energy is focused into the fractures.

The percentage increase when sonic energy is focussed into the fractures is calculated by taking the averages of the assymptotic values of Figure 5.16 and 5.17. This is shown in Figure 5. 18 (average concentration of trichloroethylene in the effluent stream at assymptotic value, from Figure 5.16) and Figure 5.19 (average rate of removal of organic at assymptotic value, from Figure 5.17). The percentage improvement in concentration for the first sound-air pair, as calculated from Figure 5.18 is 9.1% and from the second-air pair is 7%. Also, the percentage improvement in removal rate for the first

sound-air pair, as calculated from Figure 5.19 is 31% and from the second-air pair is 25%. The change in the removal rate is more consistent with Experiment 1 because the effluent flow rate changed between the runs (Table 6.1). These results, like the results from experiment 1, show that there is a remarkable improvement when sonic energy is used as an enhancement technique to remove volatile organic compounds.

6.3 Experimental Run 3 (Saturday, November 14, 1998)

The results of experimental run 3 are presented in Figure 5.20 to Figure 5.25. The procedure for this experiment was reversed, compared to Experiment 1 and 2. In this experiment the first run was performed using no sound. In Figure 5.20, initially the experiments are performed using no sound (whistles shut). At the start of the experiment the concentration of trichloroethylene in the effluent stream decreases rapidly from a value of about 30 PPMv until it stabilizes at a constant value of about 18 PPMv. The drop in the concentration of trichloroethylene in the effluent stream (from 30PPMv to 18 PPMv) took about 15 minutes. This initial decline observed was very similar to the decline observed in Figure 5.8. Thus the method of operation does not affect the initial period. The concentration of the organic removed then stayed at this value of 18 PPMv until the whistles were turned on. When the whistles were turned on, there was an observed sharp increase in the concentration of trichloroethylene in the effluent stream, from 18 PPMv to 20 PPMv and an increase in the effluent flowrate from 9.5 scfm to 12.8 scfm (Table 6.1). Thus, the increase in the concentration was underestimated due to the dilution effect of the increased air flowrate. The concentration of trichloroethylene in the effluent stream stayed constant at 20 PPMv for about 20 minutes.

When the method of operation was switched from sound to air, Figure 5.20, the concentration of trichloroethylene in the effluent stream drops to a lower constant value of about 18 PPMv. The effluent flowrate also dropped to 10.4 scfm. This drop in effluent flowrate indicates that the drop in concentration was underestimated.

Figure 5.21 is similar to Figure 5.20 but with the concentrations of TCE in Figure 5.20 converted to removal rates (Appendix B). Figure 5.21 also takes into account the variation in the effluent flowrates. In Figure 5.21, initially the experiments are performed using air (whistles shut). At the start of the experiment the removal rate of trichloroethylene decreases rapidly from a value of about 0.0004 cubic feet per minute until it stabilizes at a constant value of about 0.000171 cubic feet per minute. The drop in the removal rate of trichloroethylene (from 0.0004cubic feet per minute to 0.000171 cubic feet per

When the whistles were turned on, there was an observed sharp increase in the removal rate of trichloroethylene, from 0.000171 cubic feet per minute to 0.000256 cubic feet per minute. The removal rate of trichloroethylene stayed constant at this value for about 20 minutes.

When the method of operation was switched from sound to air, Figure 5.21, the removal rate of trichloroethylene drops to a lower constant value of about 0.000186 cubic feet per minute (Table 6.1). The trends displayed follow the same explanation as given for Figure 5.8. Figure 5.21 unlike Figure 5.20 takes into account the variation of the effluent flowrates.

Figure 5.22 shows the assymptotic values for Figure 5.20 and Figure 5.23 shows the corresponding constant rate plot for Figure 5.21. These are the periods during which the penetration ability of the method employed limits both the amount and rate of removal of the organic contaminant. From Figure 5.22 and 5.23, it is evident that both the amount and the removal rate of the organic are increased when sonic energy is focused into the fractures. The percentage increase when sonic energy is focussed into the fractures is calculated by taking the averages of the assymptotic values of Figure 5.22 and 5.23. This is shown in Figure 5.24 (average concentration of trichloroethylene in the effluent stream at assymptotic value, from Figure 5.22) and Figure 5.25 (average rate of removal of organic at assymptotic value, from Figure 5.23). The percentage improvement for the first air-sound pair, as calculated from Figure 5.24 is 11% and from the air-sound pair is 16.2%. Also, the percentage improvement in removal rate for the first air-sound pair, as calculated from Figure 5.25 is 49.7% and from the second-air pair is 39.7%. These results, like the results of experiment 1 and 2, show that there is a remarkable improvement when sonic energy is used as an enhancement technique to remove volatile organic compounds.

6.4 Experimental Run 4 (Sunday, November 15, 1998)

The results of experimental run 4 are presented in Figure 5.26 to Figure 5.31.In Figure 5.26, initially the experiments are performed using sonic energy (whistles turned on). At the start of the experiment, the concentration of trichloroethylene in the effluent stream decreases slightly from a value of about 23 PPMv until it stabilizes at a constant value of

about 21 PPMv. After this period, Figure 5.26, the concentration of trichloroethylene in the effluent stream stays at this constant value of about 21 PPMv.

When the method of operation is switched from sound to air, Figure 5.26, the concentration of trichloroethylene in the effluent stream drops to a lower constant value of about 16 PPMv (Table 6.1). The effluent flowrate is also observed to drop slightly from 11 scfm to 10 scfm. This drop in effluent flowrate caused the drop in the concentration to be underestimated. It took about 15 minutes for the concentration to drop from 21 PPMv to about 16 PPMv.

The subsequent increase in the concentration of trichloroethylene in the effluent stream (to about 19.9 PPMv) when the sound is again switched on and the drop in the concentration of trichloroethylene in the effluent stream (to about 16 PPMv) when the sound is again switched off (Figure 5.26) follow the same explanation as given for Figure 5.8. For this last sound and no sound pair, the effluent flow rate was again observed to drop from 19.9 scfm to 16 scfm, when the mode of operation was switched from sound to no sound (Table 6.1).

Figure 5.27 is similar to Figure 5.26 but with the concentrations of Figure 5.26 converted to removal rates in Figure 5.27. Thus, Figure 5.27 takes into account the variation in effluent flowrates observed, in Table 6.1, whereas Figure 5.26 does not. In Figure 5.27, initially the experiments are performed using sonic energy (whistles turned on). At the start of the experiment, the removal rate of trichloroethylene decreases slightly from a value of about 0.00025 cubic feet per minute until it stabilizes at a constant value of about 0.000233 cubic feet per minute.

After this period, Figure 5.27, the removal rate of trichloroethylene stays at this constant value of about 0.000233 cubic feet per minute (Table 6.1).

When the method of operation is switched from sound to air, Figure 5.27, the removal rate of trichloroethylene drops to a lower constant value of about 0.000159 cubic feet per minute. It took about 15 minutes for the removal rate to drop from 0.000233 cubic feet per minute to about 0.000159 cubic feet per minute.

The subsequent increase in the removal rate of trichloroethylene (to about 0.000222 cubic feet per minute) when the sound is again switched on and the drop in the removal rate of trichloroethylene (to about 0.000155 cubic feet per minute) when the sound is again switched off (Figure 5.27) follow the same explanation as for Figure 5.8.

Figure 5.28 shows the assymptotic values for Figure 5.26 and Figure 5.29 shows the corresponding constant rate plot for Figure 5.27. These are the periods during which the penetration ability of the method employed limits both the amount and rate of removal of the organic contaminant. From Figure 5.28 and 5.29, it is evident that both the amount and the removal rate of the organic are increased when sonic energy is focused into the fractures.

The percentage increase when sonic energy is focussed into the fractures is calculated by taking the averages of the assymptotic values of Figure 5.28 and 5.29. This is shown in Figure 5. 30 (average concentration of trichloroethylene in the effluent stream at assymptotic value, from Figure 5.28) and Figure 5.31 (average rate of removal of organic at assymptotic value, from Figure 5.29).

The percentage improvement in concentration for the first sound-air pair, as calculated from Figure 5.30 is 33.3% and from the second-air pair is 22%. Also, the

percentage improvement in removal rate for the first sound-air pair, as calculated from Figure 5.31 is 46.5% and from the second-air pair is 43.22%. These results, like the results from experiment 1,2 and 3, confirm that there is a remarkable improvement when sonic energy is used as an enhancement technique to remove volatile organic compounds.

Table 6.2 gives the average of the percentage improvements for each of the experimental runs and the average over the entire field study.

	95% Confidence Interval	95% Confidence Interval
	Standard Deviation =	Standard Deviation = 8.33
	Average = 20.8375%	Average = 37.925 %
4b	22.0	43.2
4a	33.3	46.5
3b	16.5	39.7
3a	11.0	49.7
2b	7.0	25.1
2a	9.1	31.0
1b	33.0	33.3
la	34.8	34.9
	Concentration/PPMv	Cubic Feet per Minute
Experiment*	Percent Improvement	Percent Improvement

 Table 6.2 Average Percentage Improvement for Each of the Experimental Runs

*a = First pair (of no sound and Sound), b= second pair (of no sound and Sound)

The probability is 0.95 that the interval 11.138 to 30.5495 will contain the true mean percentage improvement of the concentration of TCE in the effluent stream and the probability is 0.95 that the interval 30.96 to 44.89 will contain the true mean percentage improvement of the removal rate of TCE.

6.5 Additional Experiment (Saturday November 8, 1998)

In Figure 5.32, with no sonic energy, initially concentration of trichloroethylene dropped from a value of 50 PPMv to a value of about 30 PPMv. This observation can be explained by the fact that the contaminants on the surface of the fractures were being removed. After this period the removal rate of trichloroethylene in effluent stream remains at this constant value of 30 PPMv for about 30 minutes. During this entire period the vacuum pump reading was set to 27 inches of water vacuum.

When the sonic energy was turned on, for a considerably lower vacuum of 20 inches of water vacuum, the concentration of trichloroethylene in effluent stream remains fairly constant. This can be explained by the fact that due to the vibrations occurring within the fractures when sonic energy is focussed into them both the concentration and the removal rate of organic is increased and one could afford to set the vacuum pump to a lower reading of 20 inches of water vacuum and still have the same amount of organic removed, as in air at a higher vacuum reading of 27 inches of water vacuum. The procedure was then repeated for another air –sonic energy pair. The results obtained are as explained above (Figure 5.32). Finally, a last run was performed using no sound and setting the vacuum pump to 20 inches of water vacuum (same vacuum pump reading as in the sonic energy experiments). This led to a rapid decline in the removal rate of

trichloroethylene in effluent stream (Figure 5.32). This proved that the sonic energy enhanced the removal rate of trichloroethylene in effluent stream.

Figure 5.33 is similar to Figure 5.32, with all concentrations converted to removal rates (Refer to appendix B for sample calculation). Figure 5.34 shows the assymptotic values for Figure 5.32. Only the runs with a vacuum of 20 inches of water vacuum are shown. This was done in order to facilitate a comparison of the effect of the sonic energy at a given vacuum. Figure 5.35 is the corresponding at assymptotic value for Figure 5.36 represents the average concentration of Trichloroethylene in the effluent stream, at assymptotic value. It is corresponds to the average value of Figure 5.34. Finally, Figure 5.37 represents the average removal rate of trichloroethylene, at assymptotic value. It is corresponds to the average of Figure 5.35. Thus, this test showed that for about the same removal rate a vacuum of 20 inches of water is needed with sonic energy compared to 27 inches of water with no sonic energy. Furthermore, a comparison of sonic energy and no sonic energy at 20 inches of water vacuum showed a marked decline in the removal rate when sonic energy was not used.

CHAPTER 7

CONCLUSIONS AND RECOMMENDATIONS

The conclusions of the study are;

- 1. The field study, like the previous laboratory studies, show that sonic energy significantly improves the rate of removal and the amount of the organics removed.
- 2. In every experiment the rate of removal increased when sonic energy is focussed into the fractures
- 3. The average percent increase achieved (over the entire field study) on the concentration of the organic in the effluent stream, when sonic energy is focussed into the fractures is 20.8%
- 4. The average percent increase achieved (over the entire field study) on the rate of removal of organic contaminant is 37.9%.
- 5. These values are very significant and shows that sonic energy can be used to reduce the time requirements of a site remediation exercise.
- 6. Also the results of the additional study shows that for the same removal rate less vacuum and therefore, less vacuum pump energy will be consumed using sound, compared to using air alone. This is seen in the vacuum requirements needed to achieve the same removal rate. Thus in situations where the cost of energy is critical using sonic energy could lead to multiple gains.
- 7. The probability is 0.95 that the interval 11.14 to 30.54 percent will contain the true mean percentage improvement of the concentration of TCE in the effluent stream.
- 8. The probability is 0.95 that the interval 30.95 to 44.89 percent will contain the true mean percentage improvement of the removal rate of trichloroethylene.

Recommendations:

- 1. It is recommended that more tests should be made to investigate other parameters that may greatly affect the outcome of the results. A key factor in this category is the attenuation of the sonic field within the formation. Tests should also be run to find the attenuation coefficients in different types of soils and also to establish the radius of influence of the whistle.
- 2. It is also recommended that the optimum power levels at which the whistle should operate be established.
- 3. Also the effect of other parameters such as the orientation and location of the fractures can be studied. Eventually all these parameters can be built into a mathematical model to predict the expected amount of enhancement and the effective range of sonic field. These factors will enhance the design of future field decontamination studies.

APPENDIX A FIELD DATA

Preliminary Experiment1(Saturday,26 September, 1998)

Morning Session

No Sound-sound -No Sound

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Table A.1 Flow Manifold Data

Electronic flowmeter reading: 9.9 scfm (Air1)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Table A.2 Flow Manifold Data

Electronic flowmeter reading: 11.9 scfm (Sound)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
Look an order which developed even dependent of the second s	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Table A.3 Flow Manifold Data

Electronic flowmeter reading: 9.6 scfm (Air2

Time/sec	Conc. of TCE/PPMv	ft^3(organic)/min (flowrate)	Experiment Type, Air
0	340	0.003366	No Sound 99 SCFM
4	115	0.0011385	
6	110	0.001089	
8	110	0.001089	
10	105	0.0010395	
12	97	0.0009603	
17	98.7	0.00097713	
22	98	0.0009702	
27	91.6	0.00090684	
32	94.5	0.00093555	
37	88.1	0.00087219	
42	85	0.0008415	
47	89.5	0.00088605	······································
52	83.1	0.00082269	
57	86.1	0.00085239	
62	81	0.0008019	
67	80.3	0.00079497	
72	79.6	0.00078804	
77	76.8	0.00076032	
82	77.5	0.00076725	
87	73.9	0.00073161	
92	76.1	0.00075339	
102	70.8	0.00070092	
112	70.4	0.00069696	
132	70.4	0.00069696	
142	66.9	0.00066231	
152	60.5	0.00059895	
162	58.4	0.00057816	
172	54.1	0.00053559	
182	52	0.0005148	
192	49.2	0.00048708	
202	47.7	0.00047223	
212	47	0.0004653	
222	45.6	0.00045144	
232	44.9	0.00044451	
242	43.5	0.00043065	
252	41.2	0.00040788	
262	43.5	0.00043065	
272	44.9	0.00044451	
282	47.7	0.00047223	
292	49.4	0.00048906	

Table A.4 Field Data for Preliminary Experiment 1

Time/sec	Conc. of TCE/PPMv	ft^3(organic)/min (flowrate)	Experiment Type, Air flow
302	47.7	0.00047223	
312	47	0.0004653	
322	45.6	0.00045144	
332	44.2	0.00043758	
342	42.3	0.00041877	
352	39.9	0.00039501	
362	37.8	0.00037422	
372	35.7	0.00035343	
412	32.2	0.00031878	
422	31.5	0.00031185	
432	30.7	0.00030393	
442	30	0.000297	
452	29.3	0.00029007	
462	28.6	0.00028314	
472	27.9	0.00027621	
482	27.9	0.00027621	
492	27.9	0.00027621	
502	27.9	0.00027621	
512	27.9	0.00027621	
522	27.9	0.00027621	
532	27.9	0.00027621	
542	27.2	0.00026928	
552	27.2	0.00026928	
562	26.5	0.00026235	
592	26.5	0.00026235	
622	27.2	0.00026928	
652	26.5	0.00026235	
682	26.5	0.00026235	
712	26.5	0.00026235	
742	26.5	0.00026235	
772	26.5	0.00026235	
802	26.5	0.00026235	
832	26.5	0.00026235	
1432	26.5	0.00026235	
1552	24.4	0.00024156	
1612	24.4	0.00024156	
1672	. 22.3	0.00022077	
1732	21.5	0.00021285	
1792	20.8	0.00020592	
1852	20.1	0.00019899	

 Table A.4 Field Data for Preliminary Experiment 1

 (Continued)

Time/sec	Conc. of TCE/PPMv	ft^3(organic)/min	Experiment Type, Air
		(flowrate)	flow
1912	20.1	0.00019899	
1972	19.4	0.00019206	
2032	19.4	0.00019206	
2092	18.7	0.00018513	
2152	18.7	0.00018513	
2212	18.7	0.00018513	
2272	18.7	0.00018513	
2332	18.7	0.00018513	
2392	18.7	0.00018513	
2452	18.7	0.00018513	
2512	18.7	0.00018513	
2572	18.7	0.00018513	
2632	18	0.0001782	
2692	18	0.0001782	
2812	18	0.0001782	
2932	18	0.0001782	
3052	17.3	0.00017127	
3232	17.3	0.00017127	
3412	17.3	0.00017127	
3592	16.6	0.00016434	
3772	16.6	0.00016434	
3774	17.3	0.00020587	Sound,11.9 SCFM
3776	17.3	0.00020587	
3778	18	0.0002142	
3780	17.3	0.00020587	
3782	17.3	0.00020587	
3784	18	0.0002142	
3786	18	0.0002142	
3788	18	0.0002142	
3790	18	0.0002142	
3792	. 17.3	0.00020587	
3794	. 17.3	0.00020587	
3799	17.3	0.00020587	
3804	17.3	0.00020587	
3809	18	0.0002142	
3814	17.3	0.00020587	
3819	17.3	0.00020587	
3824	18	0.0002142	
3829	17.3	0.00020587	

Table A.4 Field Data for Preliminary Experiment 1
(Continued)

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Time/sec	Conc. of TCE/PPMv	ft^3(organic)/mi	Experiment Type, Air
		(flowrate)	flow
3834	17.3	0.00020587	
3839	18	0.0002142	
3844	17.3	0.00020587	
3854	18	0.0002142	
3864	17.3	0.00020587	
3874	18	0.0002142	
3884	17.3	0.00020587	
3894	18	0.0002142	
3904	18	0.0002142	
3914	18	0.0002142	
3944	18	0.0002142	
3974	18	0.0002142	
4004	18	0.0002142	
4034	18.7	0.00022253	
4064	18.7	0.00022253	
4094	18.7	0.00022253	
4124	19.4	0.00023086	
4154	19.4	0.00023086	
4184	19.4	0.00023086	
4214	19.4	0.00023086	
4244	19.4	0.00023086	
4274	19.4	0.00023086	
4304	19.4	0.00023086	
4364	19.4	0.00023086	
4424	18.7	0.00022253	
4484	18.7	0.00022253	
4544	18.7	0.00022253	
4604	18.7	0.00022253	
4664	18.7	0.00022253	
4784	18.7	0.00022253	
4904	18.7	0.00022253	
5024	18.7	0.00022253	
5144	. 18.7	0.00022253	
5264	18.7	0.00022253	
5564	18.7	0.00022253	
5864	. 18.7	0.00022253	
5869	20.8	0.00019968	No Sound, 9.6 SCFM
5874	20.8	0.00019968	
5879	20.8	0.00019968	
5884	20.1	0.00019296	

Table A.4 Field Data for Preliminary Experiment 1(Continued)

Time/sec	Conc. of TCE/PPMv	ft^3(organic)/min	Experiment Type, Air
		(flowrate)	flow
5889	20.8	0.00019968	
5894	20.8	0.00019968	
5899	20.8	0.00019968	
5904	20.8	0.00019968	
5909	20.8	0.00019968	
5914	21.5	0.0002064	
5919	21.5	0.0002064	
5924	20.8	0.00019968	
5929	20.8	0.00019968	
5934	20.8	0.00019968	
5939	20.8	0.00019968	
5944	20.8	0.00019968	
5954	20.8	0.00019968	
5964	20.8	0.00019968	
5974	20.8	0.00019968	
5984	20.8	0.00019968	
5994	20.8	0.00019968	
6024	20.8	0.00019968	
6054	20.8	0.00019968	
6084	20.8	0.00019968	
6114	20.8	0.00019968	
6144	20.8	0.00019968	
6204	20.1	0.00019296	
6264	20.1	0.00019296	
6324	19.4	0.00018624	
6384	19.4	0.00018624	
6444	18.7	0.00017952	
6504	18.7	0.00017952	
6564	18	0.0001728	
6624	18	0.0001728	
6684	18	0.0001728	
6744	. 18	0.0001728	
6804	18	0.0001728	
6924	. 17.3	0.00016608	End

Table A.4 Field Data for Preliminary Experiment 1(Continued)

Preliminary Experiment 2(Saturday ,26 September, 1998)

Afternoon Session:

Air – Sound

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Table A.5 Flow Manifold Data

Electronic flowmeter reading: 9.75 scfm(Air),

Table A.6 Flow Manifold Data

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Electronic flowmeter reading: 12 scfm(Sound)

Table A.7	Field	Data	for	Preliminary	Experiment	2
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Time/sec	Conc. of TCE/PP M v	ft^3(organic)/min (flowrate)	Experiment type, Air flow
2	8.8	0.000858	No Sound, 9.75 SCFM
4	8.8	0.000858	
6	17.3	0.000168675	
8	60.5	0.000589875	
10	73.9	0.000720525	
12	73.2	0.0007137	
14	65.5	0.000638625	-
16	57.9	0.000564525	
18	48.4	0.0004719	
20	43.5	0.000424125	
25	38.5	0.000375375	

Time/sec	Conc. of	ft^3(organic)/min	Experiment type, Air
	ICE/PPMV	(flowrate)	flow
30	33.6	0.0003276	
35	32.9	0.000320775	
40	29.3	0.000285675	
45	27.9	0.000272025	
50	30	0.0002925	
55	30	0.0002925	
60	30	0.0002925	
65	30.7	0.000299325	
70	30	0.0002925	
75	31.5	0.000307125	
80	28.6	0.00027885	
85	30	0.0002925	
90	28.6	0.00027885	
95	30	0.0002925	
100	27.9	0.000272025	
105	27.9	0.000272025	
110	27.9	0.000272025	
115	27.2	0.0002652	
120	28.6	0.00027885	
125	27.2	0.0002652	
135	25.8	0.00025155	
145	25.8	0.00025155	
155	25.8	0.00025155	
165	25.8	0.00025155	
185	25.8	0.00025155	
205	23.7	0.000231075	
225	21.5	0.000209625	
245	21.5	0.000209625	
265	22.3	0.000217425	alaan na maada a dhaannaa maannaa ku dhaha alama maan da aha a shiinaanna maana dha ahaannaanna
325	20.8	0.0002028	
385	19.4	0.00018915	
445	18.7	0.000182325	
505	18.7	0.000182325	
565	18	0.0001755	
625	17.3	0.000168675	
685	17.3	0.000168675	,
985	17.3	0.000168675	
1285	18	0.0001755	ay daga ay Aliyong a tantan yang ang kanan nya tang ang kanan nya tang barang barang ang kanan nya Manan Milan

Table A.7 Field Data for Preliminary Experiment 2 (Continued)

Time/sec	Conc. of	ft^3(organic)/min	Experiment type, Air
0.405	ICE/PPMV	(flowrate)	flow
2485	18	0.0001755	
2785	17.3	0.000168675	
3085	17.3	0.000168675	
3095	16.6	0.0001992	Sound, 12 SCFM
3105	16.6	0.0001992	
3115	17.3	0.0002076	
3125	17.3	0.0002076	
3135	17.3	0.0002076	
3145	16.6	0.0001992	
3155	17.3	0.0002076	
3165	17.3	0.0002076	
3175	17.3	0.0002076	
3185	17.3	0.0002076	
3195	17.3	0.0002076	
3205	17.3	0.0002076	
3235	17.3	0.0002076	
3265	17.3	0.0002076	
3295	18	0.000216	
3325	18	0.000216	
3355	18	0.000216	
3385	18	0.000216	
3415	18	0.000216	
3445	18.7	0.0002244	
3475	18.7	0.0002244	
3505	18.7	0.0002244	
3535	18.7	0.0002244	
3565	18.7	0.0002244	
3625	18.7	0.0002244	
3685	18.7	0.0002244	
3745	18.7	0.0002244	
3805	18.7	0.0002244	
3865	18.7	0.0002244	
3925	19.4	0.0002328	
4225	18.7	0.0002244	
4525	18.7	0.0002244	
4825	19.4	0.0002328	
5125	19.4	0.0002328	
5425	19.4	0.0002328	
5725	19.4	0.0002328	
6025	19.4	0.0002328	END

Table A.7 Field Data for Preliminary Experiment 2 (Continued)

Experimental Studies

Experimental Run 1 : Sunday September 27, 1998.

Sound- No Sound- Sound- No Sound

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	
Electronic flour	neter reading: 11	2 cofm(Sound)		I

Electronic flowmeter reading: 11.3 scfm(Sound)

Table A.9 Flow Manifold Data

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 11.3 scfm(Air)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Table A.10 Flow Manifold Data

Electronic flowmeter reading: 11.3 scfm (Sound)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Table A.11 Flow Manifold Data

Electronic flowmeter reading: 11.3 scfm (Air)

Time/sec	Conc. of TCE/PPMv	Ft^3 organic/min	Experiment type, Air
		(flowrate)	flow
0	284	0.003209	Sound, 11.3 SCFM
10	388	0.004384	
15	446	0.00504	
20	484	0.005469	
25	499	0.005639	
30	489	0.005526	
35	443	0.005006	
40	420	0.004746	
45	377	0.00426	
50	233	0.002633	
55	151	0.001706	
60	119	0.001345	
65	91	0.001028	
70	74.3	0.00084	
75	64.9	0.000733	
80	56.5	0.000638	
85	52.3	0.000591	
90	46	0.00052	
95	41.9	0.000473	
100	39.7	0.000449	
105	36.6	0.000414	
110	35.6	0.000402	
115	34.5	0.00039	
120	33.5	0.000379	
125	31.4	0.000355	
130	30.3	0.000342	
135	30.3	0.000342	
140	29.3	0.000331	
145	29.3	0.000331	
150	28.2	0.000319)
155	28.2	0.000319)
160	27.2	. 0.000307	
165	28.2	0.000319)
170	27.2	0.000307	
175	26.1	0.000295	5
180	26.1	0.000295	5
185	26.1	0.000295	5
190	26.1	0.000295	5
200	26.1	0.000295	5
210	26.1	0.000295	5
220	26.1	0.000295	5

Table	A.12	Field	Data	for	Experimental	Run1
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Time/sec	Conc. of TCE/PPMv	Ft^3 organic/min	Experiment type,
		(flowrate)	Air flow
230	25.1	0.000284	
240	25.1	0.000284	
250	25.1	0.000284	
260	25.1	0.000284	
270	25.1	0.000284	
280	25.1	0.000284	
290	25.1	0.000284	
300	25.1	0.000284	
330	25.1	0.000284	
360	25.1	0.000284	
390	24	0.000271	
420	24	0.000271	
450	24	0.000271	
480	24	0.000271	
510	24	0.000271	
570	24	0.000271	
630	24	0.000271	
690	23	0.00026	
750	23	0.00026	
930	23	0.00026	
1230	21.9	0.000247	
1530	21.9	0.000247	
1830	21.9	0.000247	
2130	23	0.00026	
2135	23	0.00026	No Sound,
2140	25.1	0.000284	11.3 SCFM
2145	24	0.000271	
2150	24	0.000271	
2155	24	0.000271	
2160	24	0.000271	
2165	24	0.000271	
2170	24	0.000271	
2175	24	0.000271	
2180	24	0.000271	
2185	24	0.000271	
2190	24	0.000271	
2195	24	0.000271	
2200	24	0.000271	
2230	24	0.000271	

Table A.12 Field Data for Experimental Run1(Continued)
Time/sec	Conc. of TCE/PPMv	Ft^3 organic/min	Experiment type,
		(flowrate)	Air flow
2260	24	0.000271	
2290	24	0.000271	
2320	23	0.00026	
2350	21.9	0.000247	
2380	20.9	0.000236	
2410	20.9	0.000236	
2440	20.9	0.000236	
2470	19.8	0.000224	
2500	19.8	0.000224	
2530	19.8	0.000224	
2560	19.8	0.000224	
2590	19.8	0.000224	
2620	19.8	0.000224	
2650	18.8	0.000212	
2680	18.8	0.000212	
2710	18.8	0.000212	
3010	19.8	0.000224	
3310	19.8	0.000224	
3610	17.8	0.000201	
3910	16.7	0.000189	
4210	16.7	0.000189	
4510	16.7	0.000189	
4810	16.7	0.000189	
5110	16.7	0.000189	
5115	16.7	0.000189	Sound,
5120	16.7	0.000189	11.3 SCFM
5125	16.7	0.000189	
5130	16.7	0.000189	
5135	16.7	0.000189	
5140	16.7	0.000189	
5145	5 17.8	0.000201	
5150) 17.8	0.000201	
5155	5 17.8	0.000201	
5160	17.8	0.000201	
5165	5 17.8	0.000201	
5170) 17.8	0.000201	
5180) 17.8	0.000201	
5190) 17.8	0.000201	
5200) 17.8	0.000201	

Table A.12 Field Data for Experimental Run1 (Continued)

Time/sec	Conc. of TCE/PPMv	Ft ³ organic/min	Experiment type.
		(flowrate)	Air flow
5230	18.8	0.000212	
5260	18.8	0.000212	
5290	18.8	0.000212	
5320	18.8	0.000212	
5380	18.8	0.000212	
5440	19.8	0.000224	
5500	19.8	0.000224	
5560	19.8	0.000224	
5620	19.8	0.000224	
5680	19.8	0.000224	
5740	20.9	0.000236	
5800	20.9	0.000236	
5860	20.9	0.000236	
5920	20.9	0.000236	
6220	20.9	0.000236	
6520	20.9	0.000236	
6820	20.9	0.000236	
7120	21.9	0.000247	
7420	21.9	0.000247	
7720	21.9	0.000247	
8020	21.9	0.000247	
8030	21.9	0.000247	No Sound,
8040	21.9	0.000247	11.3 SCFM
8050	21.9	0.000247	
8060	23	0.00026	
8070	23	0.00026	
8080	23	0.00026	
8090	23	0.00026	
8100	23	0.00026	
8110	21.9	0.000247	
8120	21.9	0.000247	
8130	21.9	0.000247	
8160	20.9	0.000236	
8190	19.8	0.000224	
8220	19.8	0.000224	
8250	19.8	0.000224	
8280	19.8	0.000224	•
8310	19.8	0.000224	
8370	19.8	0.000224	

Table A.12 Field Data for Experimental Run1 (Continued)

Time/sec	Conc. of TCE/PPMv	Ft ³ organic/min	Experiment type,
		(flowrate)	Air flow
8430	18.8	0.000212	
8490	18.8	0.000212	
8550	19.8	0.000224	
8610	18.8	0.000212	
8670	19.8	0.000224	
8970	16.7	0.000189	
9270	16.7	0.000189	
9570	15.7	0.000177	
9870	15.7	0.000177	End

Experimental Run 2 : Saturday November 7, 1998.

Sound- No Sound- Sound- No Sound

Flow meter	Outlet	Inlet	Reading/SCFM		
	pressure/psi	pressure/psi			
1	30-40	95-115	5		
2	25-30	95-115	5		
3	95-115	95-115	0		
4	95-115	95-115	0		
N°-3-4 E 752		0 (0 1)			

 Table A.13 Flow Manifold Data

Electronic flowmeter reading: 12.1 scfm (Sound)

Table	A.14	Flow	Manifold	Data
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Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 10.1 scfm(Air)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi	_	
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Table A.15 Flow Manifold Data

Electronic flowmeter reading: 12.5 scfm (Sound)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 10.7scfm (Air)

CummulativeTime/sec	Conc. of TCE (PPMV)	Ft^3 organic/min (flowrate)	Experiment type, Air flow
0	31.8	0.00038478	Sound1,
5	24.2	0.00029282	12.1scfm
10	24.6	0.00029766	
15	23.3	0.00028193	
20	24.2	0.00029282	
25	22.3	0.00026983	
30	23.7	0.00028677	
35	23.3	0.00028193	
40	23.7	0.00028677	
45	22.8	0.00027588	
50	24.2	0.00029282	
55	23.3	0.00028193	
60	23.7	0.00028677	
65	23.3	0.00028193	
70	23.7	0.00028677	
75	23.3	0.00028193	
80	23.7	0.00028677	
85	22.8	0.00027588	
95	23.3	0.00028193	

Table.	A.17	Field	Data	for	Experimental	Run	2
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CummulativeTime/sec	Conc. of	Ft^3 organic/min	Experiment type,
	TCE/PPMv	(flowrate)	Air flow
105	23.3	0.00028193	
115	23.7	0.00028677	
125	23.7	0.00028677	· · · · · · · · · · · · · · · · · · ·
135	23.7	0.00028677	
145	23.7	0.00028677	
155	23.7	0.00028677	
165	23.7	0.00028677	
175	23.7	0.00028677	
195	24.2	0.00029282	
215	24.6	0.00029766	
235	25.1	0.00030371	
255	25.5	0.00030855	
275	25.5	0.00030855	
295	25.5	0.00030855	
315	25.5	0.00030855	
335	26	0.0003146	
355	26	0.0003146	
385	26.9	0.00032549	
415	27.3	0.00033033	
445	27.3	0.00033033	
475	27.8	0.00033638	
505	28.2	0.00034122	
535	28.2	0.00034122	
565	28.7	0.00034727	
595	28.7	0.00034727	
625	28.7	0.00034727	
655	29.1	0.00035211	
715	29.1	0.00035211	
775	29.1	0.00035211	
835	29.1	0.00035211	
895	30	0.000363	
955	30	0.000363	
1015	30.5	0.00036905	
1075	30.9	0.00037389	
1135	30.9	0.0003/389	
1195	30.9	0.00037389	
1255	30.9	0.0003/389	
1315	31.4	0.0003/994	•
1375	30.9	0.00037385	1

CummulativeTime/sec	Conc. of	Ft^3 organic/min	Experiment type,
	TCE/PPMv	(flowrate)	Air flow
1435	30.9	0.00037389	
1495	30.9	0.00037389	
1555	30.9	0.00037389	
1615	30.9	0.00037389	
1675	30.9	0.00037389	
1735	31.4	0.00037994	
2155	32.8	0.00039688	
2455	31.4	0.00037994	
2755	32.8	0.00039688	
3055	32.8	0.00039688	
3360	32.8	0.00039688	No Sound,
3365	31.4	0.00031714	10.1scfm
3370	35.5	0.00035855	
3375	37.7	0.00038077	
3380	39.1	0.00039491	
3385	40.4	0.00040804	
3390	40.4	0.00040804	
3395	40.4	0.00040804	
3400	40	0.000404	
3405	40	0.000404	
3410	39.5	0.00039895	
3415	38.6	0.00038986	
3420	38.6	0.00038986	
3425	38.2	0.00038582	
3430	37.7	0.00038077	
3435	37.3	0.00037673	
3440	37.3	0.0003/6/3	
3445	36.4	0.00036764	
3450	35.9	0.00036259	
3455	35.5	0.00035855	
3460	35	0.0003535	
3465	34.6	0.00034946	
3475	33.7	0.00034037	
3485		0.00033532	
3495		0.00033532	
3505	32.8	0.00033128	
3515		0.00032623	)
3525	32.3	0.00032623	
3535	31.8	0.00032118	5

CummulativeTime/sec	Conc. of	Ft^3 organic/min	Experiment type.
an contra materia hage providente della contra della	TCE/PPMv	(flowrate)	Air flow
3545	31.8	0.00032118	· · · · · · · · · · · · · · · · · · ·
3555	31.4	0.00031714	
3565	31.4	0.00031714	
3575	31.4	0.00031714	
3585	30.5	0.00030805	
3595	30.5	0.00030805	
3605	30	0.000303	
3615	30	0.000303	
3625	30	0.000303	
3645	29.6	0.00029896	
3665	29.6	0.00029896	
3685	29.6	0.00029896	
3705	29.1	0.00029391	
3725	29.1	0.00029391	
3745	29.1	0.00029391	
3765	29.1	0.00029391	
3785	29.1	0.00029391	
3805	29.1	0.00029391	
3835	29.1	0.00029391	
3865	29.1	0.00029391	
3895	28.7	0.00028987	
3925	28.7	0.00028987	
3955	28.7	0.00028987	
3985	28.7	0.00028987	
4045	29.1	0.00029391	
4105	29.1	0.00029391	
4165	29.6	0.00029896	
4225	30	0.000303	
4285	30	0.000303	
4345	29.6	0.00029896	
4405	29.6	0.00029896	
4465	30	0.000303	
4525	29.6	0.00029896	
4585	29.6	0.00029896	
4645	30	0.000303	
4705	30	0.000303	
4765	30	0.000303	
4825	29.6	0.00029896	

CummulativeTime/sec	tiveTime/sec Conc. of		Experiment type,	
	TCE/PPMv	(flowrate)	Air flow	
4885	29.6	0.00029896		
5185	30	0.000303		
5485	29.6	0.00029896		
5785	29.6	0.00029896		
6385	29.6	0.00029896		
6985	30	0.000375	sound2,	
6995	27.2	0.00034	12.5scfm	
7005	28.2	0.0003525		
7015	28.2	0.0003525		
7025	28.2	0.0003525		
7035	28.7	0.00035875		
7045	29.1	0.00036375		
7055	29.1	0.00036375		
7065	29.1	0.00036375		
7095	28.7	0.00035875		
7125	29.1	0.00036375		
7155	29.6	0.00037		
7185	29.6	0.00037		
7215	30	0.000375		
7245	30	0.000375		
7305	30.5	0.00038125		
7365	30.5	0.00038125		
7425	30.5	0.00038125		
7545	31.4	0.0003925		
7665	31.8	0.0003975		
7785	31.8	0.0003975		
7905	31.8	0.0003975		
8205	32.3	0.00040375		
8505	32.8	0.00041		
8805	32.8	0.00041		
9105	33.2	0.000415		
9405	33.2	0.000415		
9705	33.2	0.000415		
10005	33.2	0.000415		
10015	38.6	0.00041302	No Sound,	
10025	38.2	0.00040874	10.7scfm	
10035	38.2	0.00040874	•	
10045	37.7	0.00040339	)	
10055	36.8	0.00039376		

CummulativeTime/sec Conc. of		Ft^3 organic/min	Experiment type,
	TCE/PPMv	(flowrate)	Air flow
10065	36.8	0.00039376	
10085	36.4	0.00038948	
10105	35.9	0.00038413	
10125	35.5	0.00037985	
10145	35	0.0003745	
10165	35	0.0003745	
10185	39.6	0.00042372	
10245	39.1	0.00041837	
10305	33.2	0.00035524	
10365	33.2	0.00035524	
10425	32.8	0.00035096	
10725	31.4	0.00033598	
11025	30.9	0.00033063	
11325	30.9	0.00033063	
11625	30.9	0.00033063	
11925	30.9	0.00033063	
12225	30.9	0.00033063	
12525	30.9	0.00033063	End

Experimental Run 3 : Saturday November 14, 1998.

Air-Sound-Air-Sound

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Table A.18 Flow Ma	anifold Data
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Electronic flowmeter reading: 9.5 scfm(Air1)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi	_	
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Table A.19 Flow Manifold Data

Electronic flowmeter reading: 12.8 scfm (Sound)

T	able	A.20Fl	ow l	Manifold	Data

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 10.4 scfm(Air)

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Electronic flowmeter reading: 12.5 scfm (Sound)

Table A.22 Field Data for Experimental Run 3

time/sec	Conc. of	ft^3 organic/min	Type of experiment,
	ICE/FFMV	(IIUWIALE)	nowiate
0	59.1	0.000561	No Sound1, 9.5 scfm
5	48.2	0.000458	
10	40.4	0.000384	
15	36.7	0.000349	
25	33.1	0.000314	
35	31	0.000295	
45	29.4	0.000279	
55	27.3	0.000259	
65	25.8	0.000245	
75	25.2	0.000239	
85	24.7	0.000235	

time/sec	Conc. of TCE/PPMv	ft^3 organic/min (flowrate)	Type of experiment, flowrate
95	24.2	0.00023	
105	23.7	0.000225	
115	24.7	0.000235	
125	23.7	0.000225	
135	23.7	0.000225	
145	24.2	0.00023	
155	24.7	0.000235	
165	23.7	0.000225	
175	23.2	0.00022	
185	23.7	0.000225	
195	23.2	0.00022	
205	22.1	0.00021	
215	22.1	0.00021	
225	23.2	0.00022	
235	22.6	0.000215	
245	21.6	0.000205	
255	22.1	0.00021	
265	21.6	0.000205	
275	20.5	0.000195	
285	21.1	0.0002	
295	21.1	0.0002	
305	20.5	0.000195	
335	19.5	0.000185	
365	19.5	0.000185	
395	20.5	0.000195	
425	19	0.000181	
455	17.9	0.00017	
485	19	0.000181	
515	20	0.00019	
545	21.1	0.0002	
575	21.1	0.0002	
605	21.1	0.0002	
635	20	0.00019	
665	19	0.000181	
695	18.5	0.000176	
725	20	0.00019	)
755	20.5	0.000195	8
815	19.5	0.000185	)
875	5 18.5	0.000176	) 

time/sec	Conc. of	ft^3 organic/min	Type of experiment,
	ICE/PPMv	(flowrate)	flowrate
935	17.9	0.00017	
995	18.5	0.000176	
1055	18.5	0.000176	
1235	18	0.000171	
1415	17.9	0.00017	
1595	17.9	0.00017	
1775	18.5	0.000176	
2375	17.9	0.00017	
2975	17.9	0.00017	
2985	20.8	0.000266	Sound1, 12.8scfm
2995	21.2	0.000271	
3005	21.1	0.00027	
3015	20.5	0.000262	
3025	20	0.000256	
3035	20	0.000256	
3045	19.5	0.00025	
3055	20	0.000256	
3065	20	0.000256	
3075	19.5	0.00025	
3085	19.5	0.00025	
3095	19	0.000243	
3105	19	0.000243	
3115	19	0.000243	
3125	19	0.000243	
3135	19	0.000243	
3145	19	0.000243	
3155	19	0.000243	
3165	19	0.000243	
3175	19.5	0.00025	
3185	19.5	0.00025	
3195	19.5	0.00025	
3210	19	0.000243	
3240	19	0.000243	
3270	19.5	0.00025	
3300	19.5	0.00025	
3330	19.5	0.00025	
3360	19.5	0.00025	
3390	19.5	0.00025	
3420	19.5	0.00025	
3450	19.5	0.00025	

time/sec	Conc. of TCF/PPMv	ft^3 organic/min (flowrate)	Type of experiment,
3480	19.5	0.00025	nowiate
3510	19.5	0.00025	
3540	19	0.000243	
3570	20	0.000256	
3600	19.5	0 00025	
3630	19.5	0.00025	
3660	19.5	0.00025	
3690	19.5	0.00025	
3720	19	0.000243	
3750	19.5	0.00025	
3780	19.5	0.00025	
3810	20	0.000256	
3870	19.5	0.00025	
3930	19.5	0.00025	
3990	19.5	0.00025	
4050	19.5	0.00025	
4110	19.5	0.00025	
4170	19.5	0.00025	
4230	19.5	0.00025	
4290	20	0.000256	
4350	19.5	0.00025	
4410	19.5	0.00025	
4470	19.5	0.00025	
4770	19.5	0.00025	
5070	20	0.000256	
5370	20	0.000256	
5670	20	0.000256	
5970	20	0.000256	
5975	21	0.000218	No Sound 2, 10.4scfm
5980	21	0.000218	
5985	21	0.000218	
5990	21	0.000218	
5995	21	0.000218	
6000	21	0.000218	
6005	21.6	0.000225	
6010	21.6	0.000225	
6015	21.6	0.000225	
6020	21.1	0.000219	1
6025	5 21.1	0.000219	

time/sec	Conc. of	ft^3 organic/min	Type of experiment.
	TCE/PPMv	(flowrate)	flowrate
6030	21.1	0.000219	
6035	21.6	0.000225	
6040	21.1	0.000219	
6045	21.1	0.000219	
6050	21.1	0.000219	
6055	21.1	0.000219	
6060	20.5	0.000213	
6065	20.5	0.000213	
6070	20.5	0.000213	
6075	20	0.000208	
6080	20	0.000208	
6110	20	0.000208	
6140	20	0.000208	
6170	20	0.000208	
6200	19.5	0.000203	
6230	19.5	0.000203	
6260	19.5	0.000203	
6290	19	0.000198	
6320	19	0.000198	
6350	19	0.000198	
6410	19	0.000198	
6470	19	0.000198	
6530	18.5	0.000192	
6590	18.5	0.000192	
6650	18.5	0.000192	
6710	18.5	0.000192	
7310	17.9	0.000186	
7910	17.9	0.000186	
8510	17.9	0.000186	
8515	20.1	0.000251	Sound2,12.5scfm
8520	20.1	0.000251	
8525	20.1	0.000251	
8530	19.5	0.000244	
8535	19.5	0.000244	
8540	19.5	0.000244	
8550	19.5	0.000244	
8560	18.5	0.000231	
8570	18.5	0.000231	
8580	18.5	0.000231	

time/sec	Conc. of	ft^3 organic/min	Type of experiment,
0.500	ICE/PPMv	(flowrate)	flowrate
8590	18	0.000225	
8600	18	0.000225	
8610	17.9	0.000224	
8620	17.9	0.000224	
8630	17.9	0.000224	
8660	17.9	0.000224	
8690	17.9	0.000224	
8720	17.9	0.000224	
8750	17.9	0.000224	
8780	17.9	0.000224	
8810	18.5	0.000231	
8840	18.5	0.000231	
8870	18.5	0.000231	
8900	18.5	0.000231	
8930	18.5	0.000231	
8990	19	0.000238	
9050	19	0.000238	
9110	19	0.000238	
9170	19	0.000238	
9230	19	0.000238	
9290	19	0.000238	
9350	19.5	0.000244	
9410	19.5	0.000244	
9710	20	0.00025	
10010	20	0.00025	
10310	20.5	0.000256	
10610	20.5	0.000256	
10910	21.1	0.000264	
11510	21.1	0.000264	
12110	21.1	0.000264	End

### Experimental Run 4 : Sunday November 15, 1998.

#### Sound-Air-Sound-Air

Flow meter	Outlet	Inlet	Reading/SCFM	l
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	
Electronic flour	actor readings 11 (	action (Cound)	- J	I

 Table A.23 Flow Manifold Data

Electronic flowmeter reading: 11.0 scfm (Sound)

#### Table A.24 Flow Manifold Data

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 10.0 scfm(Air)

Table A.25 H	Flow Mar	ifold Data
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Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Electronic flowmeter reading: 11.3 scfm (Sound)

Table A.26 Flow Manifold Data

		7		
Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 9.6 scfm (Air2)

time/sec	Conc. of	ft^3 organic/min	Experiment Type, flowrate
Ω	226	0.000240	Sound1 11 Ocofm
5 5	22.0	0.000249	Sound I, I I.USCIM,
10	22.0	0.000249	
10	21.1	0.000243	
20	21.0	0.000238	
25	21.0	0.000238	
30	21.0	0.000230	
35	21.0	0.000230	
40	21.0	0.000230	
40	21.1	0.000232	
	21.1	0.000232	
65	21.1	0.000232	
75	21.1	0.000232	
85	21.1	0.000232	
05		0.000232	
105	20.5	0.000232	
105	20.0	0.000220	
100		0.00022	
100		0.000230	
190		0.000230	
200		0.000230	
313		0.000230	
3/3		0.000230	
430		0.000232	
/ 30		0.000232	
1335		0.000232	
1935		0.000232	No Sound1 10cofm
1945		0.000215	No Sound 1, Tuscim
1955		0.000203	
1965	20.5	0.000203	)
1975	20.5	0.000203	)
1985	20.5		
1995		0.0002	
2005			
2015	20.5		
2028	$\frac{20}{20}$		-
2035	$\frac{20}{100}$		-
2045	19.5		2
2055	19.5		2
2065	5 19.8	0.000195	
2075	5  20	0.0002	-

 Table A.27 Field Data for Experimental Run 4

time/sec	Conc. of	ft^3 organic/min	Experiment Type, flowrate
	TCE/PPMv	(flowrate)	
2085	20	0.0002	
2095	19.5	0.000195	
2125	21.1	0.000211	
2155	20	0.0002	
2185	19	0.00019	
2215	19	0.00019	
2245	17.9	0.000179	
2275	17.9	0.000179	
2305	17.9	0.000179	
2335	17.9	0.000179	
2395	16.9	0.000169	
2455	15.3	0.000153	
2515	15.9	0.000159	
2575	16.9	0.000169	
2635	15.9	0.000159	
2695	15.9	0.000159	
2755	16.9	0.000169	
2815	15.3	0.000153	
2875	14.8	0.000148	
2935	15.3	0.000153	
2995	15.9	0.000159	
3055	15.9	0.000159	
3115	15.9	0.000159	
3175	15.9	0.000159	
3235	16.4	0.000164	
3295	15.9	0.000159	
3355	15.9	0.000159	
3415	15.9	0.000159	
3475	14.9	0.000149	
3535	15.9	0.000159	
3595	15.9	0.000159	
3655	15.9	0.000159	
3715	16.4	0.000164	
3775	15.9	0.000159	
3835	15.9	0.000159	
3840	15.3	0.000173	Sound2,11.3 scfm
3845	14.8	0.000167	
3850	14.8	0.000167	
3855	14.8	0.000167	
3860	14.8	0.000167	

time/sec	Conc. of	ft^3 organic/min	Experiment Type, flowrate
3865	15.2		
3870	11.0		
3875	14.0	0.000167	
3880	14.0	0.000167	
3885	14.0	0.000167	
3895	14.0	0.000107	
3905	15.3	0.000173	
3915	15.3	0.000173	
3945	15.3	0.000173	
3975	10.0	0.000173	
4005	14.0	0.000167	
4005	14.0	0.000107	
4065	15.0	0.000107	
4000	16.0	0.000173	
4000	16.4	0.000185	
4185	17.4	0.000103	
4245	17.0	0.000202	
4305	17.0	0.000202	
4365	16.9	0.000202	
4425	17.9	0.000202	
4485	19.9	0.000225	
4545	17.9	0.000202	
4605	17.9	0.000202	
4665	17.9	0.000202	
4965	19.9	0.000225	
5265	18.9	0.000214	
5565	19.9	0.000225	
5865	19.9	0.000225	
5870	20	0.000192	No Sound 2, 9.6 scfm
5875	20	0.000192	
5880	20	0.000192	
5885	19.5	0.000187	
5890	19.5	0.000187	
5895	19.5	0.000187	
5900	19.5	0.000187	
5905	19.5	0.000187	
5910	19.5	0.000187	
5915	19.5	0.000187	
5920	19.5	0.000187	

 

 Table A.27 Field Data for Experimental Run 4 (Continued)

time/sec	Conc. of	ft^3 organic/min	Experiment Type,
	ICE/PPINIV	(flowrate)	flowrate
5930	19.5	0.000187	
5940	19.5	0.000187	
5950	19.5	0.000187	
5960	19.5	0.000187	
5970	19.5	0.000187	
5980	19.5	0.000187	
5990	19.5	0.000187	
6000	19	0.000182	
6010	19.5	0.000187	
6020	19.5	0.000187	
6030	19.5	0.000187	
6060	19.5	0.000187	
6090	19	0.000182	
6120	18.5	0.000178	
6150	17.9	0.000172	
6180	17.9	0.000172	
6210	17.9	0.000172	
6240	17.9	0.000172	
6270	17.9	0.000172	
6300	17.9	0.000172	
6330	17.4	0.000167	
6630	15.3	0.000147	
6930	16.4	0.000157	
7230	15.9	0.000153	
7530	15.9	0.000153	
7830	16.4	0.000157	
8130	15.9	0.000153	End

#### Additional Experiment : Sunday November 8, 1998.

#### Air-Sound-Air-Sound-Air

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi	_	
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

#### Table A.28 Flow Manifold Data

Electronic flowmeter reading: 12.3 scfm(Air)

Vacuum Pump Reading: 27 inches of water vacuum

Table.	A.29	Flow	Manifold	Data
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Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Electronic flowmeter reading: 12.3scfm (Sound)

Vacuum Pump Reading: 20 inches of water vacuum

Table A.30 Flow Manifold Data

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	
4	0	95-115	5	

Electronic flowmeter reading: 12.3scfm(Air)

Vacuum Pump Reading: 27 inches of water vacuum

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi		
1	30-40	95-115	5	
2	25-30	95-115	5	
3	95-115	95-115	0	
4	95-115	95-115	0	

Table A.31 Flow Manifold Data

Electronic flowmeter reading: 13.6 scfm (Sound) Vacuum Pump Reading: 20 inches of water vacuum

Flow meter	Outlet	Inlet	Reading/SCFM	
	pressure/psi	pressure/psi	_	
1	90-115	95-115	0	
2	90-115	95-115	0	
3	0	95-115	5	**************************************
4	0	95-115	5	

Table A.32 Flow Manifold Data

Electronic flowmeter reading: 10.5 scfm(Air) Vacuum Pump Reading: 20 inches of water vacuum

Cummulative	Conc. of	ft^3 organic/min	Type of Experiment,
time/seconds	TCE PPMv	(flowrate)	Flowrate
(	140	0.001722	No Sound1, 12.3scfm
1(	277	0.003407	
20	121	0.001488	
30	88.9	0.001093	
4(	80.3	0.000988	
5(	79.4	0.000977	
60	73.9	0.000909	
7(	67.7	0.000833	
80	64.4	0.000792	
90	60.8	0.000748	
10	58.5	0.00072	
11	56.3	0.000692	
12	54	0.000664	
13	51.8	0.000637	
14	49.5	0.000609	
15	48.1	0.000592	
16	46.3	0.000569	
17	0 45	0.000554	
18	43.2	0.000531	
19	0 43.2	0.000531	
20	0 41.8	0.000514	
22	0 40.4	0.000497	
24	38.6	0.000475	

Table A. 33 Field Data for Additional Experiment

Cummulative	Conc. of	ft^3 organic/min	Type of Experiment.
time/seconds	TCE/ PPMv	(flowrate)	Flowrate
260	37.3	0.000459	
280	37.7	0.000464	
300	36.4	0.000448	
320	35.5	0.000437	
340	34.6	0.000426	
360	34.1	0.000419	
390	33.7	0.000415	
420	33.2	0.000408	
450	32.8	0.000403	
480	32.8	0.000403	
510	32.8	0.000403	
540	32.8	0.000403	
570	33.2	0.000408	
630	32.8	0.000403	
690	31.8	0.000391	
750	31.4	0.000386	
810	31.4	0.000386	
870	31.4	0.000386	
930	30.5	0.000375	
990	30.5	0.000375	
1050	30.5	0.000375	
1110	30.9	0.00038	
1170	30.5	0.000375	
1470	30.5	0.000375	
1770	30.5	0.000375	
2070	29.1	0.000358	
2370	30.5	0.000375	
2380	53	0.000652	sound1,12.3 scfm
2390	30	0.000369	
2400	29.6	0.000364	
2410	29.1	0.000358	
2420	29.1	0.000358	
2430	28.7	0.000353	
2440	28.7	0.000353	
2450	30	0.000369	
2460	27.3	0.000336	
2470	25.5	0.000314	
2480	26	0.00032	
2490	28.2	0.000347	

 Table A. 33 Field Data for Additional Experiment (Continued)

Cummulative		Conc. of	ft^3 organic/min	Type of Experiment,
time/seconds		TCE PPMv	(flowrate)	Flowrate
25	00	28.2	0.000347	
25	10	28.2	0.000347	
25	20	28.2	0.000347	
25	30	28.7	0.000353	
25	40	28.2	0.000347	
25	60	27.8	0.000342	
25	80	28.2	0.000347	
26	00	28.2	0.000347	
26	20	29.1	0.000358	
26	40	30.5	0.000375	
26	60	30	0.000369	
26	80	29.6	0.000364	
27	00	29.1	0.000358	
27	30	28.7	0.000353	
27	60	28.7	0.000353	
27	90	29.6	0.000364	
28	20	27.8	0.000342	
28	50	27.3	0.000336	
28	80	27.3	0.000336	
29	10	28.7	0.000353	
29	40	30.5	0.000375	
29	70	30.5	0.000375	
30	000	29.6	0.000364	
30	)30	29.1	0.000358	
30	60	28.7	0.000353	
31	20	27.3	0.000336	
3*	80	26.9	0.000331	
32	240	27.8	0.000342	
33	300	28.2	0.000347	
36	300	30	0.000369	
39	900	30	0.000369	
42	200	30.5	0.000375	
4	500	30.9	0.00038	
4	300	30	0.000369	
5	100	31.4	0.000386	
5	400	30.5	0.000375	
54	41(	32.3	0.000397	No Sound 2,12.3 scfm
5.	420	32.3	0.000397	
5.	430	31.8	0.000391	L

# Table A. 33 Field Data for Additional Experiment (Continued)

Cummulative	Conc. of	ft^3 organic/ min	Type of Experiment
time/seconds	TCE/PPMv	(flowrate)	Flowrate
544	0 31.8	0.000391	
545	0 31.8	0.000391	
546	0 31.4	0.000386	
547	0 31.4	0.000386	
548	0 31.4	0.000386	
549	0 31.4	0.000386	
550	0 31.4	0.000386	
551	0 31.4	0.000386	
552	0 30.9	0.00038	
553	0 30.9	0.00038	
554	0 30.9	0.00038	
555	0 30.9	0.00038	
556	0 31.4	0.000386	
558	0 31.4	0.000386	
560	0 30.9	0.00038	
562	0 30.5	0.000375	
564	0 30.5	0.000375	
566	0 30.5	0.000375	
568	0 30.5	0.000375	
571	0 30	0.000369	
574	0 30	0.000369	
577	0 30	0.000369	
580	0 29.6	0.000364	
583	0 29.6	0.000364	
586	0 29.6	0.000364	
592	0 30	0.000369	
598	0 30	0.000369	-
604	0 30	0.000369	
610	0 29.1	0.000358	
616	0 29.6	0.000364	
622	0 29.6	0.000364	
628	0 30	0.000369	
634	0 30	0.000369	nationana ar ist Audoren waar i Nikolaana in Things nar ar bindu ar ar 18 dit ar 18 dit ar 18 dit ar 18 dit ar
664	0 30.5	0.000375	anan kanan atari katalan matukan kana katara katara katara katara katara di katara di katara di katara di kata
694	0 30.5	0.000375	
724	0 30	0.000369	
754	0 30	0.000369	
755	0 42.7	0.000581	sound2, 13.6 sctm
756	0 30.9	0.00042	

## Table A. 33 Field Data for Additional Experiment (Continued)

Cummulative	Conc. of	ft^3 organic /min	Type of Experiment.
time/seconds	TCE/PP <b>M</b> v	(flowrate)	Flowrate
7570	30	0.000408	
7580	29.6	0.000403	
7590	29.1	0.000396	, 894 <u></u>
7600	28.7	0.00039	
7610	28.2	0.000384	
7620	28.7	0.00039	an an an an Anna Anna Anna Ann
7630	28.2	0.000384	
7640	28.2	0.000384	
7650	28.2	0.000384	
7680	28.2	0.000384	
7710	27.8	0.000378	
7740	28.8	0.000392	
7770	28.2	0.000384	
7800	28.2	0.000384	
7830	28.2	0.000384	
7860	28.2	0.000384	
7890	28.2	0.000384	
7920	28.7	0.00039	
7950	28.7	0.00039	
7980	28.7	0.00039	
8010	28.7	0.00039	
8070	28.7	0.00039	
8130	29.1	0.000396	
8190	29.1	0.000396	
8250	29.6	0.000403	
8310	29.1	0.000396	
8370	29.1	0.000396	
8430	29.1	0.000396	
8490	29.1	0.000396	
8550	28.7	0.00039	
8610	29.1	0.000396	
8670	29.1	0.000396	
8730	29.1	0.000396	
8790	28.7	0.00039	
9090	28.2	0.000384	
9390	28.2	0.000384	
9690	27.8	0.000378	
9700	29.1	0.000306	No Sound 3, 10.5
9710	30	0.000315	

# Table A. 33 Field Data for Additional Experiment (Continued)

Cummulative	Conc. of	ft^3 organic/min	Type of Experiment,
time/seconds	TCE/PPMv	(flowrate)	Flowrate
9720	30.5	0.00032	
9730	30.5	0.00032	
9740	30.5	0.00032	
9750	30.5	0.00032	
9760	30.5	0.00032	
9770	20.5	0.000215	
9780	20.1	0.000211	
9810	20.1	0.000211	
9840	19.6	0.000206	
9870	19.6	0.000206	
9900	19.2	0.000202	
9930	19.2	0.000202	
9960	19.2	0.000202	
9990	19.2	0.000202	
10020	19.2	0.000202	
10050	19.2	0.000202	
10080	19.2	0.000202	
10110	19.2	0.000202	
10140	19.2	0.000202	
10170	19.2	0.000202	
10230	18.7	0.000196	
10290	18.3	0.000192	
10350	18.3	0.000192	
10410	17.8	0.000187	
10470	17.8	0.000187	
10770	17.8	0.000187	
11070	17.8	0.000187	
11370	16.9	0.000177	
11670	16.9	0.000177	End

 Table A. 33 Field Data for Additional Experiment (Continued)

Table A.34 gives a summary of some of the results obtained from the first part of the field preliminary studies. It shows the pressure through the sonic device at a flow rate of 5 SCFM and 11 SCFM. Only one whistle is used for this study.

Table A.34One Whistle Study

Flowrate/SCFM	Pressure/psi		
5	32-40		
11	105		

Table A.35 gives a summary of the rest of the results obtained from the first part of the field preliminary studies. It shows the removal rates of some of the samples that were analyzed by the GC (for 3 runs). For each sample a brief description of the run and the contaminants detected are shown. The last two rows of Table A.35 (run 3) also show a comparison of the GC results and the PID results, for a point chosen randomly during the preliminary study.

Contaminant	Run	Description of run	GC/PPMv	
DCE	1	Vacuum pump running	12 PPMv	
TCE	1	Vacuum pump running	700 PPMv	
DCE	2	Vacuum Pump stopped	2 PPMv	
ТСЕ	2	Vacuum Pump Stopped	417 PPMv	
			PID/PPMv	GC/PPMv
TCE	3	point during experiments	250	211.8
DCE	3	point during experiments	0	0

**Table A.35** GC Data Obtained for a Particular Day

#### **APPENDIX B**

#### **CALCULATIONS**

#### **Removal Rate of Organic:**

Calculation of removal rate of organic: (based on first data point in Experimental Run 1)

Removal Rate of Organic (cubic feet of organic/ minute) = (cubic feet of organic/cubic feet air)* (cubic feet air/minute)

Sample Calculation: 284 PPMV = 284 (cubic feet of organic/ 10^6 *cubic feet air)

Removal Rate of Organic = 284 (cubic feet of organic/ 10^6 *cubic feet air) *11.3 (cubic feet air / minute) = 0.003209 cubic feet of organic/ minute

Statistics:

Concentration	Removal Rate	Concentration	Removal Rate
Percent Improvement/	Percent Improvement /	Percent Improvement ²	Percent Improvement ²
PPMv	Cubic Feet Per Minute	/PPMV ²	Cubic Feet Per Minute ²
34.8	34.9	1211.04	1218.01
33	33.3	1089	1108.89
9.1	31	82.81	961
7	25.1	49	630.01
11	49.7	121	2470.09
16.5	39.7	272.25	1576.09
33.3	46.5	1108.89	2162.25
22	43.2	484	1866.24
Sum = 166.7	Sum = 303.4	Sum = 4417.99	Sum = 11992.58
$Sum^2 = 27788.89$	Sum ² = 92051.56		

 Table B.1 Statistical Treatment of the Average Percent Improvement

Standard Deviation for the Percent Improvement (based on Concentration/PPMv):

Standard Deviation =  $[(8 (4417.99) - 27788.89)/(8*7)]^{0.5} = 11.6$ 

Standard Deviation for the Percent Improvement (based on Cubic Feet per Minute):

Standard Deviation =  $[(8 (11992.58) - 92051.56)/(8*7)]^{0.5} = 8.333$ 

#### t-test and Confidence interval calculations:

For small samples (n < 30), the 95% confidence interval for the true mean of the population is given by:

 $X - t_{0.025}$ * standard deviation/(n) ^{0.5} < Mean <  $X + t_{0.025}$ * standard deviation/(n) ^{0.5} -----equation 6.1, (Freund, 1970).

Where: X = Sample Mean. n = Sample Size. Mean = true population mean.  $t_{0.025}$  = t distribution factor for 95% confidence interval  $t_{0.025}$  = 2.365 (for n = 8), (Freund, 1970). Standard deviation = Sample standard deviation.

From equation 6.1, the confidence interval for the true population mean is calculated as follows:

#### 95% Confidence interval for the mean Percent Improvement

( based on Concentration/PPMV):

 $20.8375-2.365*11.6/8^{0.5} < Mean < 20.8375+2.365*11.6/8^{0.5}$ 

11.138 < Mean < 30.5495

The probability is 0.95 that the interval 11.138 to 30.5495 will contain the true mean

percentage improvement.

#### 95% Confidence interval for the mean Percent Improvement

#### (based on Cubic Feet per Minute)

37.925 -2.365*8.333/ 8  $^{0.5}$  < Mean < 37.925+2.365*8.333/ 8  $^{0.5}$ 

30.95688< Mean < 44.89312

The probability is 0.95 that the interval 30.95688 to 44.89312 will contain the true mean percentage improvement.

#### **APPENDIX C**

### PARTS DESCRIPTION AND COSTS

## Table C.1 Assembly Parts for Ultrasonic Field Apparatus

Part	Description	Catalog	Quantity	Price/	Total
Number		No. McMaster		unit	Price
1	¹ / ₂ " × 1/8" Sch. 40 Hex	4638K636	4	\$0.46	\$1.84
	Bushing Galv.				
2	1/8" ID Pressure Gages	3847K2	4	\$7.21	\$28.84
3	1 3/8" Nipple ¾" Sch. 40 Galv.	4549K591	8	\$0.64	\$5.12
4	¹ /2 ^{°°} ×3/4 ^{°°} Sch.40 Bushing Galv.	4638K455	8	\$1.57	\$12.56
5	³ / ₄ " ID Flowmeter (2.4-24.0) SCFM	4111K33	4	\$314.6 5	\$1258. 60
6	½" Sch.40 Bronze Globe Valve	4695K53	4	\$21.90	\$87.60
7	¹ / ₂ " Sch. 40 Union Galv.	4638K733	4	\$3.34	\$13.36
8	³ / ₄ " Nipple ¹ / ₂ " Sch. 40 Galv.	4549K593	4	\$0.72	\$2.88
9	1"×1/2" Sch.40 Hex Bushing Galv.	4638K663	6	\$1.19	\$7.14
10	1" Sch. 40 Plug Galv.	4638K516	1	\$0.69	\$0.69
11	Twin Self-Extracting Nylon Air-Hose 25', ½"OD	5612K33	2	\$85.42	\$170.8 4
12	¹ / ₂ " Sch.40 Elbow 90° Galv.	4638K133	4	\$0.85	\$3.40
13	2" Nipple 1" Sch.40 Galv.	4549K611	4	\$0.93	\$3.72
14	½" Sch.40 Tee Galv.	4638K123	4	\$1.10	\$4.40
15	1" Sch.40 Union Galv.	4638K735	3	\$5.02	\$15.06
16	1" Sch.40 Tee Galv.	4638K125	4	\$2.86	\$11.44
17	2.5" Nipple 1" Sch.40 Galv.	4549K613	4	\$1.19	\$4.76

Part	Description	Catalog No.	Quantity	Price/unit	Total
Number		McMaster			Price
18	1/2" IPS Forged-Brass	45085K33	4	\$7.23	\$28.9
	Ball valves				2
19	¹ / ₂ " Sch.40 Female Plug	6534K76	8	\$3.91	\$31.2
	Hose Connector				8
20	¹ / ₂ " Sch.40 male Hose	6534K83	8	\$10.33	\$82.6
	Connection				4
22	¹ / ₂ "OD × ¹ / ₂ "ID Tubing-	5182K268	2	\$16.63	\$33.2
	to-Pipe Adapter				6
23	1" Sch.40 Tee Steel	4443K666	1	\$14.83	\$14.8
					3
24	1.5" Sch.40 Galv.	4549K661	1	\$3.10	\$3.10
25	$1' \frac{1}{4}$ " sch 40 union				
	1/01/00/07/11/11/	5100770.15			
26	1/2" OD Tubing Union	5182K345	4	\$12.66	\$50.6
27	D'and last	4(20)74((	1		4
27	Pipe adapter	4638K466	I		
	1" Sah 10 Latarala 150	442012264	1	\$19.62	¢10.6
20	Too Stool	44295304	1	\$18.02	\$18.0 2
20	$1"\times 11/4"$ Sob 40	16291/150		\$2.61	\$5.22
29	Bushing Galy	40368439	2	\$2.01	\$3.22
31	1' 1" perforated pipe	16388737		09.92	\$176
51	i i periorated pipe	40301(737	2	\$0.00	φ17.0 0
32	3' 11/4" Sch 40 Pine	4549K 664	6		
52		+3+9100+	0		
33	1/2" OD Teflon Tubing	52355K15	100	\$1.17	\$117
					00
35	4' Packer 1" Sch.40				
36	Ultrasonic Whistle				
	Array				
37	1' Packer 1" Sch.40				
Total	Total				\$
1					

## Table C.1 Assembly Parts for Ultrasonic Field Apparatus (Continued)

McMaster - Supply Co., 473 Ridge Rd., Dayton, NJ 08810-0317. Catalog 101.



Figure C.1 Flow Manifold for the Ultrasonic Experiment (Fernandez, 1997)



Figure C.2 Well-Head Assembly for Ultrasonic Field Test



Figure C.3 Bore-Hole Extension Pipe and Fittings

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Figure C.4a Ultrasonic Transducer (Whistle)

.



Figure C. 4b Detail Schematic of Whistle (Fernandez, 1997)



Figure C.5 Bore-Hole Extension Pipe (Extraction System)



Figure C.6 Packers (Extraction System)

## REFERENCES

- 1. J. G Henry and G. W. Heinke, *Environmental Science and Engineering*, Prentice Hall, Englewood Cliffs, NJ, 1989.
- R.J Calkins, W.J Bishop and J. Borgerding, "Institutional Approach Toward Combined Industrial Waste Treatment," Journal Water Pollution Control Federation, vol. 51, pp. 612-618, April 1979.
- 3. G.F Bloom, "The Hidden Liability of Hazardous Waste Cleanup," Technical Review, vol. 1 pp. 59-65, February March 1986.
- 4. T.K Sherwood, Industrial Chemistry, Prentice Hall, Englewood Cliffs, NJ, 1985.
- 5. Warren L. Mc Cabe, J.C Smith, Unit Operations of Chemical Engineering, NY 1993.
- 6. F.A Lowenheim, M.K Moran, Industrial Chemicals, Rahway, NJ 1975.
- 7. Ibert Mellan, Industrial Solvents, NY 1939.
- 8. Hendrickson, Organic Chemistry, Prentice Hall, Englewood Cliffs, NJ 1970.
- 9. T.K Sherwood and E.W Comings, Trans AICHE, vol. 118, 1932.
- 10. O.A Hougen, Dryer Calculations and Design, in "Teaching of Chemical Engineering," Process Chemical Engineering, June 1939.
- 11.U.S Environmental Protection Agency, State Decision Makers' Guide for Hazardous Waste Management, U.S EPA OSW SW 412, Washington D.C 1977.
- 12. Henry J.G., Leachate from Hazardous Waste Landfills, Toronto, University of Toronto, Solid and Hazardous Waste Publication No. WM82- 02,1982.
- 13.Accutech Pneumatic Fracturing Extraction and Hot Gas Injection Phase 1, Applications Analysis Report by EPA/540/AR-93/509, July 1993, Cincinnati, OH 45268.
- 14. H. Fernandez, "An Investigation into the Feasibility of Utilizing Pneumatic Ultrasonic Devices Coupled with Pneumatic Fracturing in Enhancing Removal of Volatile Organic Compounds from Soil", May 1997.

- 15. T.Boland, D. Hanesian, A. J. Perna, J. R. Schuring, H. Fernandez, H. Kaleem, J. Lin, J. Alvarez, C. Chamion, M, Robinson, D. Delgado, M. Cubias, G. Salinas, J. Singh, "Ultrasonic Field Demonstration Work Plan", Department of Chemical Engineering and Environmental Science and Department of Civil and Environmental Engineering, N.J.I.T, Newark N.J, February 1998.
- 16. Sentex Systems inc., Scentograph "Plus II" Instruction/Operation Manual, Ridgefield, NJ, July 1993.
- 17. John E. Freund, Statistics a First Course, Prentice Hall, Englewood Cliffs, NJ, 1970.