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**THE ENTRAINER EFFECT ON AZEOTROPIC DISTILLATION COLUMN
DESIGN**

New Jersey Institute of Technology

D.ENG.SC.

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THE ENTRAINER EFFECT ON AZEOTROPIC
DISTILLATION COLUMN DESIGN

by

CHARLES MING-HSIAO TSAI

Dissertation submitted to the Faculty of the Graduate School
of the New Jersey Institute of Technology in partial
fulfillment of the requirements for the degree of
Doctor of Engineering Science
1982

APPROVAL SHEET

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ABSTRACT

Title of Dissertation: The Entrainer Effect on Azeotropic Distillation Column Design

Charles Ming-Hsiao Tsai, Doctor of Engineering Science, 1982

Dissertation directed by:

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A rigorous multistage distillation model was developed for use with non-ideal multicomponent systems. A subsidiary computer program was also developed to handle the problem of two liquid phases after condensation of a vapor.

Using the program developed, entrainer entry location and rate effects have been investigated via computer simulation for azeotropic distillation columns. The specific example used to verify the procedure was an ethanol dehydration column using benzene as the entrainer. The results indicate that the entrainer should enter the column two to five theoretical stages above the feed. The optimum entrainer rate was also determined.

The overhead vapor product from the azeotropic column, upon condensing to a subcooled liquid, forms two liquid phases in the accumulator. A two-liquid phase separation model was used to deal with the liquid phase splitting. This model has been validated using published data for nonideal ternary and quaternary systems. A study of the accumulator temperature effect on liquid phase splitting was also accomplished for the ethanol

dehydration column. The results indicate that at lower accumulator temperatures the benzene loss in the heavy phase can be reduced.

Finally, the azeotropic distillation model and the two-liquid phase flash model were included in a representation of a complete industrial ethanol purification plant. Incorporated in the plant simulation is the azeotropic column, two-liquid phase accumulator, a benzene stripping column and an ethanol recovery/water purge column. The ability of this model to converge on the benzene entrainer stream was verified.

PREFACE

Azeotropic distillation refers to those processes where a component is used to modify the relative volatility of the key components to some value greater than or less than unity. The added component is called an entrainer, and forms an azeotrope with one or more of the feed components. The azeotrope is removed as part of the desired overhead product. The entrainer is recovered to a significant degree with the distillate or the overhead product.

It is commonly accepted that the entrainer in azeotropic distillation forms a minimum boiling azeotrope with one of the key components and not with the other, or it forms a ternary azeotrope having a widely different composition from that of an azeotrope with one of both of the keys. The change of the relative volatility is due to the tendency toward immiscibility of the entrainer and the associated key component with respect to the other key component. If immiscibility does occur or is approached, one phase will contain the entrainer and light key, and the other the heavy key. The entrainer in this case will act as a stripping agent.

If a heterogeneous phase azeotrope (second liquid phase) is formed upon condensing the overhead vapor, a decanter (liquid gravity separator) which also serves as the distillate accumulator can be used to separate the entrainer from the key component. The entrainer rich layer is recycled back to the column as the second feed. The other liquid (net overhead

product) is processed further to separate the light key from the entrainer. The loss of the entrainer in the heavy key rich stream, if it can not be recovered, will require a make up stream to maintain the proper concentration level of the entrainer to the azeotropic column. The recycled entrainer stream can mix with the feed to column, either above or below the feed entry. The entrainer entry point and rate plays an important role in the operation of the azeotropic distillation column.

This research proposes to study the entrainer effect on the azeotropic distillation process which forms two liquid phases when the column overhead vapor is condensed. The optimum entrainer rate, composition, and entry location are the main subjects of this dissertation.

DEDICATION

Dedicated to Cheryl,
my lovely and intelligent daughter.

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The author wishes to express his sincere appreciation to his faculty advisor, Dr. Edward C. Roche, Jr. for his guidance and help. The management of the Bayway Refinery of Exxon Company, USA has been very supportive of this work. The author is in debt to his family for the encouragement and support he received during the preparation of this work.

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CHAPTER 1

INTRODUCTION

This research proposes to study the entrainer effect on azeotropic distillation from the standpoint of rate, composition, and entry location. A ternary system, ethanol-water-benzene, with benzene as the entrainer was chosen for the in depth analysis.

The selection of the entrainer is not included in this study, this having been discussed by Ewell and Berg (1944). The entrainer rate effect was investigated by Hachmuth (1952) for the ternary system, MEK-n-heptane-toluene, with MEK used as the entrainer. Hachmuth concluded that the entrainer must be presented in an appreciable concentration, at least as much as 30 mole percent and preferably more than 50 mole percent of the feed rate. Recently, Black (1980) evaluated the entrainer selection, and entrainer to feed ratio for the ethanol dehydration process. Black concluded that pentane was superior to either benzene or diethyl ether as the entrainer in the ethanol dehydration process. Unfortunately, the detailed approaches used to arrive at the conclusion were not revealed in the publication.

Rubin and Benedict (1945) studied the entrainer entry location using a graphical method for the ternary system, toluene-n-heptane-methanol (entrainer). They concluded that the entrainer can be added with, above or below the feed depending upon

the volatility of the entrainer with respect to the non-entrainer components.

The two-liquid phase handling in a distillation column has been reported by Block and Hegner (1976), Ross and Seider (1979), and Shah-Boston (1979). Since in this research, the two-liquid phase distillate is the primary concern, the two-liquid phase flash calculation methods by Deam (1969), and Roche (1977) were used to determine the distillate splitting.

To facilitate the azeotropic distillation simulation, the algorithm of Fredenslund (1977) was modified and enhanced for the reboiled absorber stripper type simulation. In Chapter 2, this azeotropic distillation model, which is the first part of this research, is discussed. As part of the initial phase of this research, the effect of entrainer location and rate was evaluated. This is the first accomplishment of this dissertation.

The next focal point in this research endeavor is the two-liquid phase flash which occurs in the distillate accumulator. In Chapter 3, a two-liquid phase flash model was utilized for this purpose. The simulation is based on Deam (1967) and Roche (1977). The sub-cooling effect on liquid phase splitting is investigated in this segment of the study. This is the second accomplishment of this research.

The third part of the research effort was the combining of the previously mentioned models using a specific executive. The details of this executive are discussed in Chapter 4, where the

entrainer rate and composition are converged to yield the desired separation. The ability to converge the entrainer rate and composition is the third accomplishment of this research.

Finally, the conclusions reached as a result of this research and the recommendations for further studies are presented in Chapter 5.

CHAPTER 2

THE AZEOTROPIC DISTILLATION MODEL

In this chapter, the azeotropic distillation model used in this research is described. The model is presented by Fredenslund et al. (1977) which is based on the work of Naphtali and Sandholm, and has been extensively modified and enhanced for this research. Since the systems which form a heterogeneous azeotrope in the overhead distillate are the primary subject of this research, the two-liquid phase condition is assumed to occur only in the overhead accumulator upon condensation and subcooling. The capabilities of the azeotropic distillation model are summarized as follows:

1. Multiple feeds (maximum, 10 feed streams)
2. Vapor and liquid sidestreams (maximum, 10 streams of each phase)
3. Multicomponent system (maximum, 10 components)
4. Multiple stages (maximum, 50 stages)
5. Vapor-liquid equilibrium package can handle ideal and most non-ideal vapor-liquid systems.
6. Multiple stage heat exchangers (maximum, 10 heat exchangers).

With the multistage vapor liquid model validated in its various forms, the reboiled absorber mode was used to conduct the entrainer location study. The ethanol-water-benzene system was verified using the developed model with information published by the Chem Share Corporation (1980) on the design of the ethanol recovery section of a 'Gasohol Process.' With continuing

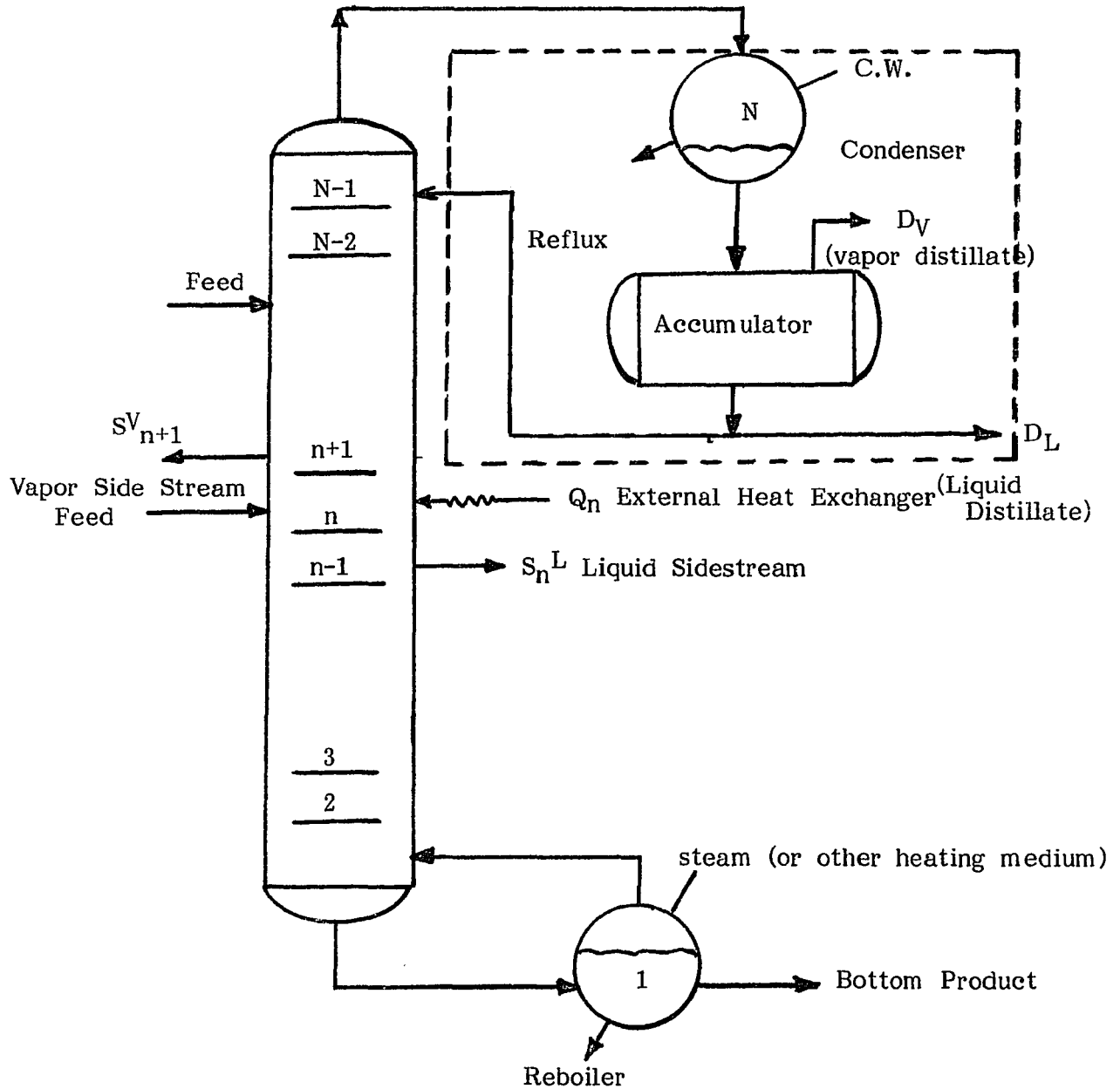
interest in gasohol as a gasoline octane improver, the ethanol-water-benzene (entrainer) system was used to conduct the entrainer effect study. The results of this study are discussed in Section 2.5.

2.1 Equations and Column Configuration

The model developed by Fredenslund et al. (1977), is based on the Naphtali-Sandholm algorithm (1971) and has been enhanced to accommodate the strong non-ideal behavior of the systems chosen for this research. Modifications proposed by Roche (1978) were used in this work. The newer Vapor-Liquid-Equilibrium (VLE) correlations by Prausnitz et al. (1977, 1980, 1981) and Tsonopoulos (1972) have been incorporated in the model and used in this investigation. These newer correlations are chosen since the topic of interest is a low pressure separation involving polar compounds. In addition, the stagewise Newton-Raphson convergence technique has been enhanced.

The distillation column for this research is illustrated in Figure 2.1 and the following column specifications must be specified to execute the rigorous multicomponent stage to stage calculations.

- 1) The number of stages, with the reboiler designated as the first stage, while the partial condenser (with vapor and liquid distillates) is the Nth.
- 2) Murphree stage efficiencies (default value is 100%).
- 3) Sidestream location and withdrawal rate.

Figure 2.1 Distillation Column Configuration

Note: For reboiled absorber type operations the overhead stream contains only the vapor distillate, condenser and accumulator are not included.

- 4) Feed composition, rate, thermal state, and location.
- 5) Total overhead flow rate.
- 6) Reflux ratio or boilup ratio.
- 7) Column pressure distribution.

This model with appropriate specifications can be applied to the following simulations:

- a) Absorber/stripper (end feed column).
- b) Reboiled absorber/refluxed stripper.
- c) Distillation column with product rate and reflux ratio.
- d) Distillation column with product rate and boilup ratio.

The product distillate can be either dew point vapor, bubble point liquid, subcooled liquid, or an equilibrium vapor/liquid mixture. All but the dew point vapor require the presence of a condenser.

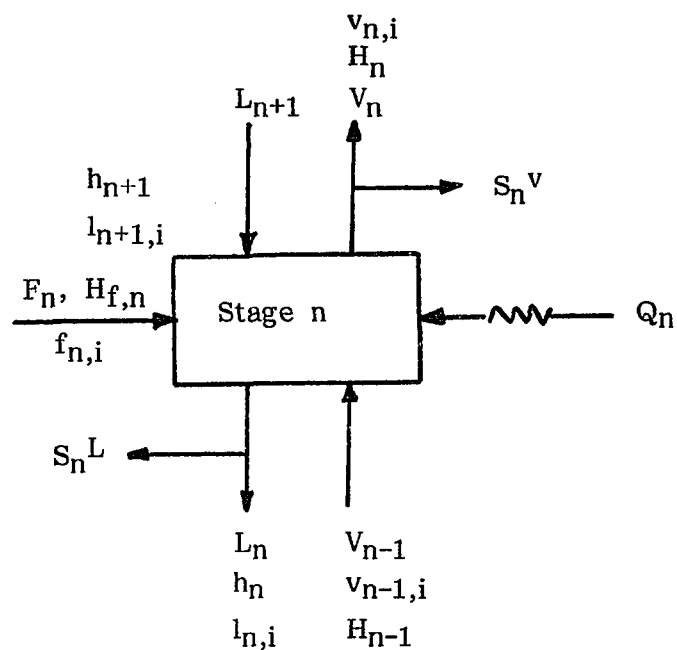
For an arbitrary stage n , the configuration and nomenclature are given in Figure 2.2.

The threefold set of discrepancy functions describing stage n are listed below:

Component Material Balance (one for each component)

$$F_1(n,i) = \left(1 + \frac{S_n^L}{L_n}\right)l_{n,i} + \left(1 + \frac{S_n^V}{V_n}\right)v_{n,i} - v_{n-1,i} - l_{n+1,i} - f_{n,i} = 0 \quad (2-1)$$

Equilibrium Condition coupled with the Murphree stage efficiency, η_n



Subscript n = stage, $n = 1, 2, \dots N$

i = component, $i = 1, 2, \dots c$

H = vapor phase enthalpy

h = liquid phase enthalpy

H_f = feed enthalpy

V = total vapor flow

v = component vapor flow

L = total liquid flow

l = component liquid flow

F = total feed

f = component feed

S^L = liquid sidestream

S^V = vapor sidestream

Q_n = external heat exchanger

Figure 2.2 Stage Configuration

$$F_2(n,i) = \eta_n \frac{K_{n,i} V_n}{L_n} l_{n,i} - v_{n,i} + (1 - \eta_n) \frac{V_n}{V_{n-1}} = 0 \quad (2-2)$$

Where the Murphree stage efficiency is defined by

$$\eta_n = \frac{Y_{n,i} - Y_{n-1,i}}{K_{n,i} x_{n-1,i} - Y_{n-1,i}} \quad (2-3)$$

and the Equilibrium Constant is defined by

$$K_{n,i} = \left(\frac{y_i}{x_i} \right)_n = \left(\frac{\gamma_i f_i^O}{\phi_i \pi} \right)_n \quad (2-4)$$

Equation 2-4 will be discussed in more detail in Section 2.3.

Energy Balance

$$F_3(n) = \left(1 + \frac{S_n L}{L_n} \right) h_n + \left(1 + \frac{S_n V}{V_n} \right) H_n - H_{n-1} -$$

$$h_{n+1} - h_n^F - Q_n + h_{n+1}^{\text{mix}} = 0 \quad (2-5)$$

The enthalpy calculations associated with Eq. 2-5 are calculated as follows:

Liquid Enthalpy

$$h_n = l_i \int_{T_{\text{ref}}}^T C_{pL}(T) dT + h_n^{\text{mix}} \quad (2-5A)$$

where, $C_{pL}(T)$ = the component specific heat capacity which is a polynomial function of temperature.

T_{ref} = the reference time used in enthalpy calculations, normal boiling point is used.

h_n^{mix} = the heat of mixing, which is described in Appendix B.

Vapor Enthalpy

$$H_v = v_i \left\{ \lambda_i + \int_{T_{ref}}^T C_{p_v}(T) dt + \Delta H \right\} \quad (2-5B)$$

where, $C_{p_v}(T)$ = the component specific heat capacity, which
is a polynomial function of temperature

λ = the component heat of vaporization

ΔH = the enthalpy correction due to temperature,
pressure effects

2.2 Iteration Algorithm

The discrepancy functions, $F_i(N_i)$, are a quantitative measure of the failure of the independent variable, $l_{n,i}$, $v_{n,i}$, and T_n to satisfy the physical condition of the column.

These relationships comprise a vector of test functions,

$$\underline{F}(\underline{x}) = \begin{pmatrix} \underline{F}_1 \\ \underline{F}_2 \\ \underline{F}_3 \end{pmatrix} = 0 \quad (2-6)$$

Which can be solved for equally many unknowns:

$$\underline{x} = \begin{pmatrix} \underline{l} \\ \underline{v} \\ \underline{T} \end{pmatrix} \quad (2-7)$$

Where the vector \underline{l} contains all the elements $l_{n,i}$, \underline{v} all the elements $v_{n,i}$, and \underline{T} all the elements T_n . Once all $l_{n,i}$, $v_{n,i}$, and T_n 's are known, the product flow rates, composition, and concentration profile can be evaluated.

Eq. (2-6) is solved by utilizing the Newton-Raphson method which employs the simultaneous convergence of all the independent variables \underline{x} .

In a Newton-Raphson iteration, a new set of values of the independent variables, \underline{x}_r , is generated from a previous estimate, \underline{x}_{r-1} , in the following fashion:

$$\underline{x}_r = \underline{x}_{r-1} - \underline{F}_{r-1}(\underline{x}_{r-1}) \left(\frac{\partial \underline{F}}{\partial \underline{x}} \right)^{-1} \Big|_{\underline{x}_{r-1}} \quad (2-8)$$

When $(\underline{x}_r - \underline{x}_{r-1})$ is sufficiently small, the correct set of values of \underline{x} has been found, and the iteration stops.

The variations between subsequent iterations are arbitrarily limited as follows:

- 1) Negative component molar flow rates are not permitted.
- 2) Componential flow rates exceeding L_n are set equal to L_n .
- 3) The maximum change in the temperature at each stage, T_n is 10°K .

The linearization of the discrepancy function, evaluation of Jacobian correction matrix, $(\partial \underline{F} / \partial \underline{X})$, the structure of the Jacobian, and the iteration algorithm proposed by Roche (1977) are described in Appendix A.

2.3 VLE Predication Method

Vapor-liquid equilibrium requires that the fugacity of a component in the vapor be equal to the fugacity of that component in the liquid. The vapor-liquid equilibrium has been discussed

by Prausnitz (1967, 1980) and Gmehling et al. (1977) in detail for systems involving separations at or below essentially atmospheric pressure.

$$f_i^V = f_i^L \quad (2-8)$$

The vapor phase fugacity can be represented by:

$$f_i^V = \phi_i y_i \pi \quad (2-9)$$

Where

y_i = is the component mole fraction in the vapor phase.

π = is the system pressure.

ϕ_i = is the fugacity coefficient of the component in the multicomponent vapor phase.

The fugacity in the liquid phase can be represented by:

$$f_i^L = \gamma_i X_i f_i^{OL} \quad (2-10)$$

where

X_i = is the component mole fraction in the liquid phase.

γ_i = is the activity coefficient at the system temperature.

f_i^{OL} = is the standard state fugacity of the pure component at system temperature and pressure.

$$f_i^{OL} = f_i^S \text{Exp} \int_{p_i^S}^{\pi} \frac{\bar{V}_i dp}{RT} = \phi_i^S p_i^S \text{Exp} \int_{p_i^S}^{\pi} \frac{\bar{V}_i dp}{RT} \quad (2-11)$$

Where

$\int_{p_i^S}^{\pi} \bar{V}_i dp$ = is the poynting correction factor.

f_i^S = fugacity of component at saturation ($p = p_i^S$)

\bar{V}_i = partial molar volume of component.

ϕ_i^S = fugacity coefficient of component at saturation pressure.

p_i^S = saturation pressure of component.

For convenience, \bar{V}_i is assumed to be independent of pressure, and if system conditions are remote from the critical, then \bar{V}_i is set equal to V_i^L , the molar volume of component.

Combine Eqs. 2.10 and 2.11 and eliminate the subscript L for simplicity, then

$$f_i^L = \gamma_i x_i p_i^S \phi_i^S \text{Exp} \left[\frac{(\pi - p_i^S) V_i}{RT} \right] \quad (2-12)$$

Substituting Eqs. 2-9 and 2-11 into Eq. 2-8 one obtains

$$K_i = \frac{y_i}{x_i} = \frac{\gamma_i p_i^S \phi_i^S \text{Exp} \left[\frac{(\pi - p_i^S) V_i}{RT} \right]}{\phi_i \pi} \quad (2-13)$$

For a system with an ideal vapor, Eq. 2-13 becomes

$$K_i = \frac{\gamma_i p_i^S}{\pi} \quad (2-14)$$

The saturation pressure or the vapor pressure data are most conveniently related to temperature by an equation having the general form

$$\ln p_i^S = c_1 + \frac{c_2}{c_3 + T} + c_4 T + c_5 T^2 + c_6 \ln T \quad (2-15)$$

The empirical constants, c_1 , c_2 , c_3 , c_4 , c_5 , and c_6 , as reported by Gmehling et al. (1977) have been used for the systems involved with in this investigation.

Systems chosen for this research contain polar components

like ethanol which exhibit strong non-ideality in the vapor phase and this requires special handling in the fugacity coefficient calculations. The Tsonopoulos' (1972) correlation is therefore used for the virial coefficient predictions in this study.

The fugacity coefficient obtained from the virial equation of state truncated after the second virial coefficient is

$$\ln \phi_i = \left[2 \sum_j^m y_j B_{ij} - B_{mix} \right] \frac{\pi}{RT} \quad (2-16)$$

Where

B_{ij} = the second cross virial coefficient for binary i-j.

B_{mix} = the second virial coefficient of the vapor mixture.

$$B_{mix} = \sum_{i=1}^m \sum_{j=1}^n y_i y_j B_{ij}$$

For a pure component, the equation (2-16) becomes

$$\phi_i^S = \text{Exp} \frac{B_{ii} P_i^S}{RT} \quad (2-17)$$

The fugacity coefficient can be expressed in either a pressure or a volume series, the existence of real root in the following expression is the determining factor for selecting which method to use.

$$\text{Root} = \left(0.25 + B_{mix} \frac{\pi}{RT} \right)^{1/2}$$

For a negative root the following pressure series must be used:

$$\phi_i = \text{Exp} \left\{ \left[2 \sum_j^m y_j B_{ij} - B_{mix} \right] \right\} \frac{\pi}{RT} \quad (2-18)$$

For a positive root the following volume series is used:

$$\phi_i = \text{Exp}\left[\frac{2}{V_{\text{mix}}}\left(\sum_{j=1}^m y_j B_{ij}\right)\right] - Z_{\text{mix}} \quad (2-19)$$

Where

$$V_{\text{mix}} = B_{\text{mix}} + \frac{RT}{\pi}$$

$$Z_{\text{mix}} = \frac{V_{\text{mix}}}{RT}$$

The calculation of the fugacity coefficient of a component in the vapor phase precipitates the problem of obtaining the second virial coefficients. Tsonopoulos (1972) extended the well known Pitzer-Curl correlation for polar compounds' second virial coefficients predictions. This method is discussed in Appendix B together with other thermodynamic property calculation methods used.

The liquid phase activity coefficient for components with strong non-ideal behavior must be handled carefully. Also, due to the two-liquid formation in the heterogeneous azeotropic distillate reflux accumulator. The NRTL by Renon et al. (1968) and UNIQUAC by Abrams et al. (1975) were used in this research. For a solution of m components, the activity coefficient for a given component can be calculated using:

NRTL Equation

$$\ln \gamma_i = \frac{\sum_{j=1}^m \tau_{ji} G_{ji} X_j}{\sum_{l=1}^m G_{l,i} X_l} + \sum_{j=1}^m \frac{X_j G_{ij}}{\sum_{l=1}^m B_{lj} X_l} \left(\tau_{ij} - \frac{\sum_{n=1}^m X_n \tau_{nj} G_{nj}}{\sum_{l=1}^m G_{lj} X_l} \right) \quad (2-20)$$

UNIQUAC Equation

$$\ln \gamma_i = \ln \gamma_i^C + \ln \gamma_i^R \quad (2-21)$$

$$\ln \gamma_i^C = \ln \frac{\phi_i}{x_i} + \frac{z}{2} q_i \ln \frac{\theta_i}{\phi_i} + l_i - \frac{\phi_i}{x_i} \sum_{j=1}^m x_j l_j \quad (2-22)$$

$$\ln \gamma_i^R = q_i \left[1 - \ln \left(\sum_{j=1}^m \theta_j \tau_{ji} \right) - \frac{\sum_{j=1}^m \theta_j \tau_{ij}}{\sum_{k=1}^m \theta_k \tau_{kj}} \right] \quad (2-23)$$

The details of the above equations are described in Appendix B.

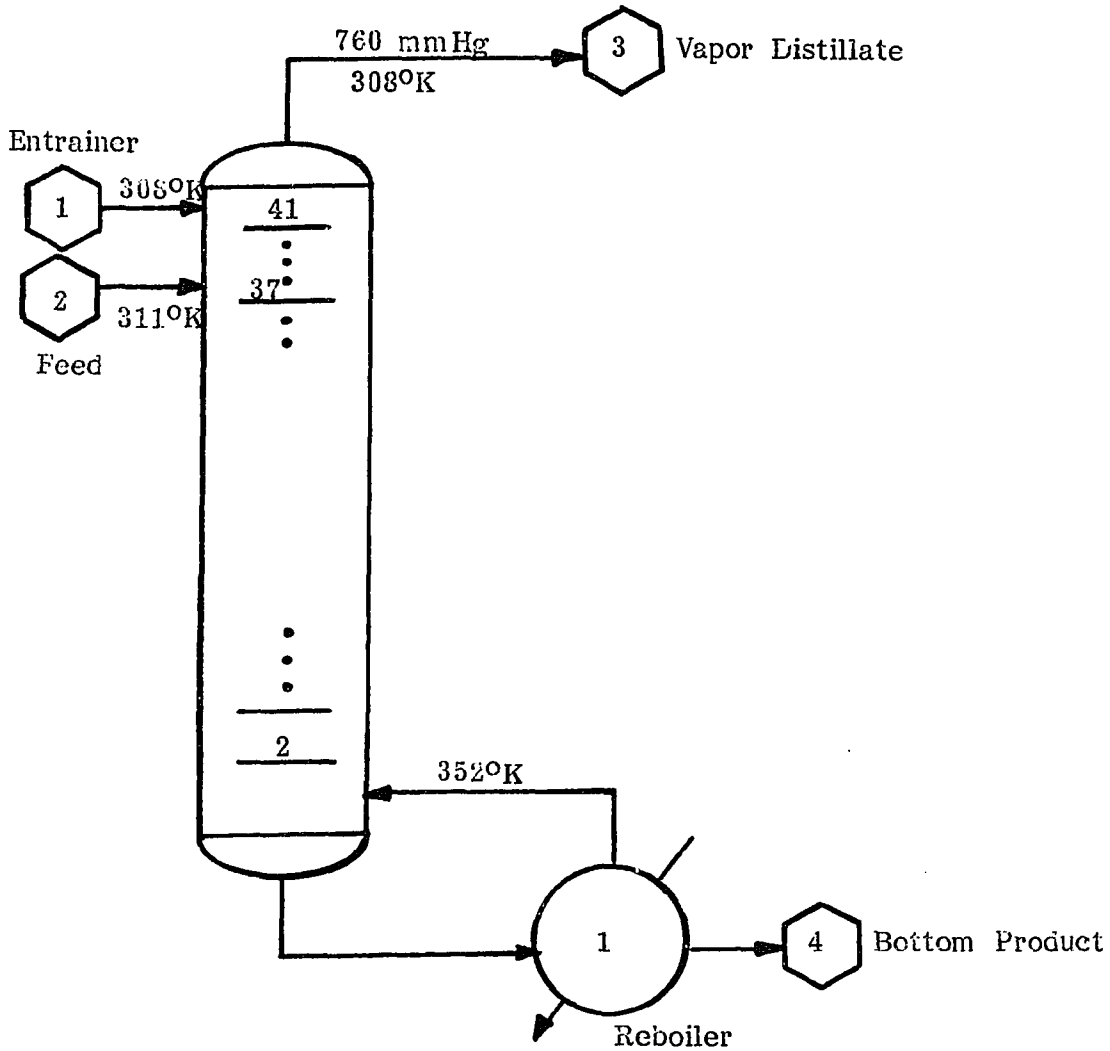
2.4 Research Problem Definition

It has been known that during the concentrating process of ethanol, an entrainer is required to break the ethanol-water azeotrope to achieve high purity ethanol for both industrial and Gasohol uses. Black (1980) evaluated different entrainers for the ethanol dehydration process, however, the details used in his work are unavailable to the general public. Recently, the Chem Share Corporation used its Design 2000 flowsheet simulation for commercial representation of the Gasohol dehydration process. Chem Share (1980) used a system with benzene as the entrainer in their publication entitled "GASOHOL."

In this research, a system similar to Chem Share's is illustrated in Figure 2.3. This problem, as described, has been studied using the extended Naphthali-Sandholm algorithm. In this segment of the research, the two-liquid phase handling in the distillate accumulator has been ignored intentionally. The

Figure 2.3 The Research Problem Configuration (Reboiled Absorber Type)

System: Ethanol (1) - Water (2) - Benzene (3)



Component	1	2	3	4
Ethanol	28.6 mol %	74.2 mol %	32.7 mol %	99.8 mol %
Water	3.8	25.8	12.3	0.15
Benzene	67.6	0.0	55.0	3×10^{-5}
Total moles/hr	591.85 mole/hr	727.0 mole/hr	727.0 mole/hr	123.35 mole/hr

entrainer rate and location have been adjusted manually to enable the studies of its effects on the azeotropic distillation.

2.3.1 The Dehydration of Ethanol Using Benzene as the Entrainer

The Chem Share azeotropic distillation column consists of 41 stages which can be configured as the conventional distillation column with very small reflux ratio or as a reboiled absorber. The feed stream contains 74.2 mol percent of ethanol and 25.8 mol percent of water as obtained from an ethanol concentrator of crude biomass ethanol. It is desired to produce a bottom stream which is 99.8 mol percent ethanol. The entrainer stream enters the top of the column containing 67.6 mol percent benzene, 28.6 mol percent water and 3.8 mol percent ethanol. In Chapter 3, the overhead vapor stream from the azeotropic column will be used as the feed to a two-liquid phase flash model to generate the entrainer stream (organic rich) as used above. In Chapter 4, the complete azeotropic model is used to study the entrainer rate and composition, coupled with a make up stream to compensate for benzene losses in the water rich overhead product and the small amount of benzene removed with the ethanol product.

The thermodynamic data of the components of the research problem are indicated in Table 2.1 and Table 2.2.

2.5 Results of Simulation--The Research Problem

The research problem, an ethanol dehydration column with

TABLE 2.1

Research Problem

Component Physical Properties and Thermodynamic Data

Part I

1. Vapor pressure equation constants:

temperature in °K
 pressure in mmHg
 from Gmehling et al. (1977)

	c ₁	c ₂	c ₃
ETOH	18.912	-3803.98	-41.68
Water	18.304	-3816.44	-46.13
Benzene	15.901	-2788.51	-52.36

2. Tsouopoulos' K_{ij} , and A_i , B_i for Polar Component from Tsouopoulos (1972)

$$K_{ii} = 0$$

	A_i	B_i	K_{ij}	K_{ji}
ETOH	0.0878	0.0572	0.20	0.15
Water	0.0279	0.0229	0.20	0.40
Benzene	0.0	0.0	0.15	0.40

3. NRTL binary parameters:

from Gmehling (1977)

ETOH(1) - Water(2) - Benzene(3)

ij	A_{ij}	A_{ji}	α_{ij}
12	-86.76	1289.91	0.270
13	282.33	1096.90	0.295
23	3820.95	2213.35	0.267

Note: A_{ij} and A_{ji} in cal/gmole; $\alpha_{ij} = \alpha_{ji}$

TABLE 2.2

Research Problem

Component Physical Properties and Thermodynamic Data

Part II

UNIQUAC Equation parameters:

Binary Parameters: cal/gmole

ij	A _{ij}	A _{ji}
12	110.01	200.05
13	-156.57	850.61
23	428.45	1072.79

Pure Component Parameters:

	r _i	q _i
ETOH	2.1055	1.9720
Water	0.9200	1.4000
Benzene	3.1878	2.4000

benzene as the entrainer has been simulated using the azeotropic distillation model completed in this segment of this research. Three different cases are investigated. The first two cases employing the NRTL equation for component activity coefficient calculations using two different operational modes, a reboiled absorber mode, and a conventional distillation with small reflux ratio. The third case used the UNIQUAC equation for activity coefficient calculation in a conventional distillation column with small reflux ratio is also investigated. The input and output data for the above three cases is attached in Appendix C.

2.5.1 Comparison of the Research Problem and Chem Share's Example

This research problem uses the reboiled absorber solution and is compared with the Chem Share Gasohol example. The configuration of both columns is similar, a 41-stage column with vapor distillate as the overhead product and operated under atmospheric pressure. The feeds (normal feed and entrainer feed) are identical in rate, composition, location, and thermal condition. In the vapor-liquid equilibrium prediction method, the research problem uses the virial equation for fugacity coefficient calculation, while Chem Share assumed ideal vapor and failed to correct the non-ideal behavior of the polar components, water and ethanol in the vapor phase. The NRTL equation is used in both. The following important simulation results are compared and tabulated:

- 1) Mainstream Composition--Table 2.3

- 2) Column Temperature Profile--Table 2.4
- 3) Selected Stage Component Equilibrium Constant--Table 2.5
- 4) VLE Prediction Methods--Table 2.6.

Also, the column composition profiles of the research problem is shown in Figure 2.4.

2.5.2 Simulation Using UNIQUAC Equation

The UNIQUAC equation is used as an alternative to calculate the liquid phase activity coefficients in addition to the previously mentioned NRTL equation. To achieve the same degree of separation as in the NRTL case, the pressure drop per stage in the simulation was assumed to be 45 mmHg which is excessive for a column operation. As a result, the bottom temperature is 33°K higher than that obtained by using the NRTL equation. Also, from the standpoint of convergence, the NRTL case required only nine iterations, while in the UNIQUAC twelve iterations were necessary.

The computer input and output of the UNIQUAC case run with stage $\Delta P = 45$ mmHg is attached in Appendix C. Another investigation with stage ΔP set equal to zero was conducted, the converged results are tabulated in Table 2.7. From this table of main stream composition, it is obvious that the bottom and distillate product qualities were all off spec. Also, 40 iterations were required to converge the simulation.

One explanation for this is that with two interaction

TABLE 2.3

Main Stream Composition Comparison

Stream: (1) Entrainer
 (2) Feed
 (3) Vapor Distillate
 (4) Bottom Product

Composition: in mole percent

Component/Stream	<u>Research Problem</u>				<u>Chem Share's</u>			
	(1)	(2)	(3)	(4)	(1)	(2)	(3)	(4)
Ethanol	28.588	74.236	32.724	99.869	28.588	74.236	32.736	99.76
Water	3.827	25.764	12.276	3×10^{-4}	3.827	25.764	12.278	5.1×10^{-4}
Benzene	67.584	0.0	54.998	0.13	67.585	0.0	54.986	0.27

By comparing the compositions of the bottom product and the overhead distillate, Table 2.3 indicates that it is an average deviation of 0.15%.

TABLE 2.4

Column Temperature Profile (°K)

<u>Stage No.</u>	<u>Research Problem</u>	<u>Chem Share</u>
41 (Distillate)	338.0	338.8
37	339.4	338.8
33	340.8	340.1
29	342.2	340.7
25	343.6	340.5
21	345.0	340.5
17	346.4	340.5
13	347.8	340.6
11	348.5	340.7
9	349.2	341.3
5	350.6	343.3
3	351.3	349.7
1 (Reboiler)	352.0	351.3

Note: Chem Share's problem with stages numbered from top down.

TABLE 2.5
Selected Component Equilibrium Constants

	<u>Research Problem</u>	<u>Chem Share</u>	<u>Dev.</u>
(1) Stage No. 1 (Bottom)			
Ethanol	0.99649	0.99325	0.3%
Benzene	3.6817	3.7697	2.3
Water	1.1510	1.1802	2.4
(2) Stage No. 2			
Ethanol	0.987	0.9778	0.9
Benzene	3.679	3.681	0.05
Water	1.145	1.169	2.0
(3) Stage No. 39			
Ethanol	1.119	1.115	0.3
Benzene	0.8311	0.8327	0.2
(4) Stage No. 41 (Vapor Distillate)			
Ethanol	1.1249	1.1175	0.6
Benzene	0.82846	0.8309	0.3
Water	2.7155	2.7098	0.2

TABLE 2.6

VLE Prediction Method

	<u>Research Problem</u>	<u>Chem Share</u>
Vapor phase - ϕ_i	Virial Eq.-- Tsonopoulos Correlation	Ideal--Antoine Eq.
Liquid phase - γ_i	NRTL & UNIQUAC	NRTL Eq.

Note: (1) In most azeotropic distillation publications, the ideal vapor phase is assumed. In this research the Tsonopoulos method is applied to the virial equation to calculate the vapor fugacity coefficient of component. This is believed to be a better approach due to the polar compound's existence in the system.

(2) In the liquid phase activity coefficient calculation, the newer UNIQUAC prediction was also used on a separate run. The results indicated that the NRTL equation is superior to the UNIQUAC equation in both the predictions and convergence.

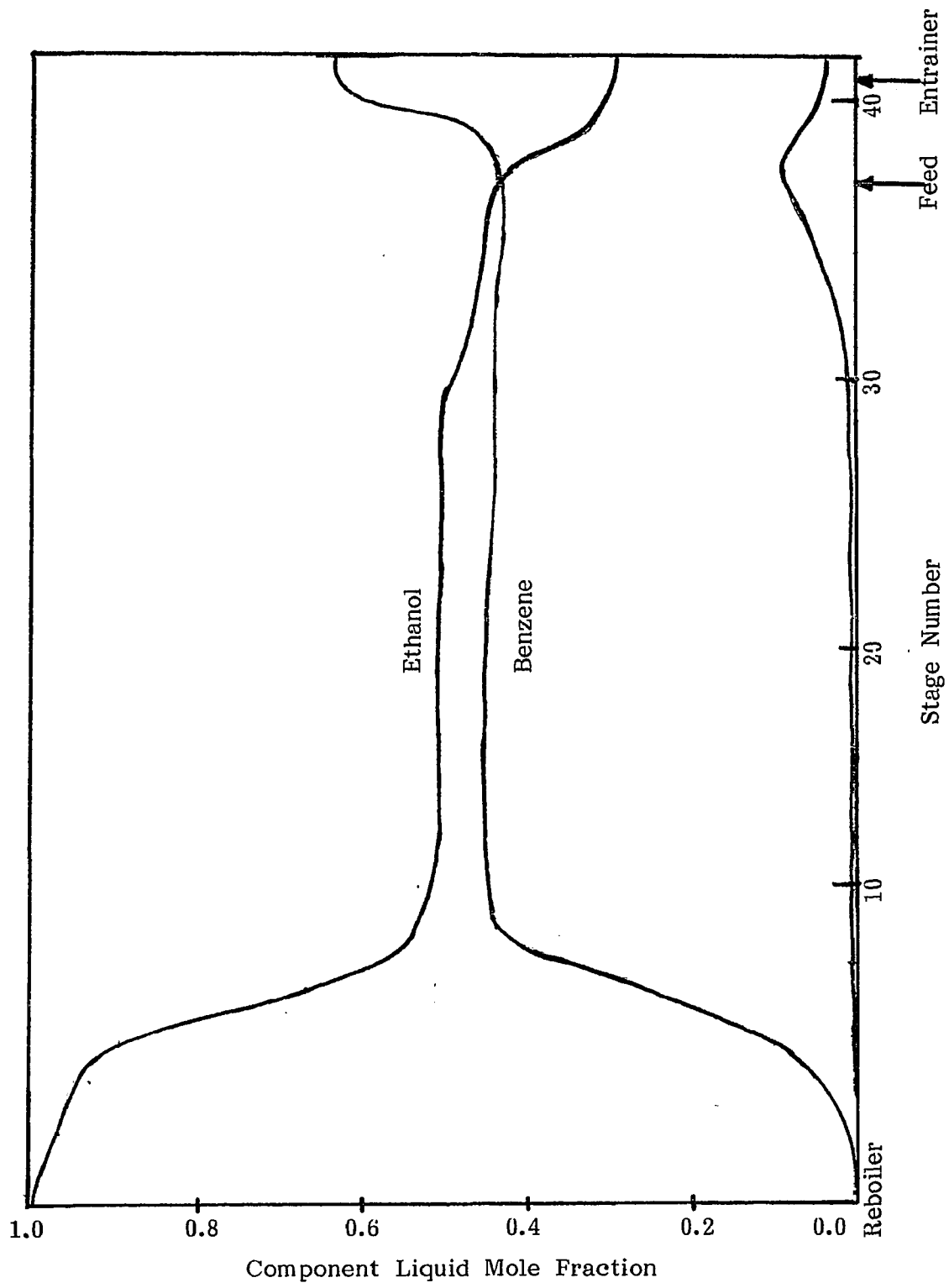


Figure 2.4 Column Composition Profiles--The Research Problem

TABLE 2.7

Main Stream Composition Using UNIQUAC

Component/Stream	Research Problem using NRTL				Research Problem using UNIQUAC			
	(1)	(2)	(3)	(4)	(1)	(2)	(3)	(4)
Ethanol	28.588	74.236	32.724	99.869	28.588	74.236	38.18	78.23
Water	3.827	25.764	12.276	3×10^{-4}	3.827	25.764	4.26	21.77
Benzene	67.585	0.0	54.998	0.13	67.585	0.0	57.55	0.0

Note: Composition in mol percent

parameters per binary for the UNIQUAC Equation versus three parameters per binary as required by the NRTL Equation. The NRTL equation will predict the VLE better.

2.6 Results of Studies on Entrainer Effect

With the azeotropic distillation model for this research segment validated, the entrainer entry and rate effects on the separation of the research problem were evaluated. The results confirm the previous study by Rubin et al. (1945).

2.6.1 The Entrainer Entry Effect

The entry of an entrainer to the column, especially its location in relation to the feed is of importance in this study. The results indicated that the entrainer must be introduced above the feed, and at least two, preferably three, stages above the feed. Table 2.8 indicates that in this research, since the entrainer, benzene is more volatile than one of the key components--water; it should be introduced above the feed. During the evaluation of the entrainer entry effect, the feed location was fixed at the 37th stage of the column. This study confirms the report by Rubin et al. (1945) that a volatile entrainer should be introduced above the feed. Mixing the entrainer with feed or introducing it at a point below the feed will result in either poorer separations or failure in program convergence.

Also, studies were made by varying the feed introducing point while the entrainer entry is fixed at the 41st stage.

In all cases, the entrainer stays above the feed. If the

TABLE 2.8

The Entrainer Entry Effect--Entrainer Location Varied

Feed enters at 37th stage in all cases

<u>Entrainer Entry Stage No.</u>	<u>Ethanol Mol % in Bottoms</u>	<u>Benzene Mol % in Bottoms</u>	<u>H₂O Mol % in Bottoms</u>
41 (base case)	99.86	0.139	0.00003
40 (3 above feed)	99.85	0.148	0.0003
39 (2 above feed)	99.76	0.237	0.00003
38 (1 above feed)	98.93	<u>1.0687</u>	0.00001
37 (mixed with feed)	90.52	<u>9.4837</u>	0
36 (below feed)	48.03	51.96	0
35 (below feed)	35.0	65.0	0
32 (below feed)	15.6	84.0	0

number of stages between the feed and entrainer is reduced to less than two, the benzene concentration level in the bottom product will begin to increase to an unacceptable level as indicated in Table 2.9.

The entrainer entry effect was further evaluated by varying both entrainer and feed locations with the entrainer always above the feed. This study confirms the previous findings in that the entrainer must stay at least two stages above the feed irregardless of the feed point. The results are indicated in Table 2.10.

2.6.2 The Entrainer Rate Effect

The next focal point is the entrainer rate effect on the separation. The benzene rate in the entrainer stream of the research problem was varied, however, the entrainer and feed locations remain unchanged.

The research problem with 400 moles/time benzene rate seems to yield the best separation. Increasing the benzene rate does not improve the separation further. Decreasing the benzene rate more than 10 moles/time resulted in very poor separations as evidenced in Table 2.11.

2.7 Conclusions

The azeotropic distillation model for this research was completed and validated in various forms. A research problem was defined and simulated using the stagewise model in a reboiled absorber mode. The results compared favorably with the Gasohol

TABLE 2.9

The Entrainer Entry Effect--Feed Location Varied

Entrainer enters at 41th stage in all cases

<u>Feed Introduced Stage No.</u>	<u>Ethanol Mol % in Bottoms</u>	<u>Benzene Mol % in Bottoms</u>	<u>H₂O Mol % in Bottoms</u>
40	98.9314	<u>1.685</u>	0.00003
39	99.129	<u>0.8701</u>	0.00002
38	99.859	0.140	0.0002
37 (base case)	99.86	0.139	0.00003
36	99.861	0.1392	0.0004
35	99.862	0.1379	0.00005
30	99.862	0.1375	0.00006
24	99.855	0.1359	0.00007
20	99.85	0.1487	0.00008

TABLE 2.10

The Entrainer Entry Effect--Feed and Entrainer Locations Varied

<u>Case</u>	<u>Bottom Product Ethanol Mol %</u>	<u>Bottom Product Benzene Mol %</u>	<u>Bottom Product H₂O Mol %</u>
(1) Feed = 37 Entrainer = 41 (Base) Δ Stage = 4	99.86	0.139	0.00003
(2) Feed = 39 Entrainer = 39 Δ Stage = 0	90.52	<u>9.48</u>	0.0000
(3) Feed = 38 Entrainer = 39 Δ Stage = 1	98.93	<u>1.068</u>	0.0001
(4) Feed = 35 Entrainer = 38 Δ Stage = 3	99.85	0.1487	0.00008
(5) Feed = 33 Entrainer = 38 Δ Stage = 5	99.86	0.138	0.00019
(6) Feed = 26 Entrainer = 36 Δ Stage = 10	99.86	0.137	0.00373

TABLE 2.11
The Entrainer Rate Effect

Feed at 37th Stage

Entrainer at 41st Stage

<u>Entrainer Rate</u>	<u>Bottom Product Ethanol Mol %</u>	<u>Bottom Product Benzene Mol %</u>	<u>Bottom Product H₂O Mol %</u>
400 moles/time (base case)	99.86	0.139	0.00003
405	98.34	1.660	0.00001
410	96.83	3.170	0.00001
420	93.87	6.128	0.0000
395	96.75	3.24	0.0001
390	97.64	0.1211	2.2398
385	96.75	3.247	0.0001
380	77.208	0.5223	12.269

design simulation of Chem Share, and, in fact, is a better process representation since the simulation model is capable of correcting non ideal vapor behavior in the vapor-liquid equilibrium.

The in-depth studies of entrainer entry effect on the azeotropic distillation leads to the conclusion that the entrainer must always enter the column above the feed with at least two stages separating the feeds. Better separation is achievable only if three or more stages are provided between the feed and entrainer. If the entrainer enters the column below the feed or mixes with the feed, poor separations will result.

The entrainer rate for this research is probably at its optimum, 400 moles/time, for the research problem. Further increase in the entrainer rate will theoretically drive the water out of the bottom product stream, however, the benzene concentration will begin to buildup in the bottom ethanol product.

Reducing the entrainer rate to the column will result in poor separations as indicated by the water concentration level increase in the bottoms, since less than the required amount of entrainer is presented in the column to break the water-ethanol azeotrope to accomplish the separation.

CHAPTER 3

THE TWO-LIQUID PHASE FLASH MODEL

In this chapter, the second part of the research dealing with the two-liquid phase separation in the overhead distillate accumulator is discussed. The literature concerning "Three-Phase Flash" (or Two-Liquid Phase Separation) have been covered by Deam et al. (1969), Osbrone (1964), Heidemann (1974), Leach (1977), and Roche (1977). A stand alone model has been developed to study the two-liquid phase separation that results when the overhead vapor from the reboiled absorber column is condensed.

3.1 Background and Theory

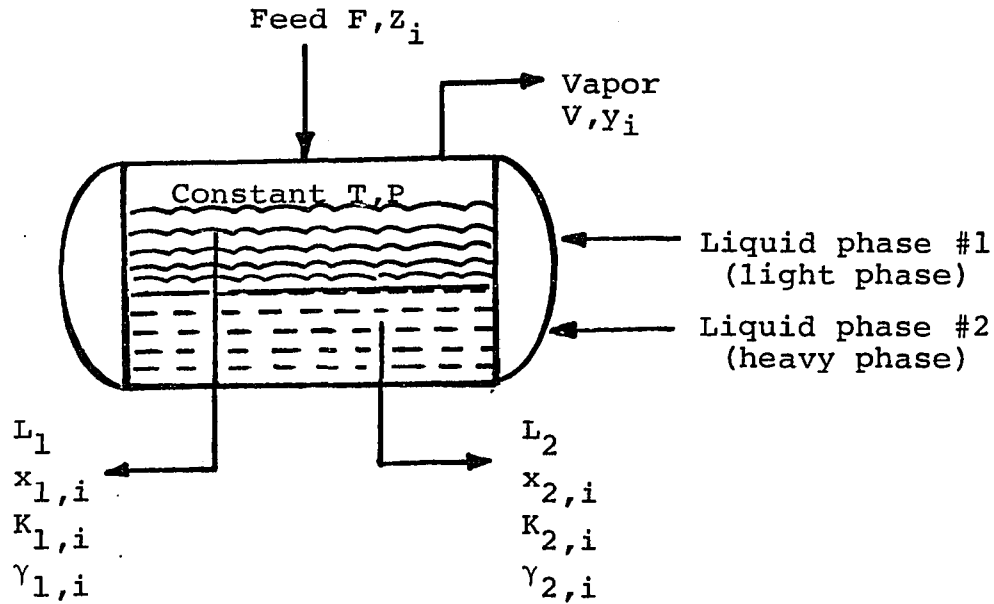
The two-liquid phase flash model is illustrated in Figure 3.1. Feed enters the accumulator which is at constant temperature and pressure. Depending on the conditions and the compositional structure, a single liquid phase or two liquid phases can exist in the accumulator.

3.1.1 Dew Point Temperature

The temperature at which a gas mixture begins to condense is the dew point, and is established using the following relationship

$$F = \sum_i \frac{y_i}{K_i} - 1.0 = 0.0 \quad (3.0)$$

where y_i is the given vapor feed composition. The vapor liquid equilibrium constant, K_i , is functionally dependent on the

Figure 3.1 Two-Liquid Phases Flash Model

where,

- F - Total Feed moles/time
- Z_i - Component mole fraction in feed
- L_1 - Liquid phase #1 rate, moles/time
- $x_{1,i}$ - Component mole fraction in liquid phase #1
- $K_{1,i}$ - Component vapor and liquid phase #1 equilibrium constant
- $\gamma_{1,i}$ - Component activity coefficient in liquid phase #1
- L_2 - Liquid phase #2 rate, moles/time
- $x_{2,i}$ - Component mole fraction in liquid phase #2
- $K_{2,i}$ - Component vapor and liquid phase #2 equilibrium constant
- $\gamma_{2,i}$ - Component activity coefficient in liquid phase #2
- V - Vapor phase rate, moles/time
- y_i - Component mole fraction in vapor phase

temperature, the equilibrium liquid composition, X_i , and the vapor phase composition. Because of the functional dependence of the VLE constant, the evaluation of the dew point temperature, T_{DP} , is iterative. The solution method requires the use of the Secant Method to solve Eq. (3-0). Wegstein's algorithm is used to enhance the convergence of liquid composition. The details of solution is described in Appendix D.

3.1.2 Bubble Point Temperature--Two Liquid Phases

With the two-liquid phase, the unknowns associated with the bubble point temperature are two fold, namely the temperature, T_{BP} , and the relative amount of material in the two liquid phases. The basic relationships governing two-liquid phase equilibrium are:

Material Balance:

$$F = L_1 + L_2 + V \quad (3-1a)$$

$$(L_1 + L_2)Z_i = L_1X_{1,i} + L_2X_{2,i} \quad (3-1b)$$

$$V \equiv 0 \quad (3-1c)$$

Equilibrium Relationship:

$$Y_{1,i}X_{1,i} = Y_{2,i}X_{2,i} \quad (3-2a)$$

$$Y_i = K_{1,i}X_{1,i} = K_{2,i}X_{2,i} \quad (3-2b)$$

Constraints:

$$\sum_i X_{1,i} = 1.0 \quad (3-3a)$$

$$\sum_i X_{2,i} = 1.0 \quad (3-3b)$$

$$\sum_i Y_i = 1.0 \quad (3-4)$$

When Eq. 3-3a and Eq. 3-3b are combined

$$F_1 = \sum_i (X_{2,i} - X_{1,i}) = 0 \quad (3-5)$$

From Eq. 3-1a

$$Z_i = \frac{L_1}{L_1 + L_2} X_{1,i} + \frac{L_2}{L_1 + L_2} X_{2,i} \quad (3-6)$$

Let $R = \frac{L_1}{L_1 + L_2}$ be the liquid distribution factor (3-7)

then $1 - R = \frac{L_2}{L_1 + L_2}$ (3-8)

Combining Eq. 3-6, Eq. 3-7, and Eq. 3-8, then

$$Z_i = R X_{1,i} + (1 - R) X_{2,i}$$

or

$$Z_i = [R + (1 - R) \frac{X_{2,i}}{X_{1,i}}] X_{1,i} \quad (3-9)$$

Substituting Eq. 3-2b into Eq. 3-9

$$Z_i = [R + (1 - R) \frac{\gamma_{1,i}}{\gamma_{2,i}}] X_{1,i} \quad (3-10)$$

let $W_i = R + (1 - R)$

the resulting phase compositions are then defined by

$$X_{1,i} = \frac{Z_i}{W_i}$$

$$X_{2,i} = \frac{Z_i}{W_i} \frac{\gamma_{1,i}}{\gamma_{2,i}}$$

$$Y_i = \frac{Z_i}{W_i} K_{1,i}$$

Substituting $X_{1,i}$ and $X_{2,i}$ into Eq. 3-5, then

$$F_1 = \sum_i (X_{1,i} - X_{2,i}) = \sum_i \frac{Z_i (1 - \frac{\gamma_{1,i}}{\gamma_{2,i}})}{W_i} = 0 \quad (3-11)$$

Eq. 3-11 and the bubble point temperature condition,

$$F_2 = \sum_i (X_{1,i} - Y_i) = \sum_i \frac{Z_i(1 - K_{1,i})}{W_i} = 0 \quad (3-12)$$

In Eq. 3-11, the liquid phase distribution factor R , is one of the unknown and has the following range of value:

$$0.0 < R < 1.0 \quad (3-13)$$

The solution of Eqs. 3-11 and 3-12 is described in Appendix D. The solution method requires the use of the Secant and Newton methods to solve the functions F_1 and F_2 . Wegstein's algorithm is used to converge the phase compositions.

3.1.3 Bubble Point Temperature--Single Liquid Phase

The temperature at which a liquid mixture begins to boil at a fixed pressure is the bubble point, and is obtained from

$$F = \sum_i K_i X_i - 1.0 = 0.0 \quad (3-14)$$

Where X_i is the given feed composition and K_i is the vapor-liquid equilibrium constant for component i . A dual convergence methodology involving the Secant Method and Wegstein's Algorithm is utilized in the computation. The detailed procedure of the solution method is discussed in Appendix D.

To properly use this model during the bubble point temperature calculation, a two-liquid phase condition is always assumed at the beginning. If the value of R approaches either one or zero, then the calculation will use the single liquid phase design equation, Eq. 3-14.

3.1.4 Subcooled Liquid--Two Liquid Phases

If the system temperature is lower than the two liquid phase bubble point temperature, the liquid phase distribution factor must still be determined. The material balance equation, Eq. 3-11, is used, with the liquid phase distribution factor R as the only independent variable. The computational procedure is the same as in the bubble point temperature calculation, except that it is an isothermal evaluation. During the R calculation, if its value approaches zero or one, a single liquid phase will result.

3.1.5 Isothermal Flash

A feed composition, specified by mole fraction Z_i , having a temperature T satisfying the inequality

$$T_{BP} < T < T_{DP} \quad (3-15)$$

will separate into a vapor and at least one liquid phase. The relative amounts of material in the different phases can be calculated as follows:

Single liquid phase--For a single liquid phase flash the criteria of an overall material balance between the equilibrium vapor and liquid yields the following equation,

$$F_1 = \sum_i (X_i - Y_i) = \sum_i \frac{Z_i(1.0 - K_i)}{W_i} \quad (3-16)$$

where

$$W_i = 1.0 + (K_i - 1.0)U \quad (3-17)$$

and $U = V/F$, the molar fraction of the total feed which resides in the vapor phase.

$$X_i = \frac{Z_i}{W_i} \quad (3-18)$$

$$Y_i = K_i \frac{Z_i}{W_i} \quad (3-19)$$

The solution of Eq. 3-16 is described in Appendix D, and involves Newton's method to obtain the distribution factor and the Wegstein algorithm to assist in the phase composition convergence.

Two-Liquid Phases--For a two-liquid phase flash, the criteria of an overall material balance between an equilibrium vapor and two-liquid phase results in two equations,

$$F_1 = \sum_i (X_{1,i} - Y_i) = \sum_i \frac{Z_i (1.0 - K_{1,i})}{W_i} = 0.0 \quad (3-20)$$

$$F_2 = \sum_i (X_{1,i} - X_{2,i}) = \sum_i \frac{Z_i (1.0 - \frac{K_{1,i}}{K_{2,i}})}{W_i} = 0.0 \quad (3-21)$$

Where

$$W_i = R(1 - U) + (1 - R)(1 - U) \frac{K_{1,i}}{K_{2,i}} + UK_{1,i} \quad (3-22)$$

and

$$X_{1,i} = \frac{Z_i}{W_i} \quad (3-23a)$$

$$X_{2,i} = \frac{Z_i}{W_i} \cdot \frac{K_{1,i}}{K_{2,i}} \quad (3-23b)$$

$$Y_i = \frac{Z_i}{W_i} \cdot K_{1,i} \quad (3-23c)$$

The definitions of R and U remain as before. The solution of Eq. 3-20 and Eq. 3-21 is described in Appendix D. In the calculation of the liquid phase distribution factor R, if its value approaches either zero or one, then a single liquid phase is assumed. Also, to calculate the molar vapor fraction of the

feed, U , only Eq. 3-20 is required in the calculation. The composition convergence again is enhanced using the Wegstein algorithm.

3.1.6 Adiabatic Flash

When the total enthalpy of the system is specified, the equations associated with the isothermal flash are solved with the enthalpy balance. The general enthalpy balance based on one mole of feed is

$$H_F = UH_V + (1 - U)R H_{L1} + (1 - U) H_{L2} \quad (3-24)$$

where $H_V = \sum_i h_{v,i} y_i + h_d \quad (3-25a)$

$$H_{L1} = \sum_i h_{L,i} X_{1,i} + h_1^{\text{mix}} \quad (3-25b)$$

$$H_{L2} = \sum_i h_{L,i} X_{2,i} + h_2^{\text{mix}} \quad (3-25c)$$

where

$h_{v,i}$ = ideal gas enthalpy per mole of component i

$h_{L,i}$ = enthalpy per mole of pure liquid i

h_d = enthalpy departure from ideal vapor enthalpy

h^{mix} = heat of mixing

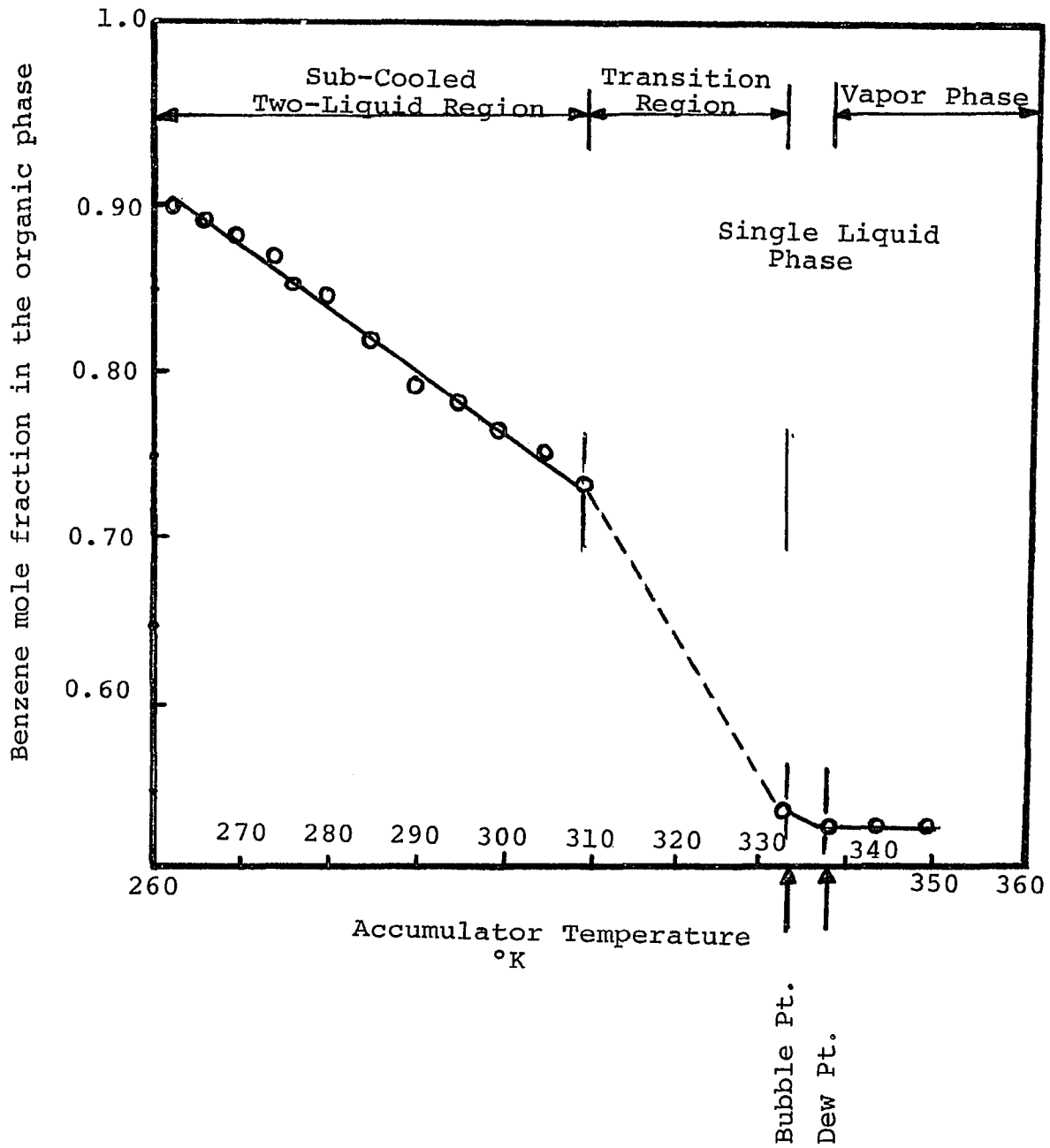
The adiabatic flash solution is achieved by decoupling the enthalpy equation from the equations which define the isothermal flash rather than solving the total set of equations simultaneously. Using temperature as the prime variable, a series of isothermal flash calculations are performed until the enthalpy balance (Eq. 3-24) is satisfied. The Secant Method is used to promote convergence of the overall enthalpy balance. The ideal

vapor enthalpy departure is evaluated via the appropriate equation of state (in this study the virial equation of state is used). The heat of mixing is evaluated from the appropriate activity coefficient relationship, i.e. the NRTL or the UNIQUAC equation. The method of solution is described in Appendix D.

3.2 Results and Conclusions

A specific stream, the overhead distillate as generated by the distillation column for the research problem, is used to demonstrate the capabilities of the two-liquid phase flash model. Also, two examples from Roche (1977) were simulated using this model. The results of the above sample runs indicated that this model is a reliable tool for two-liquid phase flash simulation, the computer inputs and outputs are included in Appendix E.

In the context of this dissertation, the focal point is the two-liquid phase separation at the bubble point and at various levels of subcooled liquid. The liquid phase activity coefficients were calculated by the NRTL equation in the above study. The results indicate that at a lower drum (or accumulator) temperature, the benzene (entrainer) concentration level in the light phase is increased. This is a favorable separation in terms of reducing the benzene loss left in the heavy phase to be recovered by further processing. The benzene concentration level in the light phase is plotted vs. the drum temperature over a range of 100°K, and illustrated in Figure 3.2.

Figure 3.2 Benzene Concentration vs. Accumulator Temperature

The two-liquid phase is stable at a temperature below 315°K, however, a single liquid phase will form between 315°K and 338°K (the dew point temperature). At a drum temperature above the dew point, only the vapor phase exists.

The results of simulating the overhead distillate stream using this two-liquid phase model also compared with the entrainer stream as reported by Chem Share. In the above two cases, the feed conditions and activity coefficient calculation method are identical, however, the resulting liquid phase splitting is different. From the available information, Chem Share seems to use the dew point liquid composition as the entrainer composition in their Gasohol simulation instead of the light phase composition at the reported drum temperature (308°K). The results of the research problem versus Chem Share's is compared and tabulated in Table 3.1.

In conclusion, a reliable two-liquid phase flash model based on Deam (1969) and Roche (1977) has been developed and validated. The desirable subcooling in the overhead distillate drum has been investigated, a drum temperature lower than the bubble point is desirable since it reduces the benzene loss. In the next chapter, this model will be combined with the reboiled absorber/stripping model via a new executive.

TABLE 3.1

Results of Two-Liquid Phase FlashResearch Problem vs. Chem Share's

- I. Feed Conditions: Pressure = 760 mmHg
Temperature = 95°F (308°K)
- II. Bubble Point Calculated = 337.8°K (148°F)
(for two-liquid phases)
- III. Dew Point Calculated = 338.9°K (150°F)
- IV. Liquid Phase Distribution factor, R = 0.4265
- V. Composition: unit in mol fraction

	<u>Feed Stream</u>	<u>Research Problem Light Phase</u>	<u>Research Problem Dew Point Liquid</u>	<u>Chem Share's Entrainer</u>
Benzene	0.5498	0.7729	0.6710	0.6759
Ethanol	0.3275	0.1855	0.2848	0.2859
Water	0.1267	0.0416	0.0432	0.0382

CHAPTER 4

THE COMPLETE AZEOTROPIC DISTILLATION MODEL

The standalone azeotropic distillation model as described in Chapter 2, and the two-liquid phase flash model which has been discussed in Chapter 3 were combined in this part of the research using an executive program. Also, two distillation columns, one simulating the benzene stripping, the other simulating the ethanol concentrating process (also the water purge) were investigated in this chapter. The complete azeotropic distillation model as illustrated in Figure 4.1 includes the azeotropic distillation column, the two-liquid phase flash decanter, the benzene stripping column, and the ethanol concentration column. This complete mode represents a typical ethanol recovery plant, using benzene as the entrainer.

4.1 Background and Theory

The executive program developed in this chapter performs the following functions:

1. Call the azeotropic distillation program to generate the overhead vapor stream and pass to the two-liquid phase flash model as the feed.
2. Call the two-liquid phase program to generate the two liquid streams.
3. Perform necessary component material balances to calculate the benzene entrainer stream.

4. Test for overall convergence.

4.2 Results

The Combination of the azeotropic distillation model and two-liquid phase flash model was successfully implemented. Sample input and output data are attached in Appendix F for the significant unit operations in the ethanol dehydration process. The scope of this chapter was modified to simulate a typical industrial ethanol purification plant which includes the above mentioned azeotropic column and two-liquid phase separation.

The typical ethanol purification plant as described by Keister (1982), Black (1980), and Van Winkle (1967) is illustrated in Figure 4.1. The simulation of the azeotropic distillation column and accumulator has been accomplished in this chapter via the executive program. The benzene stripping column which serves the purpose of recovering benzene in the feed (heavy phase stream from the accumulator) was simulated as a reboiled absorbed column. The main streams of interest associated with this column are indicated in Figure 4.1. The case input/output data are attached in Appendix F.

The bottom stream from the benzene stripping column is then fed into a conventional distillation column, the ethanol recovery column, to recover the ethanol with small amounts of entrainer from the water (bottoms). The main streams of this column are shown in Figure 4.1, and the sample input/output data are attached in Appendix F.

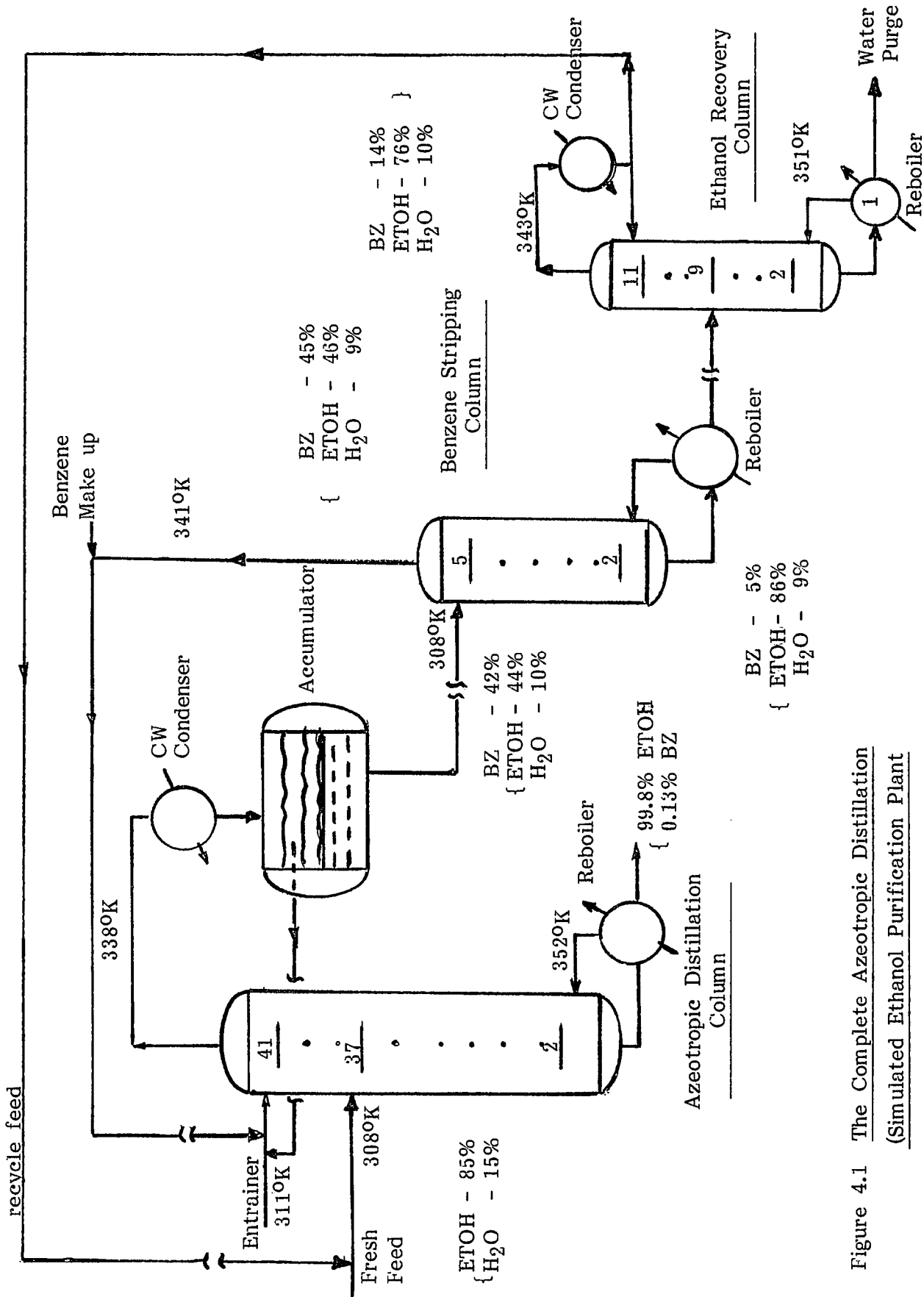


Figure 4.1 The Complete Azeotropic Distillation (Simulated Ethanol Purification Plant)

CHAPTER 5

CONCLUSIONS AND RECOMMENDATIONS

5.1 Conclusions

A rigorous multi-stage distillation model based on Naphtali (1972) has been enhanced to handle non ideal systems. Results of an industrial ethanol dehydration column simulated on this model indicate that it yields a better simulation when compared with Chem Share's gasohol simulation results. The in depth studies of the entrainer entry location and rate effects on the azeotropic distillation have determined that the entrainer should enter the column above the feed with at least two theoretical stages separating the feed. Introducing the entrainer below the feed or mixing with the feed will result in poor separation. The optimum entrainer rate was determined via a parametric study.

In the second part of this dissertation, a two-liquid phase flash model based on Roche (1977) and Deam (1967) was used to deal with the two-liquid phase in the distillate accumulator. The temperature effect on the liquid phase splitting was investigated. It is concluded that lower accumulator temperatures are preferred, since it reduces the benzene in the heavy phase (significant water present).

A typical industrial ethanol purification plant was simulated in the last part of the research using the models de-

veloped. The azeotropic distillation and two-liquid phase separation were investigated using the combined mode. The benzene stripping column and the ethanol recovery/water purge column were simulated using the stand alone model developed in the first part of this research.

5.2 Recommendations for Future Research

The following areas are recommended for further investigations in the azeotropic distillation research.

5.2.1 Vapor-Liquid Equilibrium (VLE) and Liquid-Liquid Equilibrium (LLE)

At near completion of this research, Keister (1982) presented a similar study on ethanol dehydration simulation to the 1982 AIChE Winter Meeting. According to W. Keister, the Local Effective Molar Fraction (LEMF) method is superior to UNIQUAC and NRTL methods in predicting liquid liquid equilibrium (LLE) in the distillate drum. Future investigators should verify this, as well as the modified UNIQUAC method by Anderson (1978) and UNIFAC method by Fredenslund et al. (1977) in the VLE/LLE package. The modified UNIQUAC method in theory will predict the LLE of a system containing alcohol better than the original UNIQUAC method. The UNIFAC method does not require experimentally determined binary parameters in predicting the VLE, adding this to the existing VLE package should enhance this model's capability.

5.2.2 Two-Liquid Phase Formation in the Column

It has been assumed that two liquid phases do not occur in

the column in this research. However, according to Prokopakis et al. (1980), the tendency to form two liquid phases near the feed stage or on the top section of an ethanol dehydration column can not be overlooked. The future investigator should examine the possibility of two liquid phases formation as well as developing a method to predict or avoid it. Ross-Seider (1979), and Boston-Shah (1979) have investigated this problem, but an in depth study of this subject is recommended.

5.2.3 Different Alcohol Dehydration and Entrainer Systems

Using different types of entrainer in the ethanol dehydration process has been reported by Black (1980). The future studies of comparing different entrainer effects on the ethanol dehydration process should be investigated. Also, other alcohol dehydration processes, like isopropanol dehydration, which uses cyclohexane as the entrainer as reported by Prokopakis et al. (1980) should also be investigated for entrainer effect.

5.2.4 Overall Ethanol Purification Process

In Chapter 4 of this dissertation, an investigation to simulate an industrial ethanol purification plant was made. Future studies should concentrate on the following topics:

1. Connect all the major process units together using a flow sheet simulator program (like ASPEN).
2. Determine entry locations for the overhead of the benzene stripping column and the ethanol recovery column.

3. Determine the need for make-up entrainer.
4. Determine the optimum ethanol concentration level in the feed to the azeotropic distillation column, as well as the separation requirements in the subsequent columns.

A schematic is presented in Figure 5.1 to illustrate this future system.

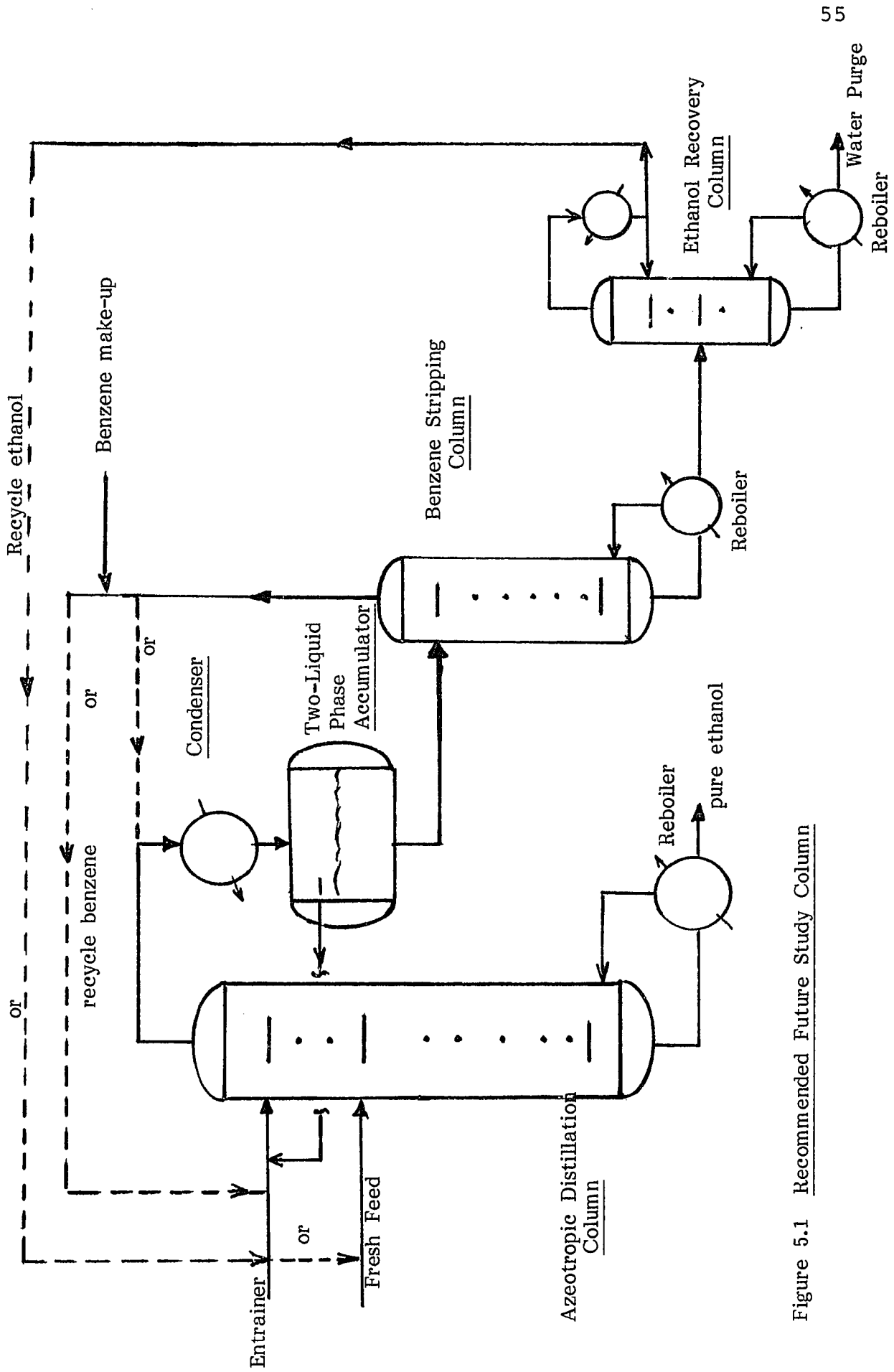


Figure 5.1 Recommended Future Study Column

APPENDIX A

DISTILLATION MODEL SOLUTION ALGORITHM

A.1 Linearization of the Discrepancy Functions

A.1.1 Component Material Balance

A.1.2 Equilibrium Condition

A.1.3 Energy Balance

A.2 Solution Algorithm

A.1 Linearization of the Discrepancy Functions

The discrepancy functions (Eq. 2-1 through 2-5) can be linearized using a two term Taylor's expansion about a known set of variables. The variables of interest are the independent variables, $l_{n,i}$, $v_{n,i}$, and T_n . The linearization process generates a new set of variables, namely $l_{n,i}$, $v_{n,i}$, and T_n . For each of the discrepancy functions, the resulting equations in the restructured mode will have the following form according to Roche (1978):

•Original equation

$$F(x_1, x_2, x_3) = 0 \quad (A1-1)$$

•Linearized equation in the restructured mode

$$\frac{\partial F}{\partial X_1} \Delta X_1 + \frac{\partial F}{\partial X_2} \Delta X_2 + \frac{\partial F}{\partial X_3} \Delta X_3 + \dots = -F \quad (A1-2)$$

The linearization of each of the discrepancy functions is given in the following summary:

A.1.1 Component Material Balance

$$F_1(n, i) = \left(1 + \frac{S_n^L}{L_n}\right) l_{n,i} + \left(1 + \frac{S_n^V}{V_n}\right) v_{n,i} - v_{n-1,i} - l_{n-1,i} - f_{n,i} = 0 \quad (2-1)$$

Linearizing the discrepancy equation, Eq. 2-1, one obtains

$$\begin{aligned} & - \Delta v_{n-1,i} + \left(1 + \frac{S_n^L}{L_n}\right) \Delta l_{n,i} - \frac{S_n^V}{V_n^2} l_{n,i} \sum_k \Delta l_{n,k} + \left(1 + \frac{S_n^V}{V_n}\right) \Delta v_{n,i} \\ & - \frac{S_n^V}{V_n^2} v_{n,i} \sum_k \Delta v_{n,k} - \Delta l_{n+1,i} = -F_1(n, i) \end{aligned} \quad (A1-3)$$

A.1.2 Equilibrium Condition

The equilibrium condition is described by the following discrepancy equation:

$$F_2(n,i) = \eta_{n,i} \frac{K_{n,i} V_n}{L_n} l_{n,i} - v_{n,i} + (1 - \eta_{n,i}) \frac{V_n}{V_{n-1}} = 0 \quad (2-2)$$

The above equation can be linearized as follows:

$$\begin{aligned} & (1 - \eta_n) \frac{V_n}{V_{n-1}} \Delta v_{n-1,i} - (1 - \eta_n) \frac{V_n}{V_{n-1}^2} v_{n-1,i} \sum_k \Delta v_{n-1,k} \\ & + \eta_n K_{n,i} \frac{V_n}{L_n} \Delta l_{n,i} - \eta_n K_{n,i} \frac{V_n}{L_n^2} l_{n,i} \sum_k l_{n,k} \\ & + \eta_n k_{n,i} \frac{l_{n,i}}{L_n} \sum_k \Delta v_{n,k} + (1 - \eta_n) \frac{v_{n-1,i}}{V_{n-1}} \sum_k \Delta v_{n,k} \\ & + \eta_n \frac{V_n}{L_n} l_{n,i} \sum_k \frac{\partial K_{n,i}}{\partial l_{n,k}} \Delta l_{n,k} + \eta_n \frac{V_n}{L_n} l_{n,i} \sum_k \frac{\partial K_{n,i}}{\partial v_{n,k}} v_{n,k} \\ & + \eta_n \frac{V_n}{L_n} l_{n,i} \frac{\partial K_{n,i}}{\partial T_n} \Delta T_n - \Delta v_{n,i} = -F_2(n,i) \end{aligned} \quad (A1-4)$$

A.1.3 Energy Balance

The discrepancy equation of energy balance is described by the following equation:

$$\begin{aligned} F_3(n) = & (1 + \frac{S_n^L}{L_n}) h_n + (1 + \frac{S_n^V}{V_n}) H_n - H_{n-1} - h_{n+1} - h_n^F - Q_n + \\ & h_n^{\text{mix}} - h_{n+1}^{\text{mix}} = 0 \end{aligned} \quad (2-5)$$

Again, the Eq. 2-5 can be linearized as follows:

$$\begin{aligned}
& - \sum_k \frac{H_{n-1}}{v_{n-1,k}} \Delta v_{n-1,k} - \frac{\partial H_{n-1}}{\partial T_{n-1}} \Delta T_{n-1} + \left(1 + \frac{S_n^L}{L_n}\right) \sum_k \frac{\partial h_n}{\partial l_{n,k}} \Delta l_{n,k} \\
& + \left(1 + \frac{S_n^L}{L_n}\right) \frac{\partial h_n}{\partial T_n} \Delta T_n + \left(1 + \frac{S_n^V}{V_n}\right) \sum_k \frac{\partial H_n}{\partial v_{n,k}} \Delta v_{n,k} \\
& - \frac{S_n^V}{V_n^2} H_n \sum_k \Delta v_{n,k} + \left(1 + \frac{S_n^V}{V_n}\right) \frac{\partial H_n}{\partial T_n} \Delta T_n \\
& - \sum_k \frac{\partial h_{n+1}}{\partial l_{n+1,k}} \Delta l_{n+1,k} - \frac{\partial h_{n+1}}{\partial T_{n+1}} \Delta T_{n+1} \\
& + \sum_k \frac{\partial h_n^{\text{mix}}}{\partial l_{n,k}} \Delta l_{n,k} + \frac{\partial h_n^{\text{mix}}}{\partial T_n} \Delta T_n \\
& - \sum_k \frac{\partial h_n^{\text{mix}}}{\partial l_{n+1,k}} \Delta l_{n+1,k} - \frac{\partial h_{n+1}^{\text{mix}}}{\partial T_n} \Delta T_n = -F_3(n) \tag{A1-5}
\end{aligned}$$

When the collective set of linearized discrepancy functions are ordered stagewise, the result is a tridiagonal super matrix as shown in Figure A1-1. The solution of the $2c + 1$ linearized equations yields the desired corrections which are used to revise the column's flow and temperature maps.

A.2 Solution Algorithm

The linearized set of discrepancy functions arranged stage-wise are schematically represented by Figure A1-1. The coefficient matrix is of the order of 5% density, and thus the unique structure is exploited in determination of the flow and temperature map corrections.

The method of solution employed is the Thomas Procedure.

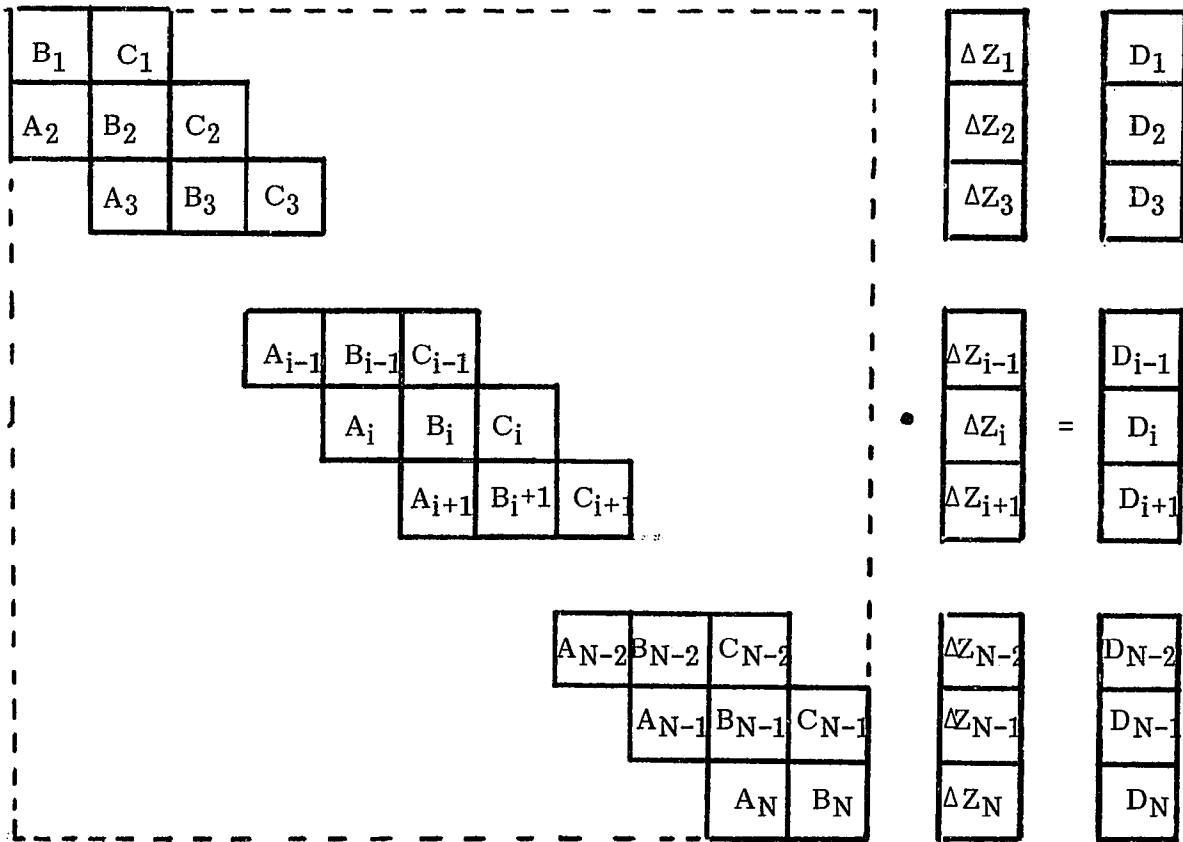


Figure A1-1 Schematic Representation of the Column's Linearized Discrepancy Functions

The Thomas procedure requires that B-matrix and D-matrix for the entire column be generated before the solution methodology be initiated. When this is completed, the inversion procedure is initiated to eliminate the A-matrix, and the C-matrix, starting with stage 2:

- Perform Gaussian elimination on

$$B_{m-1}^T * S^T = A_m^T \quad (A1-6)$$

and retain the results in S after transposing.

- Eliminate A_m (by induction) by modifying B_m and D_m (subtract from B_m the product of $S \cdot C_{m-1}$ and subtract from D_m the product of $S \cdot D_{m-1}$).

The back substitution is initiated with $m = N$, by performing Gaussian elimination on:

$$B_N * S = D_N$$

The result of this operation is the set of flow and temperature corrections for stage N. The back substitution continues for $m = N - 1, N - 2$, etc. with the preliminary requirement that the elimination of C_m be accomplished.

The two fold process is thus:

- Eliminate C_m by subtracting from D_m the product of

$$C_m * D_{m+1}$$

- Perform Gaussian elimination on

$$B_m * S = D_m$$

to obtain the set of corrections associated with stage m.

It should be noted that C_m is not explicitly defined. Rather, the unique structure of this part of the Jacobian is exploited to minimize the iterative computational effort.

APPENDIX B

THE VLE PREDICTION METHODS

The VLE Package used for this research can be summarized as follows:

B.1 Vapor Phase Component--Fugacity Coefficient Calculation

- B.1.1 Ideal Vapor
- B.1.2 Redlich-Kwong Equation of State
- B.1.3 Virial Equation with Tsonopoulos Correlation

B.2 Liquid Phase Component Activity Coefficient Calculation

- B.2.1 Ideal Liquid
- B.2.2 Wilson Equation (for single liquid phase only)
- B.2.3 NRTL Equation (for two liquid phases)
- B.2.4 UNIQUAC Equation (for two liquid phases)

B.3 Vapor Phase Enthalpy Correction

- B.3.1 Redlich-Kwong Equation of State
- B.3.2 Virial Equation of State

B.4 Liquid Phase Enthalpy of Mixing

- B.4.1 Using Wilson Equation
- B.4.2 Using NRTL Equation
- B.4.3 Using UNIQUAC Equation

B.1 Vapor Phase Component Fugacity Coefficient Calculation

B.1.1 Ideal Vapor

For ideal vapor, the fugacity coefficient of component i is set equal to 1, then the equilibrium constant can be expressed as

$$K_i = \frac{\gamma_i P_i^S}{\pi} \quad (2-14)$$

the saturation vapor pressure P_i^S is calculated using Eq. 2-15.

B.1.2 Redlich-Kwong Equation for Real Vapor

The Redlich-Kwong equation of state can be expressed as follows:

$$\ln \phi_i = (Z - 1) \frac{B_i}{B} - \ln(Z - Bp) - \frac{A^2}{B} \left(\frac{2A_i}{A} - \frac{B_i}{B} \right) \ln \left(1 + \frac{Bp}{Z} \right)$$

where: (B1-7)

$$A^2_i = a \frac{T_{ri}^{-2.5}}{P_{ci}} \quad (B1-8)$$

$$B_i = b/T_{ri}P_{ci} \quad (B1-9)$$

$$A = \sum_i y_i A_i \quad (B1-10)$$

$$B = \sum_i y_i B_i \quad (B1-11)$$

Where the subscript r stands for reduced property and c for the critical point. The compressibility factor Z is obtained by solving the following R-K equation of state, which is a cubic equation of Z .

$$z^3 - z^2 + Dz - A^2P^2B = 0 \quad (\text{B1-12})$$

where $D = A^2P - BP - B^2P^2$

$$A = \sum_i A_i Y_i$$

$$B = \sum_i B_i Y_i$$

$$D = D_i$$

$$A_i = a/D$$

$$B_i = b/T$$

Eq. (B1-12) can be further simplified as

$$z = \frac{1}{1-h} \frac{A^2}{B} \frac{h}{1+h} \quad (\text{B1-12A})$$

where $h = \frac{BP}{Z}$

The constants a and b are, respectively, 0.478 and 0.0867, if the first and the second partial derivatives of pressure with respect to volume are set equal to zero at the critical point. If experimental PVT data are available, these constants may be treated as adjustable parameters to obtain a more accurate representation of the vapor phase.

B.1.3 Virial Equation with Tsonopoulos' Correlation

Tsonopoulos (1972) extended the Pitzer-Curl equation for polar gases. In reduced form, the Pitzer-Curl equation is

$$\frac{B_{ii}P_{ci}}{RT_{ci}} = f_{PC}^{(0)}(T_R) + \omega_i f_{PC}^{(1)}(T_R) \quad (\text{B1-13})$$

where

$$f_{PC}^{(0)}(T_R) = 0.1445 - \frac{0.330}{T_R} - \frac{0.1385}{T_R^2} - \frac{0.0121}{T_R^3}$$

$$f_{PC}^{(1)}(T_R) = 0.073 + \frac{0.46}{T_R} - \frac{0.50}{T_R^2} - \frac{0.097}{T_R^3} - \frac{0.0073}{T_R^8}$$

P_{Ci} - is the critical pressure

T_{Ci} - is the absolute critical temperature

ω_i - is the acentric factor

Tsonopoulos suggested that a third function in reduced temperature $f^{(2)}(T_R)$ be added to the right hand side of Eq. B1-13 to be used for polar gases.

$$f^{(2)}(T_R) = \frac{a_i}{T_R^6} - \frac{b_i}{T_R^8} \quad (B1-14)$$

For nonhydrogen bonded compounds, e.g. ketones, aldehydes, and ethers, "a" is required, and can be correlated as a function of reduced dipole moment, μ_R .

$$\mu_R = \frac{10^5 \mu^2 P_{Ci}}{T_{Ci}^2} \quad (B1-15)$$

For hydrogen bonded compounds, e.g. alcohols and water, "b" is also required. A few available values for "a" and "b" along with methods of estimation were provided by Tsonopoulos (1972).

The binary second virial cross coefficient, B_{ij} , can be obtained from the same correlations using Eq. B1-13 by replacing P_{Ci} , T_{Ci} , ω_i , a_i , and b_i with the appropriate P_{Cij} , T_{Cij} , ω_{ij} , a_{ij} and b_{ij} .

$$\frac{B_{ij} P_{Cij}}{RT_{Cij}} = f_{PC}^{(0)}(T_R) + \omega_{ij} f_{PC}^{(1)}(T_R) \quad (B1-13A)$$

The mixing rules provided by Tsonopoulos (1972) are:

$$T_{Cij} = (T_{Ci}T_{Cj})^{1/2}(1 - K_{ij}) \quad (B1-16)$$

$$P_{Cij} = \frac{4T_{Cij} \left(\frac{P_{Ci}v_{Ci}}{T_{Ci}} + \frac{P_{Cj}v_{Cj}}{T_{Cj}} \right)}{(v_{Ci}^{1/3} + v_{Cj}^{1/3})^3} \quad (B1-17)$$

$$\omega_{ij} = 0.5(\omega_i + \omega_j) \quad (B1-18)$$

The value for T_{Cij} requires a "characteristic constant" for each binary, K_{ij} , to account for nonideal mixing in the vapor. The "characteristic constant" for selected compounds were given by Tsonopoulos. For polar/polar binaries, the polar contribution to B_{ij} is calculated by assuming that

$$a_{ij} = 0.5(a_i + a_j) \quad (B1-19)$$

$$b_{ij} = 0.5(b_i + b_j) \quad (B1-20)$$

For polar/nonpolar binaries, B_{ij} is assumed to have no polar term:

$$a_{ij} = 0$$

$$b_{ij} = 0$$

B.2 The Liquid Phase Activity Coefficient Calculations

The activity coefficients, γ_i , can be obtained from G^E , the molar excess Gibbs energy of mixing, by partial differentiation:

$$\left(\frac{\partial n \Delta G^E}{\partial n_i} \right)_{T,p,n_j \neq n_i} = RT \ln \gamma_i$$

There exists a large number of equations for the representation of the molar excess Gibbs energy as a function of

concentration at constant temperature. The Wilson, NRTL and UNIQUAC equations are cited here due to their newness and advantages over Margules and VanLaar equations.

For multicomponent systems, the Wilson, NRTL and UNIQUAC require only the binary parameters. This is due to the modern concept of local composition, which has been used in the derivation of these equations, either on the basis of local volume fraction, as with Wilson's equation or of local mole fraction as in the case of NRTL or UNIQUAC.

The Wilson equation is not applicable for a system containing immiscible liquid phases. In this research, the NRTL and UNIQUAC equations are used since the research problem is dealing with heterogeneous azeotropes (two liquid phases). For the completeness of activity coefficient calculation methods survey, the Wilson Equation is also described in this Appendix.

B.2.1 Ideal Liquid

For ideal liquid, the activity coefficient, γ_i , for a given component i is set equal to one in Equation 2-13.

$$\gamma_i = 1.0$$

B.2.2 Wilson Equation

For a solution of m components Wilson's equation is:

$$\frac{q^E}{RT} = - \sum_{i=1}^m x_i \ln \left[\sum_{j=1}^m x_j A_{ij} \right] \quad (B2-1)$$

where

$$A_{ij} = \frac{v_j}{v_i} \text{Exp} - \left[\frac{(\lambda_{ij} - \lambda_{ii})}{RT} \right] \quad (\text{B2-2})$$

$$A_{ji} = \frac{v_i}{v_j} \text{Exp} - \left[\frac{(\lambda_{ji} - \lambda_{jj})}{RT} \right] \quad (\text{B2-3})$$

and

$$A_{ii} = A_{jj} = 1 \quad (\text{B2-4})$$

The activity coefficient for any component i is given by:

$$\ln \gamma_i = -\ln \left[\sum_{j=1}^m x_j A_{ij} \right] + 1 - \frac{\sum_{k=1}^m x_i A_{ki}}{\sum_{j=1}^m x_j A_{kj}} \quad (\text{B2-5})$$

Equation (B2-3) requires only parameters which can be obtained from binary data; for each possible binary pair in the multicomponent solution, two parameters, A_{ij} and A_{ji} are needed.

B.2.3 NRTL (Non Random Two Liquid) Equation

For a solution of m component, Renon's NRTL equation is:

$$\frac{g^E}{RT} = \sum_{i=1}^m x_i \frac{\sum_{j=1}^m \tau_{ji} G_{ji} x_j}{\sum_{l=1}^m G_{li} x_l} \quad (\text{B2-6})$$

where

$$\tau_{ji} = \left(\frac{g_{ji} - g_{ii}}{RT} \right) \text{ and } \tau_{ij} = \left(\frac{g_{ij} - g_{jj}}{RT} \right) \quad (\text{B2-7})$$

$$G_{ji} = \text{Exp}(-\alpha_{ji} \tau_{ji}) \quad (\text{B2-8})$$

$$\tau_{ii} = \tau_{jj} = 0$$

$$G_{ii} = G_{jj} = 1$$

$$\alpha_{ij} = \alpha_{ji}$$

The activity coefficient for a component i is given by:

$$\ln \gamma_i = \frac{\sum_{j=1}^m \tau_{ji} G_{ji} x_j}{\sum_{l=1}^m G_{li} x_l} + \sum_{j=1}^m \frac{x_j G_{ij}}{\sum_{l=1}^m G_{lj} x_l} \left(\tau_{ij} - \frac{\sum_{n=1}^m x_n n_j G_{nj}}{\sum_{l=1}^m G_{lj} x_l} \right) \quad (\text{B2-9})$$

For a given binary, three parameters, τ_{ij} , τ_{ji} and α_{ij} are needed to calculate the activity coefficient.

B.2.4 UNIQUAC (Universal Quasi-Chemical) Equation

The activity coefficient for a given i component in the m component solution can be expressed by:

$$\ln \gamma_i = \ln \gamma_i^C + \ln \gamma_i^R \quad (\text{B2-10})$$

where

$$\ln \gamma_i^C = \ln \frac{\phi_i}{x_i} + \frac{z}{2} q_i \ln \frac{\theta_i}{\phi_i} + l_i - \frac{\phi_i}{x_i} \sum_{j=1}^m x_j l_j \quad (\text{B2-11})$$

$$\ln \gamma_i^R = q_i \left[1 - \ln \left(\sum_{j=1}^m \theta_j \tau_{ji} \right) - \sum_{j=1}^m \frac{\theta_j \tau_{ij}}{\sum_{k=1}^m \theta_k \tau_{kj}} \right] \quad (\text{B2-12})$$

$$l_i = \frac{z}{2} (r_i - q_i) - (r_i - 1) \quad (\text{B2-13})$$

$$\phi_i = \frac{r_i x_i}{\sum_k r_k x_k} = (\text{area fraction of component}) \quad (\text{B2-14})$$

$$Q_i = \frac{q_i x_i}{\sum_k q_k x_k} = (\text{volume fraction of component}) \quad (\text{B2-15})$$

Notations:

- 1) A_{ij} and A_{ji} adjustable Wilson's binary parameters.
- 2) V_i^L , molar volume of pure liquid component.
- 3) λ_{ij} , an empirically determined interaction energy between components i and j

$$\lambda_{ij} = \lambda_{ji}$$

- 4) g_{ij} , parameter for interaction between components i and j

$$g_{ij} = g_{ji}$$

- 5) α_{ij} , nonrandomness parameter for NRTL Equation

$$\alpha_{ij} = \alpha_{ji}$$

- 6) τ_{ij}, τ_{ji} , adjustable binary parameters in NRTL and UNIQUAC equations.

- 7) q_i , area parameter of component i .

- 8) r_i , volume parameter of component i .

- 9) U_{ij} , parameter of interaction between components i and j ,

$$U_{ij} = U_{ji}$$

- 10) Z , coordination number

- 11) γ_i^C , combinatorial part of activity coefficient of component i .

- 12) γ_i^R , residual part of activity coefficient of component i .

and

$$\tau_{ji} = \text{Exp} - \left(\frac{U_{ji} - U_{ii}}{RT} \right) \quad (\text{B2-16})$$

B.3 Vapor Phase Enthalpy Correction

The molar enthalpy correction, ΔH , for a vapor at temperature T and pressure π relative to the ideal vapor at the same T and composition is given exactly by

$$\Delta H = \int_0^{\pi} \left[v - T \left(\frac{\partial v}{\partial T} \right)_{p,y} \right] dP \quad (B3-1)$$

where v is the vapor phase molar volume.

B.3.1 Redlich-Kwong Equation of State

The vapor enthalpy correction for Redlich-Kwong equation of state is given by Prausnitz et al. (1977) as

$$\Delta H = \frac{bRT}{v-b} - \frac{a}{T^{0.5}(v+b)} - \frac{3a}{2bT^{0.5}} \ln \frac{v+b}{v} \quad (B3-2)$$

B.3.2 Virial Equation of State

The enthalpy correction for the virial equation can be expressed as

$$\Delta H = \pi \sum_{i=1}^m \sum_{j=1}^m Y_i Y_j \left[B_{ij} - T \frac{dB_{ij}}{dT} \right] \quad (B3-3)$$

B.4 Liquid Phase Enthalpy of Mixing

The liquid enthalpy of mixing can be expressed as:

$$\left(\frac{\partial g^E}{\partial T} \right)_{p,x} = h^{mix} \quad (B4-1)$$

or

$$\left[\frac{\partial (\sum x_i \ln i)}{\partial T} \right]_{p,x} = - \frac{h^{mix}}{RT^2} \quad (B4-2)$$

B.4.1 Wilson Equation

The enthalpy of mixing calculated from Wilson's equation is

$$h^{mix} = R \left[\frac{-x_i (x_j \frac{\partial A_{ji}}{\partial (1/T)})}{x_i + x_j A_{ji}} - \frac{x_j (x_i \frac{\partial A_{ij}}{\partial (1/T)})}{x_j + x_i A_{ij}} \right] \quad (B4-3)$$

where:

$$\frac{\partial A_{ij}}{\partial (1/T)} = -A_{ij} \tau'_{ij} + \frac{A_{ij}}{(V_j/V_i)} \frac{\partial (v_j/V_i)}{\partial (1/T)} \quad (B4-4)$$

$$\tau'_{ij} = \frac{\partial \left[\frac{(\lambda_{ji} - \lambda_{ij})}{RT} \right]}{\partial (1/T)} \quad (\text{B4-5})$$

B.4.2 NRTL Equation

The enthalpy of mixing using the NRTL equation can be expressed as

$$\begin{aligned} \frac{h_{\text{mix}}}{R} = & x_i x_j \left[\frac{G_{ji} \tau'_{ji}}{x_i + x_j G_{ji}} + \tau_{ji} \left(\frac{G_{ji}^2 x_j (\alpha_{ji} \tau'_{ji} + \tau_{ji} \alpha'_{ji})}{(x_i + x_j G_{ji})^2} \right. \right. \\ & \left. \left. - \frac{G_{ji} (\alpha_{ji} \tau'_{ji} + \tau_{ji} \alpha'_{ji})}{x_i + x_j G_{ji}} \right) + \frac{G_{ij} \tau'_{ij}}{x_j + x_i G_{ij}} \right. \\ & \left. + \tau_{ij} \left(\frac{G_{ij}^2 x_i (\alpha_{ij} \tau'_{ij} + \tau_{ij} \alpha'_{ij})}{(x_j + x_i G_{ij})^2} - \frac{G_{ij} (\alpha_{ij} \tau'_{ij} + \tau_{ij} \alpha'_{ij})}{x_j + x_i G_{ij}} \right) \right] \end{aligned} \quad (\text{B4-1})$$

where

$$\tau'_{ji} = \frac{\partial \tau_{ji}}{\partial (1/T)} = \frac{A_{ij}}{RT}$$

$$\alpha'_{ij} = \frac{\partial \alpha_{ij}}{\partial (1/T)} = -B' T^2$$

$$(\lambda_{ij} - \lambda_{jj}) = A_{ij} + B_{ij} T$$

$$\alpha_{ij} = A' + B' T$$

$$\tau_{ij} = \frac{A_{ij}}{RT} + \frac{B_{ij}}{R}$$

G_{ji} , $(\lambda_{ij} - \lambda_{jj})$, α_{ij} and τ_{ij} are the same as these defined in Sections B.2.3 and B.2.4.

B.4.3 UNIQUAC Equation

The enthalpy of mixing using the UNIQUAC equation as described by Prausnitz et al. (1980) is

$$h^{\text{mix}} = -RT \sum_i^m \left[\frac{\sum_{j=1}^m \theta_j \tau_{ji} \ln \tau_{ji}}{\sum_{j=1}^m \theta_j \tau_{ji}} \right] \quad (\text{B4-3})$$

In the above equation, if $(U_{ji} - U_{ii})$ is assumed to be temperature dependent, then

$$\tau_{ji} = \text{Exp} \left[- \frac{a_{ji} + b_{ji}/T}{T} \right]$$

and Equation (B4-3) becomes

$$h^{\text{mix}} = R \sum_{i=1}^m \left\{ \frac{q_i x_i}{\sum_{j=1}^m \theta_j \tau'_{ji}} \left[\sum_{j=1}^m \theta_j \tau'_{ji} (a_{ji} + 2b_{ji}/T) \right] \right\} \quad (\text{B4-4})$$

The parameters are the same as those defined in Section B.2.4.

APPENDIX C

THE RESEARCH PROBLEM--SAMPLE COMPUTER INPUTS
AND RESULTS

- C.1 A Reboiled Absorber Case Using NRTL Equation for Activity Coefficient Calculation
- C.2 A Conventional Distillation Column Case Using the NRTL Equation for Activity Coefficient Calculation
- C.3 A Conventional Distillation Column Case Using the UNIQUAC Equation for Activity Coefficient Calculation

C.1 Inputs

```

0.1 //BAYWAY JOB (JA00008-SAI,NJIT,-30,10),'ROCHE-TSAI',
0.2 // MSGLEVEL=(1,1),TIME=(2,30)
0.3 /*FRIDRITY -2

?
10055. //STEP6 EXEC PGM=BAYWAY2,REGION=230K
10056. //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
10057. //FT06F001 DD SYSOUT=A
10058. //FT05F001 DD *
10059. CHEMSHARE'S ETOH DEHYDRATION --VIRFUG VAPOR--NRTL LIQUID
10060. 3 1 2 2 0 0 1
10061. ETHANOL 0 46.07 516.2 63.0 167.0 64.0
10062. 0.0878 0.0572 0.248 0.635
10063. 202.06 351.1 0.58 0.38
10064. 18.9119 -3803.98 -41.68
10065. WATER 0 18.01 647.3 217.3 56.0 18.84
10066. 0.0279 0.0229 0.229 0.344
10067. 538.7 373.2 1.0 0.454
10068. 18.3036 -3816.44 -46.13
10069. BENZENE 0 78.114 562.1 48.3 259.0 92.26
10070. 0.0 0.0 0.544 0.251
10071. 94.118 353.3 0.41 0.249
10072. 15.9008 -2788.51 -52.36
10073. 0.0 0.20 0.15
10074. 0.20 0.0 0.40
10075. 0.15 0.40 0.0
10076. -86.763 0.0 1289.908 0.0 282.332 0.0 1096.899
10077. 3820.954 0.0 2213.355 0.0
10078. 0.270 0.295 0.267
10079. 3 042 2 0 020 2 0
10080. 1.0 0.0 0.0
10081. 727.00 0.0 .000005 338.0 352.0 760.0 0.001
10082. 37 310.97 0.0 191.9 66.6 0.0
10083. 41 308.2 0.0 169.2 22.65 400.0
10084. /*
10085. //
?

```

C.1. Results

2472. *** STAGewise SIMULATION PROGRAM ***

2473.

2474. CHEMSHARE'S ETOH DEHYDRATION --VIRFUG VAPOR--NRTL LIQUID

2475. NUMBER OF COMPONENTS = 3

2476. VAPOR-LIQUID EQUILIBRIUM DATA CODES:

2477. LIQUID = 1

2478. VAPOR = 2

2479. ENTHALPY = 2

2480.

2481. NOTE: ALL FLOWS ARE PER UNIT TIME, AND ENERGY

2482. VALUES ARE IN UNITS OF K-BTU PER UNIT TIME.

2483. COMPONENT NO	1	2	3
2484. NAME	ETHANOL	WATER	BENZENE
2485. MOLECULAR WEIGHT	46.1	18.0	78.1
2486. CRITICAL PROPERTIES			
2487. TEMPERATURE (DEG K)	516.2	647.3	562.1
2488. PRESSURE (ATM)	63.0	217.3	48.3
2489. VOLUME (CC/GMOL)	167.0	56.0	259.0
2490. COMPRESSIBILITY	0.248	0.229	0.544
2491. ACENTRIC FACTOR	0.635	0.344	0.251
2492. YSONOPOULOS -A-	0.0878	0.0279	0.0
2493. TSONOPOULOS -B-	0.0572	0.0229	0.0
2494. AK(1,1)=	0.0		
2495. AK(1,2)=	0.200		
2496. AK(1,3)=	0.150		
2497. AK(2,1)=	0.200		
2498. AK(2,2)=	0.0		
2499. AK(2,3)=	0.400		
2500. AK(3,1)=	0.150		
2501. AK(3,2)=	0.400		
2502. AK(3,3)=	0.0		
2503. VAPOR PRESSURE CONSTANTS			
2504. (MMHG AND DEG K)			
2505. C(1)	18.912	18.304	15.901
2506. C(2)	-3803.98	-3816.44	-2788.51
2507. C(3)	-41.68	-46.13	-52.36
2508. C(4)	0.0	0.0	0.0
2509. C(5)	0.0	0.0	0.0
2510. C(6)	0.0	0.0	0.0
2511. LIQUID MOLAR VOLUME CONSTANTS			
2512. C(1)	64.000	18.840	92.260
2513. ENTHALPY CONSTANTS (CAL/GM)			
2514. LATENT HEAT	202.1	538.7	94.1
2515. AT DEG K	351.1	373.2	353.3
2516. LIQUID SP. HEAT CONSTANTS			
2517. C(1)	0.580	1.000	0.410
2518. C(2)	0.0	0.0	0.0
2519. C(3)	0.0	0.0	0.0
2520. C(4)	0.0	0.0	0.0
2521. VAPOR SP. HEAT CONSTANTS			
2522. C(1)	0.380	0.454	0.249
2523. C(2)	0.0	0.0	0.0
2524. C(3)	0.0	0.0	0.0
2525. C(4)	0.0	0.0	0.0

NRTL CONSTANTS, CAL/GMOL

A 12 = -86.76 B 12 = 0.0 A 21 = 1289.91 B 21 = 0.0
 A 13 = 282.33 B 13 = 0.0 A 31 = 1096.90 B 31 = 0.0
 A 23 = 3820.95 B 23 = 0.0 A 32 = 2213.35 B 32 = 0.0

ALPHA12 = 0.2700 ALPHA13 = 0.2950 ALPHA23 = 0.2670 ALPHA

CHEMSHARE'S ETOH DEHYDRATION --VIRFUG VAPOR--NRTL LIQUID
 DISTILLATION COLUMN; PRODUCT RATES & REFLUX RATIO SPECIFIED

FEED STREAMS:

STAGE	37.	41.
TEMP	310.97	308.20
FRACV	0.0	0.0
H	-2.8042E 05	-7.9823E 05
COMP 1 ETHANOL	191.900	169.200
COMP 2 WATER	66.600	22.650
COMP 3 BENZENE	0.0	400.000
TOTAL	258.500	591.850
NUMBER OF STAGES	42	

361.100
 89.250
 400.000
 850.350

ESTIMATED OVERHEAD PRODUCT RATE 727.0000
 ESTIMATED BOTTOMS PRODUCT RATE 123.3496

PRESSURE AT TOP OF COLUMN 759.99072
 CONSTANT DELTA P PER STAGE 0.00100
 REFLUX RATIO 0.000005

2598.	FINAL COLUMN PROFILES:					
2599.	STAGE	TEMP	PRES	EFF(%)	LIQUID(MOLS)	VAPOR(MOLS)
2600.	1	351.37	760.041	100.00	123.350	742.934
2601.	2	351.15	760.040	100.00	866.284	743.371
2602.	3	350.49	760.038	100.00	866.721	745.182
2603.	4	348.74	760.037	100.00	868.531	752.778
2604.	5	345.47	760.036	100.00	876.127	774.207
2605.	6	342.35	760.035	100.00	897.556	800.991
2606.	7	341.17	760.033	100.00	924.341	814.621
2607.	8	340.94	760.032	100.00	937.970	818.529
2608.	9	340.89	760.031	100.00	941.878	819.462
2609.	10	340.89	760.030	100.00	942.811	819.675
2610.	11	340.88	760.029	100.00	943.024	819.724
2611.	12	340.88	760.027	100.00	943.073	819.735
2612.	13	340.88	760.026	100.00	943.084	819.737
2613.	14	340.88	760.025	100.00	943.086	819.737
2614.	15	340.88	760.024	100.00	943.086	819.736
2615.	16	340.88	760.022	100.00	943.085	819.734
2616.	17	340.88	760.021	100.00	943.083	819.732
2617.	18	340.88	760.020	100.00	943.081	819.727
2618.	19	340.88	760.019	100.00	943.076	819.720
2619.	20	340.88	760.018	100.00	943.070	819.710
2620.	21	340.88	760.016	100.00	943.060	819.696
2621.	22	340.87	760.015	100.00	943.045	819.671
2622.	23	340.87	760.014	100.00	943.021	819.635
2623.	24	340.86	760.013	100.00	942.985	819.580
2624.	25	340.85	760.011	100.00	942.929	819.496
2625.	26	340.84	760.010	100.00	942.845	819.368
2626.	27	340.81	760.009	100.00	942.718	819.177
2627.	28	340.77	760.008	100.00	942.526	818.890
2628.	29	340.72	760.007	100.00	942.239	818.469
2629.	30	340.64	760.005	100.00	941.818	817.864
2630.	31	340.53	760.004	100.00	941.213	817.022
2631.	32	340.37	760.003	100.00	940.371	815.899
2632.	33	340.15	760.002	100.00	939.249	814.492
2633.	34	339.89	760.000	100.00	937.841	812.856
2634.	35	339.57	759.999	100.00	936.205	811.127
2635.	36	339.24	759.998	100.00	934.476	809.481
2636.	37	338.91	759.997	100.00	932.831	788.169
2637.	38	338.90	759.996	100.00	653.018	791.221
2638.	39	338.92	759.994	100.00	656.071	791.480
2639.	40	338.91	759.993	100.00	656.329	791.442
2640.	41	338.89	759.992	100.00	656.291	727.005
2641.	42	338.89	759.991	100.00	0.004	727.001

2642.	FINAL VAPOR COMPOSITION PROFILE (MOLE%):			
2643.	STAGE	ETHANOL	WATER	BENZENE
2644.	1	99.48808	0.00004	0.51191
2645.	2	98.33203	0.00004	1.66794
2646.	3	94.92275	0.00005	5.07721
2647.	4	86.22646	0.00005	13.77351
2648.	5	70.69477	0.00005	29.30522
2649.	6	55.95709	0.00006	44.04287
2650.	7	49.11826	0.00008	50.88168
2651.	8	47.14054	0.00012	52.85934
2652.	9	46.66149	0.00019	53.33833
2653.	10	46.55068	0.00031	53.44902
2654.	11	46.52515	0.00048	53.47440
2655.	12	46.51917	0.00074	53.48011
2656.	13	46.51745	0.00114	53.48141
2657.	14	46.51656	0.00176	53.48168
2658.	15	46.51558	0.00271	53.48170
2659.	16	46.51424	0.00417	53.48162
2660.	17	46.51213	0.00641	53.48147
2661.	18	46.50898	0.00985	53.48119
2662.	19	46.50413	0.01512	53.48077
2663.	20	46.49662	0.02321	53.48017
2664.	21	46.48511	0.03562	53.47927
2665.	22	46.46757	0.05465	53.47781
2666.	23	46.44064	0.08379	53.47559
2667.	24	46.39950	0.12836	53.47216
2668.	25	46.33666	0.19640	53.46695
2669.	26	46.24097	0.29996	53.45910
2670.	27	46.09584	0.45685	53.44730
2671.	28	45.87712	0.69291	53.42998
2672.	29	45.55092	1.04436	53.40473
2673.	30	45.07109	1.55954	53.36937
2674.	31	44.38018	2.29773	53.32209
2675.	32	43.41490	3.32155	53.26357
2676.	33	42.12145	4.67894	53.19964
2677.	34	40.48127	6.37442	53.14432
2678.	35	38.53958	8.34078	53.11963
2679.	36	36.41735	10.43428	53.14838
2680.	37	34.28720	12.47106	53.24174
2681.	38	33.01278	12.14383	54.84340
2682.	39	32.86877	12.12716	55.00408
2683.	40	32.81918	12.16647	55.01435
2684.	41	32.72653	12.27641	54.99706
2685.	42	32.72654	12.27645	54.99701

2686.	FINAL	LIQUID COMPOSITION PROFILE (MOLE%):		
2687.	STAGE	ETHANOL	WATER	BENZENE
2688.	1	99.86089	0.00003	0.13909
2689.	2	99.54116	0.00004	0.45882
2690.	3	98.54964	0.00004	1.45035
2691.	4	95.62410	0.00005	4.37590
2692.	5	88.14607	0.00005	11.85392
2693.	6	74.70303	0.00005	25.29696
2694.	7	61.81589	0.00005	38.18405
2695.	8	55.79129	0.00007	44.20865
2696.	9	54.04488	0.00011	45.95501
2697.	10	53.62170	0.00017	46.37814
2698.	11	53.52379	0.00027	46.47595
2699.	12	53.50122	0.00042	46.49835
2700.	13	53.49597	0.00065	46.50340
2701.	14	53.49446	0.00100	46.50455
2702.	15	53.49367	0.00153	46.50480
2703.	16	53.49283	0.00236	46.50480
2704.	17	53.49166	0.00363	46.50471
2705.	18	53.48986	0.00557	46.50457
2706.	19	53.48716	0.00856	46.50429
2707.	20	53.48299	0.01315	46.50398
2708.	21	53.47654	0.02018	46.50329
2709.	22	53.46664	0.03097	46.50239
2710.	23	53.45159	0.04750	46.50093
2711.	24	53.42844	0.07283	46.49872
2712.	25	53.39310	0.11157	46.49533
2713.	26	53.33910	0.17071	46.49020
2714.	27	53.25686	0.26071	46.48243
2715.	28	53.13216	0.39707	46.47077
2716.	29	52.94421	0.60220	46.45359
2717.	30	52.66390	0.90758	46.42853
2718.	31	52.25152	1.35516	46.39332
2719.	32	51.65766	1.99634	46.34599
2720.	33	50.82787	2.88534	46.28680
2721.	34	49.71565	4.06355	46.22082
2722.	35	48.30484	5.53457	46.16060
2723.	36	46.63395	7.23981	46.12624
2724.	37	44.80659	9.05455	46.13885
2725.	38	30.85966	4.85332	64.28702
2726.	39	29.33865	4.49412	66.16724
2727.	40	29.16642	4.47703	66.35657
2728.	41	29.10641	4.52400	66.36962
2729.	42	29.10652	4.52404	66.36944

2730.	FINAL VAPOR COMPOSITION PROFILE (MOLES):			
2731.	STAGE	ETHANOL	WATER	BENZENE
2732.	1	739.131	0.000	3.803
2733.	2	730.972	0.000	12.399
2734.	3	707.347	0.000	37.834
2735.	4	649.094	0.000	103.684
2736.	5	547.324	0.000	226.883
2737.	6	448.211	0.000	352.780
2738.	7	400.127	0.001	414.493
2739.	8	385.859	0.001	432.669
2740.	9	382.373	0.002	437.087
2741.	10	381.564	0.003	438.108
2742.	11	381.378	0.004	438.343
2743.	12	381.334	0.006	438.395
2744.	13	381.321	0.009	438.407
2745.	14	381.313	0.014	438.409
2746.	15	381.305	0.022	438.409
2747.	16	381.293	0.034	438.407
2748.	17	381.275	0.053	438.405
2749.	18	381.247	0.081	438.400
2750.	19	381.204	0.124	438.393
2751.	20	381.138	0.190	438.383
2752.	21	381.036	0.292	438.367
2753.	22	380.881	0.448	438.342
2754.	23	380.644	0.687	438.305
2755.	24	380.281	1.052	438.247
2756.	25	379.727	1.609	438.159
2757.	26	378.884	2.458	438.027
2758.	27	377.606	3.742	437.828
2759.	28	375.683	5.674	437.533
2760.	29	372.820	8.548	437.101
2761.	30	368.620	12.755	436.489
2762.	31	362.596	18.773	435.653
2763.	32	354.222	27.101	434.577
2764.	33	343.076	38.110	433.307
2765.	34	329.054	51.815	431.987
2766.	35	312.605	67.654	430.868
2767.	36	294.792	84.464	430.226
2768.	37	270.241	98.293	419.635
2769.	38	261.204	96.085	433.933
2770.	39	260.150	95.984	435.346
2771.	40	259.745	96.291	435.407
2772.	41	237.923	89.250	399.831
2773.	42	237.922	89.250	399.829

2774.	FINAL LIQUID COMPOSITION PROFILE (MOLES):			
2775.	STAGE	ETHANOL	WATER	BENZENE
2776.	1	123.178	0.000	0.172
2777.	2	862.309	0.000	3.975
2778.	3	854.150	0.000	12.571
2779.	4	830.525	0.000	38.006
2780.	5	772.271	0.000	103.855
2781.	6	670.502	0.000	227.054
2782.	7	571.389	0.001	352.951
2783.	8	523.305	0.001	414.664
2784.	9	509.037	0.001	432.840
2785.	10	505.551	0.002	437.258
2786.	11	504.742	0.003	438.280
2787.	12	504.556	0.004	438.514
2788.	13	504.512	0.006	438.566
2789.	14	504.499	0.009	438.578
2790.	15	504.491	0.014	438.580
2791.	16	504.483	0.022	438.580
2792.	17	504.471	0.034	438.578
2793.	18	504.453	0.053	438.576
2794.	19	504.425	0.081	438.571
2795.	20	504.382	0.124	438.564
2796.	21	504.316	0.190	438.554
2797.	22	504.214	0.292	438.538
2798.	23	504.060	0.448	438.513
2799.	24	503.822	0.687	438.476
2800.	25	503.459	1.052	438.418
2801.	26	502.905	1.610	438.331
2802.	27	502.062	2.458	438.198
2803.	28	500.784	3.742	437.999
2804.	29	498.861	5.674	437.704
2805.	30	495.998	8.548	437.272
2806.	31	491.798	12.755	436.660
2807.	32	485.774	18.773	435.824
2808.	33	477.400	27.101	434.748
2809.	34	466.254	38.110	433.478
2810.	35	452.232	51.815	432.158
2811.	36	435.783	67.654	431.039
2812.	37	417.970	84.464	430.397
2813.	38	201.519	31.693	419.806
2814.	39	192.482	29.485	434.104
2815.	40	191.428	29.384	435.517
2816.	41	191.023	29.691	435.578
2817.	42	0.001	0.000	0.002

2818.	STAGE #	1				
2819.	TEMP, DEG K	351.372				
2820.	PRES, MMHG	760.0408				
2821.	# NAME	K	VP	FUG	GAM	
2822.	1 ETHANOL	0.99626	756.57	1.0008	1.0000	
2823.	2 WATER	1.1508	330.50	0.97409	2.7169	
2824.	3 BENZENE	3.6804	716.99	0.97208	4.0135	
2825.	STAGE #	2				
2826.	TEMP, DEG K	351.150				
2827.	PRES, MMHG	760.0396				
2828.	# NAME	K	VP	FUG	GAM	
2829.	1 ETHANOL	0.98785	749.92	1.0012	1.0000	
2830.	2 WATER	1.1460	327.50	0.97430	2.7298	
2831.	3 BENZENE	3.6353	712.03	0.97293	3.9883	
2832.	STAGE #	3				
2833.	TEMP, DEG K	350.486				
2834.	PRES, MMHG	760.0383				
2835.	# NAME	K	VP	FUG	GAM	
2836.	1 ETHANOL	0.96319	730.37	1.0021	1.0002	
2837.	2 WATER	1.1324	318.68	0.97490	2.7701	
2838.	3 BENZENE	3.5007	697.39	0.97544	3.9112	
2839.	STAGE #	4				
2840.	TEMP, DEG K	348.743				
2841.	PRES, MMHG	760.0371				
2842.	# NAME	K	VP	FUG	GAM	
2843.	1 ETHANOL	0.90172	681.04	1.0042	1.0021	
2844.	2 WATER	1.1020	296.48	0.97616	2.8940	
2845.	3 BENZENE	3.1476	660.06	0.98173	3.6918	
2846.	STAGE #	5				
2847.	TEMP, DEG K	345.466				
2848.	PRES, MMHG	760.0359				
2849.	# NAME	K	VP	FUG	GAM	
2850.	1 ETHANOL	0.80201	595.85	1.0069	1.0160	
2851.	2 WATER	1.0791	258.24	0.97741	3.2494	
2852.	3 BENZENE	2.4722	594.16	0.99257	3.1861	
2853.	STAGE #	6				
2854.	TEMP, DEG K	342.345				
2855.	PRES, MMHG	760.0347				
2856.	# NAME	K	VP	FUG	GAM	
2857.	1 ETHANOL	0.74906	523.23	1.0076	1.0799	
2858.	2 WATER	1.1879	225.79	0.97724	4.0918	
2859.	3 BENZENE	1.7410	536.34	1.0021	2.4620	
2860.	STAGE #	7				

2860.	STAGE #	7				
2861.	TEMP, DEG K	341.165				
2862.	PRES, MMHG	760.0334				
2863.	# NAME	K	VP	FUG	GAM	
2864.	1 ETHANOL	0.79459	497.78	1.0068	1.2050	
2865.	2 WATER	1.4652	214.44	0.97665	5.3171	
2866.	3 BENZENE	1.3325	515.67	1.0060	1.9523	
2867.	STAGE #	8				
2868.	TEMP, DEG K	340.936				
2869.	PRES, MMHG	760.0322				
2870.	# NAME	K	VP	FUG	GAM	
2871.	1 ETHANOL	0.84494	492.96	1.0063	1.2945	
2872.	2 WATER	1.6709	212.30	0.97641	6.1264	
2873.	3 BENZENE	1.1957	511.73	1.0070	1.7636	
2874.	STAGE #	9				
2875.	TEMP, DEG K	340.894				
2876.	PRES, MMHG	760.0310				
2877.	# NAME	K	VP	FUG	GAM	
2878.	1 ETHANOL	0.86338	492.09	1.0062	1.3253	
2879.	2 WATER	1.7425	211.91	0.97634	6.4011	
2880.	3 BENZENE	1.1607	511.01	1.0072	1.7140	
2881.	STAGE #	10				
2882.	TEMP, DEG K	340.885				
2883.	PRES, MMHG	760.0298				
2884.	# NAME	K	VP	FUG	GAM	
2885.	1 ETHANOL	0.86813	491.90	1.0062	1.3331	
2886.	2 WATER	1.7608	211.83	0.97633	6.4709	
2887.	3 BENZENE	1.1525	510.86	1.0072	1.7023	
2888.	STAGE #	11				
2889.	TEMP, DEG K	340.883				
2890.	PRES, MMHG	760.0286				
2891.	# NAME	K	VP	FUG	GAM	
2892.	1 ETHANOL	0.86924	491.86	1.0062	1.3349	
2893.	2 WATER	1.7650	211.81	0.97633	6.4871	
2894.	3 BENZENE	1.1506	510.83	1.0072	1.6996	
2895.	STAGE #	12				
2896.	TEMP, DEG K	340.883				
2897.	PRES, MMHG	760.0273				
2898.	# NAME	K	VP	FUG	GAM	
2899.	1 ETHANOL	0.86949	491.85	1.0062	1.3353	
2900.	2 WATER	1.7660	211.80	0.97633	6.4909	
2901.	3 BENZENE	1.1501	510.82	1.0072	1.6990	

3056.	STAGE #	35				
3057.	TEMP, DEG K	339.574				
3058.	PRES, MMHG	759.9993				
3059.	# NAME	K	VP	FUG	GAM	
3060.	1 ETHANOL	0.79784	465.14	1.0065	1.2952	
3061.	2 WATER	1.5070	199.92	0.98483	5.8172	
3062.	3 BENZENE	1.1508	488.82	1.0091	1.7730	
3063.	STAGE #	36				
3064.	TEMP, DEG K	339.240				
3065.	PRES, MMHG	759.9980				
3066.	# NAME	K	VP	FUG	GAM	
3067.	1 ETHANOL	0.78091	458.50	1.0064	1.2862	
3068.	2 WATER	1.4412	196.97	0.98689	5.6347	
3069.	3 BENZENE	1.1522	483.31	1.0095	1.7948	
3070.	STAGE #	37				
3071.	TEMP, DEG K	338.910				
3072.	PRES, MMHG	759.9968				
3073.	# NAME	K	VP	FUG	GAM	
3074.	1 ETHANOL	0.76522	452.04	1.0062	1.2786	
3075.	2 WATER	1.3773	194.10	0.98887	5.4534	
3076.	3 BENZENE	1.1539	477.93	1.0099	1.8170	
3077.	STAGE #	38				
3078.	TEMP, DEG K	338.902				
3079.	PRES, MMHG	759.9956				
3080.	# NAME	K	VP	FUG	GAM	
3081.	1 ETHANOL	1.0698	451.89	1.0055	1.7892	
3082.	2 WATER	2.5022	194.04	0.98829	9.9165	
3083.	3 BENZENE	0.85310	477.81	1.0104	1.3429	
3084.	STAGE #	39				
3085.	TEMP, DEG K	338.917				
3086.	PRES, MMHG	759.9944				
3087.	# NAME	K	VP	FUG	GAM	
3088.	1 ETHANOL	1.1203	452.18	1.0055	1.8728	
3089.	2 WATER	2.6984	194.16	0.98824	10.688	
3090.	3 BENZENE	0.83129	478.04	1.0105	1.3079	
3091.	STAGE #	40				
3092.	TEMP, DEG K	338.912				
3093.	PRES, MMHG	759.9932				
3094.	# NAME	K	VP	FUG	GAM	
3095.	1 ETHANOL	1.1252	452.08	1.0054	1.8814	
3096.	2 WATER	2.7175	194.12	0.98828	10.765	
3097.	3 BENZENE	0.82907	477.96	1.0105	1.3046	
3098.	STAGE #	41				
3099.	TEMP, DEG K	338.894				
3100.	PRES, MMHG	759.9919				
3101.	# NAME	K	VP	FUG	GAM	
3102.	1 ETHANOL	1.1244	451.74	1.0054	1.8814	
3103.	2 WATER	2.7136	193.97	0.98839	10.757	
3104.	3 BENZENE	0.82864	477.68	1.0105	1.3047	
3105.	STAGE #	42				
3106.	TEMP, DEG K	338.894				
3107.	PRES, MMHG	759.9907				
3108.	# NAME	K	VP	FUG	GAM	
3109.	1 ETHANOL	1.1244	451.74	1.0054	1.8814	
3110.	2 WATER	2.7136	193.97	0.98839	10.757	
3111.	3 BENZENE	0.82865	477.68	1.0105	1.3047	

C.2 Inputs

```

0.1 //BAYWAY JOB (JA00008-SAI,NJIT,-30,10),'ROCHE-TSAI',
0.2 // MSGLEVEL=(1,1),TIME=(2,30)
0.3 /*PRIORITY -2

?
10025. //STEP5 EXEC FGM=BAYWAY2,REGION=230K
10026. //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
10027. //FT06F001 DD SYSOUT=A
10028. //FT05F001 DD *
10029. CHEMSHARE'S ETOH DEHYDRATIO -VIR VAP-NRTL LIQ--REBOILED ABSOB
10030. 3 1 2 2 1 0 1
10031. ETHANOL 0 46.07 516.2 63.0 167.0 64.0
10032. 0.0878 0.0572 0.248 0.635
10033. 202.06 351.1 0.58 0.38
10034. 18.9119 -3803.98 -41.68
10035. WATER 0 18.01 647.3 217.3 56.0 18.84
10036. 0.0279 0.0229 0.229 0.344
10037. 538.7 373.2 1.0 0.454
10038. 18.3036 -3816.44 -46.13
10039. BENZENE 0 78.114 562.1 48.3 259.0 92.26
10040. 0.0 0.0 0.544 0.251
10041. 94.118 353.3 0.41 0.249
10042. 15.9008 -2788.51 -52.36
10043. 0.0 0.20 0.15
10044. 0.20 0.0 0.40
10045. 0.15 0.40 0.0
10046. -86.763 0.0 1289.908 0.0 282.332 0.0 1096.899
10047. 3820.954 0.0 2213.355 0.0
10048. 0.270 0.295 0.267
10049. 5 041 2 0 020 2 0
10050. 1.0 0.0 0.0
10051. 727.00 0.0 6.015 338.0 352.0 760.0 0.001
10052. 37 310.97 0.0 191.9 66.6 0.0
10053. 41 308.2 0.0 169.2 22.65 400.0
10054. /*
?

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42.  *** STAGewise SIMULATION PROGRAM ***
43.
44.  CHEMSHARE'S ETOH DEHYDRATIO -VIR VAP-NRTL LIQ--REBOILED ABSOR
45.  NUMBER OF COMPONENTS = 3
46.  VAPOR-LIQUID EQUILIBRIUM DATA CODES:
47.      LIQUID = 1
48.      VAPOR = 2
49.      ENTHALPY = 2
50.
51.  NOTE: ALL FLOWS ARE PER UNIT TIME, AND ENERGY
52.        VALUES ARE IN UNITS OF K-BTU PER UNIT TIME.
53.  COMPONENT NO          1          2          3
54.  NAME                  ETHANOL   WATER    BENZENE
55.  MOLECULAR WEIGHT      46.1     18.0     78.1
56.  CRITICAL PROPERTIES
57.  TEMPERATURE (DEG K)   516.2    647.3    562.1
58.  PRESSURE ( ATM)       63.0     217.3    48.3
59.  VOLUME (CC/GMOL)      167.0    56.0     259.0
60.  COMPRESSIBILITY       0.248    0.229    0.544
61.  ACENTRIC FACTOR       0.635    0.344    0.251
62.  TSONOPOULOS -A-      0.0878   0.0279   0.0
63.  TSONOPOULOS -B-      0.0572   0.0229   0.0
64.  AK(1,1)= 0.0
65.  AK(1,2)= 0.200
66.  AK(1,3)= 0.150
67.  AK(2,1)= 0.200
68.  AK(2,2)= 0.0
69.  AK(2,3)= 0.400
70.  AK(3,1)= 0.150
71.  AK(3,2)= 0.400
72.  AK(3,3)= 0.0
73.  VAPOR PRESSURE CONSTANTS
74.  ( MMHG AND DEG K)
75.  C(1)                  18.912    18.304    15.901
76.  C(2)                  -3803.98  -3816.44  -2788.51
77.  C(3)                  -41.68    -46.13    -52.36
78.  C(4)                   0.0        0.0        0.0
79.  C(5)                   0.0        0.0        0.0
80.  C(6)                   0.0        0.0        0.0
81.  LIQUID MOLAR VOLUME CONSTANTS
82.  C(1)                  64.000    18.840    92.260
83.  ENTHALPY CONSTANTS (CAL/GM)
84.  LATENT HEAT           202.1     538.7     94.1
85.  AT DEG K              351.1     373.2     353.3
86.  LIQUID SP. HEAT CONSTANTS
87.  C(1)                  0.580     1.000     0.410
88.  C(2)                   0.0        0.0        0.0
89.  C(3)                   0.0        0.0        0.0
90.  C(4)                   0.0        0.0        0.0
91.  VAPOR SP. HEAT CONSTANTS
92.  C(1)                  0.380     0.454     0.249
93.  C(2)                   0.0        0.0        0.0
94.  C(3)                   0.0        0.0        0.0
95.  C(4)                   0.0        0.0        0.0
96.  NRTL CONSTANTS, CAL/GMOL
97.  UNIQUAC PARAMETERS, CAL/G MOL

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NRTL CONSTANTS, CAL/GMOL

A 12 = -86.76 B 12 = 0.0 A 21 = 1289.91 B 21 = 0.0
 A 13 = 282.33 B 13 = 0.0 A 31 = 1096.90 B 31 = 0.0
 A 23 = 3820.95 B 23 = 0.0 A 32 = 2213.35 B 32 = 0.0

ALPHA12 = 0.2700 ALPHA13 = 0.2950 ALPHA23 = 0.2670 ALPHA
 CHEMSHARE'S ETH DEHYDRATIO -VIR VAP-NRTL LIQ--REBOILED ABSOR

REBOILED ABSORBER
 FEED STREAMS:

STAGE 37. 41.
 TEMP 310.97 308.20
 FRACV 0.0 0.0
 H -2.8042E 05 -7.9823E 05
 COMP 1 ETHANOL 361.100
 COMP 2 WATER 89.250
 COMP 3 BENZENE 400.000
 TOTAL 258.500 591.850
 NUMBER OF STAGES 41

ESTIMATED OVERHEAD PRODUCT RATE 727.0000
 ESTIMATED BOTTOMS PRODUCT RATE 123.3496

PRESSURE AT TOP OF COLUMN 759.99097
 CONSTANT DELTA P PER STAGE 0.00100
 BOILUP RATIO 6.014999

121.	INITIAL COLUMN PROFILES:				
122.	STAGE	TEMP	PRES	LIQUID(MOLS)	VAPOR(MOLS)
123.	1	352.00	760.040	123.350	727.251
124.	2	351.65	760.039	850.601	727.251
125.	3	351.30	760.037	850.601	727.251
126.	4	350.95	760.036	850.601	727.251
127.	5	350.60	760.035	850.601	727.251
128.	6	350.25	760.034	850.601	727.251
129.	7	349.90	760.032	850.601	727.251
130.	8	349.55	760.031	850.601	727.251
131.	9	349.20	760.030	850.601	727.251
132.	10	348.85	760.029	850.601	727.251
133.	11	348.50	760.028	850.601	727.251
134.	12	348.15	760.026	850.601	727.251
135.	13	347.80	760.025	850.601	727.251
136.	14	347.45	760.024	850.601	727.251
137.	15	347.10	760.023	850.601	727.251
138.	16	346.75	760.021	850.601	727.251
139.	17	346.40	760.020	850.601	727.251
140.	18	346.05	760.019	850.601	727.251
141.	19	345.70	760.018	850.601	727.251
142.	20	345.35	760.017	850.601	727.251
143.	21	345.00	760.015	850.601	727.251
144.	22	344.65	760.014	850.601	727.251
145.	23	344.30	760.013	850.601	727.251
146.	24	343.95	760.012	850.601	727.251
147.	25	343.60	760.010	850.601	727.251
148.	26	343.25	760.009	850.601	727.251
149.	27	342.90	760.008	850.601	727.251
150.	28	342.55	760.007	850.601	727.251
151.	29	342.20	760.006	850.601	727.251
152.	30	341.85	760.004	850.601	727.251
153.	31	341.50	760.003	850.601	727.251
154.	32	341.15	760.002	850.601	727.251
155.	33	340.80	760.001	850.601	727.251
156.	34	340.45	760.000	850.601	727.251
157.	35	340.10	759.998	850.601	727.251
158.	36	339.75	759.997	850.601	727.251
159.	37	339.40	759.996	850.601	727.251
160.	38	339.05	759.995	592.101	727.251
161.	39	338.70	759.993	592.101	727.251
162.	40	338.35	759.992	592.101	727.251
163.	41	338.00	759.991	592.101	727.000

1969.	FINAL COLUMN PROFILES:					
1970.	STAGE	TEMP	PRES	EFF(%)	LIQUID(MOLS)	VAPOR(MOLS)
1971.	1	351.38	760.040	100.00	123.351	741.948
1972.	2	351.17	760.039	100.00	865.298	742.353
1973.	3	350.54	760.037	100.00	865.703	744.018
1974.	4	348.88	760.036	100.00	867.368	751.055
1975.	5	345.67	760.035	100.00	874.406	771.619
1976.	6	342.47	760.034	100.00	894.970	798.696
1977.	7	341.19	760.032	100.00	922.047	813.115
1978.	8	340.94	760.031	100.00	936.465	817.336
1979.	9	340.90	760.030	100.00	940.687	818.347
1980.	10	340.89	760.029	100.00	941.698	818.580
1981.	11	340.88	760.028	100.00	941.930	818.633
1982.	12	340.88	760.026	100.00	941.983	818.645
1983.	13	340.88	760.025	100.00	941.996	818.647
1984.	14	340.88	760.024	100.00	941.998	818.647
1985.	15	340.88	760.023	100.00	941.998	818.646
1986.	16	340.88	760.021	100.00	941.996	818.644
1987.	17	340.88	760.020	100.00	941.994	818.641
1988.	18	340.88	760.019	100.00	941.992	818.637
1989.	19	340.88	760.018	100.00	941.988	818.630
1990.	20	340.88	760.017	100.00	941.981	818.620
1991.	21	340.88	760.015	100.00	941.971	818.605
1992.	22	340.87	760.014	100.00	941.955	818.581
1993.	23	340.87	760.013	100.00	941.932	818.544
1994.	24	340.86	760.012	100.00	941.895	818.489
1995.	25	340.85	760.010	100.00	941.839	818.404
1996.	26	340.84	760.009	100.00	941.754	818.276
1997.	27	340.81	760.008	100.00	941.626	818.082
1998.	28	340.77	760.007	100.00	941.433	817.794
1999.	29	340.72	760.006	100.00	941.145	817.372
2000.	30	340.64	760.004	100.00	940.722	816.764
2001.	31	340.52	760.003	100.00	940.115	815.920
2002.	32	340.36	760.002	100.00	939.271	814.797
2003.	33	340.15	760.001	100.00	938.147	813.389
2004.	34	339.88	760.000	100.00	936.740	811.756
2005.	35	339.57	759.998	100.00	935.107	810.032
2006.	36	339.24	759.997	100.00	933.383	808.395
2007.	37	338.91	759.996	100.00	931.746	787.088
2008.	38	338.90	759.995	100.00	651.938	790.137
2009.	39	338.92	759.993	100.00	654.988	790.397
2010.	40	338.91	759.992	100.00	655.247	790.364
2011.	41	338.89	759.991	100.00	655.214	727.000

2012.	FINAL VAPOR COMPOSITION PROFILE (MOLE%):			
2013.	STAGE	ETHANOL	WATER	BENZENE
2014.	1	99.51968	0.00004	0.48029
2015.	2	98.43357	0.00004	1.56640
2016.	3	95.21686	0.00005	4.78313
2017.	4	86.91057	0.00005	13.08941
2018.	5	71.66200	0.00005	28.33799
2019.	6	56.58866	0.00006	43.41130
2020.	7	49.33349	0.00008	50.66644
2021.	8	47.19740	0.00013	52.80247
2022.	9	46.67717	0.00020	53.32265
2023.	10	46.55664	0.00031	53.44306
2024.	11	46.52892	0.00048	53.47060
2025.	12	46.52235	0.00075	53.47690
2026.	13	46.52052	0.00116	53.47834
2027.	14	46.51964	0.00179	53.47859
2028.	15	46.51862	0.00275	53.47863
2029.	16	46.51721	0.00423	53.47857
2030.	17	46.51511	0.00650	53.47841
2031.	18	46.51191	0.00998	53.47813
2032.	19	46.50694	0.01531	53.47775
2033.	20	46.49943	0.02350	53.47710
2034.	21	46.48779	0.03605	53.47618
2035.	22	46.47005	0.05527	53.47468
2036.	23	46.44288	0.08470	53.47244
2037.	24	46.40131	0.12970	53.46899
2038.	25	46.33787	0.19835	53.46377
2039.	26	46.24136	0.30278	53.45585
2040.	27	46.09514	0.46090	53.44396
2041.	28	45.87480	0.69864	53.42656
2042.	29	45.54641	1.05236	53.40124
2043.	30	45.06384	1.57044	53.36573
2044.	31	44.36955	2.31208	53.31839
2045.	32	43.40048	3.33958	53.25994
2046.	33	42.10356	4.70018	53.19626
2047.	34	40.46110	6.39740	53.14152
2048.	35	38.51941	8.36302	53.11756
2049.	36	36.39984	10.45293	53.14722
2050.	37	34.27486	12.48391	53.24123
2051.	38	33.00119	12.15590	54.84293
2052.	39	32.85823	12.13803	55.00374
2053.	40	32.81119	12.17429	55.01451
2054.	41	32.72498	12.27647	54.99855

2055.	FINAL LIQUID COMPOSITION PROFILE (MOLE%):			
2056.	STAGE	ETHANOL	WATER	BENZENE
2057.	1	99.86953	0.00003	0.13045
2058.	2	99.56955	0.00004	0.43042
2059.	3	98.63818	0.00004	1.36180
2060.	4	95.87853	0.00005	4.12146
2061.	5	88.73867	0.00005	11.26131
2062.	6	75.54976	0.00005	24.45023
2063.	7	62.37875	0.00006	37.62119
2064.	8	55.99010	0.00008	44.00985
2065.	9	54.10424	0.00011	45.89565
2066.	10	53.64473	0.00018	46.35510
2067.	11	53.53826	0.00027	46.46147
2068.	12	53.51378	0.00043	46.48580
2069.	13	53.50800	0.00066	46.49135
2070.	14	53.50637	0.00101	46.49261
2071.	15	53.50560	0.00156	46.49284
2072.	16	53.50474	0.00240	46.49287
2073.	17	53.50353	0.00368	46.49282
2074.	18	53.50171	0.00565	46.49264
2075.	19	53.49897	0.00867	46.49237
2076.	20	53.49470	0.01331	46.49200
2077.	21	53.48823	0.02043	46.49134
2078.	22	53.47825	0.03133	46.49044
2079.	23	53.46300	0.04804	46.48896
2080.	24	53.43965	0.07361	46.48673
2081.	25	53.40396	0.11271	46.48333
2082.	26	53.34946	0.17237	46.47816
2083.	27	53.26656	0.26312	46.47033
2084.	28	53.14092	0.40052	46.45857
2085.	29	52.95162	0.60708	46.44130
2086.	30	52.66948	0.91437	46.41616
2087.	31	52.25481	1.36439	46.38081
2088.	32	51.65817	2.00845	46.33338
2089.	33	50.82526	2.90048	46.27426
2090.	34	49.71026	4.08126	46.20848
2091.	35	48.29775	5.55352	46.14875
2092.	36	46.62713	7.25781	46.11506
2093.	37	44.80242	9.06910	46.12848
2094.	38	30.84084	4.85619	64.30297
2095.	39	29.32033	4.49602	66.18366
2096.	40	29.14935	4.47749	66.37317
2097.	41	29.09241	4.52085	66.38676

2098.	FINAL VAPOR COMPOSITION PROFILE (MOLES):			
2099.	STAGE	ETHANOL	WATER	BENZENE
2100.	1	738.384	0.000	3.564
2101.	2	730.724	0.000	11.628
2102.	3	708.430	0.000	35.587
2103.	4	652.747	0.000	98.309
2104.	5	552.958	0.000	218.661
2105.	6	451.971	0.000	346.724
2106.	7	401.138	0.001	411.976
2107.	8	385.761	0.001	431.574
2108.	9	381.981	0.002	436.364
2109.	10	381.103	0.003	437.474
2110.	11	380.901	0.004	437.728
2111.	12	380.853	0.006	437.786
2112.	13	380.839	0.009	437.799
2113.	14	380.832	0.015	437.801
2114.	15	380.823	0.023	437.801
2115.	16	380.810	0.035	437.799
2116.	17	380.792	0.053	437.796
2117.	18	380.764	0.082	437.792
2118.	19	380.720	0.125	437.785
2119.	20	380.654	0.192	437.774
2120.	21	380.551	0.295	437.759
2121.	22	380.395	0.452	437.734
2122.	23	380.156	0.693	437.696
2123.	24	379.790	1.062	437.638
2124.	25	379.231	1.623	437.550
2125.	26	378.382	2.478	437.416
2126.	27	377.096	3.771	437.216
2127.	28	375.162	5.713	436.919
2128.	29	372.283	8.602	436.487
2129.	30	368.065	12.827	435.872
2130.	31	362.020	18.865	435.035
2131.	32	353.626	27.211	433.960
2132.	33	342.466	38.231	432.693
2133.	34	328.446	51.931	431.380
2134.	35	312.020	67.743	430.270
2135.	36	294.255	84.501	429.640
2136.	37	269.773	98.259	419.055
2137.	38	260.755	96.048	433.334
2138.	39	259.710	95.939	434.748
2139.	40	259.328	96.221	434.815
2140.	41	237.911	89.250	399.839

?

2141.	FINAL LIQUID COMPOSITION PROFILE (MOLES):			
2142.	STAGE	ETHANOL	WATER	BENZENE
2143.	1	123.190	0.000	0.161
2144.	2	861.574	0.000	3.724
2145.	3	853.914	0.000	11.789
2146.	4	831.620	0.000	35.748
2147.	5	775.937	0.000	98.470
2148.	6	676.147	0.000	218.822
2149.	7	575.161	0.001	346.885
2150.	8	524.328	0.001	412.137
2151.	9	508.951	0.001	431.734
2152.	10	505.171	0.002	436.525
2153.	11	504.293	0.003	437.635
2154.	12	504.091	0.004	437.888
2155.	13	504.043	0.006	437.947
2156.	14	504.029	0.010	437.959
2157.	15	504.021	0.015	437.961
2158.	16	504.013	0.023	437.961
2159.	17	504.000	0.035	437.960
2160.	18	503.982	0.053	437.957
2161.	19	503.954	0.082	437.952
2162.	20	503.910	0.125	437.946
2163.	21	503.844	0.192	437.935
2164.	22	503.741	0.295	437.919
2165.	23	503.585	0.452	437.894
2166.	24	503.345	0.693	437.856
2167.	25	502.979	1.062	437.798
2168.	26	502.421	1.623	437.710
2169.	27	501.572	2.478	437.577
2170.	28	500.286	3.771	437.376
2171.	29	498.352	5.714	437.080
2172.	30	495.474	8.602	436.647
2173.	31	491.255	12.827	436.033
2174.	32	485.210	18.865	435.196
2175.	33	476.816	27.211	434.121
2176.	34	465.656	38.231	432.853
2177.	35	451.636	51.931	431.540
2178.	36	435.210	67.743	430.430
2179.	37	417.445	84.501	429.800
2180.	38	201.063	31.659	419.216
2181.	39	192.045	29.448	433.495
2182.	40	191.000	29.339	434.908
?				
2183.	41	190.618	29.621	434.976
?				

2184.	STAGE #	1				
2185.	TEMP, DEG K	351.378				
2186.	PRES, MMHG	760.0398				
2187.	# NAME	K	VP	FUG	GAM	
2188.	1 ETHANOL	0.99649	756.75	1.0008	1.0000	
2189.	2 WATER	1.1510	330.58	0.97408	2.7166	
2190.	3 BENZENE	3.6817	717.12	0.97205	4.0142	
2191.	STAGE #	2				
2192.	TEMP, DEG K	351.169				
2193.	PRES, MMHG	760.0386				
2194.	# NAME	K	VP	FUG	GAM	
2195.	1 ETHANOL	0.98859	750.50	1.0011	1.0000	
2196.	2 WATER	1.1464	327.76	0.97428	2.7286	
2197.	3 BENZENE	3.6392	712.46	0.97286	3.9906	
2198.	STAGE #	3				
2199.	TEMP, DEG K	350.544				
2200.	PRES, MMHG	760.0374				
2201.	# NAME	K	VP	FUG	GAM	
2202.	1 ETHANOL	0.96531	732.05	1.0020	1.0002	
2203.	2 WATER	1.1335	319.44	0.97485	2.7665	
2204.	3 BENZENE	3.5124	698.65	0.97522	3.9180	
2205.	STAGE #	4				
2206.	TEMP, DEG K	348.883				
2207.	PRES, MMHG	760.0361				
2208.	# NAME	K	VP	FUG	GAM	
2209.	1 ETHANOL	0.90646	684.89	1.0040	1.0019	
2210.	2 WATER	1.1041	298.21	0.97608	2.8829	
2211.	3 BENZENE	3.1759	662.99	0.98124	3.7104	
2212.	STAGE #	5				
2213.	TEMP, DEG K	345.675				
2214.	PRES, MMHG	760.0349				
2215.	# NAME	K	VP	FUG	GAM	
2216.	1 ETHANOL	0.80756	600.99	1.0067	1.0144	
2217.	2 WATER	1.0785	260.55	0.97737	3.2189	
2218.	3 BENZENE	2.5164	598.19	0.99191	3.2233	
2219.	STAGE #	6				
2220.	TEMP, DEG K	342.471				
2221.	PRES, MMHG	760.0337				
2222.	# NAME	K	VP	FUG	GAM	
2223.	1 ETHANOL	0.74902	525.99	1.0076	1.0742	
2224.	2 WATER	1.1759	227.02	0.97728	4.0284	
2225.	3 BENZENE	1.7755	538.57	1.0017	2.5013	
2226.	STAGE #	7				
2227.	TEMP, DEG K	341.194				
2228.	PRES, MMHG	760.0325				
2229.	# NAME	K	VP	FUG	GAM	
2230.	1 ETHANOL	0.79087	498.40	1.0069	1.1978	
2231.	2 WATER	1.4488	214.72	0.97667	5.2509	
2232.	3 BENZENE	1.3467	516.17	1.0059	1.9714	
2233.	STAGE #	8				
2234.	TEMP, DEG K	340.941				
2235.	PRES, MMHG	760.0313				
2236.	# NAME	K	VP	FUG	GAM	
2237.	1 ETHANOL	0.84295	493.07	1.0064	1.2911	
2238.	2 WATER	1.6631	212.34	0.97641	6.0964	
2239.	3 BENZENE	1.1998	511.82	1.0069	1.7694	
2240.	STAGE #	9				

2422.	STAGE #	35				
2423.	TEMP, DEG K	339.571				
2424.	PRES, MMHG	759.9983				
2425.	# NAME	K	VP	FUG	GAM	
2426.	1 ETHANOL	0.79754	465.07	1.0065	1.2949	
2427.	2 WATER	1.5059	199.89	0.98485	5.8136	
2428.	3 BENZENE	1.1510	488.76	1.0091	1.7736	
2429.	STAGE #	36				
2430.	TEMP, DEG K	339.237				
2431.	PRES, MMHG	759.9971				
2432.	# NAME	K	VP	FUG	GAM	
2433.	1 ETHANOL	0.78066	458.45	1.0064	1.2860	
2434.	2 WATER	1.4402	196.95	0.98691	5.6314	
2435.	3 BENZENE	1.1525	483.26	1.0095	1.7954	
2436.	STAGE #	37				
2437.	TEMP, DEG K	338.908				
2438.	PRES, MMHG	759.9958				
2439.	# NAME	K	VP	FUG	GAM	
2440.	1 ETHANOL	0.76502	452.01	1.0062	1.2784	
2441.	2 WATER	1.3765	194.09	0.98888	5.4507	
2442.	3 BENZENE	1.1542	477.90	1.0099	1.8175	
2443.	STAGE #	38				
2444.	TEMP, DEG K	338.900				
2445.	PRES, MMHG	759.9946				
2446.	# NAME	K	VP	FUG	GAM	
2447.	1 ETHANOL	1.0700	451.85	1.0055	1.7898	
2448.	2 WATER	2.5032	194.02	0.98830	9.9211	
2449.	3 BENZENE	0.85288	477.77	1.0104	1.3427	
2450.	STAGE #	39				
2451.	TEMP, DEG K	338.915				
2452.	PRES, MMHG	759.9934				
2453.	# NAME	K	VP	FUG	GAM	
2454.	1 ETHANOL	1.1207	452.14	1.0055	1.8735	
2455.	2 WATER	2.6997	194.15	0.98825	10.694	
2456.	3 BENZENE	0.83107	478.02	1.0105	1.3076	
2457.	STAGE #	40				
2458.	TEMP, DEG K	338.911				
2459.	PRES, MMHG	759.9922				
2460.	# NAME	K	VP	FUG	GAM	
2461.	1 ETHANOL	1.1256	452.06	1.0054	1.8821	
2462.	2 WATER	2.7190	194.11	0.98828	10.772	
2463.	3 BENZENE	0.82886	477.95	1.0105	1.3043	
2464.	STAGE #	41				
2465.	TEMP, DEG K	338.894				
2466.	PRES, MMHG	759.9910				
2467.	# NAME	K	VP	FUG	GAM	
2468.	1 ETHANOL	1.1249	451.74	1.0054	1.8822	
2469.	2 WATER	2.7155	193.97	0.98839	10.765	
2470.	3 BENZENE	0.82846	477.68	1.0105	1.3044	

C.3 Inputs

```

0.1 //CHEMSHRE JOB (JA00008-SAI,NJIT,-30,10),'ROCHE-TSAI',
0.2 // MSGLEVEL=(1,1),TIME=(3,00)
0.3 /*PRIORITY +2
11145. //CHEMS EXEC PGM=BAYWAY,REGION=200K
11150. //STEPLIB DD DSN='NCE.AJAO0.S008.TEST',UNIT=OLS,DISP=SHR
11151. //FT06F001 DD SYSOUT=A
11152. //FT05F001 DD *
11153. CHEMSHARE'S ETOH DEHYDRATION-VIR VAPOR-UNIQUAC-WITH PRESS DROP
11154. 3 3 2 2 0 0 1
11155. ETHANOL 0 46.07 516.2 63.0 167.0 64.0
11155.1 0.0878 0.0572 0.248 0.635
11155.2 2.1055 1.9720
11157. 202.06 351.1 0.58 0.38
11158. 18.9119 -3803.98 -41.68
11159. WATER 0 18.01 647.3 217.3 56.0 18.84
11160. 0.0279 0.0229 0.229 0.344
11160.1 0.9200 1.4000
11161. 538.7 373.2 1.0 0.454
11162. 18.3036 -3816.44 -46.13
11163. BENZENE 0 78.114 562.1 48.3 259.0 92.26
11164. 0.0 0.0 0.544 0.251
11164.01 3.1878 2.4000
11164.1 94.118 353.3 0.41 0.249
11165. 15.9008 -2788.51 -52.36
11165.1 0.0 0.20 0.15
11165.2 0.20 0.0 0.40
11165.3 0.15 0.40 0.0
11166. 110.010 0.0 200.054 0.0 -156.567 0.0 850.609
11167. 428.452 0.0 1072.790 0.0
11168. 3 041 2 0 050 2 0
11169. 1.0 0.0 0.0
11170. 726.50 0.0 0.00001 338.0 352.0 760.0 45.0
11171. 37 310.97 0.0 191.9 66.6 0.0
11172. 40 308.2 0.0 169.2 22.65 400.0
11175. /*
?
```

C.3 Results

30. CHEMSHARE'S ETOH DEHYDRATION-VIR VAPOR-UNIQVAC-WITH PRESS DROP
 31. NUMBER OF COMPONENTS = 3
 32. VAPOR-LIQUID EQUILIBRIUM DATA CODES:
 33. LIQUID = 3
 34. VAPOR = 2
 35. ENTHALPY = 2
 36.

37. NOTE: ALL FLOWS ARE PER UNIT TIME, AND ENERGY
 38. VALUES ARE IN UNITS OF K-BTU PER UNIT TIME.

39. COMPONENT NO	1	2	3
40. NAME	ETHANOL	WATER	BENZENE
41. MOLECULAR WEIGHT	46.1	18.0	78.1
42. CRITICAL PROPERTIES			
43. TEMPERATURE (DEG K)	516.2	647.3	562.1
44. PRESSURE (ATM)	63.0	217.3	48.3
45. VOLUME (CC/GMOL)	167.0	56.0	259.0
46. UNIQVAC COMPONT R(I)	2.1055	0.9200	3.1878
47. UNIQVAC COMPONT Q(I)	1.9720	1.4000	2.4000
48. COMPRESSIBILITY	0.248	0.229	0.544
49. ACENTRIC FACTOR	0.635	0.344	0.251
50. TSONOPOULOS -A-	0.0878	0.0279	0.0
51. TSONOPOULOS -B-	0.0572	0.0229	0.0
52. AK(1,1)=	0.0		
53. AK(1,2)=	0.200		
54. AK(1,3)=	0.150		
55. AK(2,1)=	0.200		
56. AK(2,2)=	0.0		
57. AK(2,3)=	0.400		
58. AK(3,1)=	0.150		
59. AK(3,2)=	0.400		
60. AK(3,3)=	0.0		
61. VAPOR PRESSURE CONSTANTS			
62. (MMHG AND DEG K)			
63. C(1)	18.912	18.304	15.901
64. C(2)	-3803.98	-3816.44	-2788.51
65. C(3)	-41.68	-46.13	-52.36
66. C(4)	0.0	0.0	0.0
67. C(5)	0.0	0.0	0.0
68. C(6)	0.0	0.0	0.0
69. LIQUID MOLAR VOLUME CONSTANTS			
70. C(1)	64.000	18.840	92.260
71. ENTHALPY CONSTANTS (CAL/GM)			
72. LATENT HEAT	202.1	538.7	94.1
73. AT DEG K	351.1	373.2	353.3
74. LIQUID SP. HEAT CONSTANTS			
75. C(1)	0.580	1.000	0.410
76. C(2)	0.0	0.0	0.0
77. C(3)	0.0	0.0	0.0
78. C(4)	0.0	0.0	0.0
79. VAPOR SP. HEAT CONSTANTS			
80. C(1)	0.380	0.454	0.249
81. C(2)	0.0	0.0	0.0
82. C(3)	0.0	0.0	0.0
83. C(4)	0.0	0.0	0.0

UNIQUAC PARAMETERS,CAL/G MOL
 A 12 = 110.01 B 12 = 0.0 21 = 200.05
 A 13 = -156.57 B 13 = 0.0 31 = 850.61
 A 23 = 428.45 B 23 = 0.0 32 = 1072.79

CHEMSHARE'S ETOH DEHYDRATION-VIR VAPOR-UNIQUAC-WITH PRESS DROP
 DISTILLATION COLUMN; PRODUCT RATES & REFLUX RATIO SPECIFIED

FEED STREAMS:

STAGE	37.	40.
TEMP	310.97	308.20
FRACV	0.0	0.0
H	-2.8042E 05	-7.9823E 05
COMP 1 ETHANOL	191.900	169.200
COMP 2 WATER	66.600	22.650
COMP 3 BENZENE	0.0	400.000
TOTAL	258.500	591.850
NUMBER OF STAGES	41	

ESTIMATED OVERHEAD PRODUCT RATE 726.5000
 ESTIMATED BOTTOMS PRODUCT RATE 123.8496

PRESSURE AT TOP OF COLUMN 760.00000
 CONSTANT DELTA P PER STAGE 45.00000
 REFLUX RATIO 0.000010

107.	ITERATION # = 1					
108.	SUM DT**2 = 0.0					
109.	SUM (DF/F)**2 = 25832.5					
110.	41 0.0	1	2	157.29	3	2 6.2428
111.	ITERATION # = 2					
112.	SUM DT**2 = 5807.61					
113.	SUM (DF/F)**2 = 15502.5					
114.	1 24.935	1	2	124.05	1	2 2.4719
115.	ITERATION # = 3					
116.	SUM DT**2 = 3262.81					
117.	SUM (DF/F)**2 = 0.161315E 07					
118.	1 22.074	1	2	1269.6	1	2 19.646
119.	ITERATION # = 4					
120.	SUM DT**2 = 987.885					
121.	SUM (DF/F)**2 = 55108.4					
122.	1 20.112	4	2	105.67	1	2 99.790
123.	ITERATION # = 5					
124.	SUM DT**2 = 159.510					
125.	SUM (DF/F)**2 = 30528.0					
126.	27 3.5894	30	2	71.750	27	2 47.668
127.	ITERATION # = 6					
128.	SUM DT**2 = 49.8170					
129.	SUM (DF/F)**2 = 17994.6					
130.	4 3.5137	16	2	49.866	16	2 65.588
131.	ITERATION # = 7					
132.	SUM DT**2 = 21.5095					
133.	SUM (DF/F)**2 = 50.9166					
134.	6 3.2831	6	2	3.8350	5	2 3.6582
135.	ITERATION # = 8					
136.	SUM DT**2 = 8.64310					
137.	SUM (DF/F)**2 = 1.74704					
138.	7 2.1977	8	3	0.63446	7	3 0.63476
139.	ITERATION # = 9					
140.	SUM DT**2 = 0.721670					
141.	SUM (DF/F)**2 = 0.383326E-01					
142.	8 0.63337	9	3	0.96522E-01	8	3 0.96553E-0
143.	ITERATION # = 10					
144.	SUM DT**2 = 0.182898E-02					
145.	SUM (DF/F)**2 = 0.165059E-02					
146.	9 0.29642E-01	1	3	0.11050E-01	1	3 0.11039E-0
147.	ITERATION # = 11					
148.	SUM DT**2 = 0.112582E-02					
149.	SUM (DF/F)**2 = 0.104351E-02					
150.	8 0.20203E-01	1	3	0.87315E-02	1	3 0.87170E-0
151.	ITERATION # = 12					
152.	SUM DT**2 = 0.761609E-05					
153.	SUM (DF/F)**2 = 0.528009E-07					
154.	8 0.15828E-02	4	3	0.66723E-04	3	3 0.66613E-0
155.	REBOILER DUTY = 7.042080E 06					

155.	REBOILER DUTY	=	7.042080E 06	
156.	CONDENSER DUTY	=	2.001900E 04	
157.	PRODUCT STREAMS:			
158.	STAGE		1.	41.
159.	TEMP		385.38	339.09
160.	FRACV		0.0	1.0000
161.	H		1.1345E 05	5.8300E 06
162.	COMP 1 ETHANOL		123.788	237.312
163.	COMP 2 WATER		0.001	89.249
164.	COMP 3 BENZENE		0.061	399.939
165.	TOTAL		123.850	726.501

	FINAL COLUMN PROFILES:					
	STAGE	TEMP	PRES	EFF(%)	LIQUID(MOLS)	VAPOR(MOLS)
166.						
167.						
168.	1	385.38	2560.000	100.00	123.850	781.710
169.	2	384.79	2515.000	100.00	905.560	781.197
170.	3	384.15	2470.000	100.00	905.046	780.870
171.	4	383.38	2425.000	100.00	904.720	781.001
172.	5	382.37	2380.000	100.00	904.851	782.253
173.	6	380.90	2335.000	100.00	906.103	786.189
174.	7	378.76	2290.000	100.00	910.039	795.485
175.	8	376.13	2245.000	100.00	919.335	811.159
176.	9	373.79	2200.000	100.00	935.009	827.830
177.	10	372.25	2155.000	100.00	951.679	838.814
178.	11	371.25	2110.000	100.00	962.664	843.774
179.	12	370.46	2065.000	100.00	967.624	845.399
180.	13	369.72	2020.000	100.00	969.248	845.593
181.	14	369.00	1975.000	100.00	969.442	845.226
182.	15	368.26	1930.000	100.00	969.076	844.644
183.	16	367.52	1885.000	100.00	968.493	843.973
184.	17	366.76	1840.000	100.00	967.822	843.256
185.	18	365.99	1795.000	100.00	967.105	842.513
186.	19	365.20	1750.000	100.00	966.362	841.745
187.	20	364.39	1705.000	100.00	965.594	840.953
188.	21	363.56	1660.000	100.00	964.802	840.132
189.	22	362.71	1615.000	100.00	963.981	839.278
190.	23	361.84	1570.000	100.00	963.127	838.387
191.	24	360.95	1525.000	100.00	962.236	837.447
192.	25	360.03	1480.000	100.00	961.296	836.446
193.	26	359.08	1435.000	100.00	960.296	835.367
194.	27	358.09	1390.000	100.00	959.217	834.185
195.	28	357.07	1345.000	100.00	958.034	832.870
196.	29	355.99	1300.000	100.00	956.719	831.382
197.	30	354.86	1255.000	100.00	955.232	829.673
198.	31	353.67	1210.000	100.00	953.522	827.693
199.	32	352.40	1165.000	100.00	951.542	825.399
200.	33	351.05	1120.000	100.00	949.249	822.777
201.	34	349.61	1075.000	100.00	946.626	819.863
202.	35	348.08	1030.000	100.00	943.712	816.752
203.	36	346.48	985.000	100.00	940.602	813.591
204.	37	344.83	940.000	100.00	937.440	787.305
205.	38	343.46	895.000	100.00	652.655	791.227
206.	39	342.09	850.000	100.00	656.576	791.430
207.	40	340.66	805.000	100.00	656.779	726.508
208.	41	339.09	760.000	100.00	0.007	726.501

209.	FINAL VAPOR COMPOSITION PROFILE (MOLE%):			
210.	STAGE	ETHANOL	WATER	BENZENE
211.	1	99.86929	0.00062	0.13010
212.	2	99.68407	0.00070	0.31525
213.	3	99.26102	0.00077	0.73825
214.	4	98.30552	0.00084	1.69368
215.	5	96.21408	0.00091	3.78504
216.	6	91.96247	0.00097	8.03659
217.	7	84.55057	0.00103	15.44843
218.	8	74.68391	0.00113	25.31500
219.	9	65.60950	0.00132	34.38920
220.	10	59.84085	0.00170	40.15745
221.	11	56.99171	0.00233	43.00597
222.	12	55.70491	0.00329	44.29179
223.	13	55.08259	0.00470	44.91273
224.	14	54.71482	0.00672	45.27847
225.	15	54.43802	0.00959	45.55241
226.	16	54.19054	0.01367	45.79580
227.	17	53.94911	0.01949	46.03140
228.	18	53.70462	0.02779	46.26762
229.	19	53.45297	0.03964	46.50741
230.	20	53.19155	0.05659	46.75186
231.	21	52.91756	0.08089	47.00156
232.	22	52.62752	0.11573	47.25676
233.	23	52.31689	0.16576	47.51735
234.	24	51.97922	0.23763	47.78315
235.	25	51.60530	0.34084	48.05387
236.	26	51.18252	0.48887	48.32862
237.	27	50.69342	0.70061	48.60598
238.	28	50.11356	1.00200	48.88446
239.	29	49.41057	1.42763	49.16182
240.	30	48.54383	2.02144	49.43473
241.	31	47.46495	2.83509	49.69996
242.	32	46.12267	3.92155	49.95578
243.	33	44.47471	5.32150	50.20381
244.	34	42.50569	7.04216	50.45215
245.	35	40.24881	9.03501	50.71620
246.	36	37.79687	11.18638	51.01677
247.	37	35.29266	13.33579	51.37157
248.	38	33.24677	12.81634	53.93689
249.	39	32.79996	12.60751	54.59253
250.	40	32.66510	12.28474	55.05014
251.	41	32.66511	12.28481	55.05007

252.	FINAL LIQUID COMPOSITION PROFILE (MOLE%):			
253.	STAGE	ETHANOL	WATER	BENZENE
254.	1	99.95022	0.00055	0.04924
255.	2	99.88037	0.00061	0.11904
256.	3	99.72048	0.00068	0.27885
257.	4	99.35535	0.00074	0.64393
258.	5	98.53062	0.00080	1.46860
259.	6	96.72476	0.00086	3.27441
260.	7	93.04956	0.00091	6.94957
261.	8	86.62514	0.00097	13.37391
262.	9	78.03063	0.00105	21.96833
263.	10	70.07853	0.00122	29.92026
264.	11	65.00105	0.00155	34.99740
265.	12	62.49012	0.00210	37.50777
266.	13	61.35857	0.00294	38.63851
267.	14	60.81460	0.00417	39.18124
268.	15	60.49599	0.00593	39.49808
269.	16	60.25806	0.00843	39.73352
270.	17	60.04629	0.01199	39.94173
271.	18	59.84013	0.01706	40.14282
272.	19	59.63149	0.02429	40.34423
273.	20	59.41682	0.03462	40.54855
274.	21	59.19386	0.04940	40.75673
275.	22	58.96019	0.07056	40.96925
276.	23	58.71282	0.10092	41.18627
277.	24	58.44779	0.14450	41.40771
278.	25	58.15963	0.20709	41.63330
279.	26	57.84036	0.29695	41.86269
280.	27	57.47918	0.42582	42.09500
281.	28	57.06109	0.61011	42.32880
282.	29	56.56503	0.87236	42.56261
283.	30	55.96324	1.24260	42.79416
284.	31	55.22082	1.75895	43.02022
285.	32	54.29627	2.46615	43.23759
286.	33	53.14563	3.40997	43.44441
287.	34	51.73271	4.62535	43.64194
288.	35	50.04452	6.11805	43.83745
289.	36	48.10973	7.84544	44.04485
290.	37	46.00824	9.70857	44.28320
291.	38	32.13780	5.88275	61.97945
292.	39	29.69117	5.30129	65.00756
293.	40	29.15386	5.05197	65.79419
294.	41	31.60366	5.56455	62.83181

295.	FINAL VAPOR COMPOSITION PROFILE (MOLES):			
296.	STAGE	ETHANOL	WATER	BENZENE
297.	1	780.688	0.005	1.017
298.	2	778.729	0.005	2.463
299.	3	775.100	0.006	5.765
300.	4	767.768	0.007	13.228
301.	5	752.638	0.007	29.609
302.	6	722.999	0.008	63.183
303.	7	672.587	0.008	122.890
304.	8	605.805	0.009	205.345
305.	9	543.135	0.011	284.684
306.	10	501.954	0.014	336.846
307.	11	480.881	0.020	362.873
308.	12	470.929	0.028	374.442
309.	13	465.774	0.040	379.779
310.	14	462.464	0.057	382.706
311.	15	459.807	0.081	384.756
312.	16	457.354	0.115	386.504
313.	17	454.929	0.164	388.163
314.	18	452.468	0.234	389.811
315.	19	449.938	0.334	391.474
316.	20	447.316	0.476	393.161
317.	21	444.577	0.680	394.875
318.	22	441.691	0.971	396.616
319.	23	438.618	1.390	398.379
320.	24	435.298	1.990	400.158
321.	25	431.651	2.851	401.945
322.	26	427.562	4.084	403.721
323.	27	422.877	5.844	405.464
324.	28	417.381	8.345	407.144
325.	29	410.791	11.869	408.723
326.	30	402.755	16.771	410.147
327.	31	392.864	23.466	411.363
328.	32	380.696	32.368	412.335
329.	33	365.928	43.784	413.065
330.	34	348.489	57.736	413.639
331.	35	328.733	73.794	414.226
332.	36	307.512	91.011	415.068
333.	37	277.861	104.993	404.451
334.	38	263.057	101.406	426.763
335.	39	259.589	99.780	432.062
336.	40	237.315	89.250	399.944
337.	41	237.312	89.249	399.939

338.	FINAL LIQUID COMPOSITION PROFILE (MOLES):			
339.	STAGE	ETHANOL	WATER	BENZENE
340.	1	123.788	0.001	0.061
341.	2	904.476	0.006	1.078
342.	3	902.516	0.006	2.524
343.	4	898.887	0.007	5.826
344.	5	891.555	0.007	13.289
345.	6	876.426	0.008	29.670
346.	7	846.787	0.008	63.244
347.	8	796.375	0.009	122.951
348.	9	729.593	0.010	205.406
349.	10	666.923	0.012	284.745
350.	11	625.741	0.015	336.907
351.	12	604.669	0.020	362.934
352.	13	594.717	0.029	374.503
353.	14	589.562	0.040	379.839
354.	15	586.252	0.057	382.766
355.	16	583.595	0.082	384.816
356.	17	581.141	0.116	386.565
357.	18	578.717	0.165	388.223
358.	19	576.256	0.235	389.871
359.	20	573.725	0.334	391.534
360.	21	571.104	0.477	393.222
361.	22	568.365	0.680	394.936
362.	23	565.479	0.972	396.676
363.	24	562.406	1.390	398.440
364.	25	559.086	1.991	400.219
365.	26	555.438	2.852	402.006
366.	27	551.350	4.085	403.782
367.	28	546.665	5.845	405.524
368.	29	541.169	8.346	407.205
369.	30	534.579	11.870	408.783
370.	31	526.543	16.772	410.208
371.	32	516.652	23.467	411.424
372.	33	504.484	32.369	412.396
373.	34	489.716	43.785	413.126
374.	35	472.276	57.737	413.699
375.	36	452.521	73.794	414.287
376.	37	431.300	91.012	415.129
377.	38	209.749	38.394	404.512
378.	39	194.945	34.807	426.824
379.	40	191.476	33.180	432.122
380.	41	0.002	0.000	0.005
381.	STAGE #	1		

381.	STAGE #	1				
382.	TEMP, DEG K	385.380				
383.	PRES, MMHG	2560.0000				
384.	#	NAME	K	VP	FUG	GAM
385.	1	ETHANOL	0.99919	2550.8	1.0028	0.99999
386.	2	WATER	1.1350	1157.4	0.95681	2.6237
387.	3	BENZENE	2.6421	1858.2	0.96580	3.7688
388.	STAGE #	2				
389.	TEMP, DEG K	384.793				
390.	PRES, MMHG	2515.0000				
391.	#	NAME	K	VP	FUG	GAM
392.	1	ETHANOL	0.99804	2503.0	1.0029	0.99999
393.	2	WATER	1.1353	1135.1	0.95718	2.6281
394.	3	BENZENE	2.6484	1831.0	0.96609	3.7654
395.	STAGE #	3				
396.	TEMP, DEG K	384.148				
397.	PRES, MMHG	2470.0000				
398.	#	NAME	K	VP	FUG	GAM
399.	1	ETHANOL	0.99538	2451.2	1.0030	1.0000
400.	2	WATER	1.1353	1110.9	0.95761	2.6360
401.	3	BENZENE	2.6474	1801.4	0.96673	3.7550
402.	STAGE #	4				
403.	TEMP, DEG K	383.384				
404.	PRES, MMHG	2425.0000				
405.	#	NAME	K	VP	FUG	GAM
406.	1	ETHANOL	0.98943	2391.1	1.0034	1.0000
407.	2	WATER	1.1347	1082.9	0.95817	2.6519
408.	3	BENZENE	2.6302	1766.8	0.96813	3.7290
409.	STAGE #	5				
410.	TEMP, DEG K	382.375				
411.	PRES, MMHG	2380.0000				
412.	#	NAME	K	VP	FUG	GAM
413.	1	ETHANOL	0.97649	2313.5	1.0043	1.0002
414.	2	WATER	1.1327	1046.7	0.95897	2.6857
415.	3	BENZENE	2.5773	1721.8	0.97114	3.6684
416.	STAGE #	6				
417.	TEMP, DEG K	380.903				
418.	PRES, MMHG	2335.0000				
419.	#	NAME	K	VP	FUG	GAM
420.	1	ETHANOL	0.95077	2204.1	1.0061	1.0011
421.	2	WATER	1.1298	995.75	0.96010	2.7595
422.	3	BENZENE	2.4544	1657.9	0.97709	3.5379
423.	STAGE #	7				
424.	TEMP, DEG K	378.760				
425.	PRES, MMHG	2290.0000				
426.	#	NAME	K	VP	FUG	GAM
427.	1	ETHANOL	0.90865	2052.5	1.0086	1.0052
428.	2	WATER	1.1327	925.25	0.96133	2.9163
429.	3	BENZENE	2.2229	1568.0	0.98708	3.2889
430.	STAGE #	8				
431.	TEMP, DEG K	376.130				
432.	PRES, MMHG	2245.0000				
433.	#	NAME	K	VP	FUG	GAM
434.	1	ETHANOL	0.86215	1878.2	1.0106	1.0197
435.	2	WATER	1.1647	844.38	0.96189	3.2193
436.	3	BENZENE	1.8928	1462.9	0.99963	2.9059

612.	STAGE #	34				
613.	TEMP, DEG K	349.608				
614.	PRES, MMHG	1075.0000				
615.	#	NAME	K	VP	FUG	GAM
616.	1	ETHANOL	0.82164	705.16	1.0071	1.2437
617.	2	WATER	1.5225	307.33	0.97993	5.4346
618.	3	BENZENE	1.1560	678.39	1.0119	1.8103
619.	STAGE #	35				
620.	TEMP, DEG K	348.082				
621.	PRES, MMHG	1030.0000				
622.	#	NAME	K	VP	FUG	GAM
623.	1	ETHANOL	0.80425	663.08	1.0070	1.2406
624.	2	WATER	1.4768	288.40	0.98265	5.3673
625.	3	BENZENE	1.1569	646.32	1.0119	1.8221
626.	STAGE #	36				
627.	TEMP, DEG K	346.484				
628.	PRES, MMHG	985.0000				
629.	#	NAME	K	VP	FUG	GAM
630.	1	ETHANOL	0.78563	621.28	1.0068	1.2372
631.	2	WATER	1.4258	269.64	0.98544	5.2855
632.	3	BENZENE	1.1583	614.04	1.0118	1.8364
633.	STAGE #	37				
634.	TEMP, DEG K	344.834				
635.	PRES, MMHG	940.0000				
636.	#	NAME	K	VP	FUG	GAM
637.	1	ETHANOL	0.76709	580.49	1.0065	1.2342
638.	2	WATER	1.3736	251.37	0.98812	5.1984
639.	3	BENZENE	1.1601	582.06	1.0116	1.8519
640.	STAGE #	38				
641.	TEMP, DEG K	343.456				
642.	PRES, MMHG	895.0000				
643.	#	NAME	K	VP	FUG	GAM
644.	1	ETHANOL	1.0345	548.16	1.0054	1.6799
645.	2	WATER	2.1786	236.91	0.98756	8.3339
646.	3	BENZENE	0.87024	556.37	1.0120	1.3833
647.	STAGE #	39				
648.	TEMP, DEG K	342.091				
649.	PRES, MMHG	850.0000				
650.	#	NAME	K	VP	FUG	GAM
651.	1	ETHANOL	1.1047	517.64	1.0052	1.8046
652.	2	WATER	2.3782	223.29	0.98770	9.1656
653.	3	BENZENE	0.83978	531.82	1.0115	1.3269
654.	STAGE #	40				
655.	TEMP, DEG K	340.663				
656.	PRES, MMHG	805.0000				
657.	#	NAME	K	VP	FUG	GAM
658.	1	ETHANOL	1.1204	487.28	1.0051	1.8417
659.	2	WATER	2.4317	209.77	0.98780	9.4469
660.	3	BENZENE	0.83669	507.07	1.0109	1.3139
661.	STAGE #	41				
662.	TEMP, DEG K	339.087				
663.	PRES, MMHG	760.0000				
664.	#	NAME	K	VP	FUG	GAM
665.	1	ETHANOL	1.0336	455.51	1.0052	1.7156
666.	2	WATER	2.2077	195.64	0.98833	8.6774
667.	3	BENZENE	0.87615	480.82	1.0103	1.3708

APPENDIX D

THE SOLUTION METHOD OF TWO-LIQUID PHASE FLASH MODEL

- D.1 Temperature Determination via the Secant Method for Dew Point, Bubble Point, and the Adiabatic Flash
- D.2 Solution Methodology for Single Liquid Phase Isothermal Flash Design Equation
- D.3 Wegstein's Algorithm for the Acceleration of Compositional Convergence
- D.4 Solution Method of Two-Liquid Phase Systems
- D.5 Solution Method of the Two-Liquid Phases Isothermal Flash Design Equation

D.1 Temperature Determination Via the Secant Method for Dew Point, Bubble Point, and the Adiabatic Flash

The Secant method is a variation on Newton's method except that the first derivative of the function is not explicitly evaluated. Expansion of the function, $F(T) = 0.0$, yields

$$F(T) \cong F(T_i) + \frac{F(T_i) - F(T_{i-1})}{T_i - T_{i-1}} (T_{i+1} - T_i) = 0.0 \quad (D1-1)$$

Solving for the value of T_{i+1} , the next estimate of the desired temperature is obtained

$$T_{i+1} = T_i - F(T_i) \left[\frac{T_i - T_{i-1}}{F(T_i) - F(T_{i-1})} \right] \quad (D1-2)$$

The Secant method is not applicable until two iterations have been completed. For the first iteration, an estimate of the temperature is made, usually the given system temperature. For the second trial, an arbitrary estimate of the temperature is made predicted on the first temperature estimate being either too high or too low

$$T_2 = T_1 \pm 5.0 \quad (D1-3)$$

For the dew point temperature the following relationship is to be solved

$$F(T) = \sum_i \frac{Y_i}{K_i} - 1.0 = 0.0 \quad (D1-4)$$

The estimation of T_2 can be aided by evaluating the data obtained from the first trial. Therefore if

$$F(T_1) > 0.0 \quad T_2 = T_1 + 5.0 \quad (D1-5A)$$

$$F(T_1) < 0.0 \quad T_2 = T_1 + 5.0 \quad (D1-5B)$$

The bubble point temperature evaluation for the single liquid phase system and the two liquid phase system follow the same procedure except that different functional relationships are used.

Single Liquid Phase

$$F(T) = \sum_i K_i X_i - 1.0 = 0.0 \quad (D1-6)$$

Two Liquid Phases

$$F(T) = \sum_i \frac{Z_i}{W_i} \left(1.0 - \frac{K_{1i}}{K_{2i}} \right) = 0.0 \quad (D1-7)$$

where

$$W_i = R + (1.0 - R) \frac{K_{1i}}{K_{2i}} \quad (D1-8)$$

For two liquid phase systems the liquid phase distribution factor, R , is assumed to be constant. In both cases, the estimation of T_2 can be obtained as follows:

$$F(T_1) < 0.0 \quad T_2 = T_1 - 5.0 \quad (D1-9A)$$

$$F(T_1) > 0.0 \quad T_2 = T_1 + 5.0 \quad (D1-9B)$$

For the adiabatic flash calculation an initial temperature is assumed and an isothermal flash is performed including the system enthalpy at the temperature. The enthalpy function used as the design expression is

$$F(T) = H(T) - H_{\text{specified}} = 0.0 \quad (D1-10)$$

For the second temperature estimation the following is used:

$$F(T_1) < 0.0 \quad T_2 = T_1 + 10.0 \quad (D1-11A)$$

$$F(T_2) > 0.0 \quad T_2 = T_1 - 10.0 \quad (D1-11B)$$

D.2 Derivation of and the Solution of the Single Liquid Phase Isothermal Flash Design Equation

This appendix covers the derivation of the design equation and its solution.

The relationships governing the two-phase flash are:

$$\text{Material Balance: } (L + V)Z_i = LX_i + VY_i \quad (D2-1A)$$

$$\text{Equilibrium Relationship: } Y_i = K_i X_i \quad (D2-1B)$$

$$\text{Constraints: } X_i = 1.0 \quad (D2-1C)$$

$$Y_i = 1.0 \quad (D2-1D)$$

When combined, the desired design equation

$$F = \sum_i (X_i - Y_i) = 0.0 \quad (D2-2)$$

is obtained.

Using the material balance relation and the vapor phase distribution factor, U , the following is obtained:

$$(L + V)Z_i = LX_i + VY_i \quad (D2-1A)$$

$$Z_i = \frac{L}{L + V} X_i + \frac{V}{L + V} Y_i \quad (D2-3)$$

$$U = \frac{V}{L + V} \quad (D2-4A)$$

$$1 - U = \frac{L}{L + V} \quad (D2-4B)$$

$$Z_i = (1 - U)X_i + UY_i \quad (D2-5)$$

Substituting equation (D2-1B)

$$Z_i = [1 + (K_i - 1)U]X_i \quad (D2-6)$$

$$W_i = 1 + (K_i - 1)U \quad (D2-7)$$

The phase compositions are defined by

$$X_i = \frac{Z_i}{W_i} \quad (D2-8)$$

$$Y_i = \frac{Z_i}{W_i} \quad (D2-9)$$

The resulting design equation is thus

$$F = \sum_i (X_i - Y_i) = \sum_i \frac{Z_i(1 - K_i)}{W_i} = 0.0 \quad (D2-10)$$

The unknown in the design equation (D2-11) is U , the vapor phase distribution factor which has the following range of values

$$0.0 < U < 1.0 \quad (D2-11)$$

Solution of the Design Equation

The solution of equation (D2-11) is accomplished by use of Newton's method. Assuming an initial value of $U = 0.5$ for the vapor phase distribution factor, Newton's method provides the mechanism of evaluating successive values of U . The iterative procedure is repeated until both the function and the correction to the vapor phase distribution factor are essentially zero, that is

$$|F| < \delta_1 = 10^{-5} \quad (D2-12A)$$

$$|U| < \delta_2 = 10^{-5} \quad (D2-12B)$$

For a given value of $U = U^0$, the next value of U is given by

$$U = U^0 + \Delta U \quad (D2-13A)$$

$$\Delta U = - \frac{F}{\frac{dF}{du}} \quad (\text{D2-13B})$$

where

$$\frac{dF}{du} = - \sum_i \frac{Z_i(1 - K_i)}{W_i^2} \cdot \frac{dw_i}{du} = \sum_i \frac{Z_i(1 - K_i)^2}{W_i^2} \quad (\text{D2-14})$$

evaluated at $U = U^0$.

The initialization of the solution method requires:

- Liquid phase compositions...obtained from the dew point temperature evaluation.
- Vapor phase compositions...obtained from the bubble point temperature evaluation.
- Assume that the flash has $U = 0.5$, as previously mentioned.

With the above initialization data, the equilibrium data can be generated, as well as the function and the associated derivative.

D.3 Acceleration of Compositional Convergence Using the Wegstein Algorithm

In all of the equilibrium calculations encountered in the multicomponent nonideal flash program compositional convergence is essential. Thus, the Wegstein convergence promotion algorithm is used throughout the program. The Wegstein algorithm is, as with the Secant method, a variation of Newton's method except that the function to be solved is:

$$F(Z) = Z \quad (\text{D3-1})$$

Expansion of the function into a linearized form yields

$$F(Z) = F(Z_i) + \frac{F(Z_i) - F(Z_{i-1})}{Z_i - Z_{i-1}} (Z_{i+1} - Z_i) = Z_{i+1} \quad (D3-2)$$

Solving for Z_{i+1} , the next estimate of the component's mole fraction is

$$Z_{i+1} = \frac{f(Z_i) - SZ_i}{1 - S} \quad (D3-3)$$

where

$$S = \frac{F(Z_i) - F(Z_{i-1})}{Z_i - Z_{i-1}} \quad (D3-4)$$

As encountered in the Secant method, the Wegstein algorithm can not be applied until two iterations have passed. For the first iteration, the composition estimation is either based on ideality or a pre-established methodology. The second iteration is initialized via the mode of direct substitution. For the third and subsequent trials the relation described in equation (D3-3) is applied.

D.4 Solution Method of Two Liquid Phase Systems

The solution method of the design equation, Eq. 3-11, is accomplished by use of Newton's method. Assuming an initial value of $R = 0.5$ for the liquid phase distribution factor, Newton's method provides the means of obtaining successive value of R . The iterative procedure is repeated until both the function and correction to the distribution factor are essentially zero,

$$|\Delta R| < \delta = 10^{-5} \quad (D4-1)$$

For a given value of $R = R^0$, the next value of R is given by

$$R = R^0 + \Delta R \quad (D4-2)$$

$$\Delta R = - \frac{F}{dF/dR} \quad (D4-3)$$

where

$$\frac{dF}{dR} = - \sum_i \frac{Z_i}{W_i^2} \frac{dW_i}{dR} = - \sum_i \frac{Z_i}{W_i} \left(1 - \frac{Y_{1,i}}{Y_{2,i}} \right) \quad (D4-4)$$

are evaluated at $R = R^0$. The only requirement is that the liquid phase composition be properly initiated so that the liquid-liquid distribution factor can be properly evaluated.

D.5 Solution Methodology for the Two Liquid Phases Isothermal Flash Design Equation

The design equations as discussed in Chapter 3 are as follow:

$$F_1 = \sum_i (x_{1,i} - y_i) = \sum_i \frac{Z_i (1 - K_{1,i})}{i} = 0.0 \quad (3-20)$$

and

$$F_2 = \sum_i (x_{1,i} - x_{2,i}) = \sum_i \frac{Z_i (1 - \frac{K_{1,i}}{K_{2,i}})}{i} = 0.0 \quad (3-21)$$

where

$$W_i = R(1 - U) + (1 - R)(1 - U) \frac{K_{1,i}}{K_{2,i}} + U K_{1,i} \quad (3-22)$$

Substituting Eq. 3-22 into Eq. 3-20 and Eq. 3-21, the unknowns in the resulting equations are U and R , the vapor phase and liquid phase distribution factors.

These factors have the following range of values:

$$0.0 < U < 1.0 \quad (\text{D5-1})$$

$$0.0 < R < 1.0 \quad (\text{D5-2})$$

For a given set of initial values, $U = U^0$ and $R = R^0$, the next set is given by

$$U = U^0 + \Delta U \quad (\text{D5-3})$$

$$R = R^0 + \Delta R \quad (\text{D5-4})$$

The values of ΔU and ΔR are obtained by solving simultaneously the following pair of linearized equations:

$$F_1 + \frac{\partial F_1}{\partial U} \Delta U + \frac{\partial F_1}{\partial R} \Delta R = 0 \quad (\text{D5-5A})$$

$$F_2 + \frac{\partial F_2}{\partial U} \Delta U + \frac{\partial F_2}{\partial R} \Delta R = 0 \quad (\text{D5-5B})$$

in matrix form,

$$\begin{vmatrix} \frac{\partial F_1}{\partial U} & \frac{\partial F_1}{\partial R} \\ \frac{\partial F_2}{\partial U} & \frac{\partial F_2}{\partial R} \end{vmatrix} \cdot \begin{vmatrix} \Delta U \\ \Delta R \end{vmatrix} = \begin{vmatrix} -F_1 \\ -F_2 \end{vmatrix} \quad (\text{D5-6})$$

where

$$\frac{\partial F_1}{\partial U} = - \sum_i \frac{Z_i (1 - K_{1,i})}{W_i^2} \frac{\partial W_i}{\partial U} \quad (\text{D5-7A})$$

$$\frac{\partial F_1}{\partial R} = - \sum_i \frac{Z_i (1 - K_{1,i})}{W_i^2} \frac{\partial W_i}{\partial R} \quad (\text{D5-7B})$$

$$\frac{\partial F_2}{\partial U} = - \sum_i \frac{z_i (1 - \frac{K_{1,i}}{K_{2,i}})}{w_i^2} \frac{\partial w_i}{\partial U} \quad (D5-7C)$$

$$\frac{\partial F_2}{\partial R} = - \sum_i \frac{z_i (1 - \frac{K_{1,i}}{K_{2,i}})}{w_i^2} \frac{\partial w_i}{\partial R} \quad (D5-7D)$$

$$\frac{\partial w_i}{\partial U} = R \left(\frac{K_{1,i}}{K_{2,i}} - 1 \right) - \frac{K_{1,i}}{K_{2,i}} + K_{1,i} \quad (D5-8)$$

$$\frac{\partial w_i}{\partial R} = (1-U) \left(1 - \frac{K_{1,i}}{K_{2,i}} \right) \quad (D5-9)$$

The initialization of the solution method requires:

- Liquid phase composition, obtained by an arbitrary positioning of the feed.
- Vapor phase composition, obtained from the bubble point temperature evaluation.
- Assume the isothermal flash has $U = R = 0.5$

Using the above, the vapor liquid equilibrium constant can be generated as well as the functions F_1 and F_2 , and the associated derivatives.

APPENDIX E

SAMPLE PROBLEMS OF TWO-LIQUID PHASE FLASH MODEL

E.1 The Research Problem Distillate Flash Example

E.2 a. System--Benzene-Ethyl Acetate-Water

b. System--Benzene-Ethanol-Ethylacetate-Water

1

```

1. //EXXON JOB (JA00008-SAI,NJIT,-30,10), 'ROCHE-TSAI-CASES',
1.1 // MSGLEVEL=(1,1),TIME=(2,30)
1.2 /*PRIORITY +0
6000. //STEP6 EXEC PGM=EXXON,REGION=100K
6001. //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
6002. //FT06F001 DD SYSOUT=A
6003. //FT05F001 DD *
6004. CHEMSARE'S BZ-ETOH-WATER*** TWO LIQUID PHASE FLASH- FD @308 DEG K
6005. 3 1 2 2 0 0 1
6006. 2 308. 760.
6007. 31226.5 10968.2 1606.5
6008. BENZENE 0 78.114 562.1 48.3 259.0 92.26
6009. 0.0 0.0 0.544 0.251
6010. 94.118 353.3 0.41 0.249
6011. 15.9008 -2788.51 -52.36
6012. ETHANOL 0 46.07 516.2 63.0 167.0 64.0
6013. 0.0878 0.0572 0.248 0.635
6014. 202.06 351.1 0.58 0.38
6015. 18.9119 -3803.98 -41.68
6016. WATER 0 18.01 647.3 217.3 56.0 18.84
6017. 0.0279 0.0229 0.229 0.344
6018. 538.7 373.2 1.0 0.454
6019. 18.3036 -3816.44 -46.13
6020. 0.0 0.15 0.40
6021. 0.15 0.0 0.10
6022. 0.40 0.10 0.0
6023. 1096.889 0.0 282.332 0.0 2213.355 0.0 3820.95
6024. -86.763 0.0 1289.908 0.0
6025. 0.295 0.267 0.270
6026. /*
8019. //
8020.
8021.
8022.

```

?

E.1 Results

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COMPONENT NO	1	2	3	
30. NAME	BENZENE	ETHANOL	WATER	
31. MOLECULAR WEIGHT	78.1	46.1	18.0	
32. CRITICAL PROPERTIES				
33. TEMPERATURE (DEG K)	562.1	516.2	647.3	
34. PRESSURE (ATM)	48.3	63.0	217.3	
35. VOLUME (CC/GMOL)	259.0	167.0	56.0	
36. COMPRESSIBILITY	0.544	0.248	0.229	
37. ACENTRIC FACTOR	0.251	0.635	0.344	
38. TSONOPoulos -A-	0.0	0.0878	0.0229	
39. TSONOPoulos -R-	0.0	0.0572	0.0229	
40. AK(1,1) =	0.0			
41. AK(1,2) =	0.150			
42. AK(1,3) =	0.400			
43. AK(2,1) =	0.150			
44. AK(2,2) =	0.0			
45. AK(2,3) =	0.100			
46. AK(3,1) =	0.400			
47. AK(3,2) =	0.100			
48. AK(3,3) =	0.0			
49. VAPOR PRESSURE CONSTANTS (MMHG AND DEG K)				
50. C(1)	15.901	18.912	18.304	
51. C(2)	-2788.51	-3803.98	-3816.44	
52. C(3)	-52.36	-41.68	-46.13	
53. C(4)	0.0	0.0	0.0	
54. C(5)	0.0	0.0	0.0	
55. C(6)	0.0	0.0	0.0	
56. LIQUID MOLAR VOLUME CONSTANTS				
57. C(1)	92.260	64.000	18.840	
58. ENTHALPY CONSTANTS (CAL/GM)				
59. AT DEG K	94.1	202.1	538.7	
60. LIQUID SP. HEAT CONSTANTS				
61. C(1)	0.410	0.580	1.000	
62. C(2)	0.0	0.0	0.0	
63. C(3)	0.0	0.0	0.0	
64. C(4)	0.0	0.0	0.0	
65. VAPOR SP. HEAT CONSTANTS				
66. C(1)	0.249	0.380	0.454	
67. C(2)	0.0	0.0	0.0	
68. C(3)	0.0	0.0	0.0	
69. C(4)	0.0	0.0	0.0	
70. NRTL CONSTANTS, CAL/GMOL				
71. UNIQVAC PARAMETERS, CAL/G MOL				
72. A 12 =	1096.89	0.0	0.0	
73. A 13 =	2213.35	0.0	0.0	
74. A 23 =	-86.76	0.0	0.0	
75. ALPHA12 =	0.2950	ALPHA13 =	0.2670	
76. FEED =	43801. LBS	727.03 MOLES	ALPHA23 =	0.2700
77. FEED MOLEFRACTIONS	BENZENE	ETHANOL	WATER	
78. B 12 =	0.54985	0.32746	0.12269	
79. B 13 =				
80. B 23 =				
81. PRESS =	760.00 MMHG	ALPHA	TEMP =	
			308.00 DEG K	

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82. RESULTS OF DEW POINT CALCULATION
 83. DEW POINT OF FEED AT 760.00 MMHG IS 338.89 DEG K
 84. BENZENE ETHANOL WATER
 85. DEW PT LIQ MOLE FRAC 0.67180 0.28479 0.04341

86.
 87. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 1

	BENZENE	ETHANOL	WATER
89.			
90.	VAP MOLE FRAC	0.54985	0.32746
91.	LIQ MOLE FRAC	0.67180	0.28479
92.	VALUES OF K=Y/X	0.81846	1.14986
93.	VAPOR PRESSURES	477.577	451.617
94.	FUGACITY FACTORS	1.00954	1.00739
95.	ACTIVITY COEFFICIENTS	1.2902	1.9208
96.	ACTIVITY	0.8667	0.5470

97.
 98.
 99.

100.

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118. TWOLIQ CALCULATION TERMINATED
 119. CALCULATION CONTINUING

120.

121.

122.

123. TWOLIQ CALCULATION TERMINATED
 124. CALCULATION CONTINUING

125.

126.

127.

128. TWOLIQ CALCULATION TERMINATED
 129. CALCULATION CONTINUING

130.

131.

132.

133. TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO 0.423069

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133. TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO 0.423069
 134. BUBBLE POINT OF FEED AT 760.00 MMHG IS 337.85 DEG K
 135. BENZENE ETHANOL WATER
 136. BUB PT VAP MOLE FRAC 0.53936 0.27362 0.18701

137.
 138. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 1

	BENZENE	ETHANOL	WATER
140.			
141.	0.53936	0.27362	0.18701
142.	0.64933	0.26810	0.08257
143.	0.83065	1.02059	2.26494
144.	460.977	431.810	185.127
145.	1.00960	1.00829	0.99997
146.	1.3564	1.7815	9.2985
147.	0.8808	0.4776	0.7678

148.
 149. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 2

	BENZENE	ETHANOL	WATER
150.			
151.			
152.	0.53936	0.27362	0.18701
153.	0.47689	0.37099	0.15211
154.	1.13106	0.73750	1.22928
155.	460.977	431.810	185.127
156.	1.00960	1.00829	0.99997
157.	1.8470	1.2873	5.0467
158.	0.8808	0.4776	0.7677

159. FEED AT OR BELOW BUBBLE POINT
 160. TWOLIQ CALCULATION TERMINATED
 161. CALCULATION CONTINUING

162.
 163.
 164.

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164.
 165. TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO 0.404562
 166.

167. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 1
 168.

	BENZENE	ETHANOL	WATER
169.			
170.	VAP MOLE FRAC	0.0	0.0
171.	LIQ MOLE FRAC	0.77289	0.18548
172.	VALUES OF K=Y/X	0.0	0.0
173.	VAPOR PRESSURES	0.0	0.0
174.	FUGACITY FACTORS	1.00960	1.00829
175.	ACTIVITY COEFFICIENTS	1.1660	2.7285
176.	ACTIVITY	0.9012	0.5061

177.
 178. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 2
 179.

	BENZENE	ETHANOL	WATER
180.			
181.	VAP MOLE FRAC	0.0	0.0
182.	LIQ MOLE FRAC	0.39830	0.42394
183.	VALUES OF K=Y/X	0.0	0.0
184.	VAPOR PRESSURES	0.0	0.0
185.	FUGACITY FACTORS	1.00960	1.00829
186.	ACTIVITY COEFFICIENTS	2.2628	1.1937
187.	ACTIVITY	0.9013	0.5060
188.	IDEAL LIQ ENTHALPY =	-0.138381E 04	DELTA H = 0.250980E 00
189.	IDEAL LIQ ENTHALPY =	-0.127482E 04	DELTA H = 0.173376E 03

E.2.a Inputs

```

1. //EDWARD JOB (JA00008-SAI,NJIT,-30,10),'ROCHE-TSAI-CASES',
1.1 // MSGLEVEL=(1,1),TIME=(2,30)
1.2 /*PRIORITY +0
6000. //STEP4 EXEC PGM=EDWARD,REGION=100K
6001. //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
6002. //FT06F001 DD SYSOUT=A
6003. //FT05F001 DD *
6004. BZ-ETAC-WTR *** TWO LIQUID PHASE TEST ----- TEST CASE #1
6005. 3 2 1 2 0 1 0 0 0 176.0 175.0 14.696
6006. BZ 78.114 1011.8 710. 4.152 1.414 .42748 .08664
6007. 169.8 .45684 .00135 .00000148 .46295 -.00031304.000000961
6008. 5.192 -2179.9
6009. ETAC 0 88.107 341.28 555.51 4.576 1.565 .42748 .08864
6010. 157.45 0.35
6011. 5.30095 -2181.42 302.17
6012. WATER 18.016 1165.1 3918.7 .8977 .289 .42748 .08864
6013. 992.6 .94776 .0001795 .000000908 .19203 .00017491 .000000027
6014. 6.23553 -2983.444
6015. 201.0 -119.12 1736.38 4001.43 -455.96 376.636
6016. 0.3 0.2 0.082
6017. 78.114 88.107 18.016 2599.45
/*
//STEP5 EXEC PGM=EDWARD,REGION=100K
//STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
//FT06F001 DD SYSOUT=A
//FT05F001 DD *
BZ-ETOH - ETAC - WTR *** TEST CASE #2
7004. 4 2 1 2 0 1 0 0 0 176.0 160.0 14.696
7005. BZ 78.114 1011.8 710. 4.152 1.414 .42748 .08664
7006. 169.8 .45684 .00135 .00000148 .46295 -.00031304.000000961
7007. 5.192 -2179.9 365.42
7008. ETOH 46.069 923.4 892.05 2.673 0.943 .42748 .08664
7009. 362.0 0.35
7010. 6.4982 -2973.06 384.64
7011. ETAC 0 88.107 341.28 555.51 4.576 1.565 .42748 .08864
7012. 157.45 0.35
7013. 5.30095 -2181.42 302.17
7014. WATER 18.016 1165.1 3918.7 .8977 .289 .42748 .08864
7015. 992.6 .94776 .0001795 .000000908 .19203 .00017491 .000000027
7016. 6.23553 -2983.444 376.636
7017. 1192.8 561.86 201.0 -119.12 1736.38 4001.43 1064.2 -43.04
7018. 323.3 358.73 -455.96 2599.45
7019. 0.47 0.3 0.3
7020. 78.114 46.069 88.107 18.016 0.082
7021. /*
7022.

```

E.2.b Inputs

BZ-ETAC-WTR *** TWO LIQUID PHASE TEST ----- TEST CASE #1
 COMPONENT NO 1 2 3
 NAME BZ ETAC WATER
 COMPOSITION TYPE 1 1 1
 MOLECULAR WEIGHT 78.1 88.1 18.0
 CRITICAL PROPERTIES
 TEMPERATURE (DEG R) 1011.8 341.3 1165.1
 PRESSURE (PSIA) 710.0 555.5 3918.7
 VOLUME (CFT/LBMOL) 4.2 4.6 0.9
 VAPOR PRESSURE CONSTANTS
 (PSIA AND DEG F)
 C(1) 5.1920 5.3009 6.2355
 C(2) -2179.9 -2181.4 -2983.4
 C(3) 0.0 0.0 0.0
 C(4) 0.0 0.0 0.0
 C(5) 0.0 0.0 0.0
 C(6) 365.4 302.2 376.6
 LIQUID MOLAR VOLUME CONSTANTS
 C(1) 1.4140 1.5650 0.2890
 C(2) 0.0 0.0 0.0
 C(3) 0.0 0.0 0.0
 ENTHALPY CONSTANTS (BTU/LB)
 LATENT HEAT AT 176.0 DEG F
 LIQUID SP. HEAT CONSTANTS
 C(1) 169.8 157.4 992.6
 C(2) 0.4568 0.3500 0.9478
 C(3) 0.0013 0.0 0.0002
 C(4) 0.0000 0.0 0.0000
 VAPOR SP. HEAT CONSTANTS
 C(1) 0.4629 0.5000 0.1920
 C(2) -0.0003 0.0 0.0002
 C(3) 0.0000 0.0 0.0000
 RENON CONSTANTS, CAL/GMOL
 A12 = 201. A21 = -119. A13 = 1736. A31 = 4001.
 ALPHA12 = 0.3000 ALPHA13 = 0.2000 ALPHA23 = 0.0820 ALPHA
 FEED = 184. LBS = 3.00 MOLES PRESS = 14.70 PSIA
 FEED MOLE FRACTIONS BZ ETAC WATER
 0.33333 0.33333 0.33333
 RESULTS OF DEW POINT CALCULATION
 DEW POINT OF FEED AT 14.70 PSIA IS 188.28 DEG F
 DEW PT LIQ MOLE FRAC BZ ETAC WATER
 0.24285 0.69450 0.06265
 VLE DATA FOR VAPOR AND LIQUID PHASE # 1
 VAP MOLE FRAC BZ ETAC WATER
 0.33333 0.33333 0.33333
 LIQ MOLE FRAC 0.24285 0.69450 0.06265
 VALUES OF K=Y/X
 1.37260 0.47996 5.32015
 VAPOR PRESSURES 17.990 7.131 9.002
 FUGACITY FACTORS 0.98933 0.99967 1.00249
 ACTIVITY COEFFICIENTS 1.1333 0.9894 8.6640
 0.2752 0.6872 0.5428

A23 = -456. A32 = 2599.
 TEMP = 175.00 DEG F

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112. RESULTS OF BURBLE POINT CALCULATION
 113. TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO 0.718023
 114. BURBLE POINT OF FEED AT 14.70 PSIA IS 169.44 DEG F
 115. BZ ETAC WATER
 116. BUR PT VAP MOLE FRAC 0.46580 0.13969 0.39452

117. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 1
 118. BZ ETAC WATER
 119. VAP MOLE FRAC 0.46580 0.13969 0.39452
 120. LIQ MOLE FRAC 0.46397 0.45401 0.08202
 121. VALUES OF K=Y/X 1.00395 0.30767 4.81016
 122. VAPOR PRESSURES 13.073 4.737 5.918
 123. FUGACITY FACTORS 1.00069 0.99803 1.00334
 124. ACTIVITY COEFFICIENTS 1.1278 0.9564 11.9059
 125. ACTIVITY 0.5233 0.4342 0.9765

126. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 2
 127. BZ ETAC WATER
 128. VAP MOLE FRAC 0.46580 0.13969 0.39452
 129. LIQ MOLE FRAC 0.00069 0.02603 0.97328
 130. VALUES OF K=Y/X 675.17114 5.36557 0.40535
 131. VAPOR PRESSURES 13.073 4.737 5.918
 132. FUGACITY FACTORS 1.00069 0.99803 1.00334
 133. ACTIVITY COEFFICIENTS 758.4580 16.6786 1.0033
 134. ACTIVITY 0.5232 0.4342 0.9765
 135. RESULTS OF ISOTHERMAL FLASH CALCULATION

136. LIQUID HAS GONE TO A ONE PHASE SYSTEM

137. SYSTEM IS IN THE V/L PHASE REGION AT 175.00 DEG F AND 14.70 PSIA. VAPOR/FEED MOLE RATIO = 0.667825

138. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 1

139. BZ ETAC WATER
 140. VAP MOLE FRAC 0.35124 0.20779 0.44098
 141. LIQ MOLE FRAC 0.29734 0.58574 0.11692
 142. VALUES OF K=Y/X 1.18125 0.35474 3.77173
 143. VAPOR PRESSURES 14.397 5.363 6.717
 144. FUGACITY FACTORS 0.99669 0.99948 1.00355
 145. ACTIVITY COEFFICIENTS 1.2098 0.9727 8.2230
 146. ACTIVITY 0.3597 0.5697 0.9614
 147. IDEAL LIQ ENTHALPY = -0.373583E 02 DELTA H = 0.0
 148. IDEAL VAP ENTHALPY = 0.154041E 05 DELTA H = -0.345228E 02
 149. ENTHALPY OF VAPOR PHASE/LBMOL OF FEED = 10287.26 BTU
 150. ENTHALPY OF LIQUID PHASE/LBMOL OF FEED = -12.41 BTU
 151. ENTHALPY OF SYSTEM WITH REFERENCE TO PURE LIQUID AT 176.00 DEG F 10274.85 BTU/LBMOL

BZ- ETOH - ETAC - WTR *** TEST CASE #2

COMPONENT NO	1	2	3	4
163. NAME	BZ	ETOH	ETAC	WATER
164. COMPOSITION TYPE	1	1	1	1
165. MOLECULAR WEIGHT	78.1	46.1	88.1	18.0
166. CRITICAL PROPERTIES				
167. TEMPERATURE (DEG R)	1011.8	923.4	341.3	1165.1
168. PRESSURE (PSIA)	710.0	892.0	555.5	3918.7
169. VOLUME (CFT/LRMOL)	4.2	2.7	4.6	0.9
170. VAPOR PRESSURE CONSTANTS (PSIA AND DEG F)				
171. C(1)	5.1920	6.4982	5.3009	6.2355
172. C(2)	-2179.9	-2973.1	-2181.4	-2983.4
173. C(3)	0.0	0.0	0.0	0.0
174. C(4)	0.0	0.0	0.0	0.0
175. C(5)	0.0	0.0	0.0	0.0
176. C(6)	365.4	384.6	302.2	376.6
177. LIQUID MOLAR VOLUME CONSTANTS				
178. C(1)	1.4140	0.9430	1.5650	0.2890
179. C(2)	0.0	0.0	0.0	0.0
180. C(3)	0.0	0.0	0.0	0.0
181. ENTHALPY CONSTANTS (BTU/LB)				
182. LATENT HEAT AT 176.0 DEG F	169.8	362.0	157.4	992.6
183. LIQUID SP. HEAT CONSTANTS				
184. C(1)	0.4568	0.3500	0.3500	0.9478
185. C(2)	0.0013	0.0	0.0	0.0002
186. C(3)	0.0000	0.0	0.0	0.0000
187. VAPOR SP. HEAT CONSTANTS				
188. C(1)	0.4629	0.5000	0.5000	0.1920
189. C(2)	-0.0003	0.0	0.0	0.0002
190. C(3)	0.0000	0.0	0.0	0.0000
191. RENON CONSTANTS, CAL/GMOL				
192. A12 = 1193.	A21 = 562.	A13 = 201.	A31 = -119.	A41 = 4001.
193. A23 = 1064.	A32 = -43.	A24 = 323.	A42 = 359.	A43 = 2599.
194. ALPHA12 = 0.4700	ALPHA13 = 0.3000	ALPHA14 = 0.2000	ALPHA23 = 0.3000	ALPHA24 = 0.3000
195. FEED = 230. LBS	= 4.00 MOLES	PRESS = 14.70 PSIA	TEMP = 160.00 DEG F	ALPHA34 = 0.0820
196. FEED MOLEFRACTIONS	BZ 0.25000	ETOH 0.25000	ETAC 0.25000	WATER 0.25000
197. RESULTS OF DEW POINT CALCULATION				
198. DEW POINT OF FEED AT 14.70 PSIA IS 178.83 DEG F				
199. DEW PT LIQ MOLE FRAC	BZ 0.19634	ETOH 0.10190	ETAC 0.63309	WATER 0.06867
200. *VLE* DATA FOR VAPOR AND LIQUID PHASE # 1				
201. VAP MOLE FRAC	BZ 0.25000	ETOH 0.25000	ETAC 0.25000	WATER 0.25000
202. LIQ MOLE FRAC	0.19634	0.10190	0.63309	0.06867
203. VALUES OF K=Y/X	1.27332	2.45328	0.39489	3.64068
204. VAPOR PRESSURES	15.370	16.667	5.831	7.319
205. FUGACITY FACTORS	0.99509	0.99697	0.99910	1.00310
206. ACTIVITY COEFFICIENTS	1.2235	2.1697	0.9961	7.2873
207. ACTIVITY	0.2402	0.2211	0.6306	0.5004

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RESULTS OF BUBBLE POINT CALCULATION
TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO 0.739141
BUBBLE POINT OF FEED AT 14.70 FSIA IS 159.48 DEG F
BZ ETOH ETAC WATER
BUB PT VAP MOLE FRAC 0.36602 0.31140 0.08105 0.24153

VLE DATA FOR VAPOR AND LIQUID PHASE # 1

	BZ	ETOH	ETAC	WATER
VAP MOLE FRAC	0.36602	0.31140	0.08105	0.24153
LIQ MOLE FRAC	0.33484	0.23401	0.32336	0.10779
VALUES OF K=Y/X	1.09309	1.33071	0.25066	2.24087
VAPOR PRESSURES	10.940	10.820	3.764	4.684
FUGACITY FACTORS	1.00613	1.00508	0.99674	1.00338
ACTIVITY COEFFICIENTS	1.4594	1.7982	0.9818	7.0069
ACTIVITY	0.4887	0.4208	0.3175	0.7552

VLE DATA FOR VAPOR AND LIQUID PHASE # 2

	BZ	ETOH	ETAC	WATER
VAP MOLE FRAC	0.36602	0.31140	0.08105	0.24153
LIQ MOLE FRAC	0.00959	0.29531	0.04213	0.65297
VALUES OF K=Y/X	38.15723	1.05449	1.92375	0.36990
VAPOR PRESSURES	10.940	10.820	3.764	4.684
FUGACITY FACTORS	1.00613	1.00508	0.99674	1.00338
ACTIVITY COEFFICIENTS	50.9438	1.4250	7.5348	1.1566
ACTIVITY	0.4887	0.4208	0.3175	0.7553

RESULTS OF ISOTHERMAL FLASH CALCULATION

?

SYSTEM IS IN THE V/L PHASE REGION AT 160.00 DEG F AND 14.70 PSIA.
 TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO = 0.745013

VAPOR/FEED MOLE RATIO = 0.056955

VLE DATA FOR VAPOR AND LIQUID PHASE # 1

	BZ	ETOH	ETAC	WATER
VAP MOLE FRAC	0.35853	0.31080	0.08521	0.24546
LIQ MOLE FRAC	0.32364	0.23156	0.33399	0.11080
VALUES OF K=Y/X	1.10781	1.34217	0.25513	2.21525
VAPOR PRESSURES	11.044	10.951	3.811	4.742
FUGACITY FACTORS	1.00582	1.00488	0.99687	1.00342
ACTIVITY COEFFICIENTS	1.4656	1.7925	0.9870	6.8413
	0.4743	0.4151	0.3297	0.7581

VLE DATA FOR VAPOR AND LIQUID PHASE # 2

	BZ	ETOH	ETAC	WATER
VAP MOLE FRAC	0.35853	0.31080	0.08521	0.24546
LIQ MOLE FRAC	0.00913	0.28947	0.04362	0.65778
VALUES OF K=Y/X	39.25723	1.07366	1.95357	0.37317
VAPOR PRESSURES	11.044	10.951	3.811	4.742
FUGACITY FACTORS	1.00582	1.00488	0.99687	1.00342
ACTIVITY COEFFICIENTS	51.9362	1.4339	7.5577	1.1525
	0.4743	0.4151	0.3297	0.7581
IDEAL LIQ ENTHALPY =	-0.550023E 03	DELTA H =	0.0	
IDEAL LIQ ENTHALPY =	-0.294764E 03	DELTA H =	0.0	
IDEAL VAP ENTHALPY =	0.151238E 05	DELTA H =	-0.511989E 02	
ENTHALPY OF VAPOR PHASE/LBMOL OF FEED =			861.37 BTU	
ENTHALPY OF LIQUID PHASE/LBMOL OF FEED =			-386.44 BTU	
ENTHALPY OF LIQUID PHASE/LBMOL OF FEED =			-70.88 BTU	
ENTHALPY OF SYSTEM WITH REFERENCE TO PURE LIQUID AT 176.00 DEG F				404.06 BTU/LBMOL

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APPENDIX F

RESEARCH PROBLEM COMPUTER INPUTS AND RESULTS
USING THE COMPLETE MODEL

F.1 The Combined Model

F.2 The Benzene Stripping Column

F.3 The Ethanol Recovery/Water Purge Column

F.1 Inputs

```

1.      //FINAL2 JOB (JA00008,NJIT,15000,30,0),'RCOHE-TSAI-FINAL2',
1.01    //      REGION=280K,TIME=(4,30)
1.02    //*PASSWORD SAI
1.03    //*MAIN CLASS=+2,REPORT=YES,FAILURE=RESTART
1.04    //*FORMAT PR,DDNAME=,DEST=RMT20,COPIES=1
7003.   //STEP5 EXEC PGM=FINAL2,REGION=280K
7004.   //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
7005.   //FT06F001 DD SYSOUT=A
7006.   //FT05F001 DD *
7007.   COMPLETE AZEOTROPIC DISTILLATION MODEL-ETOH DEHYDRATION-REB ABS
7008.   3 1 2 2 0 0 1
7009.   2          308.0          760.0
7010.   0.77289    0.18548        0.04163
7011.   0.39830    0.42394        0.17777
7012.   BENZENE 0 78.114          562.1          48.3          259.0          92.26
7013.   0.0          0.0          0.544          0.251
7014.   94.118     353.3          0.41
7015.   15.9008     -2788.51       -52.36
7016.   ETHANOL 0 46.07          516.2          63.0          167.0          64.0
7017.   0.0878      0.0572         0.248          0.635
7018.   202.06      351.1          0.58
7019.   18.9119     -3803.98       -41.68
7020.   WATER 0 18.01          647.3          217.3          56.0          18.84
7021.   0.0279      0.0229         0.229          0.344
7022.   538.7       373.2          1.0
7023.   18.3036     -3816.44       -46.13
7024.   0.0          0.15           0.40
7025.   0.15         0.0            0.20
7026.   0.40         0.20           0.0
7027.   1096.889     0.0            282.332        0.0          2213.335        0.0          3820.95
7028.   -86.763    0.0            1289.908        0.0
7029.   0.295        0.267          0.270
7030.   5 041 2 0 020 2 0
7031.   1.0          0.0            0.0
7032.   727.00      0.0            6.015          338.0          352.0          760.0          0.001
7034.   41          308.2          0.0            400.0          169.2          22.65
7034.1  37          310.97         0.0            0.0            191.9          66.6
7035.   /*
7036.   //

```

F.1 Results

COMPLETE AZEOTROPIC DISTILLATION MODEL-ETOH DEHYDRATION-REB ABS
 REBOILED ABSORBER

FEED STREAMS:

STAGE	41.	37.		
TEMP	308.20	310.97		
FRACV	0.0	0.0		
H	-7.9823E 05	-2.8042E 05		
COMP 1 BENZENE	400.000	0.0	400.000	
COMP 2 ETHANOL	169.200	191.900	361.100	
COMP 3 WATER	22.650	66.600	89.250	
TOTAL	591.850	258.500	850.350	
NUMBER OF STAGES	41			

ESTIMATED OVERHEAD PRODUCT RATE 727.0000
 ESTIMATED BOTTOMS PRODUCT RATE 123.3496

PRESSURE AT TOP OF COLUMN 759.99097
 CONSTANT DELTA P PER STAGE 0.00100
 BOILUP RATIO 6.014999
 TRIAL # = 1

ITERATION # = 1						
SUM DT**2 = 0.0						
SUM (DF/F)**2 = 81754.1						
41 0.0	26	3	46.322	25	3	56.931
ITERATION # = 2						
SUM DT**2 = 793.034						
SUM (DF/F)**2 = 134860.						
4 8.6276	24	3	62.298	23	3	82.182
ITERATION # = 3						
SUM DT**2 = 41.5678						
SUM (DF/F)**2 = 9769.94						
1 3.1751	1	3	57.191	2	3	29.796
ITERATION # = 4						
SUM DT**2 = 25.0274						
SUM (DF/F)**2 = 8429.71						
2 4.1163	1	1	15.082	1	3	89.586
ITERATION # = 5						
SUM DT**2 = 33.3969						
SUM (DF/F)**2 = 71.1272						
3 4.8819	3	1	5.1586	2	1	5.2222
ITERATION # = 6						
SUM DT**2 = 24.1685						
SUM (DF/F)**2 = 6.29141						
4 4.3304	5	1	1.3458	4	1	1.3477
ITERATION # = 7						
SUM DT**2 = 5.21764						
SUM (DF/F)**2 = 0.308386						

201.	202.	203.	204.	205.	206.	207.	208.	209.	210.	211.	212.	213.	214.	215.	216.	217.	218.	219.	220.	221.	222.	223.	224.	225.	226.	227.	228.	229.	230.	231.	232.	233.	234.	235.	236.	237.	238.	239.	240.	241.	242.	243.
FINAL COLUMN PROFILES:																																										
STAGE	TEMP	PRES	EFF(%)	LIIQUID(MOLS)	VAFOR(MOLS)																																					
1	351.38	760.040	100.00	123.351	741.948																																					
2	351.17	760.039	100.00	865.298	742.352																																					
3	350.54	760.037	100.00	865.703	744.017																																					
4	348.88	760.036	100.00	867.368	751.055																																					
5	345.68	760.035	100.00	874.406	771.617																																					
6	342.47	760.034	100.00	894.968	798.694																																					
7	341.19	760.032	100.00	922.045	813.114																																					
8	340.94	760.031	100.00	936.465	817.336																																					
9	340.90	760.030	100.00	940.686	818.347																																					
10	340.89	760.029	100.00	941.697	818.580																																					
11	340.88	760.028	100.00	941.930	818.633																																					
12	340.88	760.026	100.00	941.983	818.645																																					
13	340.88	760.025	100.00	941.995	818.647																																					
14	340.88	760.024	100.00	941.998	818.647																																					
15	340.88	760.023	100.00	941.998	818.646																																					
16	340.88	760.021	100.00	941.996	818.644																																					
17	340.88	760.020	100.00	941.994	818.642																																					
18	340.88	760.019	100.00	941.992	818.637																																					
19	340.88	760.018	100.00	941.988	818.630																																					
20	340.88	760.017	100.00	941.981	818.620																																					
21	340.88	760.015	100.00	941.971	818.604																																					
22	340.87	760.014	100.00	941.955	818.581																																					
23	340.87	760.013	100.00	941.931	818.544																																					
24	340.86	760.012	100.00	941.895	818.489																																					
25	340.85	760.010	100.00	941.839	818.404																																					
26	340.84	760.009	100.00	941.754	818.276																																					
27	340.81	760.008	100.00	941.626	818.082																																					
28	340.77	760.007	100.00	941.433	817.794																																					
29	340.72	760.006	100.00	941.145	817.372																																					
30	340.64	760.004	100.00	940.722	816.764																																					
31	340.52	760.003	100.00	940.115	815.920																																					
32	340.36	760.002	100.00	939.271	814.796																																					
33	340.15	760.001	100.00	938.147	813.389																																					
34	339.88	760.000	100.00	936.740	811.756																																					
35	339.57	759.998	100.00	935.107	810.033																																					
36	339.24	759.997	100.00	933.383	808.395																																					
37	338.91	759.996	100.00	931.746	787.088																																					
38	338.90	759.995	100.00	651.938	790.137																																					
39	338.92	759.993	100.00	654.988	790.397																																					
40	338.91	759.992	100.00	655.247	790.365																																					
41	338.89	759.991	100.00	655.215	727.000																																					

244.	FINAL VAPOR COMPOSITION PROFILE (MOLE%):			
245.	STAGE	BENZENE	ETHANOL	WATER
246.	1	0.48026	99.51971	0.00004
247.	2	1.56628	98.43370	0.00004
248.	3	4.78276	95.21722	0.00005
249.	4	13.08850	86.91148	0.00005
250.	5	28.33657	71.66340	0.00005
251.	6	43.41020	56.58976	0.00006
252.	7	50.66597	49.33396	0.00008
253.	8	52.80228	47.19760	0.00013
254.	9	53.32252	46.67729	0.00020
255.	10	53.44294	46.55676	0.00031
256.	11	53.47053	46.52899	0.00048
257.	12	53.47678	46.52247	0.00075
258.	13	53.47822	46.52064	0.00116
259.	14	53.47850	46.51973	0.00179
260.	15	53.47854	46.51871	0.00275
261.	16	53.47846	46.51733	0.00423
262.	17	53.47830	46.51522	0.00650
263.	18	53.47801	46.51203	0.00998
264.	19	53.47760	46.50709	0.01531
265.	20	53.47702	46.49950	0.02350
266.	21	53.47604	46.48792	0.03605
267.	22	53.47453	46.47020	0.05527
268.	23	53.47229	46.44303	0.08470
269.	24	53.46887	46.40143	0.12970
270.	25	53.46362	46.33802	0.19835
271.	26	53.45573	46.24148	0.30279
272.	27	53.44390	46.09520	0.46091
273.	28	53.42646	45.87491	0.69866
274.	29	53.40113	45.54649	1.05238
275.	30	53.36561	45.06393	1.57047
276.	31	53.31833	44.36955	2.31211
277.	32	53.25983	43.40056	3.33963
278.	33	53.19610	42.10366	4.70024
279.	34	53.14136	40.46119	6.39746
280.	35	53.11745	38.51946	8.36307
281.	36	53.14713	36.39991	10.45297
282.	37	53.24111	34.27495	12.48394
283.	38	54.84291	33.00118	12.15591
284.	39	55.00373	32.85826	12.13803
285.	40	55.01451	32.81120	12.17428
286.	41	54.99857	32.72499	12.27648

287.	FINAL LIQUID COMPOSITION PROFILE (MOLE%):			
288.	STAGE	BENZENE	ETHANOL	WATER
289.	1	0.13045	99.86953	0.00003
290.	2	0.43039	99.56958	0.00004
291.	3	1.36170	98.63829	0.00004
292.	4	4.12114	95.87884	0.00005
293.	5	11.26053	88.73945	0.00005
294.	6	24.44898	75.55099	0.00005
295.	7	37.62022	62.37972	0.00006
296.	8	44.00944	55.99051	0.00008
297.	9	45.89547	54.10442	0.00011
298.	10	46.35499	53.64485	0.00018
299.	11	46.46136	53.53838	0.00027
300.	12	46.48572	53.51385	0.00043
301.	13	46.49126	53.50809	0.00066
302.	14	46.49252	53.50647	0.00101
303.	15	46.49276	53.50568	0.00156
304.	16	46.49278	53.50483	0.00240
305.	17	46.49271	53.50363	0.00368
306.	18	46.49253	53.50181	0.00565
307.	19	46.49227	53.49907	0.00867
308.	20	46.49188	53.49482	0.01331
309.	21	46.49126	53.48831	0.02043
310.	22	46.49032	53.47837	0.03133
311.	23	46.48883	53.46313	0.04804
312.	24	46.48660	53.43978	0.07362
313.	25	46.48323	53.40406	0.11272
314.	26	46.47805	53.34958	0.17238
315.	27	46.47023	53.26665	0.26313
316.	28	46.45851	53.14097	0.40053
317.	29	46.44120	52.95170	0.60709
318.	30	46.41607	52.66953	0.91439
319.	31	46.38072	52.25488	1.36441
320.	32	46.33337	51.65815	2.00848
321.	33	46.27416	50.82532	2.90053
322.	34	46.20834	49.71035	4.08131
323.	35	46.14863	48.29781	5.55357
324.	36	46.11498	46.62715	7.25786
325.	37	46.12839	44.80247	9.06914
326.	38	64.30287	30.84096	4.85623
327.	39	66.18366	29.32035	4.49602
328.	40	66.37316	29.14938	4.47750
329.	41	66.38677	29.09241	4.52086

330.	FINAL	VAPOR COMPOSITION PROFILE (MOLES):		
331.	STAGE	BENZENE	ETHANOL	WATER
332.	1	3.563	738.384	0.000
333.	2	11.627	730.725	0.000
334.	3	35.585	708.433	0.000
335.	4	98.302	652.753	0.000
336.	5	218.650	552.967	0.000
337.	6	346.715	451.979	0.000
338.	7	411.972	401.141	0.001
339.	8	431.572	385.763	0.001
340.	9	436.363	381.982	0.002
341.	10	437.473	381.104	0.003
342.	11	437.727	380.902	0.004
343.	12	437.785	380.854	0.006
344.	13	437.798	380.840	0.009
345.	14	437.800	380.832	0.015
346.	15	437.800	380.823	0.023
347.	16	437.798	380.811	0.035
348.	17	437.796	380.793	0.053
349.	18	437.791	380.765	0.082
350.	19	437.784	380.721	0.125
351.	20	437.774	380.654	0.192
352.	21	437.757	380.552	0.295
353.	22	437.732	380.396	0.452
354.	23	437.694	380.157	0.693
355.	24	437.637	379.791	1.062
356.	25	437.548	379.232	1.623
357.	26	437.415	378.383	2.478
358.	27	437.215	377.097	3.771
359.	28	436.918	375.162	5.714
360.	29	436.486	372.284	8.602
361.	30	435.871	368.066	12.827
362.	31	435.035	362.020	18.865
363.	32	433.959	353.626	27.211
364.	33	432.691	342.467	38.231
365.	34	431.378	328.446	51.932
366.	35	430.269	312.020	67.744
367.	36	429.639	294.255	84.501
368.	37	419.054	269.774	98.260
369.	38	433.334	260.755	96.048
370.	39	434.748	259.711	95.939
371.	40	434.815	259.328	96.221
372.	41	399.839	237.911	89.250

373.	FINAL LIQUID COMPOSITION PROFILE (MOLES):			
374.	STAGE	BENZENE	ETHANOL	WATER
375.	1	0.161	123.190	0.000
376.	2	3.724	861.574	0.000
377.	3	11.788	853.915	0.000
378.	4	35.745	831.623	0.000
379.	5	98.463	775.943	0.000
380.	6	218.811	676.157	0.000
381.	7	346.875	575.169	0.001
382.	8	412.133	524.331	0.001
383.	9	431.732	508.953	0.001
384.	10	436.524	505.172	0.002
385.	11	437.634	504.294	0.003
386.	12	437.888	504.092	0.004
387.	13	437.946	504.044	0.006
388.	14	437.958	504.030	0.010
389.	15	437.961	504.022	0.015
390.	16	437.960	504.013	0.023
391.	17	437.959	504.001	0.035
392.	18	437.956	503.983	0.053
393.	19	437.951	503.955	0.082
394.	20	437.945	503.911	0.125
395.	21	437.934	503.844	0.192
396.	22	437.918	503.742	0.295
397.	23	437.893	503.586	0.453
398.	24	437.855	503.346	0.693
399.	25	437.797	502.980	1.062
400.	26	437.709	502.422	1.623
401.	27	437.576	501.573	2.478
402.	28	437.376	500.287	3.771
403.	29	437.079	498.352	5.714
404.	30	436.646	495.474	8.602
405.	31	436.032	491.256	12.827
406.	32	435.196	485.210	18.865
407.	33	434.120	476.816	27.211
408.	34	432.852	465.656	38.231
409.	35	431.539	451.636	51.932
410.	36	430.429	435.210	67.744
411.	37	429.799	417.445	84.501
412.	38	419.215	201.064	31.660
413.	39	433.495	192.045	29.448
414.	40	434.908	191.000	29.339

583. ***** TWO-LIQUID FLASH PROGRAM *****
 ?
 584.
 585.
 586.
 587. FD MOL =726.9995 MOLES PRESS = 760.00 MMHG TEMP :
 588. BENZENE ETHANOL WATER
 589. FEED MOLEFRACTIONS 0.54999 0.32725 0.12276
 ?

591. DEW POINT OF FEED AT 760.00 MMHG IS 338.89 DEG K
 ?

592. BENZENE ETHANOL WATER
 593. DEW PT LIQ MOLE FRAC 0.66365 0.29109 0.04525
 594.
 ?

595. 'VLE' DATA FOR VAPOR AND LIQUID PHASE # 1
 596.
 597. BENZENE ETHANOL WATER
 598. VAP MOLE FRAC 0.54999 0.32725 0.12276
 599. LIQ MOLE FRAC 0.66365 0.29109 0.04525
 600. VALUES OF K=Y/X 0.82872 1.12420 2.71290
 601. VAPOR PRESSURES 477.685 451.746 193.972
 602. FUGACITY FACTORS 1.01047 1.00544 0.98839
 603. ACTIVITY COEFFICIENTS 1.3048 1.8811 10.7543
 604. ACTIVITY 0.8660 0.5476 0.4867
 ?

625. BUBBLE POINT OF FEED AT 760.00 MMHG IS 337.88 DEG K
 ?
 624. TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO 0.423302
 ?

626. BENZENE ETHANOL WATER
 627. BUB PT VAP MOLE FRAC 0.54064 0.27305 0.18630
 628.
 ?

FEED AT OR BELOW BUBBLE POINT

VLE DATA FOR VAPOR AND LIQUID PHASE # 1

	BENZENE	ETHANOL	WATER
650.	VAP MOLE FRAC	0.54064	0.27305
651.	LIQ MOLE FRAC	0.77247	0.18574
652.	VALUES OF K=Y/X	0.83009	1.02347
653.	VAPOR PRESSURES	461.493	432.423
654.	FUGACITY FACTORS	1.01084	1.00496
655.	ACTIVITY COEFFICIENTS	1.1667	2.7235
656.		0.9013	0.5059
657.			0.7427

VLE DATA FOR VAPOR AND LIQUID PHASE # 2

	BENZENE	ETHANOL	WATER
662.	VAP MOLE FRAC	0.54064	0.27305
663.	LIQ MOLE FRAC	0.39766	0.42413
664.	VALUES OF K=Y/X	1.13670	0.73476
665.	VAPOR PRESSURES	461.493	432.423
666.	FUGACITY FACTORS	1.01084	1.00496
667.	ACTIVITY COEFFICIENTS	2.2664	1.1928
668.		0.9013	0.5059
669.			0.7427
670.	IDEAL LIQ ENTHALPY =	-0.138368E 04	DELTA H = 0.0
671.	IDEAL LIQ ENTHALPY =	-0.127464E 04	DELTA H = 0.173270E 03
672.	ENTHALPY OF SYSTEM WITH REFERENCE TO PURE LIQUID AT	0.0	DEG K
673.			
674.			
675.			

676.	CONV CHECK FOR WTR	76.9016	66.6000	22.6500	12.3459
677.	CONV CHECK FOR BEZN	171.760	0.0	400.000	0.0
678.	CONV CHECK FOR ETOH	306.213	191.900	169.200	54.8823

TRIAL # = 2

679.
680.
681.

F.2 Inputs

```

1. //FINAL2 JOB (JA00008,NJIT,15000,30,0),'RCOHE-TSAI-FINAL2',
1.01 // REGION=280K,TIME=(0,30)
1.02 //*PASSWORD SAI
1.03 //*MAIN CLASS=+2,REPORT=YES,FAILURE=RESTART
1.04 //*FORMAT PR,DDNAME=,DEST=RMT20,COPIES=1
7003. //STEP5 EXEC PGM=BAYWAY2,REGION=280K
7004. //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
7005. //FT06F001 DD SYSOUT=A
7006. //FT05F001 DD *
7007. COMPLETE AZEOTROPIC DIST. MODEL--BENZENE STRIPPING COLUMN-REB.ABS
7008. 3 1 2 2 0 0 1
7012. BENZENE 0 78.114 562.1 48.3 259.0 92.26
7013. 0.0 0.0 0.544 0.251
7014. 94.118 353.3 0.41 0.249
7015. 15.9008 -2788.51 -52.36
7016. ETHANOL 0 46.07 516.2 63.0 167.0 64.0
7017. 0.0878 0.0572 0.248 0.635
7018. 202.06 351.1 0.58 0.38
7019. 18.9119 -3803.98 -41.68
7020. WATER 0 18.01 647.3 217.3 56.0 18.84
7021. 0.0279 0.0229 0.229 0.344
7022. 538.7 373.2 1.0 0.454
7023. 18.3036 -3816.44 -46.13
7024. 0.0 0.15 0.40
7025. 0.15 0.0 0.20
7026. 0.40 0.20 0.0
7027. 1096.889 0.0 282.332 0.0 2213.335 0.0 3820.95
7028. -86.763 0.0 1289.908 0.0
7029. 0.295 0.267 0.270
7030. 5 0 5 1 0 020 2 0
7031. 1.0 0.0 0.0
7032. 413.0 0.0 1.0 339.0 354.0 760.0 0.001
7034. 5 308.0 0.0 184.5 206.9 39.6
7035. /*
7036. //

```

?
"RUN.STRIPPING" (RE)CATALOGED AND RESAVED ON WYL106

?
3443 IS YOUR JOB NUMBER 04/03/82 Saturday 9:26:33 P.m.

?

F.2 Results

68. COMPLETE AZEOTROPIC DIST. MODEL--BENZENE STRIPPING COLUMN-REB.ABS
 69. NUMBER OF COMPONENTS = 3
 70. VAPOR-LIQUID EQUILIBRIUM DATA CODES:
 71. LIQUID = 1
 72. VAPOR = 2
 73. ENTHALPY = 2
 74.
 75. NOTE: ALL FLOWS ARE PER UNIT TIME, AND ENERGY
 76. VALUES ARE IN UNITS OF K-BTU PER UNIT TIME.

77. COMPONENT NO	1	2	3
78. NAME	BENZENE	ETHANOL	WATER
79. MOLECULAR WEIGHT	78.1	46.1	18.0
80. CRITICAL PROPERTIES			
81. TEMPERATURE (DEG K)	562.1	516.2	647.3
82. PRESSURE (ATM)	48.3	63.0	217.3
83. VOLUME (CC/GMOL)	259.0	167.0	56.0
84. COMPRESSIBILITY	0.544	0.248	0.229
85. ACENTRIC FACTOR	0.251	0.635	0.344
86. TSONOPOULOS -A-	0.0	0.0878	0.0279
87. TSONOPOULOS -B-	0.0	0.0572	0.0229
88. AK(1,1)=	0.0		
89. AK(1,2)=	0.150		
90. AK(1,3)=	0.400		
91. AK(2,1)=	0.150		
92. AK(2,2)=	0.0		
93. AK(2,3)=	0.200		
94. AK(3,1)=	0.400		
95. AK(3,2)=	0.200		
96. AK(3,3)=	0.0		
97. VAPOR PRESSURE CONSTANTS			
98. (MMHG AND DEG K)			
99. C(1)	15.901	18.912	18.304
100. C(2)	-2788.51	-3803.98	-3816.44
101. C(3)	-52.36	-41.68	-46.13
102. C(4)	0.0	0.0	0.0
103. C(5)	0.0	0.0	0.0
104. C(6)	0.0	0.0	0.0
105. LIQUID MOLAR VOLUME CONSTANTS			
106. C(1)	92.260	64.000	18.840
107. ENTHALPY CONSTANTS (CAL/GM)			
108. LATENT HEAT	94.1	202.1	538.7
109. AT DEG K	353.3	351.1	373.2
110. LIQUID SP. HEAT CONSTANTS			
111. C(1)	0.410	0.580	1.000
112. C(2)	0.0	0.0	0.0
113. C(3)	0.0	0.0	0.0
114. C(4)	0.0	0.0	0.0
115. VAPOR SP. HEAT CONSTANTS			
116. C(1)	0.249	0.380	0.454
117. C(2)	0.0	0.0	0.0
118. C(3)	0.0	0.0	0.0
119. C(4)	0.0	0.0	0.0
120. NRTL CONSTANTS, CAL/GMOL			

COMPLETE AZEOTROPIC DIST. MODEL--BENZENE STRIPPING COLUMN-REB,ABS

126. REBOILED ABSORBER
 127. FEED STREAMS:
 128. STAGE
 129. TEMP
 130. FRACV

144
 129. 5.
 130. 308.00
 131. 0.0
 132. -5.5245E 05

H
 COMP 1 BENZENE
 COMP 2 ETHANOL
 COMP 3 WATER
 TOTAL
 NUMBER OF STAGES

133. 184.500
 134. 206.900
 135. 39.600
 136. 431.000
 137. 5
 138. 413.0000
 139. 17.9998
 140. 759.99902
 141. 0.00100
 142. 1.000000
 143.
 144.

ESTIMATED OVERHEAD PRODUCT RATE
 ESTIMATED BOTTOMS PRODUCT RATE
 PRESSURE AT TOP OF COLUMN
 CONSTANT DELTA P PER STAGE
 BOILUP RATIO

181. REBOILER DUTY = 1.671029E 05
 182. CONDENSER DUTY = -3.243844E 06
 183. PRODUCT STREAMS:
 184. STAGE
 185. TEMP
 186. FRACV
 H
 COMP 1 BENZENE
 COMP 2 ETHANOL
 COMP 3 WATER
 TOTAL

181. 1.
 182. 348.48
 183. 0.0
 184. -1.9991E 03
 185. 0.652
 186. 15.528
 187. 1.820
 188. 183.848
 189. 191.372
 190. 37.780
 191. 413.000

192.	FINAL COLUMN PROFILES:					
193.	STAGE	TEMP	PRES	EFF(%)	LIQUID(MOLS)	VAPOR(MOLS)
194.	1	348.48	760.004	100.00	18.000	18.000
195.	2	345.71	760.003	100.00	36.000	18.307
196.	3	343.24	760.001	100.00	36.307	18.651
197.	4	341.80	760.000	100.00	36.651	18.872
198.	5	341.15	759.999	100.00	36.872	413.000
199.	FINAL VAPOR COMPOSITION PROFILE (MOLE%):					
200.	STAGE	BENZENE	ETHANOL	WATER		
201.	1	13.25762	76.96699	9.77547		
202.	2	25.57612	65.21938	9.20457		
203.	3	35.98048	55.15580	8.86385		
204.	4	41.90915	49.17363	8.91733		
205.	5	44.51516	46.33718	9.14771		
206.	FINAL LIQUID COMPOSITION PROFILE (MOLE%):					
207.	STAGE	BENZENE	ETHANOL	WATER		
208.	1	3.62488	86.26432	10.11089		
209.	2	8.44121	81.61567	9.94318		
210.	3	14.69342	75.65270	9.65390		
211.	4	20.09009	70.43363	9.47630		
212.	5	23.21991	67.28013	9.50000		
213.	FINAL VAPOR COMPOSITION PROFILE (MOLES):					
214.	STAGE	BENZENE	ETHANOL	WATER		
215.	1	2.386	13.854	1.760		
216.	2	4.682	11.940	1.685		
217.	3	6.711	10.287	1.653		
218.	4	7.909	9.280	1.683		
219.	5	183.848	191.372	37.780		
220.	FINAL LIQUID COMPOSITION PROFILE (MOLES):					
221.	STAGE	BENZENE	ETHANOL	WATER		
222.	1	0.652	15.528	1.820		
223.	2	3.039	29.381	3.580		
224.	3	5.335	27.468	3.505		
225.	4	7.363	25.815	3.473		
226.	5	8.562	24.808	3.503		

228.	TEMP, DEG K	348.476			
229.	PRES, MMHG	760.0039			
230.	# NAME	K	VP	FUG	GAM
231.	1 BENZENE	3.6574	654.49	0.98302	4.3204
232.	2 ETHANOL	0.89222	673.74	1.0039	1.0026
233.	3 WATER	0.96682	293.19	0.98492	2.5445
234.	STAGE #	2			
235.	TEMP, DEG K	345.714			
236.	PRES, MMHG	760.0027			
237.	# NAME	K	VP	FUG	GAM
238.	1 BENZENE	3.0299	598.96	0.99165	3.8769
239.	2 ETHANOL	0.79910	601.97	1.0062	1.0027
240.	3 WATER	0.92571	260.98	0.98587	2.7344
241.	STAGE #	3			
242.	TEMP, DEG K	343.243			
243.	PRES, MMHG	760.0015			
244.	# NAME	K	VP	FUG	GAM
245.	1 BENZENE	2.4487	552.49	0.99872	3.3728
246.	2 ETHANOL	0.72906	543.30	1.0074	1.0124
247.	3 WATER	0.91815	234.75	0.98611	3.0144
248.	STAGE #	4			
249.	TEMP, DEG K	341.795			
250.	PRES, MMHG	760.0002			
251.	# NAME	K	VP	FUG	GAM
252.	1 BENZENE	2.0861	526.62	1.0026	3.0026
253.	2 ETHANOL	0.69815	511.23	1.0076	1.0300
254.	3 WATER	0.94101	220.44	0.98617	3.2898
255.	STAGE #	5			
256.	TEMP, DEG K	341.148			
257.	PRES, MMHG	759.9990			
258.	# NAME	K	VP	FUG	GAM
259.	1 BENZENE	1.9171	515.37	1.0043	2.8148
260.	2 ETHANOL	0.68872	497.42	1.0076	1.0444
261.	3 WATER	0.96291	214.28	0.98632	3.4626

F.3 Inputs

```

1. //FINAL2 JOB (JA00008,NJIT,15000,30,0),'RCOHE-TSAI-FINAL2',
1.01 // REGION=280K,TIME=(0,30)
1.02 //*PASSWORD SAI
1.03 //*MAIN CLASS=+2,REPORT=YES,FAILURE=RESTART
1.04 //*FORMAT PR,DDNAME=,DEST=RMT20,COPIES=1
7003. //STEP5 EXEC PGM=BAYWAY2,REGION=280K
7004. //STEPLIB DD DSN='NCE.AJA00.S008.TEST',UNIT=OLS,DISP=SHR
7005. //FT06F001 DD SYSOUT=A
7006. //FT05F001 DD *
7007. COMPLETE AZEOTROPIC DISTILLATION---- ETOH RECOVERY COLUMN
7008. 3 1 2 2 0 0 1
7012. BENZENE 0 78.114 562.1 48.3 259.0 92.26
7013. 0.0 0.0 0.544 0.251
7014. 94.118 353.3 0.41 0.249
7015. 15.9008 -2788.51 -52.36
7016. ETHANOL 0 46.07 516.2 63.0 167.0 64.0
7017. 0.0878 0.0572 0.248 0.635
7018. 202.06 351.1 0.58 0.38
7019. 18.9119 -3803.98 -41.68
7020. WATER 0 18.01 647.3 217.3 56.0 18.84
7021. 0.0279 0.0229 0.229 0.344
7022. 538.7 373.2 1.0 0.454
7023. 18.3036 -3816.44 -46.13
7024. 0.0 0.15 0.40
7025. 0.15 0.0 0.20
7026. 0.40 0.20 0.0
7027. 1096.889 0.0 282.332 0.0 2213.335 0.0 3820.95
7028. -86.763 0.0 1289.908 0.0
7029. 0.295 0.267 0.270
7030. 3 011 1 0 020 2 1
7031. 1.0 0.0 0.0
7032. 16.230 0.0 4.0 351.0 373.0 760.0 0.001
7034. 9 351.0 0.0 2.386 13.854 1.760
7035. /*
7036. //

```

F.3 Results

124. COMPLETE AZEOTROPIC DISTILLATION---- ETOH RECOVERY COLUMN
 125. DISTILLATION COLUMN; PRODUCT RATES & REFLUX RATIO SPECIFIED

126. FEED STREAMS:

127.	STAGE	9.	
128.	TEMP	351.00	
129.	FRACV	0.0	
130.	H	-9.1638E 02	
131.	COMP 1 BENZENE	2.386	2.386
132.	COMP 2 ETHANOL	13.854	13.854
133.	COMP 3 WATER	1.760	1.760
134.	TOTAL	18.000	18.000
135.	NUMBER OF STAGES	11	
136.			
137.	ESTIMATED OVERHEAD PRODUCT RATE		16.2300
138.	ESTIMATED BOTTOMS PRODUCT RATE		1.7700
139.			
140.	PRESSURE AT TOP OF COLUMN		759.99756
141.	CONSTANT DELTA P PER STAGE		0.00100
142.	REFLUX RATIO	4.000000	

?

171. REBOILER DUTY = 7.470238E 05
 172. CONDENSER DUTY = 7.503814E 05

173. PRODUCT STREAMS:

174.	STAGE	1.	11.
175.	TEMP	351.26	343.20
176.	FRACV	0.0	0.0
177.	H	-4.9485E 01	1.3762E 05
178.	COMP 1 BENZENE	0.000	2.386
179.	COMP 2 ETHANOL	1.627	12.227
180.	COMP 3 WATER	0.143	1.617
181.	TOTAL	1.770	16.230

?

182.	FINAL COLUMN PROFILES:					
183.	STAGE	TEMP	PRES	EFF(%)	LIQUID(MOLS)	VAPOR(MOLS)
184.	1	351.26	760.010	100.00	1.770	79.816
185.	2	351.25	760.009	100.00	81.586	79.796
186.	3	351.25	760.007	100.00	81.566	79.778
187.	4	351.24	760.006	100.00	81.548	79.761
188.	5	351.24	760.005	100.00	81.531	79.746
189.	6	351.22	760.004	100.00	81.516	79.732
190.	7	351.16	760.002	100.00	81.502	79.723
191.	8	350.92	760.001	100.00	81.493	79.733
192.	9	349.96	760.000	100.00	81.503	80.370
193.	10	348.16	759.999	100.00	64.140	81.150
194.	11	343.20	759.998	100.00	64.920	0.0
195.	FINAL VAPOR COMPOSITION PROFILE (MOLE%):					
196.	STAGE	BENZENE	ETHANOL	WATER		
197.	1	0.00008	91.53092	8.46901		
198.	2	0.00032	91.14850	8.85119		
199.	3	0.00130	90.79516	9.20357		
200.	4	0.00536	90.46901	9.52563		
201.	5	0.02221	90.16066	9.81714		
202.	6	0.09242	89.83108	10.07650		
203.	7	0.38480	89.32071	10.29451		
204.	8	1.58694	87.98400	10.42907		
205.	9	6.22384	83.45498	10.32118		
206.	10	14.70113	75.33339	9.96550		
207.	11	0.0	0.0	0.0		
208.	FINAL LIQUID COMPOSITION PROFILE (MOLE%):					
209.	STAGE	BENZENE	ETHANOL	WATER		
210.	1	0.00002	91.94177	8.05826		
211.	2	0.00008	91.53984	8.46010		
212.	3	0.00031	91.16572	8.83399		
213.	4	0.00127	90.82003	9.17870		
214.	5	0.00524	90.50099	9.49378		
215.	6	0.02173	90.19933	9.77895		
216.	7	0.09041	89.87693	10.03266		
217.	8	0.37645	89.37765	10.24592		
218.	9	1.55248	88.06999	10.37756		
219.	10	4.07876	85.51009	10.41119		
220.	11	14.70113	75.33339	9.96550		

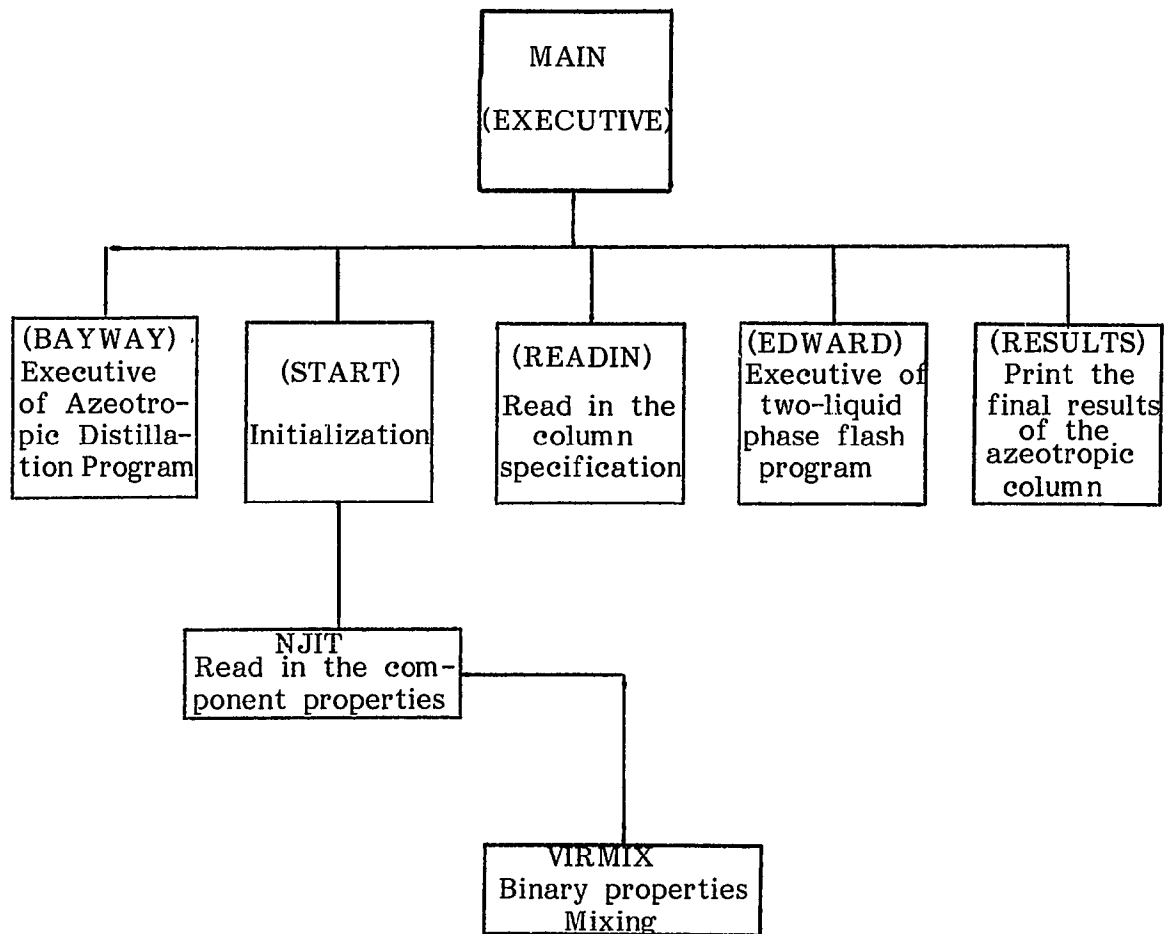
221.	FINAL VAPOR COMPOSITION PROFILE (MOLES):			
222.	STAGE	BENZENE	ETHANOL	WATER
223.	1	0.000	73.057	6.760
224.	2	0.000	72.733	7.063
225.	3	0.001	72.434	7.342
226.	4	0.004	72.159	7.598
227.	5	0.018	71.899	7.829
228.	6	0.074	71.624	8.034
229.	7	0.307	71.209	8.207
230.	8	1.265	70.152	8.315
231.	9	5.002	67.073	8.295
232.	10	11.930	61.133	8.087
233.	11	0.0	0.0	0.0
234.	FINAL LIQUID COMPOSITION PROFILE (MOLES):			
235.	STAGE	BENZENE	ETHANOL	WATER
236.	1	0.000	1.627	0.143
237.	2	0.000	74.684	6.902
238.	3	0.000	74.360	7.206
239.	4	0.001	74.062	7.485
240.	5	0.004	73.786	7.740
241.	6	0.018	73.527	7.971
242.	7	0.074	73.252	8.177
243.	8	0.307	72.836	8.350
244.	9	1.265	71.780	8.458
245.	10	2.616	54.847	6.678
246.	11	9.544	48.906	6.470

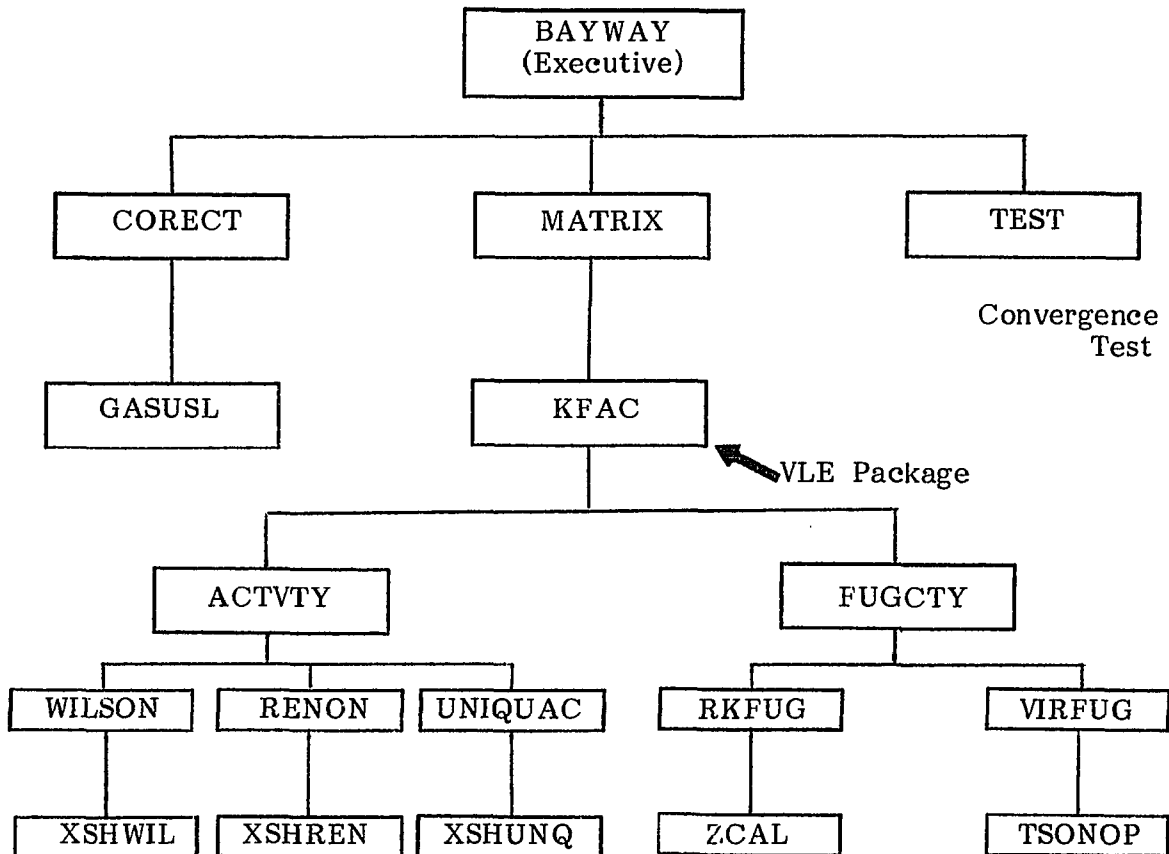
247.	STAGE #	1				
248.	TEMP, DEG K		351.258			
249.	PRES, MMHG		760.0098			
250.	# NAME		K	VP	FUG	GAM
251.	1 BENZENE		4.1447	714.43	0.97322	4.5304
252.	2 ETHANOL		0.99553	753.14	1.0007	1.0039
253.	3 WATER		1.0510	328.95	0.98144	2.4741
254.	STAGE #	2				
255.	TEMP, DEG K		351.253			
256.	PRES, MMHG		760.0085			
257.	# NAME		K	VP	FUG	GAM
258.	1 BENZENE		4.1702	714.32	0.97327	4.5588
259.	2 ETHANOL		0.99572	752.99	1.0006	1.0044
260.	3 WATER		1.0462	328.88	0.98177	2.4626
261.	STAGE #	3				
262.	TEMP, DEG K		351.248			
263.	PRES, MMHG		760.0073			
264.	# NAME		K	VP	FUG	GAM
265.	1 BENZENE		4.1942	714.22	0.97331	4.5855
266.	2 ETHANOL		0.99593	752.85	1.0006	1.0048
267.	3 WATER		1.0418	328.82	0.98206	2.4520
268.	STAGE #	4				
269.	TEMP, DEG K		351.244			
270.	PRES, MMHG		760.0061			
271.	# NAME		K	VP	FUG	GAM
272.	1 BENZENE		4.2165	714.12	0.97336	4.6103
273.	2 ETHANOL		0.99613	752.72	1.0006	1.0052
274.	3 WATER		1.0378	328.76	0.98233	2.4422
275.	STAGE #	5				
276.	TEMP, DEG K		351.238			
277.	PRES, MMHG		760.0049			
278.	# NAME		K	VP	FUG	GAM
279.	1 BENZENE		4.2366	713.98	0.97340	4.6329
280.	2 ETHANOL		0.99624	752.53	1.0006	1.0055
281.	3 WATER		1.0341	328.68	0.98258	2.4335

281.	3 WATER	1.0341	328.68	0.98258	2.4335
282.	STAGE # 6				
283.	TEMP, DEG K	351.221			
284.	PRES, MMHG	760.0037			
285.	# NAME	K	VP	FUG	GAM
286.	1 BENZENE	4.2524	713.62	0.97349	4.6522
287.	2 ETHANOL	0.99592	752.05	1.0006	1.0059
288.	3 WATER	1.0304	328.46	0.98281	2.4260
289.	STAGE # 7				
290.	TEMP, DEG K	351.160			
291.	PRES, MMHG	760.0024			
292.	# NAME	K	VP	FUG	GAM
293.	1 BENZENE	4.2562	712.26	0.97373	4.6640
294.	2 ETHANOL	0.99380	750.23	1.0006	1.0061
295.	3 WATER	1.0261	327.64	0.98305	2.4212
296.	STAGE # 8				
297.	TEMP, DEG K	350.917			
298.	PRES, MMHG	760.0012			
299.	# NAME	K	VP	FUG	GAM
300.	1 BENZENE	4.2156	706.86	0.97463	4.6505
301.	2 ETHANOL	0.98441	743.00	1.0010	1.0060
302.	3 WATER	1.0179	324.38	0.98341	2.4251
303.	STAGE # 9				
304.	TEMP, DEG K	349.965			
305.	PRES, MMHG	760.0000			
306.	# NAME	K	VP	FUG	GAM
307.	1 BENZENE	4.0090	686.05	0.97800	4.5410
308.	2 ETHANOL	0.94759	715.30	1.0022	1.0046
309.	3 WATER	0.99456	311.90	0.98422	2.4623
310.	STAGE # 10				
311.	TEMP, DEG K	348.156			
312.	PRES, MMHG	759.9988			
313.	# NAME	K	VP	FUG	GAM
314.	1 BENZENE	3.6043	647.86	0.98407	4.2966
315.	2 ETHANOL	0.88099	665.08	1.0042	1.0025
316.	3 WATER	0.95719	289.30	0.98530	2.5521
317.	STAGE # 11				
318.	TEMP, DEG K	343.201			
319.	PRES, MMHG	759.9976			
320.	# NAME	K	VP	FUG	GAM
321.	1 BENZENE	2.4545	551.72	0.99882	3.3851
322.	2 ETHANOL	0.72763	542.35	1.0074	1.0122
323.	3 WATER	0.91318	234.32	0.98634	3.0029

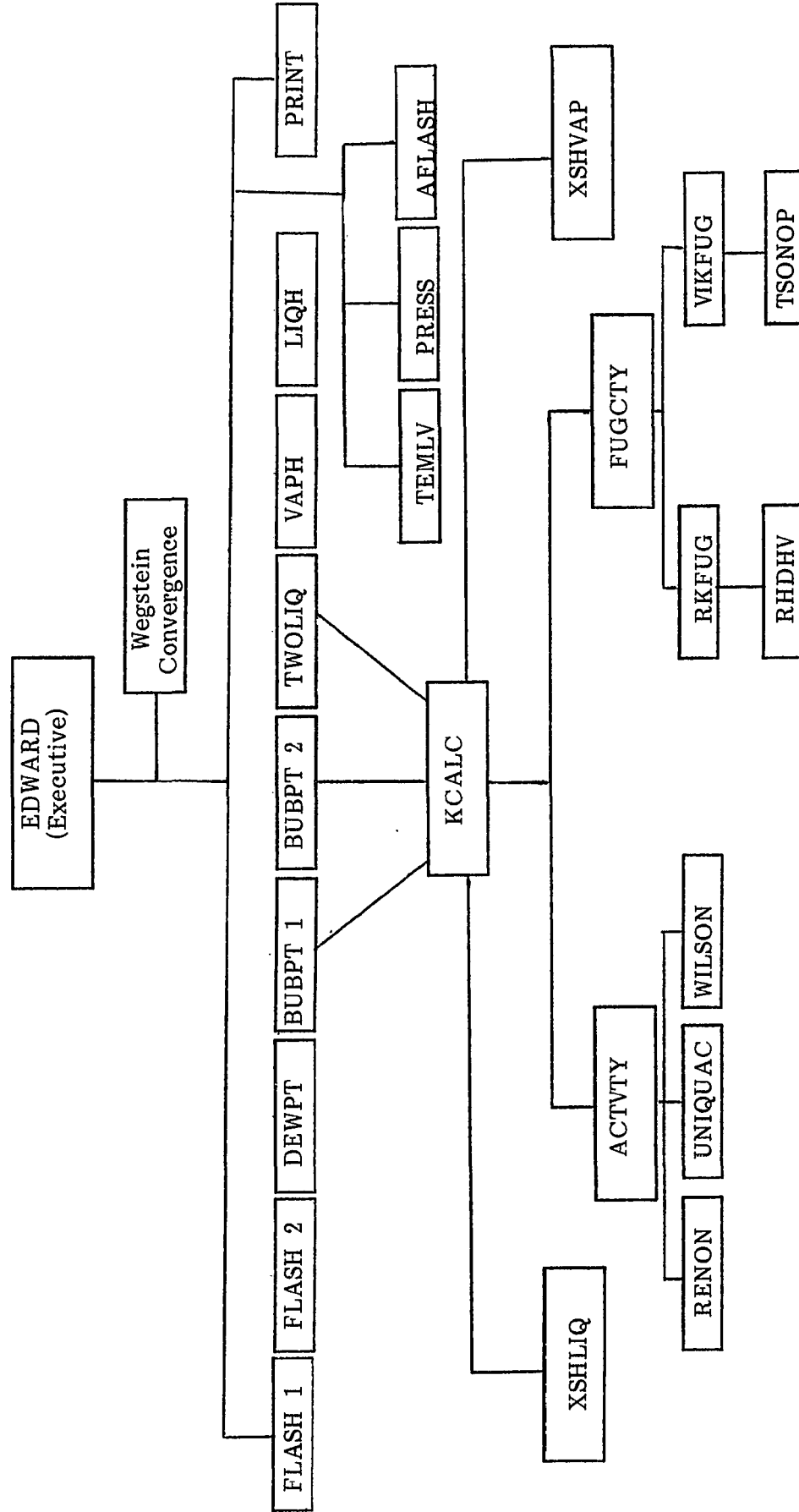
APPENDIX G

COMPUTER PROGRAMS

Structure of Computer Programs1. The Complete Azeotropic Distillation Program

2. The Azeotropic Distillation Program

3. The Two-Liquid Phase Flash Program



ORTRAN IV G LEVEL 21 MAIN DATE = 82072 14/25/17

```

C
C *****
C
C THIS IS THE EXECUTIVE PROGRAM WHICH LINKS THE AZEOTROPIC
C DISTILLATION MODEL (BAYWAY) WITH THE TWO-LIQUID PHASE
C FLASH MODEL (EDWARD).
C
C OLD MAIN IN EDWAARD BECOMES SUBROUTINE EDWARD
C THE MAIN IN BAYWAY IS REPLACED BY SUBROUTINES BAYWAY, START, TEST
C
C MARCH 10, 1982----- EDWARD C ROCHE, JR. / CHARLES M. TSAI
C *****
C
001      IMPLICIT REAL      (M)
002      INTEGER            COMPNT
003      REAL               LIQUID
C
004      COMMON /B2L/       FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
005      COMMON /CONTRL/    TDUM,P,NOBUR,NODEW,ISO,NLIQ
006      COMMON /LIQUID/    X(10,2),GAM(10,2),DHL(2),HIDL(2),HREAL(2),
+                          FRACL
007      COMMON /CTRL/     NCR, NPRT, NSTART, TITLE(20)
008      COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
+                          *
+                          NAME(2,10), L(10), NDIM, COMPNT(10)
009      COMMON /NJIT1/    CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
+                          *
+                          W(10)
010      COMMON /NJIT2/    C(180), ALPHA(45), VC(10), VOL(10), TC(10),
+                          *
+                          PC(10), OA(10), OR(10), AA(10,10), G(10,10),
+                          *
+                          THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
+                          *
+                          Q(10),R(10),XL(10),PHI(10),DUMMY(10)
011      COMMON /NJIT3/    YTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
+                          *
+                          AK(10,10), AVALUE(10,10), BVALUE(10,10),
+                          *
+                          BCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)
012      COMMON /INR0/     NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
+                          *
+                          WFEED, LF(9), ICODE, ITR, IIRMAX, ICODE, KOUNT
013      COMMON /INR1/     FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
+                          *
+                          DEST, DESTV, DESTL, BTMS, RFLX, ROIL, PP(50), T(50),
+                          *
+                          MU(50), SL(50), SV(50), FL(50), FLL(50,10),
+                          *
+                          FV(50), FVV(50,10), ZCONVT, ZCONVF
014      COMMON /INR2/     DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
+                          *
+                          DEXH(50,11), HV(50), HVV(50,10), CV(50)
015      COMMON /INR3/     AMAT(21,21), BMAT(50,21,21), D(50,21)
016      COMMON /INR4/     VAPOR, LIQUID, FEED(11)
017      COMMON /KNST/     XFKNST(10)
C
018      DATA              SMALL /0.00005/
019      KOUNT=0
C
020      NCR=5
021      NPRT=6
C
022      CALL START
C
023      25 CALL READIN
C

```

```

C
C
C   ICODE IS THE OVERALL CONVERGENCE COUNTER
C
0024 C   IF (ICODE.EQ.1) GO TO 170
C
C   KOUNT IS THE CONVERGENCE COUNTER FOR TWO LIQUID PHASES
C   SEPARATION ROUTINE, THE MAX ITERATION NO IS SET AT 20
C
0025 C   30 KOUNT=KOUNT+1
0026 C   WRITE(6,31) KOUNT
0027 C   31 FORMAT ('1 TRIAL # = ',I5//1X,100('*')//)
C
C   CALL THE AZEOTROPIC DISTILLATION COLUMN SIMULATION PROGRAM
C
0028 C   CALL BAYWAY
C
C
C   SAVE THE COLUMN SIMULATION RESULTS ,OVERHEAD DISTILLATE
C   RATES INTO THE FEED STREAM OF THE TWO-LIQ PHASES SEPARATION
C   ROUTINE
C
C
0029 C   FD(NC+1)=0
0030 C   DO 120 I=1,NC
0031 C   FD(I)=FVV(NST,I)
0032 C   120 FD(NC+1) = FD(NC+1) + FD(I)
C
C   SAVE THE ENTRAINER FEED TO THE COLUMN FOR EACH COMPONENT
C
C
0033 C   WSAVE=FSTR(1,1)
0034 C   ESAVE=FSTR(1,2)
0035 C   WSAVE=FSTR(1,3)
C
C   CALL THE TWO-LIQUID SEPARATION PROGRAM FOR PHASE SPLITTING
C
C
0036 C   CALL EDWARD
C
C
C   BEFORE CALLING SUB RESULT CHECK THE COMPOSITION IN XSAVE
C   VS. TWOLIQ'S FOR THE FOLLOWINGS
C   1A. WATER BALANCE..... WATER RATE IN THE DRUM STREAM(AQUEOUS)
C   + WATER IN ETOH PRODUCT(HUTMS) MUST EQ.
C   TO WATER RATE IN THE COLUMN FEED
C
C   1B..... WATER IN ORGANIC PHASE FROM DRUM = WATER IN ENTRAINER
C
C   NET WATER BALANCE CHECK
0037 C   WOUT=FLL(1,3) + X(3,2)*(1-FRACL)*FD(NC+1)
0038 C   WIN= FSTR(2,3)
C
C   CHECK WATER IN THE ENTRAINER
C   WSAVE= WATER IN ENTRAINER
0039 C   WCALC= X(3,1)*FRACL*FD(NC+1)

```

```

ORTRAN IV G LEVEL 21          MAIN          DATE = 82072          14/25/17

C
0040  WRITE(6,130) WOUT,WIN,WSAVE,WCALC
C      2  ETOH BALANCE ... ETOH IN BOTMS + ETOH IN PURGE(AQUEOUS)
C      MUST BE CHECKED AGAINST ETOH IN THE
C      FRESH FEED
C      2B.....ETOH IN ORGANIC PHASE FROM DRUM =WATER IN ENTRAINER
C
C      NET ETOH BALANCE
C
0041  EOUT=ELL(1,2)+X(2,2)*(1-FRACL)*FD(NC+1)
0042  EIN=FSTR(2,2)
C
C      CHECK ETOH IN THE ENTRAINER
C
C      ESAVE = ETOH IN THE ENTRAINER
0043  ECALC = X(2,1)*FRACL*FD(NC+1)
C      3  BENZENE BALANCE... BZ IN FEED #1(FROM COLUMN) MUST BE CHECKED
C      WITH BZ IN BOTMS(SMALL AMT) ,PURGE, THE SUM
C      BOTM BZ AND PURGE BZ IS THE REQUIRED MAKE UP
C
C      NET BENZENE BALANCE
C
0044  BOUT= FLL(1,1) + X(1,2)*(1-FRACL)*FD(NC+1)
0045  BIN = 0.0
0046  WRITE(6,132) BOUT,BIN,BSAVE,BCALC
C
C      CHECK BENZENE IN THE ENTRAINER---BSAVE
C
0047  BCALC= X(1,1)*FRACL*FD(NC+1) + BOUT
C
C      UPDATE THE ENTRAINER FEED RATE VIA DIRECT SUBSTITUTION
C
0048  FSTR(1,1) = BCALC
0049  FSTR(1,2) = ECALC
0050  FSTR(1,3) = WCALC
C
C      BY-PASS THE MAX ITER TENTATIVELY AND PRINT THE CONV CHECKS
C
0051  WRITE(6,131) EOUT,EIN,ESAVE,ECALC
C      CALL RESULT
C
0052  130  FORMAT(' CONV CHECK FOR WTR ',2X,4G15.6)
0053  131  FORMAT(' CONV CHECK FOR ETOH',2X,4G15.6)
0054  132  FORMAT(' CONV CHECK FOR BEZN ',2X,4G15.6)
0055  IF(KOUNT.LE.5) GO TO 30
C      160 CALL RESULT
C
0056  170 CALL EXIT
0057  STOP
0058  END

```


JRTRAN IV G LEVEL 21

RAYWAY

DATE = 82075

22/247

```

0001 SUBROUTINE RAYWAY
      C PROGRAM MAIN:RAYWAY AS MEMBER OF NCF.AJAOO.S008.TEST
      C REBOILER DUTY CALCULATION MODIFIED 3-28-81
      C WITH 'SMALL' IN KFAC CHANGED TO 0.0
      C AS OF APRIL 5,1981(NO UNIQUAC MODIFICATION)
      C REBOILED ABSORBER OPTION IMPLEMENTED NCASE=5
      C
      C READIN MODIFIED ON DEC. 17,1981
      C
      C MAIN PROGRAM MODIFIED TO MAIN PLUS START,TEST SUBROUTINES
      C ON DEC. 19,1981
      C
0002 IMPLICIT REAL (M)
0003 INTEGER COMPNT
0004 REAL LIQUID
      C
0005 COMMON /B2L/ FD(11),7(10),FK(10,2),PF,TF,HI,XSAVE(10,2)
0006 COMMON /CTRL/ NCR, NPRT, NSTART, TITLE(20)
0007 COMMON /NJIT0/ NC, NCM, IDLL, IDLV, IDH, LDRUG, ISW,
      * NAME(2,10), L(10), NDIM, COMPNT(10)
0008 COMMON /NJIT1/ CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
      * W(10)
0009 COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
      * PC(10), UA(10), OB(10), AA(10,10), G(10,10),
      * THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
      * Q(10),K(10),XL(10),PHI(10),DUMMY(10)
0010 COMMON /NJIT3/ TIC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
      * AK(10,10), AVALUE(10,10), BVALUE(10,10),
      * HCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)
0011 COMMON /INR0/ NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
      * NFEED, LF(9), ICODE, ITR, ITRMAX, ICODE
0012 COMMON /INR1/ FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
      * DEST,DESTV,DESTL, BTMS, RFLX,ROIL, PP(50), T(50),
      * NH(50), SL(50), SV(50), FL(50), FLL(50,10),
      * FV(50), FVV(50,10), ZCONVT, ZCONVF
0013 COMMON /INR2/ DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
      * DEXH(50,11), HV(50), HVV(50,10), CV(50)
0014 COMMON /INR3/ AMAT(21,21), BMAT(50,21,21), D(50,21)
0015 COMMON /INR4/ VAPOR, LIQUID, FEED(11)
0016 COMMON /VLF/ XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
0017 COMMON /KNST/ XFKNST(10)
      C
0018 DATA SMALL /0.00005/
      C
0019 NCR=5
      C
0020 ITR=0
      C
0021 30 CALL MATRIX
0022 CALL CORFCT
0023 CALL TEST
0024 IF (ICODE) 30,160,40
      C
0025 160 CALL RESULT
0026 GO TO 170
0027 40 WRITE(5,41)

```

TRAN IV G LEVEL 21

BAYWAY

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```
0028 ----- 41 FORMAT('IDCOE=1, TRIAL LIMIT REACHED PROBLEM NOT CONVERGED')
0029          CALL RESULT
          C
0030 ----- 170 RETURN
0031          END
```

ORTRAN IV G LEVEL 21

START

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```

0001-----SUBROUTINE START-----
0002      IMPLICIT REAL      (*)
0003      INTEGER            COMPNT
0004-----REAL            LIQUID-----
      C
0005      COMMON /B2L/      FD(11),7(10),FK(10,2),PF,TF,HI,XSAVE(10,2)
0006-----COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)-----
0007      COMMON /CONTRL/    TDUM,P,NORUN,NOREW,ISO,NLIW
0008-----COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
      *                     NAME(2,10), L(10), NDIM, COMPNT(10)-----
0009      COMMON /NJIT1/    CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
      *                     W(10)-----
0010-----COMMON /NJIT2/    C(180), ALPHA(45), VC(10), VOL(10), TC(10),
      *                     PC(10), OA(10), OB(10), AA(10,10), G(10,10),
      *                     THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
      *                     G(10),R(10),XL(10),PHI(10),DUMMY(10)-----
0011      COMMON /NJIT3/    TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
      *                     AK(10,10), AVALUE(10,10), BVALUE(10,10),
      *                     BCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)-----
0012-----COMMON /INR0/    NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
      *                     KFEED, LF(9), IDCODE, ITR, ITRMAX, ICODE-----
0013-----COMMON /INR1/    ESTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
      *                     DFST,DESTV,DESTL, BTMS, RFLX,BOIL, PP(50), T(50)
      *                     MH(50), SL(50), SV(50), FL(50), FLL(50,10),
      *                     FV(50), FVV(50,10), ZCONVT, ZCONVF-----
0014-----COMMON /INR2/    OFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
      *                     OFXH(50,11), HV(50), HVV(50,10), CV(50)-----
0015-----COMMON /INR3/    AMAT(21,21), BMAT(50,21,21), D(50,21)-----
0016-----COMMON /INR4/    VAPOR, LIQUID, FEED(11)-----
0017-----COMMON /VLE/    XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)-----
0018-----COMMON /KNST/    XFKNST(10)-----
      L
0019      DATA              S IALL /0.00005/
      C
0020      ICODE=0
0021      NST=0
      C
0022      WRITE (NPRT,5)
0023      5 FORMAT ('1*** AZENTROPIC DISTILLATION PROGRAM ***'/)
      C
0024      READ (NCR,10) TITLE
0025      10 FORMAT (20A4)
0026      WRITE (NPRT,11) TITLE
0027      11 FORMAT ('0',19A4,A3)
      C
      C
      C
      C      BASIC CONTROL PARAMETERS
      C
      C      NC          NUMBER OF COMPONENTS
      C      IDLL       LIQUID PHASE CLASSIFICATION CODE
      C                0  IDEAL
      C                1  NRTL
      C                2  WILSON
      C                3  UNIQUAC
      C
      C      IDLV       VAPOR PHASE CLASSIFICATION CODE

```

```

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-----
C          0      IDEAL
C          1      REDLICH-KWONG
C          2      VIRIAL-TSONOPOULOS'
C          IDH      ENTHALPY CORRECTION
C          0      CORRECT BOTH PHASES
C          1      CORRECT VAPOR PHASE ONLY
C          2      NO CORRECTION
C          LDRUG      DEBUG CODE
C          0      NO INTERMEDIATE PRINT
C          1      SUPERFICIAL TRACE
C          2      DETAILED TRACE
C
0028      READ (NCR,12) NC, IDLL, IDLV, IDH, LDRUG, NDIM
0029      12 FORMAT (20I2)
0030      NCM1=NC-1
0031      NDIR=0
0032      IF (IDLL.LT.0 .OR. IDLL.GT.4) IDLL=0
0033      IF (IDLV.LT.0 .OR. IDLV.GT.4) IDLV=0
0034      IF (IDLL.EQ.4 .AND. IDH.EQ.0) IDH=1
C
0035      WRITE (NPRT,13) NC
0036      13 FORMAT ('NUMBER OF COMPONENTS =', I3)
0037      WRITE (NPRT,14) IDLL, IDLV, IDH
0038      14 FORMAT ('OVAPOR-LIQUID EQUILIBRIUM DATA CODES: '/
*          5X, 'LIQUID =', I4 /
*          5X, 'VAPOR =', I5 /
*          5X, 'ENTHALPY =', I2 /)
C
C
C      READ IN THE PROGRAM CONTROL PARAMETERS, CALL SUBROUTINE
C      NJIT FOR COMPONENT PROPERTIES, AND NLIQ,TF,PF FOR TWO-LIQ
C      PROGRAM, ALSO THE INITIAL COMPOSITIONS IN EACH PHASE(XSAVE)
C
C
C      READ IN AZEOTROPIC DATA FOR REFLUX/PRODUCT SEPARATOR
C
C      TWO LIQUID PHASES :  1= ORGANIC RICH (REFLUX)
C                          2= AQUEOUS RICH (PURGE PRODUCT)
C
0039      READ (5, 8) NLIQ,TF,PF
0040      8  FORMAT(I2,5X,6F10.5)
C
C      COMPONENTIAL FLOW RATE IIN MOULES/HR FD(I)-- TO BE ESTABLISHED
C      BY DISTILLATION ALGORITHM , (FVV(NST,*),*=COMPONENT # )
C
C
C      READ IN THE INITIAL PHASE COMPOSITIONS 1=B7,2=ETOH,3=WTR
C      THEN NORMALIZE THE XSAVES
C
C
0041      READ (5,9) (XSAVE(I,1),I=1,NC)
0042      9  FORMAT(8F10.4)
0043      READ (5,9) (XSAVE(I,2),I=1,NC)
C

```

```
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0044          CALL NJIT
          C
0045          20 WRITE (NPRT,21) TITLE
0046          21 FORMAT ('1',19A4,A3)
0047          RETURN
0048          END
```

ORTRAM IV G LEVEL 21

TEST

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```

0001      SUBROUTINE TEST
0002      IMPLICIT REAL      (M)
0003      INTEGER            COMPNT
0004      REAL               LIQUID
          C
0005      COMMON /CTRL/     NCR, NPRI, NSTART, TITLE(20)
0006      COMMON /B2L/      FD(11),7(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
0007      COMMON /CONTRL/   TDUM,P,NORUB,NODEF,ISO,MLIQ
0008      COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
          *                NAME(2,10), L(10), NDIM, COMPNT(10)
0009      COMMON /NJIT1/    CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
          *                N(10)
0010      COMMON /NJIT2/    C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *                PC(10), QA(10), OH(10), AA(10,10), G(10,10),
          *                THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
          *                W(10),R(10),XL(10),PHI(10),DUMMY(10)
0011      COMMON /NJIT3/    ITC(10,10),PPC(10,10),VVC(10,10),ZXC(10,10),
          *                AK(10,10), AVALUE(10,10), BVALUE(10,10),
          *                BCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)
0012      COMMON /INR0/     NS1, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
          *                BFEEED, LF(9), IDCODE, ITR, ITRMAX, ICODE
0013      COMMON /INR1/     FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
          *                DEST,DESTV,DESTL, BTMS, RFLX,BOIL, PP(50), T(50)
          *                MU(50), SL(50), SV(50), FL(50), FLL(50,10),
          *                FV(50), FVV(50,10), ZCONVT, ZCONVF
0014      COMMON /IDR2/     DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
          *                DFXH(50,11), HV(50), HVV(50,10), CV(50)
0015      COMMON /INR3/     AMAT(21,21), BMAT(50,21,21), D(50,21)
0016      COMMON /INR4/     VAPOR, LIQUID, FEED(11)
0017      COMMON /VLE/     YFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
0018      COMMON /KNST/    XFKNST(10)
          C
0019      DATA             SMALL /0.00005/
          C
0020      ITR=ITR+1
          C
0021      WRITE (NPRT,75) ITR
0022      75 FORMAT (' ITERATION # =', I4)
          C
0023      IF (LDBUG.EQ.0) GO TO 110
0024      WRITE (NPRT,80)
0025      80 FORMAT (' STAGE      TEMP      PRES      LIQUID(MOLS)      VAPOR(MOL
          * )')
0026      DO 90 I=1,NST
0027      GO TO (81,85,85), KUNITS
0028      81 TDEG=(T(I)-273.16)*1.8+32.0
0029      PRES=PP(I)*14.696
0030      FLZ=FL(I)/453.59
0031      FVZ=FV(I)/453.59
0032      GO TO 90
0033      85 TDEG=T(I)
0034      PRES=PP(I)
0035      FLZ=FL(I)
0036      FVZ=FV(I)
0037      90 WRITE (NPRT,100) I,TDEG,PRES,FLZ,FVZ
0038      100 FORMAT (I5,F10.2,F12.3,F15.3,F14.3)

```

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C

C

SUM OF SQUARES OF DISCREPANCY FUNCTIONS

C

```

0039      110 REST=0.0
0040          RESF=0.0
0041          DTBIG=0.0
0042          ITBIG=0
0043          DLBIG=0.0
0044          ILBIG=0
0045          JLBIG=0
0046          DVBIG=0.0
0047          IVBIG=0
0048          JVBIG=0
0049          DO 120 I=1,NST
0050          IF (ABS(D(I,NK1)).LT.DTBIG) GO TO 112
0051          DTBIG=ABS(D(I,NK1))
0052          ITBIG=I
0053      112 REST=REST+D(I,NK1)**2
0054          DO 120 J=1,NC
0055          IF (D(I,J).EQ.0.0) GO TO 115
0056          IF (FLL(I,J)/FL(I).LT.SMALL) GO TO 115
0057          IF (ABS(D(I,J))/FLL(I,J).LT.DLBIG) GO TO 113
0058          DLBIG=ABS(D(I,J))/FLL(I,J)
0059          ILBIG=I
0060          JLBIG=J
0061      113 RESF=RESF+(D(I,J)/FLL(I,J))**2
0062      115 IF (D(I,J+NK1).EQ.0.0) GO TO 120
0063          IF (FVV(I,J)/FV(I).LT.SMALL) GO TO 120
0064          IF (ABS(D(I,J+NK1))/FVV(I,J).LT.OVBIG) GO TO 118
0065          OVBIG=ABS(D(I,J+NK1))/FVV(I,J)
0066          IVBIG=I
0067          JVBIG=J
0068      118 RESF=RESF+(D(I,J+NK1)/FVV(I,J))**2
0069      120 CONTINUE
C
0070          WRITE (NPRT,130) REST,RESF
0071      130 FORMAT (' SUM D1**2 =',      G17.6, /
*              ' SUM (DF/F)**2 =', G13.6)
0072          WRITE (NPRT,135) ITBIG,DTBIG,ILBIG,JLBIG,DLBIG,IVBIG,JVBIG,OVBIG
0073      135 FORMAT (5X,I5,G12.5,5X,215,G12.5,5X,215,G12.5)
C
0074          IF (ITR.EQ.ITRMAX) GO TO 160
C
0075          DO 150 I=1,NST
0076          IF (ABS(D(I,NK1)).GT.ZCONVT) GO TO 160
0077          DO 150 J=1,NC
0078          IF (FLL(I,J)/FL(I).LT.SMALL) GO TO 140
0079          IF (ABS(D(I,J))/FLL(I,J).GT.ZCONVF) GO TO 160
0080      140 IF (FVV(I,J)/FV(I).LT.SMALL) GO TO 150
0081          IF (ABS(D(I,J+NK1))/FVV(I,J).GT.ZCONVF) GO TO 160
0082          150 CONTINUE
0083          GO TO 190
C
C          SET ICODE=-1..... PROBLEM HAS NOT CONVERGED
0084      160 ICODE=-1
0085          GO TO 200

```

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C

C SET ICODE=1.....TRIAL LIMIT REACHED OR BAD COLUMN DATA

0086 180 ICODE=1

0087 GO TO 200

C

C SET ICODE=0.....PROBLEM CONVERGED

0088 190 ICODE=0

C

C 200 RETURN

0090 END


```

0001 SUBROUTINE NJIT
C
0002 DOUBLE PRECISION EQNW, EQNU
0003 INTEGER COMPNT, ID(10)
0004 REAL TREF(10), HVAP(10), EK(10)
C
0005 COMMON /CIRL/ NCR, NPRT, NSTART, TITLE(20)
0006 COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
* NAME(4,10), L(10), NDIM, COMPNT(10)
0007 COMMON /NJIT1/ CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
* W(10)
0008 COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
* PC(10), OA(10), OB(10), AA(10,10), G(10,10),
* THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
* Q(10), R(10), XL(10), PHT(10), DUMMY(10)
0009 COMMON /NJIT3/ TTC(10,10), PPC(10,10), VVC(10,10), ZZC(10,10),
* AK(10,10), AVALUE(10,10), BVALUE(10,10),
* BCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)
C
0010 DATA EQNR /'MRTL'/
0011 DATA EQNW /'WILSON'/
0012 DATA EQNU /'UNJQUAC'/
C
0013 DATA OMA, OMB /0.42748, 0.08664/
C
0014 WRITE (NPRT,30)
0015 30 FORMAT ('NOTE: ALL FLOWS ARE PER UNIT TIME, AND ENERGY' /
* ' VALUES ARE IN UNITS OF K-BTU PER UNIT TIME.')
```

```

C
C
C THE 1ST CARD IS THE PROBLEM TITLE CARD
C THE 2ND CARD IS THE CONTROL PARAMETERS CARD
C THE ABOVE TWO CARDS ARE SPECIFIED IN MAIN PROGRAM
C
C COMPONENT DATA, FOUR CARD(IDLV=2),THREE CARDS(IDLV=0,1)
C
0016 CONTINUE
C CARD #1
C NAME COMPONENT NAME
C L COMPONENT TYPE
C 0 NORMAL
C 1 SUPER CRITICAL
C 2 NON CONDENSIBLE
C 3 NON VOLATILE
C W MOLECULAR WEIGHT---G/G MOLE
C TC CRITICAL TEMPERATURE---DEG K
C PC CRITICAL PRESSURE----- ATM
C VC CRITICAL VOLUME-----CC/G MOLE
C VOL MOLAR VOLUME---CC/G MOLE
C OA REDLICH-KWONG PARAMETERS
C
C CARD #2 (IF VIRIAL IS CHOSEN) TSONOPOULOS' PARAMETERS
C AVALUE----TSONOPOULOS' COMPONT PARAMETER 'A'
C BVALUE---- 'B'
```

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NJIT

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C ZZO-----CRITICAL COMPRESSIBILITY OF COMPONENT
 C OMEGA-----PITZFR'S ACENTRIC FACTOR
 C
 C

C CARD #2-A(OPTIONAL, USED IF UNIQUAC IS CHOSEN)
 C COMPONENT SURFACE AREA AND VOLUME PARAMETERS

C R(I)-----COMPONENT SURFACE PARAMETER

C Q(I)-----COMPONENT VOLUME PARAMETER

017

CONTINUE

C

C CARD #3 (IF VIRIAL IS CHOSEN, OTHERWISE #2) ENTHALPY DATA
 C

C HVAP HEAT OF VAPORIZATION---CAL/G MOLE
 C TRFF REFERENCE TEMPERATURE(NORMAL B.P.)--DEG K

C CPL SPECIFIC HEAT CONSTANTS LIQUID

C CPV (CAL/G MOLE-K) VAPOR
 C

018

CONTINUE

C CARD#4(IF VIRIAL IS CHOSEN,OTHERWISE #3) VAP PRESS CONSTANTS
 C ANTOINE'S EQ. $\text{LN}(\text{VAP PRESS}) = C1 + C2/(C3+\text{TEMP}); \text{MMHG,K}$
 C
 C

C INTERNAL CODE VALUES FOR COMPONENT TYPE

C -1 NON-VOLATILE

C 0 NON-CONDENSIBLE

C 1 NORMAL

C 2 SUPERCRITICAL
 C
 C

C AK(I,J) THE TSONOPOULDS' BINARY CONSTANTS ONE PER COMPONENT
 C MUST FOLLOW THE LAST COMPONENT PROPERTIES CARD, AND IF NRTL
 C OR OTHER ACTIVITY METHODS ARE USED THEN READ IN THE BINARY
 C PARAMETERS CARD AFTER THE LAST AK(I,J) CARD
 C
 C
 C
 C

0019

DO 250 I=1,NC

0020

READ (NCR,230) (NAME(J,I),J=1,2),L(I),W(I),TC(I),PC(I),VC(I),
 * VOL(I), OA(I),OB(I)

0021

230 FORMAT (2A4,I2,7F10.0)

C

0022

VVC(I,I)=VC(I)

0023

TTC(I,I)=TC(I)

0024

PPC(I,I)=PC(I)

0025

IF (IDLV.NE.2) GO TO 81

0026

75 READ (NCR,240) AVALUE(I,I),BVALUE(I,I),ZZO(I,I),OMEGA(I)

C

C

C

READ IN UNIQUAC COMPONENT MOLECULAR SURFACE AREA AND VOLUME
 C PARAMETERS R(I) AND Q(I) , IF IDLL=3
 C
 C

C

C

0027

81 IF (IDLL.NE.3) GO TO 82

0028

READ(NCR,240) R(I),Q(I)

0029

XL(I)=5.0*(R(I)-Q(I))-(R(I)-1.0)

RTRAN IV G LEVEL 21

NJIT

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```

030      H2 IF (L(I)-1) 150,160,170
031      150 L(I)=1
032          GO TO 200
033      160 L(I)=2
034          GO TO 200
035      170 IF (L(I)-3) 180,190,150
036      180 L(I)=0
037          GO TO 200
038      190 L(I)=-1
039      200 CONTINUE

      C
040          IF (W(I).LE.0.0) W(I)=1.0
041          READ (NCR,240) HVAP(I),TREF(I),(CPL(I,J),J=1,3),(CPV(I,J),J=1,3)
042          CPL(I,4)=0.0
043          CPV(I,4)=0.0
044      240 FORMAT (8F10.5)
045          READ (NCR,240) (ANT(J,I),J=1,6)
046      250 CONTINUE

      C
      C
      C      IF VIRIAL IS CHOSEN, THEN READ IN AK(I,J) IN THE ORDER OF
      C      AK(1,1),AK(1,2),AK(1,3) PER COMPONENT #1 FOR NC=3, THEN
      C      READ IN AK(21),AK(2,2),AK(2,3) ETC.
      C
      C
      C      IF VIRIAL IS USED READ IN TSONOPOULOS' BINARY CONSTANTS
      C
047          IF (IDLV.NE.2) GO TO 259
048          DO 85 I=1,NC
049      85 READ (NCR,240) (AK(I,J),J=1,NC)

      C
      C
      C      CALL IN VIRMIX TO CALCULATE CRITICAL BINARY PROPERTIES
      C
050          CALL VIRMIX
051      259 WRITE (NPRT,260) (J,J=1,NC)

      C
052      260 FORMAT ('COMPONENT NO',23X,9(I1,9X))
053          WRITE (NPRT,270) ((NAME(J,I),J=1,2),I=1,NC)
054      270 FORMAT (' NAME',25X,9(2A4,2X))
055          WRITE (NPRT,280) (W(J),J=1,NC)
056      280 FORMAT (' MOLECULAR WEIGHT',16X,9(F5.1,5X))
057          WRITE (NPRT,290) (TC(J),J=1,NC)
058      290 FORMAT (' CRITICAL PROPERTIES'/' TEMPERATURE (DEG K)',10X,
      *          9(F7.1,3X))
059          WRITE (NPRT,300) (PC(J),J=1,NC)
060      300 FORMAT (' PRESSURE ( ATM)',11X,9(F7.1,3X))
061          WRITE (NPRT,310) (VC(J),J=1,NC)
062      310 FORMAT (' VOLUME (CC/GMOL)',6X,9(F7.1,3X))
063          IF (IDLL.NE.3) GO TO 314
064          WRITE (NPRT,291) (R(I),I=1,NC)
065      291 FORMAT(' UNIQVAC COMPONT R(I) ',6X,9(F7.4,3X))
066          WRITE (NPRT,292) (Q(I),I=1,NC)
067      292 FORMAT(' UNIQVAC COMPONT Q(I) ',6X,9(F7.4,3X))
068      314 IF (IDLV.NE.2) GO TO 311
069          WRITE(NPRT,315) (ZC(I,I),I=1,NC)

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070      315 FORMAT(' COMPRESSIBILITY ',13X,9(F7.3,3X))
071      WRITE (NPRT,316) (OMEGA(I),I=1,NC)
072      316 FORMAT(' ACENTRIC FACTOR ',13X,9(F7.3,3X))
073      WRITE (NPRT,317) (AVALUE(I,I),I=1,NC)
074      317 FORMAT(' TSONOPOULOS -A- ',13X,9(F7.4,3X))
075      WRITE (NPRT,318) (BVALUE(I,I),I=1,NC)
076      318 FORMAT(' TSONOPOULOS -B- ',13X,9(F7.4,3X))
077      DO 319 I=1,NC
078      DO 319 J=1,NC
079      319 WRITE(NPRT,321) I,J,AK(I,J)
080      321 FORMAT(' AK(',I1,',',I1,',')= ',9(F7.3,3X))
081 311 WRITE (NPRT,320) (ANT(1,J),J=1,NC)
082      320 FORMAT (' VAPOR PRESSURE CONSTANTS'/' ( MMHG AND DEG K)'/
*          ' C(1)',26X,9(F7.3,3X))
083      DO 340 I=2,6
084      WRITE (NPRT,330) I,(ANT(I,J),J=1,NC)
085      330 FORMAT (' C(',I1,',')',23X,9F10.2)
086      340 CONTINUE
087      WRITE (NPRT,350)
088      350 FORMAT (' LIQUID MOLAR VOLUME CONSTANTS')
089      I=1
090      WRITE (NPRT,390) I,(VOL(J),J=1,NC)
091      WRITE (NPRT,360) (HVAP(J),J=1,NC)
092      360 FORMAT (' ENTHALPY CONSTANTS (CAL/GM)' /
*          ' LATENT HEAT',19X,9(F7.1,3X))
093      WRITE (NPRT,370) (TREF(J),J=1,NC)
094      370 FORMAT (' AT DEG K',22X,9(F7.1,3X))
095      WRITE (NPRT,380)
096      380 FORMAT (' LIQUID SP. HEAT CONSTANTS')
097      DO 400 I=1,4
098      WRITE (NPRT,390) I,(CPL(J,I),J=1,NC)
099      390 FORMAT (' C(',I1,',')',23X,9F10.3)
100      400 CONTINUE
101      WRITE (NPRT,410)
102      410 FORMAT (' VAPOR SP. HEAT CONSTANTS')
103      DO 420 I=1,4
104      WRITE (NPRT,390) I,(CPV(J,I),J=1,NC)
105      420 CONTINUE
106      DO 440 I=1,NC
107      DO 430 J=1,4
108      CPL(I,J)=CPL(I,J)*W(I)
109      430 CPV(I,J)=CPV(I,J)*W(I)
110      HVAP(I)=HVAP(I)*W(I)
111      ENP(I,5)=CPL(I,4)/4.0
112      ENP(I,4)=CPL(I,3)/3.0
113      ENP(I,3)=CPL(I,2)/2.0
114      ENP(I,2)=CPL(I,1)
115      TR=TREF(I)
116      ENP(I,1)=-(((ENP(I,5)*TR+ENP(I,4))*TR+ENP(I,3))*TR+ENP(I,2))*TR
117      ENP(I,10)=CPV(I,4)/4.0
118      ENP(I,9)=CPV(I,3)/3.0
119      ENP(I,8)=CPV(I,2)/2.0
120      ENP(I,7)=CPV(I,1)
121      440 ENP(I,6)=HVAP(I)-(((ENP(I,10)*TR+ENP(I,9))*TR+ENP(I,8))*TR+
*          ENP(I,7))*TR

```

```

C      LIQUID-LIQUID BINARY CONSTANTS
C      C      CONSTANTS FOR NRTL, WILSON, & UNIQUAC
C
C      RENON ALPHA PARAMETERS
C      ALPHA      NRTL CONSTANTS
C
122      IF (IDLL.EQ.0) GO TO 520
C
123      450 J=NC*NCM1*2
124      DO 460 I=1,NC
125      AA(I,I)=0.0
126      460 G(I,I)=1.0
C
127      READ (NCR,240) (C(I),I=1,J)
128      IF (IDLL.EQ.3) GO TO 473
129      469 WRITE(NPRT,470) ENNR
130      470 FORMAT ('0',A6,' CONSTANTS, CAL/GMOL')
131      473 WRITE(NPRT,462) ENNU
132      462 FORMAT('0',A7,' PARAMETERS,CAL/G MOL')
133      K=0
134      DO 490 I=1,NCM1
135      II=I+1
136      DO 490 J=II,NC
137      K=K+4
138      IF (J.LE.9) ID(1)=10*I+J
139      IF (J.GE.10) ID(1)=100*I+J
140      IF (I.LE.9) ID(2)=10*J+I
141      IF (I.GE.10) ID(2)=100*J+I
142      WRITE (NPRT,480) ID(1),C(K-3),ID(1),C(K-2),ID(2),C(K-1),ID(2),C(K)
143      480 FORMAT (4X,2(5X,'A',I4,' =',F9.2,5X,'B',I4,' =',F9.2,5X))
144      490 CONTINUE
C
145      IF (IDLL.NE.1) GO TO 520
C
146      J=NC*NCM1/2
147      READ (NCR,240) (ALPHA(I),I=1,J)
148      K=1
149      KK=1
150      DO 510 I=1,NCM1
151      II=I+1
152      DO 510 J=II,NC
153      EK(KK)=ALPHA(K)
154      IF (J.LE.9) ID(KK)=10*I+J
155      IF (J.GE.10) ID(KK)=100*I+J
156      KK=KK+1
157      K=K+1
158      IF (KK.LT.5) GO TO 510
159      WRITE (NPRT,500) (ID(KK),EK(KK),KK=1,4)
160      500 FORMAT (9X,4('ALPHA',I2,' =',F7.4,5X))
161      KK=1
162      510 CONTINUE
163      IF (KK.EQ.1) GO TO 520
164      KK=KK-1
165      WRITE (NPRT,500) (ID(I),EK(I),I=1,KK)
C
166      520 IF (IDLV.EQ.0) GO TO 640

```

ORTRAN IV G LEVEL 21

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```
0167      530 DO 580 I=1,NC
0168      IF (OA(I).NE.0.0) GO TO 550
0169      OA(I)=OMA
0170      550 IF (OB(I).NE.0.0) GO TO 570
0171      OB(I)=OMB
0172      570 OA(I)=SQRT(OA(I)*TC(I)**2.5/PC(I))
0173      580 OB(I)=OB(I)*TC(I)/PC(I)
C
0174      640 RETURN
0175      END
```

```

1      SUBROUTINE READIN
2      C
3      IMPLICIT REAL      (M)
4      INTEGER           COMPNT
5      REAL              LIQUID, OUT(15,10)
6      DIMENSION         TF(9), MN(9), SLSV(9), FCALC(11)
7      C
8      COMMON /CTRL/     NCR, NPRT, NSTART, TITLE(20)
9      C
10     COMMON /NJITO/    NK, NCM1, IDLL, IDLV, IDH, LDEBUG, ISW,
11     *                NAME(4,10), L(10), NDIM, COMPNT(10)
12     COMMON /NJIT1/    CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
13     *                W(10)
14     COMMON /INRO/     NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
15     *                NFEED, LF(9), IDCODE, ITR, ITRMAX, ICODE
16     COMMON /INR1/     FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
17     *                DEST, DESTV, DESTL, BTMS, RFLX, BOIL, PP(50), T(50),
18     *                MH(50), SL(50), SV(50), FL(50), FLL(50,10),
19     *                FV(50), FVV(50,10), ZCONVT, ZCONVF
20     COMMON /INR2/     VAPOR, LIQUID, FEED(11)
21     COMMON /VLE/     XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
22     COMMON /KNST/    XFKNST(10)
23     C
24     EQUIVALENCE      (WATER,VAPOR)
25     C
26     DATA            CONVT /0.01/, CONVF /0.0001/
27     C
28     READ COLUMN SPECIFICATIONS
29     C
30     NCASE ... CALCULATION CODE
31     1  ABSORBER/STRIPPER
32     2  REBOILED ABSORBER/STRIPPER
33     3  DISTILLATION COLUMN WITH PRODUCT RATES & REFLUX RATIO
34     4  DISTILLATION COLUMN WITH PRODUCT RATES & BOILUP RATIO
35     5  REBOILED ABSORBER
36     C
37     NSTART ... CALCULATION CODE
38     -1  CHANGE MADE TO FEED AND/OR THE DESIGN PARAMETER
39     0   NEW PROBLEM
40     +1  NEW PROBLEM ... INPUT THE OVERHEAD AND BOTTOMS PRODUCTS
41     C
42     IDCODE ... DISTILLATE CODE
43     -1  SUBCOOLED LIQUID
44     0   VAPOR OR VAPOR/LIQUID
45     +1  LIQUID AT THE BUBBLE POINT
46     C
47     RFLX=0.0
48     BOIL=0.0
49     NSAVE=NST
50     C
51     READ (NCR,10,END=700,ERR=700) NCASE,NSTART,NST,NFEED,NSL,NSV,
52     *                ITRMAX,KUNITS,IDCODE
53     10 FORMAT (9I2)
54     C
55     IF (KUNITS.LE.0) KUNITS=2
56     IF (ITRMAX.LE.0) ITRMAX=50

```

```

C
3 READ (NCR,20) MUS,ZCONVT,ZCONVF
4 20 FORMAT (8F10.0)
C
5 NSTM1=NST-1
6 ZCONVT=ABS(ZCONVT)
7 ZCONVF=ABS(ZCONVF)
8 IF (ZCONVT.EQ.0.0) ZCONVT=CONVT
9 IF (ZCONVF.EQ.0.0) ZCONVF=CONVF
C
0 IF (NCASE.EQ.0) NCASE=3
1 GO TO (31,33,35,37,39), NCASE
2 31 WRITE (NPRT,32)
3 32 FORMAT ('0ABSORBER STRIPPER SIMULATION')
4 IDCODE=0
5 GO TO 45
6 33 WRITE (NPRT,34)
7 34 FORMAT ('0REBOILED ABSORBER STRIPPER SIMULATION')
8 IDCODE=0
9 GO TO 45
0 35 WRITE (NPRT,36)
1 36 FORMAT ('0DISTILLATION COLUMN; PRODUCT RATES & REFLUX RATIO SPECIF
2 *TED')
3 GO TO 45
4 37 WRITE (NPRT,38)
5 38 FORMAT ('0DISTILLATION COLUMN; PRODUCT RATES & BOILUP RATIO SPECIF
6 *TED')
7 GO TO 45
8 39 WRITE (NPRT,40)
9 40 FORMAT ('0 REBOILED ABSORBER ')
0 IDCODE=0
C
1 45 NK1=NK+1
2 NK2=2*NK+1
3 NK11=NK1+1
4 NK21=NK2-1
C
5 READ MURPHREE STAGE EFFICIENCIES. IF 'MUS' IS POSITIVE,
6 ONE CONSTANT EFFICIENCY IS READ INTO 'MU'.
C
7 IF (MUS.LT.10.0) GO TO 100
8 DO 49 I=1,NS1
9 MU(I)=0.0
C
0 50 READ (NCR,60) J,K,MUS
1 60 FORMAT (2(I2,8X),F10.0)
C
2 IF (J.EQ.0) GO TO 80
3 IF (K.EQ.0) K=J
4 IF (MUS.LE.0.0) MUS=1.0
5 DO 70 I=J,K
6 70 MU(I)=MUS
7 GO TO 50
8 80 DO 90 I=1,NST
9 IF (MU(I).EQ.0.0) MU(I)=1.0
0 90 CONTINUE

```


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```

7          GO TO 120
8          100 IF (MUS.LE.0.0) MUS=1.0
9           DO 110 J=1,NST
0           110 MU(J)=MUS
1           120 CONTINUE
           C
2           IF (NCASE.NE.1) MU(1)=1.0
3           IF (NCASE.GE.3) MU(NST)=1.0
           C
           C      TEMPERATURE AT COLUMN TOP (TT) AND COLUMN BOTTOM (TB) ARE
           C      IN DEGREES K.  PRESSURE AT TOP OF COLUMN (PP-NST) AND THE
           C      STAGE PRESSURE DROP (DPATM) ARE IN MMHG ABS.
           C
           C      REBOILER IS STAGE 1
           C
4          READ (NCR,20) DESTV,DESTL,DUMMY,TT,TR,PP(NST),DPATM
5          DPATM=ABS(DPATM)
           C
6          GO TO (121,125,125), KUNITS
7          121 TT=(TT-32.0)/1.8+273.16
8           TR=(TR-32.0)/1.8+273.16
9           PP(NST)=PP(NST)/14.696
0           DPATM=DPATM/14.696
1           DESTV=DESTV*453.59
2           DESTL=DESTL*453.59
           C
3          125 IF (NCASE.EQ.3) RFLX=DUMMY
4           IF (NCASE.EQ.4) BOIL=DUMMY
5           IF (NCASE.EQ.5) BOIL=DUMMY
           C
6          DEST=DESTV+DESTL
7          IF (ICODE.EQ.0) GO TO 129
8          DESTV=DEST
9          DESTL=0.0
0          DO 126 J=1,NK
1          ZVP=EXP(ANT(1,J)+ANT(2,J)/(ANT(3,J)+TT)+ANT(4,J)*TT+ANT(5,J)*
           *      TT*TT+ANT(6,J)*ALOG(TT))
2          126 XFKNST(J)=ZVP/PP(NST)
           C
           C      ASSUME CONSTANT PRESSURE DROP ACROSS THE COLUMN
           C
3          129 PP(1)=PP(NST)+NSTM1*DPATM
4          DO 130 I=2,NST
5          130 PP(I)=PP(I-1)-DPATM
           C
           C      GUESS OF THE TEMPERATURE PROFILE IN THE COLUMN.
           C
6          IF (NSTART.LT.0) GO TO 140
7          DO 131 I=1,NST
8          131 T(I)=TR+(I-1)*(TT-TR)/NSTM1
           C
           C
9          140 DO 141 J=1,NK1
0          141 FEED(J)=0.0
1          DO 144 I=1,NST
2          144 SL(I)=0.0

```

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```

3      144 SV(I)=0.0
4      NFTOP=0
5      NFHTM=0
      C
6      DO 145 I=1,10
7      DO 145 J=1,15
8      145 OUT(J,I)=0.0
      C
9      WRITE (NPRT,150)
0      150 FORMAT ('OFFED STREAMS:')
      C
      C      READ FEED SPECIFICATIONS ... INTERNAL UNITS
      C      FEED / EXTERNAL HEAT STREAM
      C          LF          STAGE
      C          QIN        ENERGY INPUT, K-CAL/UNIT TIME
      C          TF          TEMPERATURE, DEG K
      C          FKV        FRACTION VAPOR
      C          FSTR       FEED RATE .. COMPONENTS, MOLES/UNIT TIME
      C          HF          COMPUTED FEED STREAM ENERGY, CAL
      C
1      DO 206 I=1,NFEED
      C
2      READ (NCR,155) LF(I),QIN,TF(I),FKV(I),(FSTR(I,J),J=1,5)
3      155 FORMAT (I2,F8.0,7F10.0)
4      IF (NK.GT.5) READ (NCR,20) (FSTR(I,J),J=6,NK)
5      GO TO (156,165,165), KUNITS
6      156 TF(I)=(TF(I)-32.0)/1.8+273.16
7      QIN=QIN*252.16*1.0E3
8      DO 157 J=1,NK
9      C 157 FSTR(I,J)=FSTR(I,J)*453.59
0      165 IF (LF(I).EQ.1) NFHTM=I
1      IF (LF(I).EQ.NST) NFTOP=I
2      FSTR(I,NK1)=0.0
3      HF(I)=0.0
4      TK=TF(I)
      C
5      IF (QIN.NE.0.0) GO TO 190
      C
6      DO 169 J=1,NK
7      HF(I)=HF(I) + FSTR(I,J)*FKV(I)*((((ENP(J,10)*TK+ENP(J,9))*TK+
      *      ENP(J,8))*TK+ENP(J,7))*TK+ENP(J,6)) + FSTR(I,J)*(1.0-FKV(I))
      *      *((((ENP(J,5)*TK+ENP(J,4))*TK+ENP(J,3))*TK+ENP(J,2))*TK+
      *      ENP(J,1))
8      169 FSTR(I,NK1)=FSTR(I,NK1)+FSTR(I,J)
      C
9      IF (FKV(I).EQ.0.0) GO TO 178
      C
0      IF (IDLV.EQ.0) GO TO 188
1      IF (IDH.EQ.2) GO TO 188
2      ISTAGE=LF(I)
3      DO 170 J=1,NK
4      VP(J)=PP(ISTAGE)
5      170 Y(J)=FSTR(I,J)/FSTR(I,NK1)
6      CALL FUGCTY (TF(I),PP(ISTAGE),Y,VP,FUG,DHV)
7      HF(I)=HF(I)+DHV*FSTR(I,NK1)

```

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```

8      GO TO 188
C
9      178 IF (IDLL.EQ.0) GO TO 188
0      IF (IDH.GT.0) GO TO 188
1      DO 179 J=1,NK
2      179 X(J)=FSTR(I,J)/FSTR(I,NK1)
3      ISW=0
4      CALL XSHLIN (TF(I),X,XSH)
5      HF(I)=HF(I)+XSH*FSTR(I,NK1)
C
6      188 DO 189 J=1,NK1
7      189 FEED(J)=FEED(J)+FSTR(I,J)
C
8      190 HF(I)=HF(I)+QIN
9      GO TO (191,195,195), KUNITS
0      191 TDEG=(TF(I)-273.16)*1.8+32.0
1      H=HF(I)/252.16/1.0E3
2      QINQ=QIN/252.16/1.0E3
3      DO 192 J=1,NK1
4      192 FCALC(J)=FSTR(I,J)/453.59
5      GO TO 201
6      195 TDEG=TF(I)
7      H=HF(I)
8      QINQ=QIN
9      DO 196 J=1,NK1
0      196 FCALC(J)=FSTR(I,J)
C
1      201 IF (QIN.EQ.0.0) GO TO 202
2      OUT(1,I)=LF(I)
3      OUT(4,I)=QINQ
4      GO TO 206
C
5      202 OUT(1,I)=LF(I)
6      OUT(2,I)=TDEG
7      OUT(3,I)=FKV(I)
8      OUT(4,I)=H
9      DO 205 J=1,NK
0      205 OUT(J+4,I)=FCALC(J)
1      OUT(NK1+4,I)=FCALC(NK1)
C
2      206 CONTINUE
C
3      NFP1=NFEED+1
4      DO 207 J=1,NK
5      207 OUT(J+4,NFP1)=FEED(J)
6      OUT(NK1+4,NFP1)=FEED(NK1)
C
7      GO TO (208,210,210), KUNITS
8      208 DO 209 J=1,NK
9      209 OUT(J+4,NFP1)=FEED(J)/453.59
0      OUT(NK1+4,NFP1)=FEED(NK1)/453.59
C
1      210 WRITE (NPRT,211) (OUT(1,I),I=1,NFEED)
2      211 FORMAT (' STAGE', 15X, 10F12.0)
3      WRITE (NPRT,212) (OUT(2,I),I=1,NFEED)
4      212 FORMAT (' TEMP ', 15X, 10F12.2)

```

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```

5      WRITE (NPRT,213) (OUT(3,I),I=1,NFEED)
6      213 FORMAT (' FRACV', 15X, 10F12.4)
7      WRITE (NPRT,214) (OUT(4,I),I=1,NFEED)
8      214 FORMAT (' H      ', 15X, 1P10E12.4)
9      DO 215 J=1,NK
0      215 WRITE (NPRT,216) J,(NAME(K,J),K=1,2),(OUT(J+4,I),I=1,NFP1)
1      216 FORMAT (' COMP', 12, 2X,2A4,5X, F11.3, 9F12.3)
2      WRITE (NPRT,217) (OUT(NK1+4,I),I=1,NFP1)
3      217 FORMAT (' TOTAL', 15X, 10F12.3)

```

C

```

4      225 IF (NSL.EQ.0) GO TO 265
5      DO 235 J=1,NSL
6      235 READ (NCR,240) NN(J),SLSV(J)
7      240 FORMAT (I2,8X,F10.0)
8      GO TO 265
9      250 WRITE (NPRT,251)
0      251 FORMAT (' LIQUID SIDE STREAMS'/' STAGE',3X,' MOLES')
1      DO 264 J=1,NSL
2      N=NN(J)
3      SL(N)=SLSV(J)
4      WRITE (NPRT,260) N,SL(N)
5      260 FORMAT (I5,F11.3)
6      GO TO (261,264,264), KUNITS
7      261 SL(N)=SL(N)*453.59
8      264 CONTINUE
9      GO TO 270

```

C

```

0      265 IF (DESTL.EQ.0.0 .AND. NSL.EQ.0) GO TO 270
1      NSL=NSL+1
2      NN(NSL)=NST
3      SLSV(NSL)=DESTL/453.59
4      GO TO 250

```

C

```

5      270 IF (NSV.EQ.0) GO TO 300
6      WRITE (NPRT,280)
7      280 FORMAT (' VAPOR SIDE STREAMS'/' STAGE',3X,' MOLES')
8      DO 285 J=1,NSV
9      285 READ (NCR,240) NN(J),SLSV(J)
0      DO 299 J=1,NSV
1      N=NN(J)
2      SV(N)=SLSV(J)
3      WRITE (NPRT,260) N,SV(N)
4      GO TO (291,299,299), KUNITS
5      291 SV(N)=SV(N)*453.59
6      299 CONTINUE

```

C

C

C

FIRST GUESS OF THE COMPONENT FLOWS AND THE TOTAL FLOWS.

```

7      300 XDEST=0.0
8      DO 310 I=1,NST
9      310 XDEST=XDEST+SV(I)+SL(I)
0      BTMS=FEED(NK1)-DFSTV-XDEST
1      IF (BTMS.LE.0.0) GO TO 680

```

C

```

2      IF (NSTART.LE.0) GO TO 320

```

C

```

READ (NCR,20) (FVV(NST,J),J=1,NK)
READ (NCR,20) (FVV(1,J),J=1,NK)
READ (NCR,20) (FLL(NST,J),J=1,NK)
READ (NCR,20) (FLL(1,J),J=1,NK)
GO TO (301,320,320), KUNITS

```

```

301 DO 302 J=1,NK
    FVV (NST,J)=FVV(NST,J)*453.59

```

```

302 FLL(1,J)=FLL(1,J)*453.59

```

C

```

320 GO TO (330,330,410,410,410), NCASE

```

C

C

```

    ABSORBER / STRIPPER

```

C

C

```

    SET MAX DEVIATIONS IN COMPONENT FLOWS BETWEEN ITERATIONS

```

```

330 DO 340 J=1,NK

```

```

    FVMAX(J)=FEED(J)

```

```

340 FLMAX(J)=FEED(J)

```

C

```

    IF (NSTART) 521,380,350

```

C

```

350 FV(NST)=0.0

```

```

    FL(1)=0.0

```

```

    DO 360 J=1,NK

```

```

        FVV(1,J)=FSTR(NFRTM,J)

```

```

        FLL(NST,J)=FSTR(NFTOP,J)

```

```

        FV(NST)=FV(NST)+FVV(NST,J)

```

```

360 FL(1)=FL(1)+FLL(1,J)

```

C

```

    FV(1)=FSTR(NFRTM,NK1)

```

C

```

    FL(NST)=FSTR(NFTOP,NK1)

```

```

    DO 370 I=2,NSTM1

```

```

        FV(I)=FV(I-1)+(FV(NST)-FV(1))/NSTM1

```

```

        FL(I)=FL(I-1)+(FL(NST)-FL(1))/NSTM1

```

```

    DO 370 J=1,NK

```

```

        FVV(I,J)=FVV(I-1,J)+(FVV(NST,J)-FVV(1,J))/NSTM1

```

```

        FLL(I,J)=FLL(I-1,J)+(FLL(NST,J)-FLL(1,J))/NSTM1

```

```

370 CONTINUE

```

```

    GO TO 530

```

C

```

380 DO 390 I=1,NST

```

```

    FV(I)=FSTR(NFRTM,NK1)

```

```

    FL(I)=FSTR(NFTOP,NK1)

```

```

    DO 390 J=1,NK

```

```

        FVV(I,J)=FSTR(NFRTM,J)

```

```

        FLL(I,J)=FSTR(NFTOP,J)

```

```

        IF (FVV(I,J).LE.0.0) FVV(I,J)=1.0E-20

```

```

        IF (FLL(I,J).LE.0.0) FLL(I,J)=1.0E-20

```

```

390 CONTINUE

```

```

    DO 400 J=1,NK

```

```

        FVV(NST,J)=FVV(NST,J)/FV(NST)*DEST

```

```

400 FLL(1,J)=FLL(1,J)/FL(1)*BTMS

```

```

    FV(NST)=DEST

```

```

    FL(1)=BTMS

```

```

    GO TO 530

```

C

C

```

    DISTILLATION COLUMN (REFLUX OR BOTLUP SPECIFIED)

```

C

RAN IV G LEVEL 21

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```

C      SET MAX DEVIATIONS IN COMPONENT FLOWS BETWEEN ITERATIONS
6      410 DO 420 J=1,NK
7          FVMAX(J)=FEED(J)*(1.0+DUMMY)
8      420 FLMAX(J)=FEED(J)*(1.0+DUMMY)
C
9      IF (NSTART.LT.0) GO TO 521
C
0      FV(NST)=DESTV
1      FL(1)=BTMS
2      IF (NCASE.EQ.3) FL(NST)=DEST*RFLX
3      IF (NCASE.EQ.4) FV(1)=BTMS*BOIL
4      IF (NCASE.EQ.5) FV(1)= BTMS*BOIL
5      SS=0.0
6      S=0.0
7      DO 430 I=1,NFEED
8          IF (LF(I).EQ.NST) S=FSTR(I,NK1)*FKV(I)
9          IF (LF(1).EQ.1) SS=FSTR(I,NK1)*(1.0-FKV(I))
0      430 CONTINUE
1      FV(NST*1)=FL(NST)-S+SV(NST)+SL(NST)+DESTV
2      DO 440 JI=3,NST
3          IF (NST.LF.2) GO TO 440
4          J=NST+2-JI
5          FL(J)=FL(J+1)-SL(J)
6          FV(J-1)=FV(J)+SV(J)
7      440 CONTINUE
8      DO 460 I=1,NFEED
9          J=LF(I)
0          S=(1.0-FKV(J))*FSTR(I,NK1)
1          SSS=FKV(I)*FSTR(I,NK1)
2          DO 450 II=2,J
3              FL(II)=FL(II)+S
4      450 FV(II-1)=FV(JI-1)-SSS
5      460 CONTINUE
6          FL(1)=BTMS
C
7      IF (NSTART.GT.0) GO TO 480
C
8      DO 470 I=1,NST
9          DO 470 J=1,NK
0          FVV(I,J)=FEED(J)/FEED(NK1)*FV(I)
1      470 FLL(I,J)=FEED(J)/FEED(NK1)*FL(I)
2          GO TO 530
C
3      480 S=0.0
4          SS=0.0
5          DO 490 J=1,NK
6              S=S+FVV(NST,J)
7      490 SS=SS+FLL(1,J)
8          DO 500 J=1,NK
9              FVV(NST,J)=FVV(NST,J)/S
0              FLL(NST,J)=FVV(NST,J)
1              FLL(1,J)=FLL(1,J)/SS
2      500 FVV(1,J)=FLL(1,J)
3          DO 510 I=2,NSTM1
4              DO 510 J=1,NK
5              FVV(1,J)=FVV(I-1,J)+(FVV(NST,J)-FVV(1,J))/NSTM1

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6      510 FLL(I,J)=FLL(I-1,J)+(FLL(NST,J)-FLL(I,J))/NSTM1
7          DO 520 I=1,NST
8          DO 520 J=1,NK
9          FVV(I,J)=FVV(I,J)*FV(I)
0      520 FLL(I,J)=FLL(I,J)*FL(I)
1          GO TO 530

C
C      FOR NSTART=-1 CHECK FOR ZERO FLOWS
C
2      521 IF (NST.LE.NSAVE) GO TO 525
3          DO 522 I=NSAVE,NST
4          FL(I)=FL(NSAVE)
5          FV(I)=FV(NSAVE)
6          DO 522 J=1,NK
7          FLL(I,J)=FLL(NSAVE,J)
8      522 FVV(I,J)=FVV(NSAVE,J)
9          GO TO 530
0      525 FV(NST)=DEFSTV
1          FL(1)=BTMS

C
C      SCREEN COMPONENT TYPE AND ADJUST INITIAL FLOW MAP
C
2      530 DO 538 J=1,NK
3          IF (L(J).NE.-1) GO TO 534
C      NON-VOLATILE COMPONENT, VAPOR FLOW = 0.0
4          DO 532 I=1,NST
5      532 FVV(I,J)=0.0
6          GO TO 538
7      534 IF (L(J).NE.0) GO TO 538
C      NON-CONDENSABLE COMPONENT, LIQUID FLOW = 0.0
8          DO 536 I=1,NST
9      536 FLL(I,J)=0.0
0      538 CONTINUE
1          DO 1030 I=1,NST
2          SUMV=0.0
3          SUML=0.0
4          DO 1010 J=1,NK
5          SUMV=SUMV+FVV(I,J)
6      1010 SUML=SUML+FLL(I,J)
7          DO 1020 J=1,NK
8          FVV(I,J)=FVV(I,J)/SUMV*FV(I)
9      1020 FLL(I,J)=FLL(I,J)/SUML*FL(I)
0      1030 CONTINUE
C
1          GO TO (1031,1035,1035), KUNITS
2      1031 PRES=PP(NST)*14.696
3          DPRES=DPATM*14.696
4          DESTZ=DEST/453.59
5          BTMSZ=BTMS/453.59
6          GO TO 539
7      1035 PRES=PP(I)
8          DPRES=DPATM
9          DESTZ=DEST
0          BTMSZ=BTMS
C
1      539 WRITE (NPRT,540) NST,DESTZ,BTMSZ,PRES,DPRES

```

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2      540 FORMAT ('NUMBER OF STAGES', I9 //
*          ' ESTIMATED OVERHEAD PRODUCT RATE', F12.4 /
*          ' ESTIMATED BOTTOMS PRODUCT RATE', F13.4 //
*          ' PRESSURE AT TOP OF COLUMN', F18.5 /
*          ' CONSTANT DELTA P PER STAGE', F17.5 )
3      GO TO (590,590,550,570,570), NCASE
4      550 WRITE (NPRT,560) RFLX
5      560 FORMAT ('OREFLUX RATIO', F15.6)
6      GO TO 590
7      570 WRITE (NPRT,580) BOIL
8      580 FORMAT ('BOILUP RATIO', F15.6)
C
C      FOR NSTART=-1 CHECK FOR ZERO TEMPERATURE
C
9      590 IF (NSTART.GE.0) GO TO 610
0      IF (NST.LE.NSAVE) GO TO 610
1      DO 607 I=NSAVE,NST
2      607 T(I)=T(NSAVE)
C
3      610 IF (LDRUG.EQ.0) GO TO 800
4      WRITE (NPRT,620)
5      620 FORMAT ('INITIAL COLUMN PROFILES:' /
*          ' STAGE      TEMP      PRES      LIQUID(MOLS)      VAPOR(MOLS)' )
6      DO 630 I=1,NST
8      621 GORGB(I,621)=825.685+1.4*3210
9      PRES=PP(I)*14.696
0      FLZ=FL(I)/453.59
1      FVZ=FV(I)/453.59
2      GO TO 629
3      625 TDEG=T(I)
4      PRES=PP(I)
5      FLZ=FL(I)
6      FVZ=FV(I)
7      629 WRITE (NPRT,640) I,TDEG,PRES,FLZ,FVZ
8      630 CONTINUE
9      640 FORMAT (I5,F10.2,F12.3,F15.3,F14.3)
0      GO TO 800
C
1      680 WRITE (NPRT,690)
2      690 FORMAT (' MATERIAL BALANCE ERROR ... BOTTOMS RATE IS LT/EQ ZFRD')
C
3      700 ICODE=1
C
4      800 RETURN
5      END

```



```

0001      SUBROUTINE TEST
0002      IMPLICIT REAL      (M)
0003      INTEGER      COMPNT
0004      REAL      LIQUID

C
0005      COMMON /CTRL/ NCR, NPRT, NSTART, TITLE(20)
0006      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
*      NAME(4,10), L(10), NDIM, COMPNT(10)
0007      COMMON /NJIT1/ CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
*      W(10)
0008      COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
*      PC(10), OA(10), OB(10), AA(10,10), G(10,10),
*      THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
*      Q(10), R(10), XL(10), PHI(10), DUMMY(10)
0009      COMMON /NJIT3/ TTC(10,10), PPC(10,10), VVC(10,10), ZC(10,10),
*      AK(10,10), AVALUE(10,10), BVALUE(10,10),
*      BCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)
0010      COMMON /INR0/ NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
*      NFEED, LF(9), IDCODE, ITR, ITRMAX, ICODE
0011      COMMON /INR1/ FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
*      DEST, DESTV, DESTL, BTMS, RFLX, BOIL, PP(50), T(50),
*      MU(50), SL(50), SV(50), FL(50), FLL(50,10),
*      FV(50), FVV(50,10), ZCONVT, ZCONVF
0012      COMMON /INR2/ DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
*      DEXH(50,11), HV(50), HVV(50,10), CV(50)
0013      COMMON /INR3/ AMAT(21,21), BMAT(50,21,21), D(50,21)
0014      COMMON /INR4/ VAPOR, LIQUID, FEED(11)
0015      COMMON /VLE/ XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
0016      COMMON /KNST/ XFKNST(10)

C
0017      DATA      SMALL /0.00005/

C
0018      ITR=ITR+1

C
0019      WRITE (NPRT,75) ITR
0020      75 FORMAT (' ITERATION # =', I4)

C
0021      IF (LDBUG.EQ.0) GO TO 110
0022      WRITE (NPRT,80)
0023      80 FORMAT (' STAGE      TEMP      PRES      LIQUID(MOLS)      VAPOR(MOLS)
*      *')
0024      DO 90 I=1,NST
0025      GO TO (81,85,85), KUNITS
0026      81 TDEG=(T(I)-273.16)*1.8+32.0
0027      PRES=PP(I)*14.696
0028      FLZ=FL(I)/453.59
0029      FVZ=FV(I)/453.59
0030      GO TO 90
0031      85 TDEG=T(I)
0032      PRES=PP(I)
0033      FLZ=FL(I)
0034      FVZ=FV(I)
0035      90 WRITE (NPRT,100) I,TDEG,PRES,FLZ,FVZ
0036      100 FORMAT (I5,F10.2,F12.3,F15.3,F14.3)

C
C      SUM OF SQUARES OF DISCREPANCY FUNCTIONS

```

```

C
0037      110 REST=0.0
0038          RESF=0.0
0039          DTBIG=0.0
0040          ITBIG=0
0041          DLBIG=0.0
0042          ILBIG=0
0043          JLBIG=0
0044          DVBIG=0.0
0045          IVBIG=0
0046          JVBIG=0
0047          DO 120 I=1,NST
0048              IF (ABS(D(I,NK1)).LT.DTBIG) GO TO 112
0049              DTBIG=ABS(D(I,NK1))
0050              ITBIG=I
0051      112 REST=REST+D(I,NK1)**2
0052              DO 120 J=1,NC
0053                  IF (D(I,J).EQ.0.0) GO TO 115
0054                  IF (FLL(I,J)/FL(I).LT.SMALL) GO TO 115
0055                  IF (ABS(D(I,J))/FLL(I,J).LT.DLBIG) GO TO 113
0056                  DLBIG=ABS(D(I,J))/FLL(I,J)
0057                  ILBIG=I
0058                  JLBIG=J
0059      113 RESF=RESF+(D(I,J)/FLL(I,J))**2
0060      115 IF (D(I,J+NK1).EQ.0.0) GO TO 120
0061              IF (FVV(I,J)/FV(I).LT.SMALL) GO TO 120
0062              IF (ABS(D(I,J+NK1))/FVV(I,J).LT.DVBIG) GO TO 118
0063              DVBIG=ABS(D(I,J+NK1))/FVV(I,J)
0064              IVBIG=I
0065              JVBIG=J
0066      118 RESF=RESF+(D(I,J+NK1)/FVV(I,J))**2
0067      120 CONTINUE
C
0068          WRITE (NPRT,130) REST,RESF
0069      130 FORMAT (' SUM D1**2 =',      G17.6, /
*           ' SUM (DF/F)**2 =', G13.6)
0070          WRITE (NPRT,135) ITBIG,DTBIG,ILBIG,JLBIG,DLBIG,IVBIG,JVBIG,DVBIG
0071      135 FORMAT (5X,I5,G12.5,5X,2I5,G12.5,5X,2I5,G12.5)
C
0072          IF (ITR.EQ.ITRMAX) GO TO 160
C
0073          DO 150 I=1,NST
0074              IF (ABS(D(I,NK1)).GT.ZCONVT) GO TO 160
0075              DO 150 J=1,NC
0076                  IF (FLL(I,J)/FL(I).LT.SMALL) GO TO 140
0077                  IF (ABS(D(I,J))/FLL(I,J).GT.ZCONVF) GO TO 160
0078      140 IF (FVV(I,J)/FV(I).LT.SMALL) GO TO 150
0079                  IF (ABS(D(I,J+NK1))/FVV(I,J).GT.ZCONVF) GO TO 160
0080      150 CONTINUE
0081              GO TO 190
C
0082          SET ICODE=-1..... PROBLEM HAS NOT CONVERGED
0083      160 ICODE=-1
          GO TO 200
C
0084          SET ICODE=1.....TRIAL LIMIT REACHED OR BAD COLUMN DATA

```

```
0084      180 ICODE=1
0085      GO TO 200
      C
      C      SET ICODE=0.....PROBLEM CONVERGED
0086      190 ICODE=0
      C
0087      200 RETURN
0088      END
```

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```

0001      SUBROUTINE EDWARD
          C      THIS IS THE OLD MAIN PROGRAM OF EXXON
          C      THE READ IN PORTION FOR FD(I),TF,PF,NLIQ HAS BEEN RELOCTED
          C
          C      THE ISOTHERMAL FLASH IS THE ONLY ACTIVE SUBR IN USE ISO=0
          C
          C      VAPOR LIQUID FLASH PROGRAM FOR NON IDEAL SYSTEMS
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
0002      DIMENSION DEWC(9), BURC(9), HLIQD(2)
0003      COMMON /B2L/ FD(11),7(10),FK(10,2),PF,TF,HI,XSAVE(10,2)
0004      COMMON /CTRL/ I,P,NOBUB,NOREW,ISO,NLIQ
0005      COMMON /VAPOR/ Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRALV
0006      COMMON /LIQD/ X(10,2),GAB(10,2),DHL(2),HIOLL(2),HREAL(2),
          +      FRACL
0007      COMMON /DUMMY/ XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
          +      YC(10),RMAX,RMIN
0008      COMMON /CTRL/ NCR, NPRT, NSTART, TITLE(20)
0009      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDEBUG, ISW,
          *      NAME(2,10),L(10),COMPNT(10)
0010      COMMON /NJIT1/ CPL(10,4), CPV(10,4), ENP(10,10), ANI(6,10),
          *      W(10)
0011      COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *      PC(10), OA(10), OR(10), AZ(10,10), G(10,10),
          *      THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
          *      D(10),RR(10),XL(10),PHI(10),DUMMY(10)
0012      COMMON /NJIT3/ TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
          *      AK(10,10), AVALUE(10,10), BVALUE(10,10),
          *      BCROSS(10,10), DMFGA(10), DIPOLE(10), BZ(10,10)
          C
          C
0013      DO 15 I=1,NC
0014          Y(I)=0.0
0015          YY(I)=0.0
0016          X(I,1)=0.0
0017          X(I,2)=0.0
0018          XX(I,1)=0.0
0019      15 XX(I,2)=0.0
0020          SUM=0.0
0021          DO 25 I=1,NC
0022      25 SUM = SUM + FD(I)
0023          DO 30 I=1,NC
0024      30 Z(I)= FD(I)/SUM
          C
          C
          C
0025      WRITE (6,40)
0026      40 FORMAT('1',/// '***** TWO-LIQUID FLASH PROGRAM *****'///)
0027      WRITE (6,50)SUM,PF,TF
0028      50 FORMAT('LOFD MOL =',F8.4,' MOLES',8X,'PRESS = ',
          *      F8.2,' MMHG',8X,'TEMP =',F8.2,' DEG K')
0029      WRITE (6,60)((NAME(J,1),J=1,2),I=1,NC)
0030      60 FORMAT(25X,9(2A4,2X))
0031      WRITE (6,70)(Z(I),I=1,NC)

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0032          70 FORMAT(' FEED MOLEFRACTIONS',4X,9F10.5)
      C
0033          ISO=0
      C
0034          RMIN=0.0
0035          RMAX=1.0
0036          DO100I=1,NC
0037          IF (L(I))80,90,100
0038          80 RMAX=RMAX-Z(I)
0039          90 10 100
0040          90 RMIN=RMIN+Z(I)
0041          100 CONTINUE
      C
0042          NDEWA=0
0043          NDEWB=0
      C
0044          DO140I=1,NC
0045          IF (FD(I))140,140,110
0046          110 IF (L(I))120,130,140
0047          120 NDEWA=1
0048          GO TO 140
0049          130 NDEWB=1
0050          140 CONTINUE
      C
0051          150 WRITE (6,151)
0052          151 FORMAT ('RESULTS OF DEW POINT CALCULATION')
0053          IF (NDEWA.EQ.0) GO TO 170
0054          WRITE (6,160)
0055          160 FORMAT('NO DEW POINT SINCE NONVOLATILE COMPONENTS ARE PRESENT')
0056          TDFW=1.06
0057          DO 165 I=1,NC
0058          165 DEWC(I)=0.0
0059          GO TO 220
      C
0060          170 DO180I=1,NC
0061          180 Y(I)=Z(I)
0062          T=TF
0063          NHOLD=NLIN
0064          NLIN=1
0065          CALL DEWPT
0066          TDEW=T
0067          DO190I=1,NC
0068          190 DEWC(I)=X(I,1)
0069          WRITE (6,200) P,TDEW
0070          200 FORMAT('DEW POINT OF FEED AT',F7.2,' MMHG IS',F7.2,' DEG K')
0071          WRITE (6,60) ((NAME(J,1),J=1,2),I=1,NC)
0072          WRITE (6,210) (DEWC(I),I=1,NC)
0073          210 FORMAT (' DEW PT LIQ MOLE FRAC', 2X, 9F10.5)
0074          CALL PRINT
0075          NLIN=NHOLD
      C
      C
      C          BUBBLE POINT CALCULATION
      C
0076          220 WRITE (6,221)

```

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0077      221 FORMAT ('////RESULTS OF BUBBLE POINT CALCULATION')
0078      IF (NOBUB.EQ.0) GO TO 240
0079      WRITE (6,230)
0080      230 FORMAT('NO BUBBLE POINT SINCE NONCONDENSIBLE COMPONENTS ARE PRESE
      *NT')
0081      231 TBUB=0.1
0082      DO 235 I=1,NC
0083      235 BUBC(I)=0.0
0084      GO TO 320
      C
0085      240 IF (NLIQ.EQ.2) GO TO 280
0086      DO 241 I=1,NC
0087      241 X(1,I)=Z(I)
0088      T=TF
0089      FRAC1=1.0
0090      CALL BUBPT1
      C
0091      245 IF (NOBUB.EQ.0) GO TO 255
0092      WRITE (6,250)
0093      250 FORMAT('FEED REPRESENTS A HYPOTHETICAL LIQUID OR IS IN THE RETRO
      +GRADE REGION. BUBBLE POINT CALCULATIONS TERMINATED.')
0094      GO TO 231
      C
0095      255 TBUB=1
0096      WRITE (6,260) P,TBUB
0097      260 FORMAT('BUBBLE POINT OF FEED AT',F7.2,' MMHG IS',F7.2,' DEG K')
0098      DO 265 I=1,NC
0099      265 BUBC(I)=Y(I)
0100      WRITE (6,60)((NAME(J,I),J=1,2),I=1,NC)
0101      WRITE (6,270) (BUBC(I),I=1,NC)
0102      270 FORMAT (' BUB PT VAP MOLE FRAC', 2X, 9F10.5)
0103      CALL PRINT
0104      GO TO 320
      C
0105      280 DO 281 I=1,NC
0106      DO 281 J=1,NLIQ
0107      281 X(I,J) = XSAVE(I,J)
      C
      C THE XSAVE(I,J) VALUES ARE INITIALIZED IN THE NEW SUBROUTINE 'START'
      C FOR THE ABOVE X(I,J)
      C TWO LIQUID PHASE BUBBLE POINT INITIALIZATION
      C * COMPONENT #1(ORGANIC) PREDOMINATELY IN PHASE #1(ORGANIC)
      C * COMPONENT #NC PREDOMINATELY IN PHASE #2(AQUEOUS)
      C
      C
      C INITIALIZE LIQUID COMPOSITION XSAVE FOR NLLIQ=2
      C
      C 280 X(1,1)=Z(1)
      C X(1,2)=1.0E-12
      C X(NC,1)=1.0E-12
      C X(NC,2)=7(NC)
      C DO 281 I=2,NC#1
      C X(I,1)=Z(I)/2.0
      C 281 X(I,2)=X(I,1)
0108      DO 284 N=1,2
0109      SUM=0.0

```

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0110      DO 282 I=1,NC
0111      282 SUM=SUM+X(I,N)
0112      DO 283 I=1,NC
0113      283 X(I,N)=X(I,N)/SUM
0114      284 CONTINUE
0115      T=TF
0116      FRACL=0.5
0117      CALL SUBPT2
0118      IF (NLIQ.EQ.1) GO TO 245
0119      WRITE (6,285) FRACL
0120      285 FORMAT ('TWO LIQ PHASE REGION, LIQUID PHASE MOLE RATIO', F12.6)
0121      GO TO 245
C
0122      320 IF (TSQ.NE.0) GO TO 440
C
0123      IF (TF-TDEW) 380,340,330
0124      330 T=TF
0125      340 WRITE (6,350)
0126      350 FORMAT('OFFED AT OR ABOVE DEW POINT')
0127      CALL PRINT
0128      DO 360 I=1,NC
0129      360 Y(I)=Z(I)
0130      CALL VAPH
0131      H=HREALV
0132      WRITE (6,370)HIDL, DHV
0133      370 FORMAT('IDEAL VAP ENTHALPY =',E15.6,5X,'DELTA H =',E15.6)
0134      GO TO 650
C
0135      380 IF (TF-THUR) 390,400,440
0136      390 T=TF
0137      400 WRITE (6,410)
0138      410 FORMAT('OFFED AT OR BELOW BUBBLE POINT')
0139      GO TO (415,420), NLIQ
0140      415 DO 416 I=1,NC
0141      416 X(I,1)=Z(I)
0142      CALL LIQH
0143      H=HREAL(1)
0144      GO TO 429
0145      420 CALL TWOLIQ
0146      CALL LIQH
C
C      FRACL= PHASE#1/(PHASE#1 + PHASE#2)
C
C
0147      H=HREAL(1)*FRACL+HREAL(2)*(1.0-FRACL)
0148      WRITE (6,285) FRACL
0149      DO 428 I=1,NC
0150      428 Y(I)=0.00
0151      429 CALL PRINT
0152      WRITE (6,430) (HIDL(N),DHL(N),N=1,NLIQ)
0153      430 FORMAT('IDEAL LIQ ENTHALPY =',E15.6,5X,'DELTA H =',E15.6)
0154      GO TO 650
C
C      INITIALIZE VAPOR & LIQUID PHASES FOR FLASH
C
0155      440 FRACV=(RMAX+RMIN)/2.0

```

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C

```

0156      IF (NODIR.EQ.0) GO TO 460
0157      IF (NODEW.EQ.0) GO TO 550
0158      GO TO 480

```

C

```

0159      460 DO 470 I=1,NC
0160      470 Y(I)=HUBC(I)
0161      GO TO 570

```

C

```

C 480 GO TO (490,520), NLIQ
0162      480 GO TO (490,539), NLIQ
0163      490 DO 500 I=1,NC
0164      IF (L(I)) 492,494,496
0165      492 Y(I)=0.0
0166      X(I,1)=Z(I)
0167      GO TO 500
0168      494 Y(I)=Z(I)
0169      X(I,1)=0.0
0170      GO TO 500
0171      496 Y(I)=Z(I)/2.0
0172      X(I,1)=Y(I)
0173      500 CONTINUE
0174      HY=0.0
0175      HX=0.0
0176      DO 505 I=1,NC
0177      HY=HY+Y(I)
0178      505 HX=HX+X(I,1)
0179      DO 510 I=1,NC
0180      Y(I)=Y(I)/HY
0181      510 X(I,1)=X(I,1)/HX
0182

```

C 520 DO 540 I=1,NC

C IF (L(I)) 522,528,530

C 522 Y(I)=0.0

C IF (I.EQ.1) GO TO 524

C IF (I.EQ.NC) GO TO 526

C X(I,1)=Z(I)/2.0

C X(I,2)=X(I,1)

C GO TO 540

C 524 X(I,1)=Z(I)

C X(I,2)=1.0E-12

C GO TO 540

C 526 X(I,1)=1.0E-12

C X(I,2)=Z(I)

C GO TO 540

C 528 Y(I)=Z(I)

C X(I,1)=0.0

C X(I,2)=0.0

C GO TO 540

C 530 Y(I)=Z(I)/2.0

C IF (I.EQ.1) GO TO 532

C IF (I.EQ.NC) GO TO 534

C X(I,1)=Z(I)/4.0

C X(I,2)=X(I,1)

C GO TO 540

C 532 X(I,1)=Z(I)/2.0

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```

C      X(I,2)=1.0E-12
C      GO TO 540
C 534  X(I,1)=1.0E-12
C      X(I,2)=Z(I)/2.0
C 540  CONTINUE
0183      539  DO 543 I=1,NC
0184      Y(I) =Z(I)
0185      DO 543 J=1,NLIQ
0186      543  X(I,J)=XSAVE(I,J)
0187      SUM=0.0
0188      UY=0.0
0189      UX=0.0
0190      DO 542 I=1,NC
0191      SUM=SUM+Y(I)
0192      UY=UY+X(I,1)
0193      542  UX=UX+X(I,2)
0194      DO 544 I=1,NC
0195      Y(I)=Y(I)/SUM
0196      X(I,1)=X(I,1)/UY
0197      544  X(I,2)=X(I,2)/UX
0198      GO TO 570
C
0199      550  DO 552 I=1,NC
0200      552  Y(I)=Z(I)
0201      GO TO (554,558), NLIQ
0202      554  DO 556 I=1,NC
0203      556  X(I,1)=DFWC(I)
0204      GO TO 570
C 558  X(I,1)=DFWC(1)
C      X(I,2)=1.0E-12
C      X(NC,1)=1.0E-12
C      X(NC,2)=DFWC(NC)
C      DO 560 J=2,NCM1
C      X(I,1)=DFWC(1)/2.0
C 560  X(J,2)=X(I,1)
0205      558  DO 561 I=1,NC
0206      DO 561 J=1,NLIQ
0207      561  X(I,J)=XSAVE(I,J)
0208      DO 566 N=1,2
0209      SUM=0.0
0210      DO 562 I=1,NC
0211      562  SUM=SUM+X(I,N)
0212      DO 564 I=1,NC
0213      564  X(I,C)=X(I,N)/SUM
0214      566  CONTINUE
C
0215      570  I=TF
0216      IF (IS0)580,590,580
0217      580  WRITE (6,581)
0218      581  FORMAT ('RESULTS OF ADIABATIC FLASH CALCULATION')
0219      CALL AFLASH
0220      GO TO 600
0221      590  WRITE (6,591)
0222      591  FORMAT ('RESULTS OF ISOTHERMAL FLASH CALCULATION')
0223      GO TO (592,593), NLIQ
0224      592  CALL FLASH1

```

```

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0225      GO TO 600
0226      593 CALL FLASH2
          C
0227      600 WRITE (6,610) T,P,FRACV
0228      610 FORMAT('0SYSTEM IS IN THE V/L PHASE REGION AT',F7.2,' DEG K AND',
          * F7.2,' MMHG. VAPOR/FEED MOLE RATIO =',F9.6)
0229      IF (NLIQ.EQ.2) WRITE (6,285) FRACL
          C
0230      CALL PRINT
          C
0231      CALL LIQH
0232      CALL VAPH
0233      WRITE (6,430) (HIDL(N),DHL(N),N=1,NLIQ)
0234      WRITE (6,370) IDLV,DHV
0235      HVAPOR=HREFALV*FRACV
0236      GO TO (631,632),NLIQ
0237      631 HLIQ(1)=HREAL(1)*(1.0-FRACV)
0238      HLIQ(2)=0.0
0239      GO TO 633
0240      632 HLIQ(1)=HREAL(1)*(1.0-FRACV)*FRACL
0241      HLIQ(2)=HREAL(2)*(1.0-FRACV)*(1.0-FRACL)
0242      633 WRITE (6,640) HVAPOR,(HLIQ(N),N=1,NLIQ)
0243      640 FORMAT('0ENTHALPY OF VAPOR PHASE/LBMOL OF FEED =',F10.2,' CAL'/
          *(' ENTHALPY OF LIQUID PHASE/LBMOL OF FEED =',F10.2,' BTU'))
0244      H=HVAPOR+HLIQ(1)
0245      IF (NLIQ.EQ.2) H=H+HLIQ(2)
          C
0246      650 WRITE (6,660) TREF,H
0247      660 FORMAT('0ENTHALPY OF SYSTEM WITH REFERENCE TO PURE LIQUID AT',
          *F7.2,' DEG K ',F10.2,' BTU/LBMOL')
          C
0248      DO 670 I=1,2
0249      DO 670 J=1,NC
0250      670 XSAVE(J,I)=X(J,I)
0251      RETURN
          C
0252      END

```

```

001      SUBROUTINE      DEWPT
      C
      C      DEW POINT PROGRAM SOLVING FOR THE TEMPERATURE
      C
      C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
      C
002      COMMON /R2L/      FD(11),Z(10),EK(20),PF,TF,HI,XSAVE(10,2)
003      COMMON /CONTRL/   T,P,NOBUB,NODEW,ISO,NLIQ
004      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
005      COMMON /LIQUID/   X(20),GAM(20),DHL(2),HIDLL(2),HREALL(2),
      +                   FRACL
006      COMMON /DUMMY/    XX(20),SX(20),XC(20),YY(10),SY(10),
      +                   YC(10),RMAX,RMTN
007      COMMON /CTRL/     NCR, NPRT, NSTART, TITLE(20)
008      COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LDBG, ISW, NDIM,
      *                   NAME(2,10),L(10),COMPNT(10)
009      COMMON /NJIT1/    CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
      *                   W(10)
010      COMMON /NJIT2/    C(180), ALPHA(45), VC(10), VOL(10), TC(10),
      *                   PC(10), DA(10), OB(10), AZ(10,10), G(10,10),
      *                   THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
      *                   Q(10),RP(10),XL(10),PHI(10),DUMMY(10)
011      COMMON /NJIT3/    TTC(10,10),PPC(10,10),VVC(10,10),Zzc(10,10),
      *                   AK(10,10), AVALUE(10,10), BVALUE(10,10),
      *                   HCROSS(10,10), OMEGA(10), DIPOLE(10), BZ(10,10)
      C
012      DATA              CONV/1.0E-5/, CONV2/1.0E-4/
013      DATA              STEP/20.0/
      C
      C      CALLING PROGRAM MUST PROVIDE INITIAL GUESS FOR T
      C
014      KNT=0
015      ISTR=IDLL
016      IDLL=0
017      ISW=0
018      IDX=1
019      HD=0.0
020      DO 5 I=1,NC
021      X(I)=0.0
022      EK(I)= 0.0
023      5 XX(I)=0.0
      C
024      10 ID=1
025      IF (LDBG.GE.1) WRITE (6,20) T,(X(I),I=1,9),(EK(I),I=1,9)
026      20 FORMAT ('0DEWPT  T/X-K', F12.4/(9E14.5))
027      S=T
028      K=0
      C
029      30 CALL KCALC
030      V=-1.0
031      DO 0 I=1,NC
032      IF (L(I)) 50,40,50
033      40 XX(I)=0.0
034      GO TO 60
035      50 XX(I)=Y(I)/EK(I)
036      V=V+XX(I)

```

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```

0037      60 CONTINUE
0038      K=K+1
C
0039      IF (LDBUG.GE.2) WRITE (6,70) T,V
0040      70 FORMAT ('      T&SUM', F12.4,E14.5)
0041      IF (LDBUG.GE.2) WRITE (6,75) (X(I),I=1,9)
0042      75 FORMAT ('OX=ASS  ',9E13.5)
C
0043      IF (K.LT.20) GO TO 80
0044      71 WRITE (6,72)
0045      72 FORMAT ('ODEM POINT CALCULATION TERMINATED'/
+      ' CALCULATION CONTINUING' ///)
0046      SUM=0.0
0047      DO 74 I=1,NC
0048      74 SUM=SUM+XX(I)
0049      GO TO 175
C
0050      80 IF (ABS(V).LT.CONV) GO TO 130
C
0051      GO TO (90,100), ID
0052      90 T=S+SIGN(5.0,V)
0053      HD=V
0054      ID=2
0055      GO TO 30
0056      100 HD=V*(S-T)/(HD-V)
0057      S=T
0058      IF (ABS(HD)-STEP)120,120,110
0059      110 HD=SIGN(STEP,HD)
0060      120 T=T-HD
0061      HD=V
0062      GO TO 30
C
0063      130 IDLL=ISTR
0064      IF (IDLL)170,170,140
0065      140 GO TO (200,150,150),IDX
0066      150 DO160I=1,NC
0067      IF ( (L(I).EQ.0) .OR. (Z(I).LT. 1.0E-12) ) GO TO 160
0068      IF (ABS(X(I)-XX(I))/X(I) .GT. CONV2) GO TO 190
0069      160 CONTINUE
0070      170 SUM=V+1.0
0071      175 DO 180 I=1,NC
0072      180 X(I)=XX(I)/SUM
0073      RETURN
C
0074      190 KNT=KNT+1
0075      IF (KNT.EQ.50) GO TO 71
0076      GO TO (200,220,240), IDX
C
0077      200 IDX=2
0078      DO210I=1,NC
0079      210 X(I)=XX(I)
0080      215 SUM=V+1.0
0081      GO TO 255
0082      220 IDX=3
0083      DO230I=1,NC
0084      SX(I)=X(I)

```

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```
0085          XC(I)=XX(I)
0086          230 X(I)=XX(I)
0087          GO TO 215
0088          240 SUM=0.0
0089          DO250 I=1,NC
0090          IF (L(I).EQ.0 .OR. Z(I).LT.1.0E-12) GO TO 250
0091          CALL WEGS (X(I),XX(I),SX(I),XC(I))
0092          SUM=SUM+X(I)
0093          250 CONTINUE
          C
0094          255 DO 260 I=1,NC
0095          260 X(I)=X(I)/SUM
0096          GO TO 10
          C
0097          END
```

```

001      SUBROUTINE      BURPT1
      C
      C      BUBBLE POINT PROGRAM SOLVING FOR THE TEMPERATURE *** ONE LIQUID
      C
      C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
      C
002      COMMON /B2L/      FD(11),Z(10),EK(20),PF,TF,HI,XSAVE(10,2)
003      COMMON /CONTRL/   T,P,NQBUB,NOBEM,ISO,NLIQ
004      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
005      COMMON /LIQUID/   X(20),GAM(20),DHL(2),HIDL(2),HREAL(2),
      +                   FRACL
006      COMMON /DUMMY/    XX(20),SX(20),XC(20),YY(10),SY(10),
      +                   YC(10),RMAX,RMIN
007      COMMON /CTRL/     NCR, NPRT, NSTART, TITLE(20)
008      COMMON /NJIT1/    NC, NCM1, IDLL, IDLV, IDH, LDEBUG, ISW, NDIM,
      *                   NAME(2,10),L(10),COMPNT(10)
009      COMMON /NJIT1/    CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
      *                   W(10)
010      COMMON /NJIT2/    C(180),ALPHA(45),VC(10),VOL(10),TC(10),
      *                   PC(10),QA(10),QB(10),AZ(10,10),G(10,10),
      *                   THETA(10),THS(10),HFAC(10),HFPA(10),HM(10),
      *                   Q(10),RR(10),XL(10),PHI(10),DUMMY(10)
011      COMMON /NJIT3/    TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
      *                   AK(10,10),AVALUE(10,10),BVALUE(10,10),
      *                   BCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
      C
012      DATA              CONV/1.0E-5/, CONV2/1.0E-4/
013      DATA              STEP/20.0/
      C
      C      CALLING PROGRAM MUST PROVIDE INITIAL GUESS FOR T
      C
014      KNT=0
015      ISTR=IDLV
016      IDLV=0
017      ISW=0
018      IDX=1
019      HD=0.0
020      DO 5 I=1,NC
021      Y(I)=0.0
022      YY(I)=0.0
023      5 EK(I)=0.0
      C
024      10 ID=1
025      IF (LDEBUG.GE.1) WRITE (6,20) T,(Y(I),I=1,9),(EK(I),I=1,9)
026      20 FORMAT ('0BURPT#1 T/Y-K', F12.4/'(9E14.5)')
027      S=T
028      K=0
      C
029      30 CALL KCALC
030      V=-1.0
031      DO 60 I=1,NC
032      IF (L(I)) 40,50,50
033      40 YY(I)=0.0
034      GO TO 60
035      50 YY(I)=EK(I)*X(I)
036      V=V+YY(I)

```

```

0037      60 CONTINUE
0038      K=K+1
      C
0039      IF (LDBUG.GE.1) WRITE (6,70) T,V
0040      70 FORMAT (5X, 'TASUM', F12.4,E15.5)
0041      IF (LDBUG.GE.2) WRITE (6,75) (Y(I),I=1,9)
0042      75 FORMAT ('OY-ASS ',9E13.5)
      C
0043      IF (K.LT.20) GO TO 80
0044      71 WRITE (6,72)
0045      72 FORMAT('ORHURLE POINT (1 LIQUID PHASE) CALCULATION TERMINATED'/
+          ' CALCULATION CONTINUING' ///)
0046      SUM=0.0
0047      DO 74 I=1,NC
0048      74 SUM=SUM+YY(I)
0049      GO TO 195
      C
0050      80 IF (ABS(V).LT.CONV) GO TO 150
      C
0051      GO TO (90,100), ID
0052      90 T=S-SIGN(5.0,V)
0053      HD=V
0054      ID=2
0055      GO TO 50
0056      100 SLOPE=(HD-V)/(S-T)
0057      IF (SLOPE.GT.0.0) GO TO 120
0058      NORUR=2
0059      IDLV=ISTR
0060      RETURN
0061      120 HD=V/SLOPE
0062      S=T
0063      IF (ABS(HD)-STEP)140,140,130
0064      130 HD=SIGN(STEP,HD)
0065      140 T=T-HD
0066      HD=V
0067      GO TO 30
      C
0068      150 IDLV=ISTR
0069      IF (IDLV)190,190,160
0070      160 GO TO (220,170,170),IDX
0071      170 DO180I=1,NC
0072      IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 180
0073      IF (ABS(Y(I)-YY(I))/Y(I)-CONV2)180,180,210
0074      180 CONTINUE
0075      190 SUM=V+1.0
0076      195 DO 200 I=1,NC
0077      200 Y(I)=YY(I)/SUM
0078      IDLV=ISTR
0079      RETURN
      C
0080      210 KNT=KNT+1
0081      IF (KNT.EQ.50) GO TO 71
0082      GO TO (220,240,260), IDX
      C
0083      220 IDX=2
0084      DO230I=1,NC

```

```
0085      230 Y(I)=YY(I)
0086      235 SUM=V+1.0
0087          GO TO 275
0088      240 IDX=3
0089          DO250 I=1,NC
0090              SY(I)=Y(I)
0091              YC(I)=YY(I)
0092      250 Y(I)=YY(I)
0093          GO TO 235
0094      260 SUM=0.0
0095          DO270 J=1,NC
0096              IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 270
0097              CALL WEGS(Y(I),YY(I),SY(I),YC(I))
0098              SUM=SUM+Y(I)
0099      270 CONTINUE
C
0100      275 DO 280 J=1,NC
0101      280 Y(T)=Y(I)/SUM
0102          GO TO 10
C
0103          END
```



```

0001      SUBROUTINE      RUBPT2
          C
          C      BUBBLE POINT PROGRAM SOLVING FOR THE TEMPERATURE *** TWO LIQUIDS
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
          C
0002      COMMON /H2L/      FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
0003      COMMON /CONTRL/   T,P,NOBUB,NODEW,ISO,NLIQ
0004      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
0005      COMMON /LIQUID/   X(10,2),GAM(10,2),DHL(2),HIDLL(2),HREALL(2),
          +
          + FRACL
0006      COMMON /DUMMY/    XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
          +
          + YC(10),RMAX,RMIN
0007      COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)
0008      COMMON /NJIT0/   NC, NCM1, IDLL, IDLV, IDH, LDHUG, ISW, NDIM,
          *
          * NAME(2,10),L(10),COMPNT(10)
0009      COMMON /NJIT1/   CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
          *
          * W(10)
0010      COMMON /NJIT2/   C(180),ALPHA(45),VC(10),VOL(10),TC(10),
          *
          * PC(10),OA(10),OB(10),AZ(10,10),G(10,10),
          *
          * THETA(10),THS(10),HFAC(10),HHFA(10),HM(10),
          *
          * Q(10),RR(10),XL(10),PHI(10),DUMMY(10)
0011      COMMON /NJIT3/   TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
          *
          * AK(10,10),AVALUE(10,10),BVALUE(10,10),
          *
          * BCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
0012      DATA            CONV/1.0E-5/, CONV2/1.0E-4/
0013      DATA            STEP/20.0/
          C
          C      CALLING PROGRAM MUST INITIALIZE T & FRACL
          C
          C
0014      KNT=0
0015      ISTR=IDLV
0016      IDLV=0
0017      ISW=0
0018      IDX=1
0019      HD=0.0
0020      DO 5 I=1,NC
0021      Y(I)=0.0
0022      YY(I)=0.0
0023      DO 5 N=1,2
0024      5 EK(I,N)=0.0
          C
0025      10 ID=1
0026      IF (LDHUG.GE.1) WRITE (6,20) T,(Y(I),I=1,9),(EK(I,1),I=1,9),
          +
          + (EK(I,2),I=1,9)
0027      20 FORMAT('0RUBPT#2  T/Y-K1-K2',F12.4/(9E14.5))
0028      S=T
0029      K=0
          C
0030      30 CALL TADLTO
0031      40 CALL KCALC
0032      V=-1.0
0033      DO60I=1,NC
0034      IF (L(I).GE.0) GO TO 50
0035      YY(I)=0.0

```

```

0036      GO TO 60
0037      50 YY(1)=EK(I,1)*X(I,1)
0038      V=V+YY(I)
0039      60 CONTINUE
0040      K=K+1
      C
0041      IF (LDBUG.GE.1) WRITE (6,70) T,V
0042      70 FORMAT ('0', 4X, 'T&SUM', F12.4,E14.5)
0043      IF (LDBUG.GE.2) WRITE (6,75) (Y(I),I=1,9)
0044      75 FORMAT ('0Y=ASS ',9E13.5)
      C
0045      IF (K.LT.20) GO TO 80
0046      71 WRITE (6,72)
0047      72 FORMAT ('ORURBLE POINT (2 LIQUID PHASE) CALCULATION TERMINATED'/
+          ' CALCULATION CONTINUING' ///)
0048      SUM=0.0
0049      DO 74 I=1,NC
0050      74 SUM=SUM+YY(I)
0051      GO TO 195
      C
0052      80 IF (ABS(V).LT.CONV) GO TO 150
      C
0053      GO TO (90,100), ID
0054      90 T=S-SIGN(5.0,V)
0055      HD=V
0056      ID=2
0057      GO TO 40
0058      100 HD=(HD-V)/(S-T)
0059      HD=V/HD
0060      S=T
0061      IF (ABS(HD)-STEP)140,140,130
0062      130 HD=SIGN(STEP,HD)
0063      140 T=T-HD
0064      HD=V
0065      GO TO 30
      C
0066      150 IDLV=ISTR
0067      IF (IDLV)190,190,160
0068      160 GO TO (220,170,170),IDX
0069      170 DO180I=1,NC
0070      IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 180
0071      IF (ABS(Y(I)-YY(I))/Y(I)-CONV2)180,180,210
0072      180 CONTINUE
0073      190 SUM=V+1.0
0074      195 DO 200 I=1,NC
0075      200 Y(I)=YY(I)/SUM
0076      IDLV=ISTR
0077      RETURN
      C
0078      210 KNT=KNT+1
0079      IF (KNT.EQ.50) GO TO 71
0080      GO TO (220,240,260), IDX
      C
0081      220 IDX=2
0082      DO230I=1,NC
0083      230 Y(I)=YY(I)

```

```
0084          235 SUM=V+1.0
0085          GO TO 275
0086          240 IDX=3
0087          DO250 I=1,NC
0088          SY(I)=Y(I)
0089          YC(I)=YY(I)
0090          250 Y(I)=YY(I)
0091          GO TO 235
0092          260 SUM=0.0
0093          DO270 I=1,NC
0094          IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 270
0095          CALL WEGS(Y(I),YY(I),SY(I),YC(I))
0096          SUM=SUM+Y(I)
0097          270 CONTINUE
          C
0098          275 DO 280 I=1,NC
0099          280 Y(I)=Y(I)/SUM
0100          GO TO 10
          C
0101          END
```

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TWOLIQ

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```

0001      SUBROUTINE      TWOLIQ
C
C      TWO PHASE MATERIAL BALANCE USED FOR
C      * BUBBLE POINT
C      * SUBCOOLED LIQUID
C
C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
C
0002      DIMENSION      R(9)
C
0003      COMMON /B2L/      FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
0004      COMMON /CONTRL/   T,P,NOBUB,NODEW,ISO,NLIQ
0005      COMMON /VAPOR/   Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
0006      COMMON /LIQUID/  X(10,2),GAM(10,2),DHL(2),HIDLL(2),HREALL(2),
+      FRACL
0007      COMMON /DUMMY/   XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
+      YC(10),RMAX,RMIN
0008      COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)
0009      COMMON /NJIT0/   NC, NCM1, IDLL, IDLV, IDH, LDEBUG, ISW, NDIM,
+      NAME(2,10),I(10),COMPNT(10)
0010      COMMON /NJIT1/   CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
+      W(10)
0011      COMMON /NJIT2/   C(180),ALPHA(45),VC(10),VOL(10),TC(10),
+      PC(10),UA(10),OB(10),A7(10,10),G(10,10),
+      THETA(10),THS(10),HFAC(10),HHFA(10),HM(10),
+      Q(10),RR(10),XL(10),PHI(10),DUMMY(10)
0012      COMMON /NJIT3/   TTC(10,10),PPC(10,10),VVC(10,10),ZC(10,10),
+      AK(10,10),AVALUE(10,10),BVALUE(10,10),
+      BCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
C
0013      DATA          CONV1/1.0E-5/, CONV1/1.0E-4/
C
0014      KNT=0
0015      IDX=1
0016      ISAVE=ISW
0017      ISW=0
0018      CALL ACTVTY (T,X(1,1),GAM(1,1))
0019      CALL ACTVTY (T,X(1,2),GAM(1,2))
0020      ISW=1
0021      GO TO 30
C
0022      20 CALL ACTVTY (T,X(1,1),GAM(1,1))
0023      CALL ACTVTY (T,X(1,2),GAM(1,2))
0024      30 DO 40 I=1,NC
0025      40 R(I)=GAM(I,1)/GAM(I,2)
0026      K=0
C
0027      IF (LDEBUG.GE.2) WRITE (6,45) (X(I,1),I=1,9),(X(I,2),I=1,9)
0028      45 FORMAT ('0X1-ASS ',9E13.5/' X2-ASS ',9E13.5)
C
0029      50 F2=0.0
0030      DF2=0.0
0031      DDF2=0.0
0032      DO 60 I=1,NC
0033      IF (L(I).EQ.0 .OR. Z(I).LT.1.0E-12) GO TO 60
0034      B=Z(I)*(1.0-R(I))

```

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0035      WW=FRACL+(1.0-FRACL)*R(I)
0036      DWW=1.0-R(I)
0037      F2=F2+B/WW
0038      DF2=DF2-B/WW**2*DWW
0039      DDF2=DDF2+2.0*B/WW**3*DWW**2
0040      60 CONTINUE
0041      CC=2.0*F2*DF2
          C      FF=2.0*DF2**2+4.0*F2*DDF2
0042      FF=2.0*DF2**2-F2*DDF2
0043      DEL=-CC/FF
0044      K=K+1
          C
0045      IF (LDBUG.GE.1) WRITE (6,62) T,F2,FRACL,DEL
0046      62 FORMAT (5X, 'T=F2=FRACL=DEL', F12.4,3E14.5)
          C
0047      IF (K.LT.50) GO TO 70
0048      63 WRITE (6,64)
0049      64 FORMAT ('0TWOLIQ CALCULATION TERMINATED'/
          +      ' CALCULATION CONTINUING' ///)
0050      DO 66 I=1,NC
0051      WW=FRACL+(1.0-FRACL)*R(I)
0052      X(I,1)=Z(I)/WW
0053      66 X(I,2)=X(I,1)*R(I)
0054      DO 69 N=1,2
0055      SUM=0.0
0056      DO 67 J=1,NC
0057      67 SUM=SUM+X(I,N)
0058      DO 68 I=1,NC
0059      68 X(I,N)=X(I,N)/SUM
0060      69 CONTINUE
0061      ISN=ISAVE
0062      RETURN
          C
0063      70 NCVRG=0
0064      75 IF (FRACL+DEL.LT.1.0 .AND. FRACL+DEL.GT.0.0) GO TO 80
0065      NCVRG=1
0066      DEL=DEL/2.0
0067      GO TO 75
          C
0068      80 IF (FRACL.LT.0.99) GO TO 85
0069      FRACL=1.0-FRACL
0070      DO 81 I=1,NC
0071      HOLD=X(I,1)
0072      X(I,1)=X(I,2)
0073      X(I,2)=HOLD
0074      HOLD=XX(I,1)
0075      XX(I,1)=XX(I,2)
0076      XX(I,2)=HOLD
0077      HOLD= SX(I,1)
0078      SX(I,1)=SX(I,2)
0079      SX(I,2)=HOLD
0080      HOLD=XC(I,1)
0081      XC(I,1)=XC(I,2)
0082      XC(I,2)=HOLD
0083      HOLD=GAM(I,1)
0084      GAM(I,1)=GAM(I,2)

```

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```

0085          GAM(I,2)=HOLD
0086          81 CONTINUE
0087          85 IF (FRACL.LT.1.E-10) GO TO 166
      C
0088          90 IF (ABS(F2).LT.CONV .AND. ABS(DEL).LT.CONV .AND. NCVRG.EQ.0)
      +      GO TO 95
0089          FRACL=FRACL+DEL
0090          GO TO 50
      C
0091          95 DO 100 I=1,NC
0092          WW=FRACL+(1.0-FRACL)*R(I)
0093          XX(I,1)=Z(I)/WW
0094          100 XX(I,2)=XX(I,1)*R(I)
      C
0095          KNT=KNT+1
0096          IF (KNT.EQ.50) GO TO 63
0097          GO TO (110,150), IDX
      C
0098          110 IDX=2
0099          DO 140 N=1,2
0100          SUM=0.0
0101          DO 120 I=1,NC
0102          SX(I,N)=X(I,N)
0103          XC(I,N)=XX(I,N)
0104          X(I,N)=(X(I,N)+XX(I,N))/2.0
0105          120 SUM=SUM+X(I,N)
0106          DO 130 I=1,NC
0107          130 X(I,N)=X(I,N)/SUM
0108          140 CONTINUE
0109          GO TO 20
      C
0110          150 DO 160 I=1,NC
0111          IF (L(I).EQ.0 .OR. Z(I).LT.1.0E-12) GO TO 160
      C          IF (ABS(X(I,1)*GAM(I,1)-X(I,2)*GAM(I,2))/(X(I,2)*GAM(I,2)) .GT.
      C          +      CONV1) GO TO 190
0112          IF (ABS(X(I,1)-XX(I,1))/X(I,1) .GT. CONV1 .OR.
      *      (ABS(X(I,2)-XX(I,2))/X(I,2)) .GT. CONV1) GO TO 190
0113          160 CONTINUE
0114          GO TO 170
      C
0115          166 DO 167 I=1,NC
0116          X(I,1)=Z(I)
0117          167 X(I,2)=0.0
0118          NLIQ=1
0119          WRITE (6,168)
0120          168 FORMAT (// 5X, 'LIQUID HAS GONE TO A ONE PHASE SYSTEM' //)
0121          RETURN
      C
0122          170 DO 180 N=1,2
0123          SUM=0.0
0124          DO 172 I=1,NC
0125          172 SUM=SUM+X(I,N)
0126          DO 176 I=1,NC
0127          176 X(I,N)=X(I,N)/SUM
0128          180 CONTINUE
0129          ISW=ISAVE

```

```
0130          RETURN
          C
0131      190 DO 220 N=1,2
0132          SUM=0.0
0133          DO 200 I=1,NC
0134          IF (L(I).EQ.0 .OR. Z(I).LT.1.0E-12) GO TO 200
0135          CALL WEGS (X(I,N),XX(I,N),SX(I,N),XC(I,N))
0136          SUM=SUM+X(I,N)
0137      200 CONTINUE
0138          DO 210 I=1,NC
0139      210 X(I,N)=X(I,N)/SUM
0140      220 CONTINUE
0141          GO TO 20
          C
0142          END
```

ORTRAN IV G LEVEL 21

FLASH1

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```

0001      SUBROUTINE      FLASH1
          C
          C      ISOTHERMAL FLASH *** ONE LIQUID
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (J-N)
          C
0002      COMMON /B2L/      FD(11),Z(10),EK(20),PF,TF,HI,XSAVE(10,2)
0003      COMMON /CONTRL/   T,P,NOBUB,NOREW,ISO,NLIQ
0004      COMMON /VAPOR/   Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
0005      COMMON /LIQUID/   X(20),GAM(20),DHL(2),HIDL(2),HREAL(2),
          +                FRACL
0006      COMMON /DUMMY/    XX(20),SX(20),XC(20),YY(10),SY(10),
          +                YC(10),RMAX,RMIN
0007      COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)
0008      COMMON /NJIT0/   NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW, NDIM,
          *                NAME(2,10),L(10),COMPNT(10)
0009      COMMON /NJIT1/   CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
          *                W(10)
0010      COMMON /NJIT2/   C(180),ALPHA(45),VC(10),VOL(10),TC(10),
          *                PC(10),OA(10),OB(10),AZ(10,10),G(10,10),
          *                THETA(10),THS(10),HFAC(10),HHFA(10),HM(10),
          *                W(10),RR(10),XL(10),PHI(10),DUMMY(10)
0011      COMMON /NJIT3/   TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
          *                AK(10,10),AVALUE(10,10),BVALUE(10,10),
          *                NCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
          C
0012      EQUIVALENCE      (R,FRACV)
          C
0013      DATA            CONV1/1.0E-4/, CONV1/1.0E-5/, CONV2/1.0E-5/
          C
          C      R IS VAPOR FRACTION
          C      CALLING PROGRAM MUST SUPPLY INITIAL VALUE FOR R,X,Y
          C
0014      KNT=0
0015      IDX=1
0016      ISW=0
0017      CALL KCALC
0018      ISW=1
0019      IF (NC.GT.1) GO TO 21
          C
0020      Y(1)=1.0
0021      X(1)=1.0
0022      ISO=0
0023      RETURN
          C
0024      20 CALL KCALC
          C
0025      21 IF (LDBUG.GE.1) WRITE (6,25) T,R
0026      25 FORMAT ('0FLASH#1 T&R',F12.4,E14.5)
0027      IF (LDBUG.GE.2) WRITE (6,26) (Y(I),I=1,9),(X(I),I=1,9)
0028      26 FORMAT ('0Y-ASS ',9E13.5/' X-ASS ',9E13.5)
          C
0029      29 K=0
0030      30 AMAX=RMAX
0031      AMIN=RMIN
0032      F1=0.0

```


ORTRAN IV G LEVEL 21

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```

0033      DF=0.0
0034      DO80J=1,NC
0035      IF (L(I))40,50,60
0036      40 S=1.0/(1.0-R)
0037      GO TO 70
0038      50 S=-1.0/R
0039      GO TO 70
0040      60 S=(1.0-EK(I))/(1.0+R*(EK(I)-1.0))
0041      70 F1=F1+Z(I)*S
0042      80 DF=DF+Z(I)*S*S
0043      K=K+1
      C
0044      IF (LDBUG.GE.2) WRITE (6,90) F1,DF1,R
0045      90 FORMAT (5X, 'F1-DF1-R',3E14.5)
      C
0046      IF (K.LT.20) GO TO 99
0047      91 WRITE (6,92)
0048      92 FORMAT ('OFLASH#1 CALCULATION TERMINATED'/
+          ' CALCULATION CONTINUING' ///)
0049      GO TO 320
      C
0050      99 IF (ABS(F1).LT.CONV2) GO TO 190
      C 99 IF (F1) 100,190,110
      C 100 AMIN=R
      C GO TO 120
      C 110 AMAX=R
      C 120 IF ((AMAX-AMIN).LT.CONV1) GO TO 170
      C DF1=-F1/DF1
      C 130 F1=R+DF1
      C IF (F1-AMIN)140,150,150
      C 140 DF=DF*0.5
      C GO TO 130
      C 150 IF (F1-AMAX)160,160,140
      C 160 R=F1
0051      IF (ABS(R-AMIN).LT.CONV1 .OR. ABS(R-AMAX).LT.CONV1) GO TO 190
0052      DR=-F1/DF1
0053      130 RNEXT=R + DR
0054      IF (RNEXT.GE.AMIN) GO TO 150
0055      140 DR=DR/2.0
0056      GO TO 130
0057      150 IF (RNEXT.GT.AMAX) GO TO 140
0058      R=RNEXT
      C
      C IF (ABS(DF)-CONV1)190,30,30
0059      IF (ABS(DR).GT. CONV1) GO TO 30
      C
      C 170 DO180I=1,NC
      C 180 X(I)=(X(I)+SX(I))/2.0
      C GO TO 20
      C
0060      190 IF (LDBUG.GE.2) WRITE (6,200) R
0061      200 FORMAT (5X, 'R', E14.5)
0062      DO240I=1,NC
0063      IF (L(I))210,220,230
0064      210 XX(I)=Z(I)/(1.0-R)
0065      YY(I)=0.0

```

```

0066      GO TO 240
0067      220 XX(I)=0.0
0068          YY(I)=Z(I)/R
0069      GO TO 240
0070      230 XX(I)=Z(I)/(1.0+R*(EK(I)-1.0))
0071          YY(I)=EK(I)*XX(I)
0072      240 CONTINUE
0073      IF (IDLL+IDLV.EQ.0) GO TO 320
      C
0074      250 KNT=KNT+1
0075          IF (KNT.EQ.50) GO TO 91
0076          GO TO (340,260), IDX
      C
0077      260 IF (IDLL.EQ.0) GO TO 290
0078          DO280 I=1,NC
0079          IF ((L(I).EQ.0).OR.(Z(I).LT.1.0E-12)) GO TO 280
0080          IF (ABS(X(I)-XX(I))/X(I) .GT. CONV) GO TO 410
0081      280 CONTINUE
      C
0082      290 IF (IDLV.EQ.0) GO TO 320
0083          DO310 I=1,NC
0084          IF ((L(I).LT.0).OR.(Z(I).LT.1.0E-12)) GO TO 310
0085          IF (ABS(Y(I)-YY(I))/Y(I)-CONV)310,410,410
0086      310 CONTINUE
      C
0087      320 UX=0.0
0088          UY=0.0
0089          DO 330 I=1,NC
0090              X(I)=XX(I)
0091              Y(I)=YY(I)
0092              UX=UX+X(I)
0093      330 UY=UY+Y(I)
0094          WRITE (6,331) UX,UY
0095      331 FORMAT(' FLASH1 SUM UX/UY',2G20.6)
0096          DO 335 I=1,NC
0097              X(I)=X(I)/UX
0098      335 Y(I)=Y(I)/UY
0099          ISM=0
0100          RETURN
      C
0101      340 IF (IDLL.EQ.0) GO TO 370
0102          DO 360 I=1,NC
0103              SX(I)=X(I)
0104              XC(I)=XX(I)
0105      360 X(I)=(XX(I)+X(I))/2.0
0106      370 IF (IDLV.EQ.0) GO TO 400
0107          DO 390 I=1,NC
0108              SY(I)=Y(I)
0109              YC(I)=YY(I)
0110      390 Y(I)=(YY(I)+Y(I))/2.0
0111      400 IDX=2
0112          GO TO 20
      C
0113      410 IF (IDLL.EQ.0) GO TO 450
0114          SUM=0.0
0115          DO430 I=1,NC

```

FORTRAN IV G LEVEL 21

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0116          IF ((L(I).EQ.0).OR.(Z(I).LT.1.0E-12)) GO TO 430
0117          CALL WEGS (X(I),XX(I),SX(I),XC(I))
0118          SUM=SUM+X(I)
0119          430 CONTINUE
0120          DO440 I=1,NC
0121          440 X(I)=X(I)/SUM
          C
0122          450 IF (IDLV.EQ.0) GO TO 20
0123          SUM=0.0
0124          DO470 I=1,NC
0125          IF ((L(I).LT.0).OR.(Z(I).LT.1.0E-12)) GO TO 470
0126          CALL WEGS(Y(I),YY(I),SY(I),YC(I))
0127          SUM=SUM+Y(I)
0128          470 CONTINUE
0129          DO480 I=1,NC
0130          480 Y(I)=Y(I)/SUM
0131          GO TO 20
          C
0132          END
```

```

0001      SUBROUTINE      FLASH2
          C
          C      ISOTHERMAL FLASH *** TWO LIQUIDS
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
0002      DIMENSION      R(9), ZJ(2,2), RHS(2)
          C
0003      COMMON /B2L/      FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
0004      COMMON /CONTRL/   T,P,NOBUR,NODEW,ISO,NLIQ
0005      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
0006      COMMON /LIQUID/   X(10,2),GAM(10,2),DHL(2),HIDLL(2),HREALL(2),
          +               FRACL
0007      COMMON /DUMMY/    XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
          +               YC(10),RMAX,RMIN
0008      COMMON /CIRL/     NCR, NPRT, NSTART, TITLE(20)
0009      COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW, NDIM,
          *               NAME(2,10),L(10),COMPNT(10)
0010      COMMON /NJIT1/    CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
          *               W(10)
0011      COMMON /NJIT2/    C(180),ALPHA(45),VC(10),VOL(10),TC(10),
          *               PC(10),OA(10),OH(10),AZ(10,10),G(10,10),
          *               THETA(10),THS(10),HFAC(10),HHFA(10),HM(10),
          *               Q(10),RR(10),XL(10),PHI(10),DUMMY(10)
0012      COMMON /NJIT3/    TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
          *               AK(10,10),AVALUE(10,10),BVALUE(10,10),
          *               BCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
          C
0013      EQUIVALENCE      (U,FRACV), (V,FRACL)
          C
0014      DATA            CONV/1.0E-4/, CONV1/1.0E-5/, CONV2/1.0E-5/
          C
0015      KNT=0
0016      IDX=1
0017      ISW=0
0018      CALL KCALC
0019      ISW=1
0020      IF (NC.GT.2) GO TO 26
          C
0021      CALL TWOLIN
0022      CALL KCALC
0023      DO 11 I=1,NC
0024      11 Y(I)=EK(I,1)*X(I,1)
0025      IF (IDLV.EQ.0) GO TO 240
0026      GO TO 13
0027      12 CALL KCALC
0028      13 CALL TWOLIN
0029      GO TO (14,17), IDX
0030      14 SUM=0.0
0031      DO 15 I=1,NC
0032      YY(I)=EK(I,1)*X(I,1)
0033      SY(I)=Y(I)
0034      YC(I)=YY(I)
0035      Y(I)=(Y(I)+YY(I))/2.0
0036      15 SUM=SUM+Y(I)
0037      DO 16 I=1,NC

```

ORTRAN IV G LEVEL 21

FLASH2

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```

0038      16 Y(I)=Y(I)/SUM
0039          IDX=2
0040          GO TO 12
0041      17 DO 18 I=1,NC
0042          IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 18
0043          IF (ABS(Y(I)-YY(I))/Y(I).GT.CONV) GO TO 19
0044      18 CONTINUE
0045          GO TO 240
0046      19 SUM=0.0
0047          DO 21 I=1,NC
0048          IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 21
0049          CALL WERS (Y(I),YY(I),SY(I),YC(I))
0050          SUM=SUM+Y(I)
0051      21 CONTINUE
0052          DO 22 I=1,NC
0053      22 Y(I)=Y(I)/SUM
0054          GO TO 12
C
0055      25 CALL KCALC
0056      26 DO 27 I=1,NC
0057      27 R(I)=EK(I,1)/EK(I,2)
0058          K=0
C
0059          IF (LDBUG.GE.1) WRITE (6,28) T,U,V
0060      28 FORMAT ('OFLASH#2  T-U-V', F12.4,2E14.5)
0061          IF (LDBUG.GE.2) WRITE (6,29) (Y(I),I=1,9), (X(I,1),I=1,9),
+          (X(I,2),I=1,9)
0062      29 FORMAT ('OY-ASS ',9E13.5/' X1-ASS ',9E13.5/' X2-ASS ',9E13.5)
C
0063      30 F1=0.0
0064          F2=0.0
0065          DF1U=0.0
0066          DF2U=0.0
0067          DF1V=0.0
0068          DF2V=0.0
0069          DDF1UU=0.0
0070          DDF1VV=0.0
0071          DDF1UV=0.0
0072          DDF2UU=0.0
0073          DDF2VV=0.0
0074          DDF2UV=0.0
C
0075          DO 50 I=1,NC
C
0076          AA=Z(I)*(1.0-EK(I,1))
0077          BB=Z(I)*(1.0-R(I))
0078          WW=V*(1.0-U)+(1.0-V)*(1.0-U)*R(I)+U*EK(I,1)
0079          F1=F1+AA/WW
0080          F2=F2+BB/WW
C
0081          DWU=V*(R(I)-1.0)-R(I)+EK(I,1)
0082          DWV=(1.0-U)*(1.0-R(I))
0083          DDWUV=R(I)-1.0
C
0084          DF1U=DF1U-AA/WW**2*DWU
0085          DF2U=DF2U-BB/WW**2*DWU

```

IRTRAN IV G LEVEL 21

FLASH2

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0086      DF1V=DF1V-AA/WW**2*DWV
0087      DF2V=DF2V-BB/WW**2*DWV
          C
0088      DDF1UU=DDF1UU+2.0*AA/WW**3*DWU**2
0089      DDF2UU=DDF2UU+2.0*BB/WW**3*DWU**2
0090      DDF1VV=DDF1VV+2.0*AA/WW**3*DWV**2
0091      DDF2VV=DDF2VV+2.0*BB/WW**3*DWV**2
0092      DDF1UV=DDF1UV+2.0*AA/WW**3*DWU*DWV-AA/WW**2*DDWUV
0093      DDF2UV=DDF2UV+2.0*BB/WW**3*DWU*DWV-BB/WW**2*DDWUV
          C
0094      50 CONTINUE
          C
0095      RHS(1)=-F1
0096      RHS(2)=-F2
0097      ZJ(1,1)=DF1U
0098      ZJ(1,2)=DF1V
0099      ZJ(2,1)=DF2U
0100      ZJ(2,2)=DF2V
          C
          C      PHS(1)=2.0*(F1*DF1U+F2*DF2U)
          C      PHS(2)=2.0*(F1*DF1V+F2*DF2V)
          C      ZJ(1,1)=2.0*(DF1U**2+F1*DDF1UU+DF2U**2+F2*DDF2UU)
          C      ZJ(2,2)=2.0*(DF1V**2+F1*DDF1VV+DF2V**2+F2*DDF2VV)
          C      ZJ(1,2)=2.0*(DF1U*DF1V+F1*DDF1UV+DF2U*DF2V+F2*DDF2UV)
          C      ZJ(2,1)=ZJ(1,2)
          C
0101      DETRM=ZJ(1,1)*ZJ(2,2)-ZJ(1,2)*ZJ(2,1)
0102      DU=(RHS(1)*ZJ(2,2)-RHS(2)*ZJ(1,2))/DETRM
0103      DV=(ZJ(1,1)*RHS(2)-ZJ(2,1)*RHS(1))/DETRM
0104      K=K+1
          C
          C      IF (LDEBUG.GE.2) WRITE (6,55) F1,F2,U,DU,V,DV
0105      55 FORMAT ( 5X,'F1-F2-U-DU-V-DV', 6E14.5)
          C
          C      IF (K.LT.20) GO TO 59
0107      57 WRITE (6,58)
0108      58 FORMAT ('0FLASH2 CALCULATION TERMINATED' /
0109      +          ' CALCULATION CONTINUING' /)
0110      GO TO 240
          C
          C      COMPUTE NEW VALUES OF U & V
          C
0111      59 NCVRG=0
0112      GO TO (60,70), ID
0113      60 SDU=DU
0114      SDV=DV
0115      ID=2
0116      GO TO 80
0117      70 SSDU=SDU
0118      SSDV=SDV
0119      SDU=DU
0120      SDV=DV
0121      TOP=SSDU*SSDV+SDU*SDV
0122      BTM=(SSDU**2+SSDV**2)*(SDU**2+SDV**2)
0123      COSANG=TOP/SQRT(BTM)
0124      BETA=1.0+COSANG

```

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```

0125      BETA=1.0
0126      DU=BETA*DU
0127      DV=BETA*DV
C
C      CHECK TO INSURE THAT U & V ARE .LT. 1.0 AND .GT. 0.0
C
0128      80 IF ((U+DU.LT.1.0 .AND. U+DU.GT.0.0) .AND.
+           (V+DV.LE.1.0 .AND. V+DV.GE.0.0)) GO TO 90
0129      NCVRG=1
0130      DU=DU/2.0
0131      DV=DV/2.0
0132      GO TO 80
C
0133      90 IF (V.LT.0.99) GO TO 94
0134      V=1.0-V
0135      DO 91 I=1,NC
0136      HOLD=X(I,1)
0137      X(I,1)=X(I,2)
0138      X(I,2)=HOLD
0139      HOLD=XX(I,1)
0140      XX(I,1)=XX(I,2)
0141      XX(I,2)=HOLD
0142      HOLD=SX(I,1)
0143      SX(I,1)=SX(I,2)
0144      SX(I,2)=HOLD
0145      HOLD=XC(I,1)
0146      XC(I,1)=XC(I,2)
0147      XC(I,2)=HOLD
0148      HOLD=EK(I,1)
0149      EK(I,1)=EK(I,2)
0150      EK(I,2)=HOLD
0151      R(I)=EK(I,1)/EK(I,2)
0152      HOLD=GAM(I,1)
0153      GAM(I,1)=GAM(I,2)
0154      GAM(I,2)=HOLD
0155      91 CONTINUE
C
0156      94 IF (V.LT.1.E-10) GO TO 231
0157      IF (ABS(DU).LT.CONV1 .AND. ABS(DV).LT.CONV1 .AND. NCVRG.EQ.0)
+      GO TO 110
C
0158      95 U=U+DU
0159      V=V+DV
0160      GO TO 30
C
0161      110 IF (ABS(F1).LT.CONV2.AND.ABS(F2).LT.CONV2) GO TO 120
0162      GO TO 95
C
C      CALCULATE COMPOSITIONS
C
0163      120 DO 180 I=1,NC
0164      IF (Z(I).LT.1.0E-12) GO TO 170
0165      IF (L(I)) 130,140,150
C      NONVOLATILE
0166      130 WW=V+(1.0-V)*R(I)
0167      YY(I)=0.0

```

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```

0168      XX(I,1)=Z(I)/WW
0169      XX(I,2)=XX(I,1)*R(I)
0170      GO TO 160
      C
0171      140 YY(I)=Z(I)/U
0172      XX(I,1)=0.0
0173      XX(I,2)=0.0
0174      GO TO 180
      C
0175      150 WW=V*(1.0-U)+(1.0-V)*(1.0-U)*R(I)+U*EK(I,1)
0176      XX(I,1)=Z(I)/WW
0177      XX(I,2)=X(I,1)*R(I)
0178      YY(I)=EK(I,1)*XX(I,1)
0179      GO TO 160
0180      170 YY(I)=0.0
0181      XX(I,1)=0.0
0182      XX(I,2)=0.0
0183      180 CONTINUE
      C
0184      KNT=KNT+1
0185      IF (KNT.EQ.50) GO TO 57
0186      GO TO (280,190), IUX
      C
0187      190 IF (IDLL.EQ.0) GO TO 220
0188      DO 210 I=1,NC
0189      IF (L(I).EQ.0 .OR. Z(I).LT.1.0E-12) GO TO 210
      C
0190      IF (ABS(X(I,1)*GAM(I,1)-X(I,2)*GAM(I,2))/(X(I,2)*GAM(I,2))
      * (ABS(X(I,1)-XX(I,1))/X(I,1) .GT. CONV .OR.
      *(ABS(X(I,2)-XX(I,2))/X(I,2)) .GT. CONV) GO TO 360
      C
      *.GT.CONV) GO TO 360
0191      210 CONTINUE
0192      220 IF (IDLV.EQ.0) GO TO 240
0193      DO 230 I=1,NC
0194      IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 230
0195      IF (ABS(Y(I)-YY(I))/Y(I).GT.CONV) GO TO 360
0196      230 CONTINUE
0197      GO TO 240
      C
0198      231 DO 232 I=1,NC
0199      HOLD=X(I,1)
0200      X(I,1)=X(I,2)
0201      232 X(I,2)=HOLD
0202      WRITE (6,238)
0203      238 FORMAT (// 5X, 'LIQUID HAS GONE TO A ONE PHASE SYSTEM' //)
0204      NLIN=1
0205      CALL FLASH1
      C
0206      240 SUM=0.0
0207      SUM1=0.0
0208      SUM2=0.0
0209      DO 260 I=1,NC
0210      Y(I)=YY(I)
0211      SUM=SUM+Y(I)
0212      DO 250 N=1,2
0213      250 X(I,N)=XX(I,N)
0214      SUM1=SUM1+X(I,1)

```



```

0215      SUM2=SUM2+X(I,2)
0216      260 CONTINUE
0217      DO270 I=1,NC
0218      Y(I)=Y(I)/SUM
0219      X(I,1)=X(I,1)/SUM1
0220      270 X(I,2)=X(I,2)/SUM2
0221      ISW=0
0222      RETURN
C
0223      280 IF (IDLL.EQ.0) GO TO 320
0224      DO 310 N=1,2
0225      SUM=0.0
0226      DO290 I=1,NC
0227      SX(I,N)=X(I,N)
0228      XC(I,N)=XX(I,N)
0229      X(I,N)=(X(I,N)+XX(I,N))/2.0
0230      290 SUM=SUM+X(I,N)
0231      DO300 I=1,NC
0232      300 X(I,N)=X(I,N)/SUM
0233      310 CONTINUE
0234      320 IF (IDLV.EQ.0) GO TO 350
0235      SUM=0.0
0236      DO 330 I=1,NC
0237      SY(I)=Y(I)
0238      YC(I)=YY(I)
0239      Y(I)=(Y(I)+YY(I))/2.0
0240      330 SUM=SUM+Y(I)
0241      DO 340 I=1,NC
0242      340 Y(I)=Y(I)/SUM
0243      350 IDX=2
0244      GO TO 25
C
0245      360 IF (IDLL.EQ.0) GO TO 400
0246      DO 390 N=1,2
0247      SUM=0.0
0248      DO 370 I=1,NC
0249      IF (L(I).EQ.0 .OR. Z(I).LT.1.0E-12) GO TO 370
C
C 365 WRITE (6,365) N,I,X(I,N),XX(I,N),SX(I,N),XC(I,N)
C 365 FORMAT(/' #365', 2I5,4G20.5)
0250      CALL WEGS(X(I,N),XX(I,N),SX(I,N),XC(I,N))
0251      SUM=SUM+X(I,N)
0252      370 CONTINUE
0253      DO 380 I=1,NC
0254      380 X(I,N)=X(I,N)/SUM
0255      390 CONTINUE
0256      400 IF (IDLV.EQ.0) GO TO 420
0257      SUM=0.0
0258      DO 410 I=1,NC
0259      IF (L(I).LT.0 .OR. Z(I).LT.1.0E-12) GO TO 410
0260      CALL WEGS (Y(I),YY(I),SY(I),YC(I))
0261      SUM=SUM+Y(I)
0262      410 CONTINUE
0263      GO TO 440
0264      420 SUM=0.0
0265      DO 430 I=1,NC
0266      Y(I)=YY(I)

```

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```
267          430 SUM=SUM+Y(I)
268          440 DO 450 I=1,NC
269          450 Y(I)=Y(I)/SUM
270          GO TO 25
          C
271          END
```

ORTRAN IV G LEVEL 21

AFLASH

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```

0001      SUBROUTINE      AFLASH
          C
          C      ADIABATIC FLASH
          C
          C**** IMPLICIT REAL*4 (A-H,O-Z), INTEGER*2 (I-N)
          C
0002      COMMON /R2L/      FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
0003      COMMON /CONTRL/   T,P,NOBUB,NODEW,ISO,NLIQ
0004      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
0005      COMMON /LIQUID/   X(10,2),GAM(10,2),DHL(2),HIDL(2),HREAL(2),
          +               FRACL
0006      COMMON /DUMMY/    XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
          +               YC(10),RMAX,RMIN
0007      COMMON /CTRL/     NCR, NPRT, NSTART, TITLE(20)
0008      COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDIM,
          *               NAME(2,10),L(10),COMPNT(10)
0009      COMMON /NJIT1/    CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
          *               W(10)
0010      COMMON /NJIT2/    C(180),ALPHA(45),VC(10),VOL(10),TC(10),
          *               PC(10),GA(10),OB(10),AZ(10,10),G(10,10),
          *               THETA(10),THS(10),HFAC(10),HHFA(10),HM(10),
          *               W(10),RR(10),XL(10),PHI(10),DUMMY(10)
0011      COMMON /NJIT3/    TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
          *               AK(10,10),AVALUE(10,10),BVALUE(10,10),
          *               BCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
          C
0012      DATA STEP,TINC,CONV/30.0,10.0,1.0E-4/
          C
0013      ID=1
0014      S=T
0015      10 GO TO (11,12), NLIQ
0016      11 CALL FLASH1
0017      FRACL=1.0
0018      GO TO 15
0019      12 CALL FLASH2
0020      15 CALL LIQH
0021      CALL VAPH
0022      H=HREALV*FRACV+HREAL(1)*(1.0-FRACV)*FRACL+HREAL(2)*(1.0-FRACV)
          + * (1.0-FRACL)-HI
0023      IF (ABS(H/HI)-CONV)70,20,20
0024      20 GO TO (30,40),ID
0025      30 T=S-SIGN(TINC,H)
0026      HD=H
0027      ID=2
0028      GO TO 10
0029      40 HD=H*(S-T)/(HD-H)
0030      S=T
0031      IF (ABS(HD)-STEP)60,60,50
0032      50 HD=SIGN(STEP,HD)
0033      60 T=T-HD
0034      HD=H
0035      GO TO 10
0036      70 WRITE (6,80)
0037      80 FORMAT('0FOLLOWING IS THE RESULT OF AN ADIABATIC FLASH.')
```

C

ORTRAN IV G LEVEL 21

AFLASH

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0039

END

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LIQH

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```

001      SUBROUTINE      LIQH
          C
          C      ENTHALPY OF A REAL LIQUID
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
002      COMMON /B2L/      FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
003      COMMON /CONTRL/   T,P,NOBUB,NODEW,ISO,NLIQ
004      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
005      COMMON /LIQUID/   X(10,2),GAM(10,2),DHL(2),HIOLL(2),HREAL(2),
          +              FRACL
006      COMMON /DUMMY/    XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
          +              YC(10),RMAX,RMTN
007      COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)
008      COMMON /NJIT0/   NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDIM,
          *              NAME(2,10),L(10),COMPNT(10)
009      COMMON /NJIT1/   CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
          *              W(10)
          C
010      ISV=0
011      IF (IDLL)20,10,20
012      10 DHL(1)=0.0
013      DHL(2)=0.0
014      GO TO 50
015      20 DO 40 I=1,NLIQ
016      CALL ACTVLY (I,X(1,NLIQ),GAM(1,NLIQ))
017      40 CALL XSHLIQ(T,X(1,NLIQ),DHL(NLIQ))
018      50 DO 60 N=1,NLIQ
019      HIOLL(N)=0.0
020      DO 55 I=1,NC
021      55 HIOLL(N)=HIOLL(N)+X(I,N)*(((ENP(I,4)*T+ENP(I,3))*T + ENP(I,2))*T
          1          +ENP(I,1))
022      HREAL(N)=HIOLL(N)
023      IF (IDH.EQ.0) HREAL(N)=HREAL(N)+DHL(N)
024      60 CONTINUE
025      RETURN
          C
026      END

```

RTRAN IV G LEVEL 21

VAPH

DATE = 81357

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```

001      SUBROUTINE      VAPH
      C
      C      ENTHALPY OF REAL VAPOR
      C
      C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
      C
002      COMMON /NJIT0/  NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW, NDIM,
      *                NAME(2,10),L(10),COMPNT(10)
003      COMMON /NJIT1/  CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
      *                W(10)
004      COMMON /VAPOR/  Y(9), VP(9), FUG(9), DHV, HIDLV, HREALV, FRACV
005      COMMON /CONTRL/  T,P,NOBUB,NODEW,ISO,NLIQ
      C
006      ISW=0
007      IF (IDLV)20,10,20
008      10 DHV=0.0
009      GO TO 30
010      20 CALL XSHVAP(T,P,Y,DHV)
011      30 HIDLV=0.0
012      DO 40 I=1,NC
013      40 HIDLV=HIDLV + Y(I)*(((ENP(I,4)*T+ENP(I,3))*T + ENP(I,2))*T
      1      +ENP(I,1))
014      HREALV=HIDLV
015      IF (IDH.EQ.0)HREALV=HREALV+DHV
016      RETURN
      C
017      END

```

RTRAN IV G LEVEL 21

XSHVAP

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```
001      SUBROUTINE XSHVAP (T,P,Y,DHV)
002      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDIM,
      *      NAME(2,10),L(10),COMPNT(10)
003      GO TO (10,20),IDLV
004      10  CALL RKDHV(T,P,Y,DHV)
005      GO TO 99
006      20  CALL VIRDHV(T,P,Y,DHV)
007      99  RETURN
008      END
```

```

0001      SUBROUTINE    TEMPCV(ITF,T)
          C
          C    TEMPERATURE CONVERSION
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
0002      GO TO (20,10,30,40),ITF
0003      10 T=T*1.8+32.0
0004      20 RETURN
0005      30 T=T-459.69
0006      RETURN
0007      40 T=(T*1.8)-459.69
0008      RETURN
          C
0009      END
    
```



```
001          SUBROUTINE  PRESS(IPF,P)
      C
      C          PRESSURE CONVERSION
      C
      C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
      C
002          GO TO (20,10,30),IPF
003          10 P=P*0.0193421
004          20 RETURN
005          30 P=P*14.70
006          RETURN
      C
007          END
```

```

001      SUBROUTINE      PRINT
      C
      C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
      C
002      DIMENSION      ACT(9)
      C
003      COMMON /B2L/      FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
004      COMMON /CONTRL/   T,P,NOBUB,NDEW,ISO,NLIQ
005      COMMON /VAPOR/    Y(10),VP(10),FUG(10),DHV,HIDLV,HREALV,FRACV
006      COMMON /LIQUID/   X(10,2),GAM(10,2),DHL(2),HIDLL(2),HREALL(2),
      +                   FRACL
007      COMMON /DUMMY/    XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
      +                   YC(10),RMAX,RMIN
008      COMMON /CTRL/     NCR, NPRT, NSTART, TITLE(20)
009      COMMON /NJIT0/    NC, NCM1, IDLL, IDLV, IDH, LORUG, ISW, NDIM,
      *                   NAME(2,10),L(10),COMPNT(10)
010      COMMON /NJIT1/    CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
      *                   W(10)
011      COMMON /NJIT2/    C(180),ALPHA(45),VC(10),VOL(10),TC(10),
      *                   PC(10),OA(10),OB(10),AZ(10,10),G(10,10),
      *                   THETA(10),THS(10),HFAC(10),HHFA(10),HM(10),
      *                   N(10),RR(10),XL(10),PHI(10),DUMMY(10)
012      COMMON /NJIT3/    TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
      *                   AK(10,10),AVALHE(10,10),BVALUE(10,10),
      *                   BCROSS(10,10),OMEGA(10),DIPOLE(10),BZ(10,10)
      C
013      DO80 N=1,NLIQ
014      WRITE (6,5) N
015      5 FORMAT ('0"VLE" DATA FOR VAPOR AND LIQUID PHASE #',I3)
016      WRITE (6,1) ((NAME(J,I),J=1,2),I=1,NC)
017      1 FORMAT (/25X,9(2A4,2X))
018      WRITE (6,6) (Y(I),I=1,NC)
019      6 FORMAT (' VAP MOLE FRAC',9X,9F10.5)
020      WRITE (6,7) (X(I,N),I=1,NC)
021      7 FORMAT (' LIQ MOLE FRAC',9X,9F10.5)
022      WRITE (6,10) (EK(I,N),I=1,NC)
023      10 FORMAT(' VALUES OF K=Y/X',7X,9F10.5)
024      WRITE (6,20) (VP(I),I=1,NC)
025      20 FORMAT(' VAPOR PRESSURES',7X,9F10.3)
026      IF (IDLV)30,50,30
027      30 WRITE (6,40) (FUG(I),I=1,NC)
028      40 FORMAT(' FUGACITY FACTORS',6X,9F10.5)
029      50 IF (IDLL)60,80,60
030      60 WRITE (6,70) (GAM(I,N),I=1,NC)
031      70 FORMAT(' ACTIVITY COEFFICIENTS ',9F10.4)
032      DO 71 I=1,NC
033      71 ACT(I)=GAM(I,N)*X(I,N)
034      WRITE (6,72) (ACT(I),I=1,NC)
035      72 FORMAT (' ACTIVITY',14X,9F10.4)
036      80 CONTINUE
037      RETURN
      C
0038     END

```

ORTRAN IV G LEVEL 21

WEGS

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```
0001      SUBROUTINE      WEGS(G,F,OG,OF)
          C
          C      WEGSTEIN CONVERGENCE
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
0002      SL=(OF-F)/(OG-G)
0003      OG=G
0004      OF=F
0005      SL=1.0/(1.0-SL)
0006      IF (SL.LE.0.0)SL=0.5
0007      10 V=SL*F+(1.0-SL)*G
0008      IF (V.GT.0.0) GO TO 20
0009      SL=SL*0.5
0010      GO TO 10
0011      20 G=V
0012      RETURN
          C
0013      END
```

```

0001      SUBROUTINE      RKDHV(T,P,Y,DHV)
          C
          C      ENTHALPY DEPARTURE FROM IDEAL GAS -- EVALUATED
          C      USING REDLICH-KWONG EQUATION
          C
          C**** IMPLICIT REAL*8 (A-H,O-Z), INTEGER*2 (I-N)
          C
0002      DIMENSION      Y(20)
0003      COMMON  /NJIT0/  NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDIM,
          *              NAME(2,10),L(10),COMPNT(10)
0004      COMMON  /NJIT2/  C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *              PC(10), OA(10), OB(10), AZ(10,10), G(10,10),
          *              THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
          *              W(10),RR(10),XL(10),PHI(10),DUMMY(10)
          C
0005      TE=T
0006      A=0.0
0007      B=0.0
0008      DO10J=1,NC
0009      A=A+OA(J)*Y(J)
0010      10 B=B+OB(J)*Y(J)
0011      A=A/TE**1.25
0012      B=B/TE
0013      CALL ZCALC ( A,B,P,ZZ)
0014      DHV=1.9872*TE*(ZZ-1.0-1.5*A*A*ALOG(1.0+B*P/ZZ)/B)
0015      RETURN
          C
0016      END

```

```

RTRAN IV G LEVEL 21          KCALC          DATE = 81357          19/38/52

001          SUBROUTINE          KCALC
C
002          IMPLICIT REAL      (M)
003          INTEGER            COMPNT
C
004          COMMON /B2L/        FD(11),Z(10),EK(10,2),PF,TF,HI,XSAVE(10,2)
005          COMMON /CONTRL/     T,P,NOBUB,NODEW,ISO,NLIN
006          COMMON /VAPOR/     Y(10),VP(10),FUG(10),DHV,HIIDLV,HREALV,FRACV
007          COMMON /LIQUID/     X(10,2),GAM(10,2),DHL(2),HIDL(2),HREAL(2),
+                               FRACL
008          COMMON /DUMMY/     XX(10,2),SX(10,2),XC(10,2),YY(10),SY(10),
+                               YC(10),RMAX,RMIN
009          COMMON /CTRL/      NCR, NPRT, NSTART, TITLE(20)
010          COMMON /NJIT0/     NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDTM,
*                               NAME(2,10),L(10),COMPNT(10)
011          COMMON /NJIT1/     CPL(10,4),CPV(10,4),ENP(10,10),ANT(6,10),
*                               W(10)
C
C
012          TT=T
013          P=PF
C
014          8 DO 9 J=1,NC
015             GAM(J,1)=1.0
016             GAM(J,2)=1.0
017             FUG(J)=1.0
018             VP(J)=FXP(ANT(1,J)+ANT(2,J)/(ANT(3,J)+TT)+ANT(4,J)*TT+ANT(5,J)*
*               TI*TT+ANT(6,J)*ALOG(TI))
019          9 CONTINUE
C
C
020          11 IF (IDLL.EQ.0) GO TO 20
021             IS=0
022             DO 12 I=1,NLIN
023          12 CALL ACTVY (TI,X(1,I),GAM(1,I))
C
024          20 IF (IDLV.EQ.0) GO TO 40
025          CALL FUGCTY (TI,P,Y,VP,FUG,DHV)
C
026          40 DO 50 I=1,NLIN
027             DO 50 J=1,NC
028          50 EK(J,I)=VP(J)/P*GAM(J,I)*FUG(J)
C
C
029          250 RETURN
030          END

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0001      SUBROUTINE      VIRDHV (T,P,Y,DHV)
          C
0002      REAL            Y(1),VP(1),FUG(1)
          C
0003      COMMON /NJIT0/  NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDIM,
          *              NAME(2,10),L(10),COMPNT(10)
0004      COMMON /NJIT2/  C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *              PC(10), OA(10), OR(10), AZ(10,10), G(10,10)
0005      COMMON /NJIT3/  TTC(10,10),PPC(10,10),VVC(10,10),ZZC(10,10),
          *              AK(10,10), AVALUE(10,10), BVALUE(10,10),
          *              CROSS(10,10), OMEGA(10), DIPOLE(10), BZ(10,10)
0006      P=P/760.0
          C
          C
          C      SUBROUTINE TSONOP GENERATES B7(I,J) FOR VIRFUG
          C      CALL TSONOP(T,P,MARK)
          C
0008      RT=42.057*T
0009      PORT=P/RT
0010      SUMY=0.0
0011      BMIX=0.0
0012      DO 15 I=1,NC
0013      VP(I)=VP(I)/760.0
0014      SUMY=SUMY+Y(I)
0015      DO 10 J=1,NC
0016      BMIX=BMIX+Y(I)*Y(J)*BZ(I,J)
0017      10 CONTINUE
0018      15 CONTINUE
0019      BMIX=BMIX /SUMY**2
0020      IF (MARK) 20,20,22
0021      20 SQARG=0.25 + BMIX*PORT
          C
          C      IF BMIX*P/RT IS MORE NEGATIVE THAN 0.25, THE PRESSURE SERIES VIRIAL
          C      EQ. MUST BE USED. THE ARGUMENT MARK IS USED TO COMMUNICATE THIS
          C      TO THE MAIN PROGRAM
          C
0022      IF (SQARG) 22,25,25
0023      22 VMIX=BMIX +1. /PORT
0024      ZMIX=PORT*VMIX
0025      MARK=MARK+1
0026      GO TO 50
0027      25 VMIX=(0.5+SQRT (SQARG))/PORT
0028      ZMIX=PORT*VMIX
          C
          C      CALCULATION OF FUGACITY COEFFICIENTS
          C
0029      50  DHV=1.9872*T*(ZMIX-1.0)
          C      PRINT 400,DHV,T,P
0030      400  FORMAT(5X,'DHV=',G15.6,'TEMP=',G15.6,'PRESS=',G15.6)
          C      DO 51 I=1,NC
          C      DO 51 J=1,NC
          C 51  WRITE(6,401) I,J,B7(I,J)
0031      401  FORMAT(5X,'B(',I2,',',I2,')=',G15.5)
0032      DO 52 I=1,NC
0033      52  VP(I)=VP(I)*760.0
0034      P=P*760.0

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035 RETURN
036 END

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1      SUBROUTINE MATRIX
2      C
3      IMPLICIT REAL      (M)
4      INTEGER          COMPNT
5      REAL             LIQUID
6      C
7      COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)
8      COMMON /NJIT0/   NK, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
9      *                NAME(4,10), L(10), NDIM, COMPNT(10)
10     COMMON /INR0/    NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
11     *                NFEED, LF(9), IDCODE, ITR, ITRMAX, ICODE
12     COMMON /INR1/    FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
13     *                DEST, DESTV, DESTL, BTMS, RFLX, BOIL, PP(50), T(50),
14     *                MH(50), SL(50), SV(50), FL(50), FLL(50,10),
15     *                FV(50), FVV(50,10), ZCONVT, ZCONVF
16     COMMON /INR2/    DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
17     *                DEXH(50,11), HV(50), HVV(50,10), CV(50)
18     COMMON /INR3/    AMAT(21,21), RMAT(50,21,21), D(50,21)
19     COMMON /INR4/    VAPOR, LIQUID, FEED(11)
20     COMMON /VLE/     XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
21     C
22     EQUIVALENCE      (WATER,VAPOR)
23     C
24     C
25     D(1,J) IS THE MATRIX CONSISTING OF THE CALCULATED
26     DISCREPANCY FUNCTIONS ON PLATE I FOR COMPONENT J.
27     C
28     FOR PLATE I, D(1,J) CONSISTS OF
29     EQNS          1-NC          COMPONENTIAL MATERIAL BALANCES
30     EQNS          NC+1-2*NC     EQUILIBRIUM RELATION WITH MURPHREEFF EFF
31     EQN           2*NC+1       ENERGY BALANCE
32     C
33     NOTE THAT FOR I=1 AND I=NST (NST=NUMBER OF STAGES) THERE ARE
34     SPECIAL RELATIONS USED TO REPLACE THE ENTHALPY BALANCES.
35     C
36     PROBLEM TYPE CONTROLLED BY 'NCASE'
37     1      ABSORBER / STRIPPER
38     2      REBOILED ABSORBER / STRIPPER
39     3      FRACTIONATOR --- REFLUX GIVEN
40     4      FRACTIONATOR --- BOILUP GIVEN
41     5      FRACTIONATOR --- REBOILED ABSORBER
42     C
43     IF (IDCODE.EQ.0) GO TO 9
44     C
45     LOCATE THE LARGEST COMPONENT # IN THE LIQUID DISTILLATE
46     C
47     JBIG=0
48     BIG=0.0
49     DO 5 J=1,NK
50     IF (FLL(NST,J).LE.BIG) GO TO 5
51     JBIG=J
52     BIG=FLL(NST,J)
53     5 CONTINUE
54     C
55     9 GO TO (10,10,30,30,30), NCASE
56     10 ASSIGN 70 TO NSV1

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4      ASSIGN 170 TO NSV2
5      GO TO 50
6      30 ASSIGN 80 TO NSV1
7      ASSIGN 160 TO NSV2
      C
8      50 DO 219 I=1,NST
      C
9      CALL KFAC(I)
10     IF (I.NE.NST) GO TO 59
11     IF (IDCODE.NE.1) GO TO 59
12     IF (ITR.EQ.0) GO TO 59
      C
13     THIS IS A BUBBLE POINT CALCULATION FOR ONE LIQUID PHASE
14     TO PRE-ESTABLISH THE TEMPERATURE IN THE ACCUMULATOR(DISTILLATE
15     DRUM). METHOD OF SOLVING THE DESIGN EQUATIONS
16     FN(T)=SUM OF KX -1.0=0 WHICH(SUM OF KL)-L=0 IS SECANT METHOD
      C
17     ID=1
18     KNT=0
19     51 FN=-FL(NST)
20     DO 52 I=1,NK
21     52 FN=FN+XFK(J)*FLL(NST,J)
      C
22     WRITE (NPRT,53) JBIG,BIG,T(NST),FN
      C
23     53 FORMAT (5X,I5,G12.5,10X,2G15.5)
24     KNT=KNT+1
25     IF (KNT.EQ.20) GO TO 59
26     IF (ABS(FN)/FL(NST).LT.0.00005) GO TO 59
27     GO TO (54,55), ID
28     54 ID=2
29     SFN=FN
30     ST=T(NST)
31     T(NST)=T(NST)-SIGN(2.0,FN)
32     CALL KFAC(NST)
33     GO TO 51
34     55 DTNST=-FN/(FN-SFN)*(T(NST)-ST)
35     DTNST=SIGN(AMIN1(ABS(DTNST),10.0),DTNST)
36     SFN=FN
37     ST=T(NST)
38     T(NST)=T(NST)+DTNST
39     CALL KFAC(NST)
40     GO TO 51
      C
41     C
42     C
43     C
44     59 UA=1.0+SV(I)/FV(I)
      C
45     UA=1.0+SV(I)/FV(I)
46     UB=1.0+SL(I)/FL(I)
47     UC=MH(I)*FV(I)/FL(I)
48     UD=SL(I)/FL(I)**2
49     UE=SV(I)/FV(I)**2
      C
50     DO 60 J=1,NK
51     N=NK+J
52     D(I,J)=FVV(I,J)*UA+FLL(I,J)*UB

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3      D(I,N)=UC*XFK(J)*FLL(I,J)-FVV(I,J)
4      IF (I.GT.1) D(I,J)=D(I,J)-FVV(I-1,J)
5      IF (I.GT.1) D(I,N)=D(I,N)+(1.0-MU(I))*FVV(I-1,J)*FV(I)/FV(I-1)
6      IF (NCASE.EQ.1 .AND. I.EQ.1) D(I,N)=D(I,N)+(1.0-MU(I))*FSTR(1,J)*
      *      FV(1)/FSTR(1,NK1)
7      IF (I.LT.NST) D(I,J)=D(I,J)-FLL(I+1,J)
8      60 CONTINUE
C
C      SPECIAL EQUATIONS ... STAGES 1 & NST
C
9      GO TO NSV1, (70,80)
C
C      ABSORBER / STRIPPER
0      70 D(I,NK2)=UA*HV(I)+UB*HL(I)+(FL(I)+SL(I))*HHE(I)
1      IF (I.EQ.1) GO TO 130
2      D(I,NK2)=D(I,NK2)-HV(I-1)
3      D(I-1,NK2)=D(I-1,NK2)-HL(I)-FL(I)*HHE(I)
4      GO TO 130
C
C      DISTILLATION COLUMN
5      80 IF (I.EQ.1) GO TO 90
6      IF (I.EQ.NST) GO TO 110
7      D(I,NK2)=UA*HV(I)+UB*HL(I)-HV(I-1)+(FL(I)+SL(I))*HHE(I)
8      IF (I.GT.2) D(I-1,NK2)=D(I-1,NK2)-HL(I)-FL(I)*HHE(I)
9      GO TO 130
C
C      REBOILER ... STAGE 1
0      90 IF (NCASE.EQ.3) GO TO 91
1      IF (NCASE.EQ.4) GO TO 100
2      IF (NCASE.EQ.5) GO TO 106
C
C      REFLUX RATIO SPECIFIED ... BOTTOMS DRAWOFF PRODUCT RATE FIXED
3      91 D(1,NK2)=BTMS
4      92 DO 95 J=1,NK
5      95 D(1,NK2)=D(1,NK2)-FLL(1,J)
6      GO TO 130
C
C      BOILUP RATIO SPECIFIED ... FIX VAPOR RATE FROM REBOILER
7      100 D(1,NK2)=BOIL*BTMS
8      101 DO 105 J=1,NK
9      105 D(1,NK2)=D(1,NK2)-FVV(1,J)
0      GO TO 130
C
C      BOILUP RATIO SPECIFIED ... FIX VAPOR RATE FROM REBOILER
1      106 GO TO 100
C
C      CONDENSER ... STAGE NST
2      110 IF (NCASE.EQ.3) GO TO 111
3      IF (NCASE.EQ.4) GO TO 121
4      IF (NCASE.EQ.5) GO TO 129
C
C      DISTILLATE CODE ... "IDCODE"
C      -1  SUBCOOLED LIQUID (T-NST HELD CONSTANT)
C      0   ALL VAPOR / VAPOR-LIQUID PRODUCT
C      1   BUBBLE POINT LIQUID
C
C      REFLUX RATIO SPECIFIED ... CHECK DISTILLATE CODE

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5      111 IF (IDCODE) 116,112,116
C      VAPOR OR VAPOR/LIQUID DISTILLATE ... FIX REFLUX LIQUID RATE
6      112 D(NST,NK2)=REFLX*DEST
7      113 DO 114 J=1,NK
8      114 D(NST,NK2)=D(NST,NK2)-FLL(NST,J)
9      115 D(NSTM1,NK2)=D(NSTM1,NK2)-HL(NST)-FL(NST)*HHE(NST)
0      GO TO 130
C      LIQUID DISTILLATE ... FVV/FV=FLL/FL PLUS BUBBLE POINT EQUATION
C      FOR SUBCOOLED LIQUID BUBBLE POINT EQUATION REPLACED BY DT-NST=0.0
1      116 DO 117 J=1,NK
2      N=J+NK
3      117 D(NST,N)=FVV(NST,J)*FL(NST)-FLL(NST,J)*FV(NST)
4      N=JBIG+NK
5      D(NST,N)=-FL(NST)
6      DO 118 J=1,NK
7      118 D(NST,N)=D(NST,N)+XFK(J)*FLL(NST,J)
8      IF (IDCODE.LT.0) D(NST,N)=0.0
9      GO TO 112
C
C      BOILUP RATIO SPECIFIED ... CHECK DISTILLATE CODE
0      121 IF (IDCODE) 125,122,126
C      VAPOR OR VAPOR/LIQUID DISTILLATE ... FIX DISTILLATE VAPOR RATE
1      122 D(NST,NK2)=DESTV
2      DO 124 J=1,NK
3      124 D(NST,NK2)=D(NST,NK2)-FVV(NST,J)
4      GO TO 115
C      LIQUID DISTILLATE ... FVV/FV=FLL/FL PLUS BUBBLE POINT EQUATION
C      FOR SUBCOOLED LIQUID BUBBLE POINT EQUATION REPLACED BY DT-NST=0.0
5      126 DO 127 J=1,NK
6      N=J+NK
7      127 D(NST,N)=FVV(NST,J)*FL(NST)-FLL(NST,J)*FV(NST)
8      N=JBIG+NK
9      D(NST,N)=-FL(NST)
0      DO 128 J=1,NK
1      128 D(NST,N)=D(NST,N)+XFK(J)*FLL(NST,J)
2      IF (IDCODE.LT.0) D(NST,N)=0.0
3      GO TO 122
C
C      BOILUP RATIO SPECIFIED ... CHECK DISTILLATE CODE
4      129 GO TO 121
C
5      130 CONTINUE
C
C      DEFINITION OF THE JACOBIAN MATRIX. IT CONSISTS OF PARTIAL
C      DERIVATIVES OF THE PLATE CORRECTION FUNCTIONS WITH RESPECT
C      TO THE VARIABLES ON PLATE I.
C
C      IN THE DESCRIPTION OF THE JACOBIAN MATRIX (B-MATRIX) THE
C      CORRECTION FUNCTIONS WILL BE STATED AS F AND THE INDEPENDENT
C      VARIABLES WILL BE STATED AS V.
C
6      DO 150 J=1,NK
7      JJ=J+NK
8      DO 140 K=1,NK
9      KK=NK1+K
C

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C      F(V)=MATERIAL BALANCES
C      V=COMPONENT FLOW IN VAPOR PHASE
C
0      BMAT(I,J,KK)=-FVV(I,J)*UE
C
C      F(V)=EQUILIBRIUM RELATTONSHIPS
C      V=COMPONENT FLOW IN VAPOR PHASE
C
1      BMAT(I,JJ,KK)=MU(I)*FLL(I,J)*(XFK(J)+DFK(J,KK)*FV(I))/FL(I)
2      IF (I.GT.1) BMAT(I,JJ,KK)=BMAT(I,JJ,KK)+(1.0-MU(I))*FVV(I-1,J)/
*      FV(I-1)
3      IF (NCASE.EQ.1 .AND. I.EQ.1) BMAT(1,JJ,KK)=BMAT(1,JJ,KK)+
*      (1.0-MU(1))*FSTR(1,J)/FSTR(1,NK1)
C
C      F(V)=EQUILIBRIUM RELATIONSHIPS
C      V=COMPONENT FLOW IN LIQUID PHASE
C
4      BMAT(I,JJ,K)=UC*FLL(I,J)*(DFK(J,K)-XFK(J)/FL(I))
C
C      F(V)=MATERIAL BALANCES
C      V=COMPONENT FLOW IN LIQUID PHASE
C
5      140 BMAT(I,J,K)=-FLL(I,J)*UD
6      BMAT(I,J,J)=BMAT(I,J,J)+UB
C
C      F(V)=MATERIAL BALANCES
C      V=TEMPERATURE
C
7      BMAT(I,J,NK1)=0.0
C
C      F(V)=MATERIAL BALANCES
C      V=COMPONENT FLOW IN VAPOR PHASE
C
8      BMAT(I,J,J+NK1)=BMAT(I,J,J+NK1)+UA
C
C      F(V)=EQUILIBRIUM RELATIONSHIPS
C      V=COMPONENT FLOW IN LIQUID PHASE
C
9      BMAT(I,J+NK,J)=BMAT(I,J+NK,J)+UC*XFK(J)
C
C      F(V)=EQUILIBRIUM RELATIONSHIPS
C      V=TEMPERATURE
C
0      BMAT(I,JJ,NK1)=UC*FLL(I,J)*DFK(J,NK1)
C
C      F(V)=EQUILIBRIUM RELATIONSHIP
C      V=COMPONENT FLOW IN VAPOR PHASE
C
1     BMAT(I,J+NK,J+NK1)=BMAT(I,J+NK,J+NK1)-1.0
2     150 CONTINUE
C
C      NOTE THE SPECIAL EQUATION FOR I=1 AND I=NST
C
3     GO TO NSV2, (170,160)

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C
160 IF (I.EQ.1) GO TO 190
    IF (I.EQ.NST) GO TO 200
C
170 DO 180 K=1,NK
C
C          F(V)=ENTHALPY BALANCES
C          V=COMPONENT FLOW IN VAPOR PHASE
C
C          KK=NK1+K
C          HMAT(I,NK2,KK)=-HV(I)*UE+UA*HVV(T,K)
C
C          F(V)=ENTHALPY BALANCES
C          V=COMPONENT FLOW IN LIQUID PHASE
C
180 HMAT(I,NK2,K)=-HL(I)*UD+UR*HLL(I,K)+HHE(I)+(FL(I)+SL(I))*DEXH(I,K)
C
C          F(V)=ENTHALPY BALANCES
C          V=TEMPERATURE
C
C          HMAT(I,NK2,NK1)=CV(I)*UA+CL(I)*UR+(FL(I)+SL(I))*DEXH(I,NK1)
C          GO TO 219
C
C          SPECIAL EQUATIONS FOR STAGES 1 & NST
C
C          REBOILER ... STAGE 1
190 IF (NCASE.EQ.3) GO TO 192
    IF (NCASE.EQ.4) GO TO 196
    IF (NCASE.EQ.5) GO TO 198
C
C          REFLUX RATIO SPECIFIED ... NCASE=3
C          (1) BOTTOMS PRODUCT RATE FIXED
C          (2) PUMPBACK REFLUX RATE FIXED
192 DO 194 K=1,NK
    KK=K+NK1
    HMAT(I,NK2,KK)=0.0
194 HMAT(I,NK2,K)=-1.0
    HMAT(I,NK2,NK1)=0.0
    GO TO 219
C
C          BOILUP RATIO SPECIFIED ... NCASE=4
C          (1) VAPOR FROM REBOILER FIXED
C          (2) VAPOR DISTILLATE FIXED
196 DO 197 K=1,NK
    KK=K+NK1
    HMAT(I,NK2,KK)=-1.0
197 HMAT(I,NK2,K)=0.0
    HMAT(I,NK2,NK1)=0.0
    GO TO 219
C
C          BOILUP RATIO SPECIFIED ... NCASE=5
198 GO TO 196
C
C          CONDENSER ... STAGE NST
200 IF (NCASE.EQ.3) GO TO 201
    IF (NCASE.EQ.4) GO TO 215

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0      IF (NCASE.EQ.5) GO TO 218
C
C      REFLUX SPECIFIED ... CHECK DISTILLATE CODE
1      201 IF (IDCODE) 204,202,204
C      VAPOR OR VAPOR/LIQUID DISTILLATE ... FIX LIQUID REFLUX
2      202 GO TO 192
C      LIQUID DISTILLATE ... FVV/FV=FLL/FL PLUS BUBBLE POINT EQUATION
C      FOR SUBCOOLED LIQUID BUBBLE POINT EQUATION REPLACED BY DT-NST=0.0
3      204 DO 205 J=1,NK
4          N=J+NK
5          BMAT(NST,N,NK1)=0.0
6          DO 205 K=1,NK
7              KK=K+NK
8              BMAT(NST,N,K)=FVV(NST,J)
9      205 BMAT(NST,N,KK)=-FLL(NST,J)
10         DO 206 J=1,NK
11             K=J
12             N=J+NK
13             KK=K+NK1
14             BMAT(NST,N,K)=BMAT(NST,N,K)-FV(NST)
15      206 BMAT(NST,N,KK)=BMAT(NST,N,KK)+FL(NST)
16             N=J+16+NK
17             DO 207 K=1,NK
18                 J=K
19                 KK=K+NK1
20                 BMAT(NST,N,K)=XFK(J)-1.0
21                 BMAT(NST,N,KK)=0.0
22      207 BMAT(NST,N,NK1)=3MAT(NST,N,NK1)+DFK(J,NK1)*FLL(NST,J)
23             DO 209 J=1,NK
24                 DO 208 K=1,NK
25                     KK=K+NK1
26                     BMAT(NST,N,K)=BMAT(NST,N,K)+DFK(J,K)*FLL(NST,J)
27      209 BMAT(NST,N,KK)=BMAT(NST,N,KK)+DFK(J,KK)*FLL(NST,J)
28             IF (IDCODE.NE.-1) GO TO 214
29             DO 213 K=1,NK2
30      213 BMAT(NST,N,K)=0.0
31             BMAT(NST,N,NK1)=1.0
32      214 IF (NCASE.EQ.3) GO TO 192
33             IF (NCASE.EQ.4) GO TO 195
C
C      BOILUP SPECIFIED ... CHECK DISTILLATE CODE
4      215 IF (IDCODE) 217,216,217
C      VAPOR OR VAPOR/LIQUID DISTILLATE ... FIX DISTILLATE VAPOR RATE
5      216 GO TO 196
C      LIQUID DISTILLATE ... FVV/FV=FLL/FL PLUS BUBBLE POINT EQUATION
C      FOR SUBCOOLED LIQUID BUBBLE POINT EQUATION REPLACED BY DT-NST=0.0
6      217 GO TO 204
C
C      BOILUP SPECIFIED ... CHECK DISTILLATE CODE
7      218 GO TO 215
C
C      219 CONTINUE
C
C      FEED STREAMS ARE HANDLED HERE
C
C      DO 230 I=1,NFEED

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      J=LF(1)
      DO 220 K=1,NK
220  D(J,K)=D(J,K)-FSTR(1,K)
      IF (NCASE.EQ.5.AND.J.EQ.NST) GO TO 230
      D(J,NK2)=D(J,NK2)-HF(1)
230  CONTINUE
C
C      NOTE: SIGN BEFORE D(1,J) IS CHANGED
C           DUE TO NEWTON-RAPHSON EQUATION
C
      DO 240 I=1,NST
      DO 240 J=1,NK2
240  D(I,J)=-D(I,J)
C
C      AT THIS POINT THE H-MATRIX AND D-VECTOR ARE DEFINED FOR THE
C      ENTIRE COLUMN, STAGE=1,NST.
C
C      GENERATION OF THE A-MATRIX FOR STAGE 2 IS ACCOMPLISHED HERE.
C      A-MATRIX FOR STAGE 1 IS UNDEFINED.
C
      DO 290 J=1,NK2
      DO 290 K=1,NK2
290  AMAT(J,K)=0.0
C
C      F(V)=MATERIAL BALANCES
C      V=COMPONENT FLOW IN VAPOR PHASE
C
      DO 300 J=1,NK
300  AMAT(J,J+NK1)=-1.0
C
C      F(V)=ENTHALPY BALANCES
C      V=TEMPERATURE
C
      AMAT(NK2,NK1)=-CV(1)
C
C      F(V)=EQUILIBRIUM RELATIONSHIPS
C      V=COMPONENT FLOW IN VAPOR PHASE
C
      YX=(1.0-MU(2))*FV(2)/(FV(1)**2)
      DO 320 JJ=NK1,NK2
      J=JJ-NK
      DO 310 KK=NK11,NK2
310  AMAT(JJ,KK)=-YX*FVV(1,J)
      JJJ=J+NK1
320  AMAT(JJ,JJJ)=AMAT(JJ,JJJ)+YX*FV(1)
C
C      F(V)=ENTHALPY BALANCES
C      V=COMPONENT FLOW IN VAPOR PHASE
C
      DO 350 KK=NK11,NK2
      K=KK-NK1
350  AMAT(NK2,KK)=-HVV(1,K)
C
C      MODIFY EQUILIBRIUM RELATIONSHIP TO REFLECT NON-VOLATILE (L=-1)
C      AND NON-CONDENSABLE (L=0) COMPONENTS
C

```

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```
5      DO 370 J=1,NK
6      IF (L(J).GT.0) GO TO 370
7      JJ=J+NK
8      DO 340 K=1,NK2
9      340 AMAT(JJ,K)=0.0
0      DO 360 I=1,NST
1      D(I,JJ)=0.0
2      DO 350 K=1,NK2
3      350 BMAT(I,JJ,K)=0.0
4      C      NON-VOLATILE COMPONENT ... VAPOR FLOW = 0.0
5      IF (L(J).EQ.-1) BMAT(I,JJ,J+NK1)=1.0
6      C      NON-CONDENSABLE COMPONENT ... LIQUID FLOW = 0.0
7      IF (L(J).EQ.0) BMAT(I,JJ,J)=1.0
8      360 CONTINUE
9      370 CONTINUE
C
      RETURN
      END
```


DRTRAN IV G LEVEL 21

CORECT

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```

0001      C      SUBROUTINE CORECT
0002      IMPLICIT REAL      (M)
0003      INTEGER      COMPNT
0004      REAL*8      CM(21,42)
0005      REAL      LIQUID

      C

0006      COMMON /CTRL/      NCR, NPRT, NSTART, TITLE(20)
0007      COMMON /NJITO/      NK, NCM1, IDLL, IDLV, IDH, LDEBUG, ISW,
*                            NAME(4,10), L(10), NDIM, COMPNT(10)
0008      COMMON /INRO/      NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
*                            NFEED, LF(9), ICODE, ITR, ITRMAX, ICODE
0009      COMMON /INR1/      FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
*                            DEST, DESTV, DESTL, BTMS, RFLX, BOIL, PP(50), T(50),
*                            MH(50), SL(50), SV(50), FL(50), FLL(50,10),
*                            FV(50), FVV(50,10), ZCONVT, ZCONVF
0010      COMMON /INR2/      DEK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
*                            DFXH(50,11), HV(50), HVV(50,10), CV(50)
0011      COMMON /INR3/      AMAT(21,21), BMAT(50,21,21), O(50,21)
0012      COMMON /INR4/      VAPOR, LIQUID, FEED(11)

      C

0013      DATA      TMAX /10.0/

      C

0014      GO TO (10,10,30,30,30), NCASE
0015      10 ASSIGN 140 TO NSW1
0016      ASSIGN 230 TO NSW2
0017      ASSIGN 270 TO NSW3
0018      ASSIGN 390 TO NSW4
0019      GO TO 50
0020      30 ASSIGN 170 TO NSW1
0021      ASSIGN 220 TO NSW2
0022      ASSIGN 260 TO NSW3
0023      ASSIGN 380 TO NSW4

      C
      C      SOLUTION OF THE BLOCK TRIAGONAL EQUATION.
      C
0024      50 DO 350 I=2,NST

      C
      C      THE B-MATRIX AND THE A-MATRIX ARE STORED IN THE MATRIX CM.
      C      NOTE THAT BOTH A AND B ARE STORED IN THE TRANSPOSED FORM.
      C

0025      DO 120 K=1,NK2
0026      KK=K+NK2
0027      DO 110 J=1,NK2
0028      CM(J,K)=BMAT(1-K,K,J)
0029      CM(J,KK)=AMAT(K,J)
0030      110 CONTINUE
0031      120 CONTINUE

      C
      C      FORMATION OF THE NEXT A-MATRIX ... STAGE I+1
      C

0032      IF (I.EQ.NST) GO TO 200
0033      DO 130 J=1,NK2
0034      DO 130 K=1,NK2
0035      130 AMAT(J,K)=0.0

      C

```

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```

C          F(V)=MATERIAL BALANCES
C          V=COMPONENT FLOW IN VAPOR PHASE
C
0036      DO 140 J=1,NK
0037      140 AMAT(J,J+NK1)=-1.0
C
C          F(V)=EQUILIBRIUM RELATIONSHIPS
C          V=COMPONENT FLOW IN VAPOR PHASE
C
0038      XY=(1.0-AM(I+1))*FV(J+1)/(FV(I)**2)
0039      DO 160 JJ=NK1,NK2
0040      J=JJ-NK
0041      DO 150 KK=NK11,NK2
0042      150 AMAT(JJ,KK)=-XY*FVV(I,J)
0043      JJJ=J+NK1
0044      160 AMAT(JJ,JJJ)=AMAT(JJ,JJJ)+XY*FV(I)
C
C          NOTE: SPECIAL EQUATIONS FOR STAGE 'NST' ... THE TOP OF THE COLUMN
C
0045      GO TO NSAI, (180,170)
C
0046      170 IF (I.EQ.NSTM1) GO TO 191
C
C          F(V)=ENTHALPY BALANCES
C          V=TEMPERATURE
C
0047      180 AMAT(NK2,NK1)=-CV(I)
C
C          F(V)=ENTHALPY BALANCES
C          V=COMPONENT FLOW IN VAPOR PHASE
C
0048      DO 190 KK=NK11,NK2
0049      K=KK-NK1
0050      190 AMAT(NK2,KK)=-HVV(I,K)
C
C          MODIFY EQUILIBRIUM RELATIONSHIP TO REFLECT COMPONENT TYPE
C          L=-1  NON-VOLATILE
C          L=0   NON-CONDENSABLE
C
0051      191 DO 192 J=1,NK
0052      IF (L(J).GT.0) GO TO 198
0053      JJ=J+NK
0054      DO 194 K=1,NK2
0055      194 AMAT(JJ,K)=0.0
0056      198 CONTINUE
C
0057      200 CONTINUE
C
C          SOLUTION TO      D=A *B      (INVERSE).
C                          I  I-1
C
C          NOTE THAT THE SOLUTION IS TRANSPOSED.
C
0058      CALL GAUSL (21,42,NK2,NK2,CM)
0059      DO 210 J=1,NK2
0060      DO 210 K=1,NK2

```

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```

0061      KK=NK2+K
0062      CM(K,J)=CM(J,KK)
0063      210 CONTINUE
C
C      FORMATION OF THE MATRICES ACCORDING TO THE THOMAS PROCEDURE.
C
C      DERIVATIVES ON STAGE 'I' ARE STORED AS FOLLOWS:
C      * STAGE 'I-1' IN 'AMAT'
C      * STAGE 'I' IN 'BMAT'
C      * STAGE 'I+1' GENERATED AS NEEDED ... FTM STM 220-400
C
0064      DO 250 J=1,NK2
0065      DO 250 K=1,NK
C
0066      GO TO NSW2, (230,220)
C
0067      220 IF (I.GT.2) GO TO 230
0068      BMAT(I,J,K)=BMAT(I,J,K)+CM(J,K)
0069      GO TO 240
0070      230 BMAT(I,J,K)=BMAT(I,J,K)+CM(J,K)+CM(J,NK2)*(HLL(I,K)+HHE(I)+FL(I)*
*          DEXH(I,K))
0071      240 CONTINUE
0072      250 CONTINUE
C
0073      GO TO NSW3, (270,260)
C
0074      260 IF (I.EQ.2) GO TO 290
0075      270 DO 280 J=1,NK2
0076      280 BMAT(I,J,NK1)=BMAT(I,J,NK1)+CM(J,NK2)*(CL(I)+FL(I)*DEXH(I,NK1))
0077      290 CONTINUE
C
0078      DO 300 J=1,NK2
0079      DO 300 K=1,NK2
0080      300 D(I,J)=D(I,J)-CM(J,K)*D(I-1,K)
C
0081      350 CONTINUE
C
C      CORRECTION OF THE VARIABLES ACCORDING TO THE NEWTON-RAPHSON METHOD.
C      THE SIMPLE STRUCTURE OF THE COMPLETE JACOBIAN MATRIX IS UTILIZED.
C
0082      DO 550 KK=1,NST
0083      I=NST+1-KK
C
0084      DO 360 J=1,NK2
0085      CM(J,NK2+1)=D(I,J)
0086      DO 360 K=1,NK2
0087      360 CM(I,K)=BMAT(I,J,K)
0088      IF (I.EQ.NST) GO TO 410
C
0089      DO 370 J=1,NK
0090      370 CM(J,NK2+1)=CM(J,NK2+1)+D(I+1,J)
C
0091      GO TO NSW4, (390,380)
C
0092      380 IF (I.EQ.1) GO TO 410
0093      390 DO 400 J=1,NK

```

```

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0094      400 CM(NK2,NK2+1)=CM(NK2,NK2+1)+D(I+1,J)*(HLL(I+1,J)+HHE(I+1)+FL(I+1)*
          * DEXH(I+1,J))
0095      CM(NK2,NK2+1)=CM(NK2,NK2+1)+D(I+1,NK1)*(CL(I+1)+FL(I+1)*DEXH(I+1,N
          * K1))
          C
0096      410 CALL GAUSL (21,42,NK2,1,CM)
0097      DO 460 J=1,NK2
0098      460 D(I,J)=CM(J,NK2+1)
          C
          C      EVALUATE SIZE OF COMPUTED CORRECTIONS
          C      * MAX CHANGE SET BY FLMAX, TMAX, & FVMAX
          C      * NEGATIVE FLOWS ARE NOT PERMITTED
          C
0099      DO 490 J=1,NK
0100      IF (D(I,J).EQ.0.0) GO TO 485
0101      SAVE=FLL(I,J)
0102      QQ=1.0
0103      IF (ABS(D(I,J)).LT.FLMAX(J)) GO TO 480
0104      QQ=FLMAX(J)/ABS(D(I,J))
0105      480 FLL(I,J)=FLL(I,J)+QQ*D(I,J)
0106      IF (FLL(I,J).LE.0.0) FLL(I,J)=SAVE/10.0
0107      485 IF (L(J).EQ.0) FLL(I,J)=0.0
0108      490 CONTINUE
          C
0109      IF (ITR.EQ.0) D(I,NK1)=0.0
0110      IF (D(I,NK1).EQ.0.0) GO TO 510
0111      QQ=1.0
0112      IF (ABS(D(I,NK1)).LT.TMAX) GO TO 500
0113      QQ=TMAX/ABS(D(I,NK1))
0114      500 T(I)=T(I)+D(I,NK1)*QQ
          C
0115      510 IF (I.EQ.NST .AND. FV(I).EQ.0.0) GO TO 550
0116      DO 540 JJ=NK1,NK2
0117      J=JJ-NK1
0118      IF (D(I,JJ).EQ.0.0) GO TO 535
0119      SAVE=FVV(J,J)
0120      QQ=1.0
0121      IF (ABS(D(I,JJ)).LT.FVMAX(J)) GO TO 530
0122      QQ=FVMAX(J)/ABS(D(I,JJ))
0123      530 FVV(I,J)=FVV(I,J)+D(I,JJ)*QQ
0124      IF (FVV(I,J).LE.0.0) FVV(I,J)=SAVE/10.0
0125      535 IF (L(J).EQ.-1) FVV(I,J)=0.0
0126      540 CONTINUE
          C
0127      550 CONTINUE
          C
          C      GENERATE THE REVISED TOTAL FLOW
          C
0128      DO 560 I=1,NST
0129      FL(I)=0.0
0130      FV(I)=0.0
0131      DO 560 J=1,NK
0132      FL(I)=FL(I)+FLL(I,J)
0133      560 FV(I)=FV(I)+FVV(I,J)
          C
0134      RETURN

```

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END

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GAUSSL

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```

0001      SUBROUTINE GAUSSL (ND,NCOL,N,NS,A)
          C
0002      IMPLICIT REAL*8  (A-H,O-Z)
0003      DIMENSION        A(21,42)
          C
          C      GAUSS SOLVES  $A \cdot X = B$ , WHERE A IS  $N \times N$ , BY GAUSSIAN ELIMINATION
          C      WITH ROW PIVOTING. THE VECTOR B IS PLACED IN COLUMN N+1 OF A
          C      WITH A BEING DESTROYED.
          C
0004      NI=N+1
0005      NT=N+NS
0006      IF (N.EQ.1) GO TO 50
          C      START ELIMINATION
          DO 40 I=2,N
0007          IP=I-1
0008          I1=IP
0009          X=DABS(A(I1,I1))
0010          DO 10 J=I,N
0011              IF (DABS(A(J,I1)).LT.X) GO TO 10
0012              X=DABS(A(J,I1))
0013              IP=J
0014          10 CONTINUE
0015          IF (IP.EQ.I1) GO TO 30
          C      ROW INTERCHANGE
0017          DO 20 J=I1,NT
0018              X=A(I1,J)
0019              A(I1,J)=A(IP,J)
0020          20 A(IP,J)=X
0021          30 DO 40 J=I,N
0022              X=A(J,I1)/A(I1,I1)
0023              DO 40 K=I,NT
0024          40 A(J,K)=A(J,K)-X*A(I1,K)
          C      ELIMINATION FINISHED, NOW COMPLETE THE BACK SUBSTITUTION
0025          50 DO 70 IP=1,N
0026              I=NI-IP
0027              DO 70 K=NI,NT
0028                  A(I,K)=A(I,K)/A(I,I)
0029                  IF (I.EQ.1) GO TO 70
0030                  I1=I-1
0031                  DO 60 J=1,I1
0032          60 A(J,K)=A(J,K)-A(I,K)*A(J,I)
0033          70 CONTINUE
          C
0034      RETURN
0035      END

```

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SUBROUTINE RESULT

```

C
  IMPLICIT REAL    (M)
  INTEGER          COMPNT
  DIMENSION       OUT(15,10), PROD(10)

C
  COMMON /CTRL/   NCR, NPRT, NSTART, TITLE(20)

C
  COMMON /NJITO/  NK, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
  *              NAME(4,10), L(10), NDM, COMPNT(10)
  COMMON /INRO/   NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
  *              NFEED, LF(9), ICODE, ITR, ITRMAX, ICODE
  COMMON /INR1/   FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
  *              DEST, DESTV, DESTL, RTMS, RFLX, ROIL, PP(50), T(50),
  *              MH(50), SL(50), SV(50), FL(50), FLL(50,10),
  *              FV(50), FVV(50,10), ZCONVT, ZCONVF
  COMMON /INR2/   DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
  *              DEXH(50,11), HV(50), HVV(50,10), CV(50)
  COMMON /VLE/    XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)

C
  DO 7 I=1,NST
  7 HL(I)=HL(I)+FL(I)*HHE(I)

C
  GO TO (11,15,15), KUNITS
 11 TDEG=(T(1)-273.16)*1.8+32.0
  C1=453.59
  C2=252.16*1.0E3
  GO TO 17
 15 TDEG=T(1)
  C1=1.0
  C2=1.0

C
 17 ISTR=1
  OUT(1,1)=1
  OUT(2,1)=TDEG
  OUT(3,1)=0.0
  OUT(4,1)=HL(1)/C2
  DO 18 J=1,NK
 18 OUT(J+4,1)=FLL(1,J)/C1
  OUT(NK1+4,1)=FL(1)/C1

C
  DO 60 I=1,NST
  IF (SL(I)+SV(I) .EQ. 0.0) GO TO 60
  GO TO (22,23,23), KUNITS
 22 TDEG=(T(I)-273.16)*1.8+32.0
  GO TO 25
 23 TDEG=T(I)
 25 IF (SL(I) .EQ. 0.) GO TO 40
  ISTR=ISTR+1
  DO 30 J=1,NK
 30 PROD(J)=FLL(I,J)*SL(I)/FL(I)
  H=HL(I)/FL(I)*SL(I)/C2
 35 OUT(1,ISTR)=I
  OUT(2,ISTR)=TDEG
  OUT(3,ISTR)=0.0
  OUT(4,ISTR)=H

```

```

      DO 39 J=1,NK
39  OUT(J+4,ISTR)=PROD(J)/C1
      OUT(NK1+4,ISTR)=SL(I)/C1
40  IF (SV(I).EQ.0.) GO TO 60
      ISTR=ISTR+1
      DO 50 J=1,NK
50  PROD(J)=FVV(I,J)*SV(I)/FV(I)
      H=HV(I)/FV(I)*SV(I)/C2
55  OUT(1,ISTR)=I
      OUT(2,ISTR)=TDEG
      OUT(3,ISTR)=1.0
      OUT(4,ISTR)=H
      DO 59 J=1,NK
59  OUT(J+4,ISTR)=PROD(J)/C1
      OUT(NK1+4,ISTR)=SV(I)/C1
60  CONTINUE
C
      GO TO (61,62,62), KUNITS
61  TDEG=(T(NST)-273.16)*1.8+32.0
      GO TO 65
62  TDEG=T(NST)
C
65  ISTR=ISTR+1
      OUT(1,ISTR)=NST
      OUT(2,ISTR)=TDEG
      OUT(3,ISTR)=1.0
      IF (IDCODE.NE.0) OUT(3,ISTR)=0.0
      OUT(4,ISTR)=HV(NST)/C2
      DO 66 J=1,NK
66  OUT(J+4,ISTR)=FVV(NST,J)/C1
      OUT(NK1+4,ISTR)=FV(NST)/C1
C
70  IF (NCASE.EQ.1) GO TO 90
      IF (IDCODE.NE.0) GO TO 71
      NC=HV(NSTM1)-HL(NST)*(1.0+SL(NST)/FL(NST))-HV(NST)
      GO TO 72
71  NC=HV(NSTM1)-HL(NST)/FL(NST)*(FL(NST)+DEST)
72  NR=HV(1)+HL(1)-HL(2)
C
      NC=NC/C2
      NR=NR/C2
C
79  WRITE (NPRT,84) NR
      WRITE (NPRT,86) NC
84  FORMAT ('ORFROTHER DUTY      =',1PE14.6)
86  FORMAT ('CONDENSER DUTY      =',1PE14.6)
C
90  WRITE (NPRT,100)
100 FORMAT ('PRODUCT STREAMS:')
      WRITE (NPRT,101) (OUT(1,I),I=1,ISTR)
101 FORMAT (' STAGE', 15X, 10F12.0)
      WRITE (NPRT,102) (OUT(2,I),I=1,ISTR)
102 FORMAT (' TEMP ', 15X, 10F12.2)
      WRITE (NPRT,103) (OUT(3,I),I=1,ISTR)
103 FORMAT (' FRACV', 15X, 10F12.4)
      WRITE (NPRT,104) (OUT(4,I),I=1,ISTR)

```



```

104 FORMAT (' H      ', 15X, 1P10E12.4)
    DO 105 J=1,NK
105 WRITE (NPRT,106) J, (NAME(K,J),K=1,2), (OUT(J+4,I),I=1,ISTR)
106 FORMAT (' COMP', 12, 2X,2A4,5X, F11.3, 9F12.3)
    WRITE (NPRT,107) (OUT(NK1+4,I),I=1,ISTR)
107 FORMAT (' TOTAL', 15X, 10F12.3)

```

C

```

    WRITE (NPRT,200)
200 FORMAT (' OFINAL COLUMN PROFILES: ' /
* ' STAGE      TEMP      PRES      EFF (%)      LIQUID(MOLS)      VAPOR
*(MOLS) ' )
    DO 210 I=1,NST

```

C

```

    GO TO (201,202,202), KUNITS
201 TDEG=(T(I)-273.15)*1.8+32.0
    PRES=PP(I)*14.696
    GO TO 205
202 TDEG=T(I)
    PRES=PP(I)

```

C

```

205 FLZ=FL(I)/C1
    FVZ=FV(I)/C1
    IF (IDCODE.NE.0.AND.I.EQ.NST) FVZ=0
    WRITE (NPRT,206) I,TDEG,PRES,MU(I),FLZ,FVZ
206 FORMAT (15,F10.2,F12.3,2PF12.2,0PF15.3,F14.3)
210 CONTINUE

```

C

```

    DO 310 J1=1,NK,5
        J2=J1+4
        IF (J2.GT.NK) J2=NK
        WRITE (NPRT,301) ((NAME(K,J),K=1,2),J=J1,J2)
301 FORMAT (' OFINAL VAPOR COMPOSITION PROFILE (MOLE%): ' /
* ' STAGE', 3X,2A4,4(5X,2A4) )
        DO 303 I=1,NST
            DO 302 J=J1,J2
                Y(J)=FVV(I,J)/FV(I)
                IF (IDCODE.NE.0.AND.J.EQ.NST) Y(J)=0
302 CONTINUE
303 WRITE (NPRT,304) I, (Y(J),J=J1,J2)
304 FORMAT (15,2P5F13.5)
310 CONTINUE

```

C

```

    DO 410 J1=1,NK,5
        J2=J1+4
        IF (J2.GT.NK) J2=NK
        WRITE (NPRT,401) ((NAME(K,J),K=1,2),J=J1,J2)
401 FORMAT (' OFINAL LIQUID COMPOSITION PROFILE (MOLE%): ' /
* ' STAGE', 3X,2A4,4(5X,2A4) )
        DO 403 I=1,NST
            DO 402 J=J1,J2
                X(J)=FLL(I,J)/FL(I)
403 WRITE (NPRT,304) I, (X(J),J=J1,J2)
410 CONTINUE

```

C

```

    DO 429 J1=1,NK,5
        J2=J1+4

```

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      IF (J2.GT.NK) J2=NK
      WRITE (NPRT,421) ((NAME(K,J),K=1,2),J=J1,J2)
421  FORMAT ('FINAL VAPOR COMPOSITION PROFILE (MOLES):'/
*    ' STAGE', 3X,2A4,4(5X,2A4))
      DO 423 I=1,NST
      DO 422 J=J1,J2
      Y(J)=FVV(I,J)/C1
      IF (IDCOEF.NE.0.AND.I.EQ.NST) Y(J)=0.0
422  CONTINUE
423  WRITE (NPRT,424) I, (Y(J),J=J1,J2)
424  FORMAT (I5,5F13.3)
429  CONTINUE
C
      DO 439 J1=1,NK,5
      J2=J1+4
      IF (J2.GT.NK) J2=NK
      WRITE (NPRT,431) ((NAME(K,J),K=1,2),J=J1,J2)
431  FORMAT ('FINAL LIQUID COMPOSITION PROFILE (MOLES):'/
*    ' STAGE', 3X,2A4,4(5X,2A4))
      DO 433 I=1,NST
      DO 432 J=J1,J2
432  X(J)=FLL(I,J)/C1
433  WRITE (NPRT,424) I, (X(J),J=J1,J2)
439  CONTINUE
C
      IF (NK.GT.0) GO TO 700
      DO 550 I=1,NST
      CALL KFAC1(I)
C      TDEG=(T(I)-273.16)*1.8+32.0
C      PRES=PP(I)*14.696
      TDEG=T(I)
      PRES=PP(I)
      WRITE (NPRT,501) I,TDEG,PRES
501  FORMAT ('STAGE #', I5/
*    ' TEMP, DEG K', F10.3/
*    ' PRES, MMHG ', F10.4)
      WRITE (NPRT,502)
502  FORMAT (' #',2X,'NAME',4X,2X,6X,'K',14X,
*    'VP',13X,'FUG',12X,'GAM')
      DO 510 J=1,NK
510  WRITE (NPRT,520) J, (NAME(K,J),K=1,2),XFK(J),
*    VP(J),FUG(J),GAM(J)
520  FORMAT (I3,2X,2A4,2X,1P6G15.5)
550  CONTINUE
C
700  RETURN
      END

```

IN IV 6 LEVEL 21

KFAC

DATE = 81357

19/38/32

SUBROUTINE KFAC(I)

C

```

IMPLICIT REAL      (M)
INTEGER            COMPNT
REAL               XX(10), GAMX(10), YY(10), FIX(10), XFKX(10),
*                 VSAVE(10)

```

C

```

COMMON /CTRL/      NCR, NPRT, NSTART, TITLE(20)
COMMON /NJIT0/     NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
*                 NAME(4,10), L(10), NDIM, COMPNT(10)
COMMON /NJIT1/     CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
*                 W(10)
COMMON /INR0/      NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
*                 WFEED, LF(9), IDCODE, ITR, ITRMAX, ICODE
COMMON /INR1/      FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
*                 DEST, DESTV, DESTL, BTMS, RFLX, BOJL, PP(50), T(50),
*                 MU(50), SL(50), SV(50), FL(50), FLL(50,10),
*                 FV(50), FVV(50,10), ZCONVT, ZCONVF
COMMON /INR2/      DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
*                 DEXH(50,11), HV(50), HVV(50,10), CV(50)
COMMON /VLE/      XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
COMMON /KNST/     XFKNST(10)

```

C

```

DATA              SMALL /0.0/

```

C

```

HL(I)=0.0
HV(I)=0.0
CL(I)=0.0
CV(I)=0.0
DHV=0.0
HHE(I)=0.0
DO 1 J=1,NC
DO 1 K=1,NK2
1 DFK(J,K)=0.0
DO 2 J=1,NK1
2 DEXH(I,J)=0.0

```

C

```

TT=T(I)
TS=TT+5.0
P=PP(I)
XT=FL(I)
YT=FV(I)
DO 3 J=1,NC
Y(J)=FVV(I,J)/YT
3 X(J)=FLL(I,J)/XT

```

C

```

IF (I.NE.NST) GO TO 8
IF (IDCODE.EQ.0) GO TO 8
VSAVE(NK1)=FV(NST)
YT=0.0
DO 5 J=1,NC
Y(J)=XFKNST(J)*X(J)
5 YT=YT+Y(J)
DO 6 J=1,NC
Y(J)=Y(J)/YT
VSAVE(J)=FVV(NST,J)

```

```

6 FVV(NST,J)=Y(J)*DEST
  YT=DEST
  FV(NST)=DEST

```

C

```

8 DO 9 J=1,NC
  FIX(J)=1.0
  FUG(J)=1.0
  GAM(J)=1.0
  HLL(I,J)=(((FNP(J,5)*TT+ENP(J,4))*TT+ENP(J,3))*TT+FNP(J,2))*
*          TT+ENP(J,1)
  HVV(I,J)=(((FNP(J,10)*TT+ENP(J,9))*TT+ENP(J,8))*TT+
*          ENP(J,7))*TT+ENP(J,6)
  HL(I)=HL(I)+HLL(I,J)*FLL(I,J)
  HV(I)=HV(I)+HVV(I,J)*FVV(I,J)
  CL(I)=CL(I)+FLL(I,J)*(((CPL(J,4)*TT+CPL(J,3))*TT+CPL(J,2))*TT+
*          CPL(J,1))
  CV(I)=CV(I)+FVV(I,J)*(((CPV(J,4)*TT+CPV(J,3))*TT+CPV(J,2))*TT+
*          CPV(J,1))
  VP(J)=EXP(ANT(1,J)+ANT(2,J)/(ANT(3,J)+TT)+ANT(4,J)*TT+ANT(5,J)*
*          TT*TT+ANT(6,J)*ALOG(TT))
9 CONTINUE

```

C

```

10 IF (NDIM.EQ.0) GO TO 11
  CALL OMEDI (P,TT,Y(NDIM),AKA,ZDIM,ZNODIM)
  VP(NDIM)=(SQRT(1.0+4.0*AKA*VP(NDIM))-1.0)/2.0/AKA

```

C

```

11 IF (IDLL.EQ.0) GO TO 20
  ISW=0
  CALL ACTVTY (TT,X,GAM)

```

C

```

20 IF (IDLV.EQ.0) GO TO 40
  CALL FUGCTY (TT,P,Y,VP,FUG,DHV)
  IF (IDH.EQ.2) GO TO 35
  HV(I)=HV(I)+DHV*FV(I)
  DO 30 J=1,NC
30 HVV(I,J)=HVV(I,J)+DHV
35 IF (IDLV.EQ.4) GO TO 60

```

C

```

40 DO 50 J=1,NC
50 XFK(J)=VP(J)/P*GAM(J)*FUG(J)

```

C

C

CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
WITH RESPECT TO COMPONENT FLOW IN THE LIQUID PHASE.

C

```

60 IF (IDLL.EQ.0) GO TO 110
  ISW=1
  DO 105 K=1,NC
  IF (L(K).EQ.0 .OR. X(K).LT.SMALL) GO TO 105
  FLOWK=FLL(I,K)*0.01
  DO 70 J=1,NC
70 XX(J)=FLL(J,K)/(XT+FLOWK)
  XX(K)=(FLL(I,K)+FLOWK)/(XT+FLOWK)
91 CALL ACTVTY (TT,XX,GAMX)
99 DO 100 J=1,NC
100 DFK(J,K)=VP(J)/P*FUG(J)*(GAMX(J)-GAM(J))/FLOWK
105 CONTINUE

```

AN IV 6 LEVEL 21

KFAC

DATE = 81357

19/36/32

C
C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
C WITH RESPECT TO COMPONENT FLOW IN THE VAPOR PHASE.

C
110 IF (IDLV.EQ.0.0) GO TO 140
DO 135 K=1,NC
IF (L(K).EQ.-1 .OR. Y(K).LT.SMALL) GO TO 135
FLOWK=(FV(I,K)+FV(I,K)*0.01)
120 YY(J)=FVV(I,J)/(YI+FLOWK)
YY(K)=(FVV(I,K)+FLOWK)/(YI+FLOWK)
CALL FUGCTY (TT,P,YY,VP,FIX,DHVEK)
IF (IDH.EQ.2) GO TO 125
HV(I,K)=HV(I,K)+(DHVEK-DHV)/FLOWK*FV(I)
125 DO 140 J=1,NC
130 DEK(J,K+NK1)=VP(J)/P*GAM(J)*(FIX(J)-FUG(J))/FLOWK
135 CONTINUE

C
C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
C WITH RESPECT TO TEMPERATURE.

C
140 DO 152 J=1,NC
152 DEK(J,NK1)=XFK(J)*(-ANT(2,J)/(TT+ANT(3,J))**2+ANT(4,J)+
* 2.0*ANT(5,J)*TT+ANT(6,J)/TT)

C
C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
C WITH RESPECT TO TEMPERATURE FOR A DIMERIZING COMPONENT.

C
160 IF (NDIM.EQ.0) GO TO 161
CALL ONEDI (P,TS,Y(NDIM),AKAS,ZDIMM,ZNO)
VPNDIM=EXP(ANT(1,NDIM)+ANT(2,NDIM)/(ANT(3,NDIM)+TS)+
* ANT(4,NDIM)*TS+ANT(5,NDIM)*TS*TS+ANT(6,NDIM)*ALOG(TS))
VPNDIM=(SQRT(1.0+4.0*AKAS*VPNDIM)-1.0)/2.0/AKAS
DEK(NDIM,NK1)=(VPNDIM-VP(NDIM))/5.0/P*GAM(NDIM)*FUG(NDIM)

C
161 IF (IDLV.EQ.0) GO TO 180
CALL FUGCTY (TS,P,Y,VP,FIX,DHVT)
IF (IDH.EQ.2) GO TO 165
CV(I)=CV(I)+(DHVT-DHV)/5.0*FV(I)
165 DO 170 J=1,NC
170 DEK(J,NK1)=DEK(J,NK1)+VP(J)/P*GAM(J)*(FIX(J)-FUG(J))/5.0

C
180 IF (IDLL.EQ.0) GO TO 200
ISA=0
CALL ACTVTY (TS,Y,GAMX)
DO 190 J=1,NC
190 DEK(J,NK1)=DEK(J,NK1)+VP(J)/P*FUG(J)*(GAMX(J)-GAM(J))/5.0

C
200 IF (IDLL.EQ.0) GO TO 230
IF (IDH.GT.0) GO TO 230
C
ISA=0
CALL XSHLID (TT,X,HHE(I))
ISA=1
DO 220 K=1,NC
IF (L(K).EQ.0 .OR. X(K).LT.SMALL) GO TO 220
FLOWK=FLL(I,K)*0.10

```
      DO 210 J=1,NC
210  XX(J)=FLL(I,J)/(XT+FLOWK)
      XX(K)=(FLL(I,K)+FLOWK)/(XT+FLOWK)
      CALL XSHLIQ (TT,XX,XSH)
      DEXH(I,K)=(XSH-HHE(I))/FLOWK
220  CONTINUE
      ISW=0
      CALL XSHLIQ (TT+20.0,X,XSH)
      DEXH(I,NK1)=(XSH-HHE(I))/20.0
C
230  IF (I.NE.NST) GO TO 250
      IF (IDCODE.EQ.0) GO TO 250
      FV(NST)=VSAVE(NK1)
      DO 240 J=1,NC
      XFKNST(J)=XFK(J)
240  FVV(NST,J)=VSAVE(J)
C
250  RETURN
      END
```

AN IV G LEVEL 21

FUGCTY

DATE = 81357

19/38/32

SUBROUTINE FUGCTY (T,P,Y,VP,F,DH)

C

REAL Y(1), VP(1), F(1)
INTEGER COMPNT

C

COMMON /NJITO/ NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
* NAME(4,10), L(10), NDIM, COMPNT(10)

C

GO TO (10,20), IDLV

10 CALL KKFUG (T,P,Y,VP,F,DH)

GO TO 99

20 CALL VIRFUG (T,P,Y,VP,F,DH)

C

99 RETURN
END

```

RTRAN IV 6 LEVEL 21          ACTVTY          DATE = 81357          19/38/32

001          SUBROUTINE ACTVTY (T,X,G)
----- C
002          REAL          X(1), G(1)
003          INTEGER      COMPNT
----- C
004          COMMON /NJITO/ NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
*          NAME(4,10), L(10), NDIM, COMPNT(10)
----- C
005          GO TO (10,20,30), IDLL
006          10 CALL RENOC (T,X,G)
007          GO TO 99
008          20 CALL WILSON (T,X,G)
009          GO TO 99
010          30 CALL UNIQAC (T,X,G)
----- C
011          99 RETURN
012          END

```


ORTRAN IV G LEVEL 21

XSHLIQ

DATE = 81357

19/38/32

```
0001      SUBROUTINE XSHLIQ (T,X,XSH)
----- C
0002      REAL          X(1)
0003      INTEGER       COMPNT
----- C
0004      COMMON /NJITO/ NC, NCM1, IDL, IDLV, IDH, LDBUG, ISW,
----- *
          NAME(4,10), L(10), NDIM, COMPNT(10)
----- C
0005      GO TO (10,20,30), IDLL
0006      10 CALL XSHREN (T,X,XSH)
-----
0007      GO TO 99
0008      20 CALL XSHWIL (T,X,XSH)
0009      GO TO 99
0010      30 CALL XSHUNQ (T,X,XSH)
----- C
0011      99 RETURN
0012      END
```

RTRAN IV G LEVEL 21

RKFUG

DATE = 81357

19/38/32

```

001      SUBROUTINE RKFUG (T,P,Y,VP,FUG,DHV)
          C
002      INTEGER      COMPNT
003      REAL          Y(1), VP(1), FUG(1), A(10), B(10)
          C
004      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISK,
          *          NAME(4,10), L(10), NDTM, COMPNT(10)
005      COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *          PC(10), OA(10), OB(10), OQ(10,10), W(10,10)
          C
          C      P = ATM
          C      T = DEG K
          C      VOL = MOLAR VOLUME (CC/G-MOL)
          C      OA AND OB = W-K PARAMETERS
          C
006      D=T**1.25
007      AA=0.0
008      BB=0.0
009      DO 30 J=1,NC
010      IF (Y(J).EQ.0.0) GO TO 30
011      Z=J
012      A(J)=OA(J)/D
013      B(J)=OB(J)/T
014      AA=AA+A(J)*Y(J)
015      BB=BB+B(J)*Y(J)
016      IF (L(J)-1) 30,20,10
017      10 FUG(J)=EXP(VOL(J)*P/(82.05*T))
018      GO TO 30
019      20 CALL ZCALC (A(J),B(J),VP(J),Z)
020      FUG(J)=EXP(VOL(J)*(P-VP(J))/(82.05*T))
021      FUG(J)=EXP(Z-1.0)/((Z-B(J)*VP(J))*((1.0+B(J)*VP(J)/Z)**
          *          (A(J)*A(J)/B(J))))*FUG(J)
022      30 CONTINUE
          C
023      Z=NC+1
024      CALL ZCALC (AA,BB,P,Z)
025      ZMTX=Z
026      FN=(Z-1.0)/BB
027      DF=ALOG(Z-BB*P)
028      DZ=(AA*AA/BB)*ALOG(1.0+BB*P/Z)
029      DHV=1.9872*T*(Z-1.0-1.5*DZ)
030      DO 50 J=1,NC
031      IF (Y(J).EQ.0.0) GO TO 50
032      IF (L(J).LE.0) GO TO 50
033      40 FUG(J)=FUG(J)/EXP(FN*B(J)-DF-DZ*(2.0*A(J)/AA-B(J)/BB))
034      50 CONTINUE
          C
035      IF (NDIM.EQ.0) GO TO 70
036      CALL ONEFI (P,TE,Y(NDIM),AK,ZDIM,ZNOD)
037      DO 60 J=1,NC
038      60 FUG(J)=FUG(J)/ZNOD
039      FUG(NDIM)=1.0/ZDIM
          C
040      70 RETURN
041      END

```

ORTRAN IV G LEVEL 21

ZCALC

DATE = 81357

19/38/32

```

0001      SUBROUTINE ZCALC (A,B,P,Z)
          C
0002      INTEGER          COMPNT
          C
0003      COMMON /CTRL/   NCR, NPRT, NSTART, TITLE(20)
0004      COMMON /NJIT0/  NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
          *              NAME(4,10), L(10), NDIM, COMPNT(10)
          C
0005      K=7
          C
0006      J=0
0007      D=(A*A-B-B*B*P)*P
0008      Z=1.0
0009      10 FN=Z**3-7*7+D*Z-A*A*P*P*B
0010      DF=3.0*Z*Z-7-7+D
0011      DZ=-FN/DF
0012      J=J+1
0013      IF (J.GT.50) GO TO 30
0014      IF (ABS(DZ)/Z.LT.0.0001) GO TO 50
0015      DZ=SIGN(AMIN1(ABS(DZ),0.1*Z,0.1),DZ)
0016      Z=Z+DZ
0017      GO TO 10
          C
0018      30 Z=AMAX1(Z,0.3)
          C
0019      50 IF (LDRUG.EQ.2) WRITE (NPRT,60) K,A,B,P,J,Z
0020      60 FORMAT (' ** ZCALC ** K,A,B,P,J,RZ:', I5,3G15.6, I5,G15.6)
          C
0021      RETURN
0022      END

```

```

0001      SUBROUTINE XSHREV (T,X,XSH)
          C
0002      INTEGER      COMPNT
0003      REAL          X(1)
          C
0004      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDBG, ISW,
          *          NAME(4,10), L(10), NDIM, COMPNT(10)
0005      COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *          PC(10), OA(10), OB(10), A(10,10), G(10,10),
          *          THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
          *          W(10), R(10), XL(10), PHI(10), DUMMY(10)
          C
0006      IF (ISW.NE.0) GO TO 20
          C
0007      PT=1.9872*T
0008      K=0
0009      DO 15 I=1,NCM1
0010      IF (X(I).NE.0.0) GO TO 5
0011      K=K+NC-I
0012      GO TO 15
0013      5  IJ=I+1
0014      DO 10 J=IJ,NC
0015      K=K+1
0016      IF (X(J).EQ.0.0) GO TO 10
0017      II=4*K
0018      S=(C(II-3)+C(II-2)*T)/PT
0019      A(I,J)=S
0020      G(I,J)=EXP(-S*ALPHA(K))
0021      S=(C(II-1)+C(II)*T)/RT
0022      A(J,I)=S
0023      G(J,I)=EXP(-S*ALPHA(K))
0024      10 CONTINUE
0025      15 CONTINUE
          C
0026      20 XSH=0.0
0027      DO 70 J=1,NC
0028      IF (X(I).EQ.0.0) GO TO 70
0029      SA=0.0
0030      SB=0.0
0031      SC=0.0
0032      SD=0.0
0033      DO 60 J=1,NC
0034      IF (X(J).EQ.0.0) GO TO 60
0035      S=G(J,I)*X(J)
0036      SB=SB+S
0037      SC=SC+S*A(J,I)
0038      TF (I-J)30,60,40
0039      30 K=NC*(I-1)+J-I*(I+1)/2
0040      IJ=4*K-1
0041      GO TO 50
0042      40 K=NC*(J-1)+I-J*(J+1)/2
0043      IJ=4*K-3
0044      50 SA=SA+(1.0-ALPHA(K)*A(J,I))*C(IJ)*S
0045      SD=SD+ALPHA(K)*C(IJ)*S
0046      60 CONTINUE
0047      XSH=XSH+X(I)*(SC*SD/SB+SA)/SB

```

ORTRAN IV G LEVEL 21

XSHREN

DATE = 81357

19/38/37

0048 70 CONTINUE

C

0049 RETURN

0050 END

ORTRAN IV G LEVEL 21

RENON

DATE = 81357

19/38/3;

```

0001      SUBROUTINE RENON (T,X,GAM)
          C
0002      INTEGER      COMPNT
0003      REAL          X(1), GAM(1)
          C
0004      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
          *          NAME(4,10), L(10), NDIM, COMPNT(10)
0005      COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
          *          PC(10), OA(10), OB(10), A(10,10), G(10,10),
          *          THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
          *          Q(10), R(10), XL(10), PHT(10), DUMMY(10)
          C
0006      IF (ISW.NE.0) GO TO 20
          C
0007      RT=1.9872*T
0008      K=0
0009      DO 15 I=1,NCM1
0010      IF (X(I).NE.0.0) GO TO 5
0011      K=K+NC-1
0012      GO TO 15
0013      5 IJ=I+1
0014      DO 10 J=IJ,NC
0015      K=K+1
0016      IF (X(J).EQ.0.0) GO TO 10
0017      II=4*K
0018      S=(C(II-3)+C(II-2)*T)/RT
0019      A(I,J)=S
0020      C(T,J)=EXP(-S*ALPHA(K))
0021      S=(C(II-1)+C(II)*T)/RT
0022      A(J,J)=S
0023      G(J,I)=EXP(-S*ALPHA(K))
0024      10 CONTINUE
0025      15 CONTINUE
          C
0026      20 DO 50 I=1,NC
0027      IF (X(I).EQ.0.0) GO TO 50
0028      S=0.0
0029      SA=0.0
0030      SB=0.0
0031      DO 40 J=1,NC
0032      IF (X(J).EQ.0.0) GO TO 40
0033      SC=0.0
0034      SD=0.0
0035      DO 30 K=1,NC
0036      IF (X(K).EQ.0.0) GO TO 30
0037      SE=G(K,J)*X(K)
0038      SC=SC+SF
0039      SD=SD+SE*A(K,J)
0040      30 CONTINUE
0041      SE=G(J,I)*X(J)
0042      SH=SH+G(I,J)*X(J)*(A(I,J)-SD/SC)/SC
0043      SA=SA+SE
0044      S=S+SE*A(J,I)
0045      40 CONTINUE
0046      GAM(I)=EXP(S/SA+SH)
0047      50 CONTINUE

```

ORTRAN IV G LEVEL 21

RENON

DATE = 81357

19/38/38

0048
0049

C

RETURN
END

N IV G LEVEL 21

XSHWIL

DATE = 81357

19/38/32

SUBROUTINE XSHWIL (TE,X,XSH)

C

```

INTEGER      COMPNT
REAL         X(1)

```

C

```

COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
*        NAME(4,10), L(10), NDIM, COMPNT(10)
COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
*        PC(10), OA(10), OH(10), A(10,10), G(10,10)

```

C

IF (ISW.NE.0) GO TO 20

C

```

RT=1.9472*TE
K=0
DO 10 I=1,NCM1
  IJ=I+1
  DO 10 J=IJ,NC
    K=K+4
    S=VOL(I)/VOL(J)
    G(I,J)=EXP(-(C(K-3)+C(K-2)*TE)/RT)*S
    G(J,I)=EXP(-(C(K-1)+C(K)*TE)/RT)/S

```

10 CONTINUE

C

```

20 XSH=0.0
DO 70 I=1,NC
  IF (X(I).EQ.0.0) GO TO 70
  SA=0.0
  SB=0.0
  DO 60 J=1,NC
    IF (X(J).EQ.0.0) GO TO 60
    S=G(J,I)*X(J)
    SB=SB+S
    IF (1-J) 30,60,40
30  K=NC*(I-1)+J-I*(I+1)/2
    JJ=4*K-1
    GO TO 50
40  K=NC*(J-1)+I-J*(J+1)/2
    IJ=4*K-3
50  SA=SA+C(IJ)*S
60  CONTINUE
    SB=1.0/SB
    XSH=XSH+X(I)*SA*SB
70  CONTINUE

```

C

```

RETURN
END

```


AN IV G LEVEL 21

WILSON

DATE = 81357

19/38/32

SUBROUTINE WILSON (TE,X,GAM)

C

```

INTEGER          COMPNT
REAL             X(1), GAM(1)

```

C

```

COMMON /NJIT0/  NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
*              NAME(4,10), L(10), NDIM, COMPNT(10)
COMMON /NJIT2/  C(180), ALPHA(45), VC(10), VOL(10), TC(10),
*              PC(10), OA(10), OB(10), A(10,10), G(10,10)

```

C

```

IF (ISW.NE.0) GO TO 20

```

C

```

RT=1.9472*TE
K=0
DO 10 I=1,NCM1
  IJ=1+1
  DO 10 J=IJ,NC
    K=K+4
    S=VOL(I)/VOL(J)
    G(I,J)=EXP(-(C(K-3)+C(K-2)*TF)/RT)*S
    G(J,I)=EXP(-(C(K-1)+C(K)*TE)/RT)/S

```

```

10 CONTINUE

```

C

```

20 DO 60 I=1,NC
  S=0.0
  SA=0.0
  DO 50 J=1,NC
    IF (X(J).EQ.0.0) GO TO 50
30 SH=0.0
  DO 40 K=1,NC
    IF (X(K).EQ.0.0) GO TO 40
    SH=G(K,J)*X(K)+SH
40 CONTINUE
  S=S+X(J)*G(J,I)
  SA=SA+X(J)*G(I,J)/SH
50 CONTINUE
  GAM(I)=EXP(1.0-SA)/S
60 CONTINUE

```

C

```

RETURN
END

```

AN IV G LEVEL 21

UNIQAC

DATE = 81357

19/38/32

SUBROUTINE UNIQAC (TE,X,GAM)

C

```

      INTEGER          COMPNT
      REAL             X(1), GAM(1)

```

C

```

      COMMON /NJIT0/  NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW,
*                   NAME(4,10), L(10), NDIM, COMPNT(10)
      COMMON /NJIT2/  C(180), ALPHA(45), VC(10), VOL(10), TC(10),
*                   PC(10), OA(10), OB(10), A(10,10), G(10,10),
*                   THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
*                   Q(10), R(10), XL(10), PHI(10), DUMMY(10)

```

C

```

      IF (IS*.NF.0) GO TO 20

```

C

```

      RT=1.9872*TE
      K=0
      DO 10 I=1,NCM1
      IJ=I+1
      DO 10 J=IJ,NC
      K=K+4
      G(I,J)=EXP(-(C(K-3)+C(K-2)*TE)/RT)
      G(J,I)=EXP(-(C(K-1)+C(K)*TE)/RT)
10 CONTINUE

```

C

```

20 THETS=0.0
   PHS=0.0
   XLS=0.0
   DO 30 I=1,NC
   THETS=THETS+N(I)*X(I)
   PHS=PHS+R(I)*X(I)
30 XLS=XLS+XL(I)*X(I)
   DO 40 I=1,NC
   THETA(I)=Q(I)*X(I)/THETS
40 PHI(I)=X(I)*R(I)/PHS
   DO 50 I=1,NC
   THS(I)=0.0
   DO 50 J=1,NC
50 THS(I)=THS(I)+THETA(J)*G(J,I)
   DO 70 I=1,NC
   GA=ALOG(R(I)/PHS)+5.0*Q(I)*ALOG(Q(I)/R(I)*PHS/THETS)
   GA=GA+XL(I)-R(I)/PHS*XLS
   GB=1.0-ALOG(THS(I))
   DO 60 J=1,NC
60 GB=GB-THETA(J)*G(I,J)/THS(J)
70 GAM(I)=EXP(GA+Q(I)*GB)

```

C

```

      RETURN
      END

```

AN IV G LEVEL 21

XSHUNQ

DATE = 81357

19/38/32

```

C      SUBROUTINE XSHUNQ (TE,X,XSH)
      INTEGER      COMPNT
      REAL         X(1)
C
      COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDBUG, ISW,
*            NAME(4,10), L(10), NDIM, COMPNT(10)
      COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
*            PC(10), OA(10), OR(10), A(10,10), G(10,10),
*            THETA(10), THS(10), HFAC(10), HHFA(10), HM(10),
*            Q(10), R(10), XL(10), PHT(10), DUMMY(10)
C
      IF (ISW.NE.0) GO TO 20
C
      RT=1.9872*TF
      K=0
      DO 10 I=1,NCM1
      IJ=I+1
      DO 10 J=IJ,NC
      K=K+4
      G(I,J)=EXP(-(C(K-3)+C(K-2)*TE)/RT)
      G(J,I)=EXP(-(C(K-1)+C(K)*TE)/RT)
10  CONTINUE
20  XSH=0.0
      THETS=0.0
      DO 30 I=1,NC
30  THETS=THETS+Q(I)*X(I)
      DO 40 J=1,NC
40  THETA(J)=Q(J)*Y(J)/THETS
      DO 50 I=1,NC
      HFAC(I)=0.0
      HHFA(I)=0.0
      DO 50 J=1,NC
      ARG=G(J,I)
      HFAC(I)=HFAC(I)+THETA(J)*ARG*ALOG(ARG)
50  HHFA(I)=HHFA(I)+THETA(J)*ARG
      DO 60 J=1,NC
60  HM(I)=Q(I)*X(I)*HFAC(I)/HHFA(I)
      DO 70 I=1,NC
70  XSH=XSH-1.9872*TF*HM(I)
C
      RETURN
      END

```

AN IV G LEVEL 21

VIRFUG

DATE = 81357

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```

SUBROUTINE VIRFUG (T,P,Y,VP,FUG,DHV)
C
C REAL Y(1),VP(1),FUG(1)
C
COMMON /NJIT0/ NC, NCM1, IDLL, JDLV, IDH, LDBUG, ISW, NDIM,
* NAME(4,10), L(10)
COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
* PC(10), UA(10), OR(10), AA(10,10), G(10,10)
COMMON /NJIT3/ TTC(10,10), PPC(10,10), VVC(10,10), ZZC(10,10),
* AK(10,10), AVALUE(10,10), HVALUE(10,10),
* MCROSS(10,10), OMEGA(10), DIPOLE(10), R(10,10)
P=P/760.0

```

```

C
C SUBROUTINE TSONOP GENERATES B(I,J) FOR VIRFUG
C CALL TSONOP(T,P,MARK)
C

```

```

RT=82.057*T
PORT=P/RT
SUMY=0.0
BMIX=0.0
DO 15 I=1,NC
VP(I)=VP(I)/760.0
SUMY=SUMY+Y(I)
DO 10 J=1,NC
BMIX=BMIX+Y(I)*Y(J)*B(I,J)
10 CONTINUE
15 CONTINUE
BMIX=BMIX /SUMY**2
IF (MARK) 20,20,22
20 SQARG=0.25 + BMIX*PORT

```

```

C
C IF BMIX*P/RT IS MORE NEGATIVE THAN 0.25, THE PRESSURE SERIES VIRIAL
C EQ. MUST BE USED. THE ARGUMENT MARK IS USED TO COMMUNICATE THIS
C TO THE MAIN PROGRAM
C

```

```

IF (SQARG) 22,25,25
22 VMIX=BMIX +1. /PORT
ZMIX=PORT*VMIX
DO 24 I=1,NC
SUMB=0.0
DO 23 J=1,NC
SUMB =SUMB +Y(J)*B(I,J)
23 CONTINUE
SUMB=SUMB / SUMY
PHI=EXP ((2.0*SUMB-BMIX)*PORT)
PHIS=EXP (R(I,I)*VP(I)/RT)
FACTOR=(P-VP(I))*VOL(I)
PONTY= EXP(FACTOR/RT)
FUG(I) = PHIS*PONTY/PHI
C PRINT 300,I,PHIS,PONTY,PHI,FUG(I),Y(I),VP(I)
24 CONTINUE
MARK=MARK+1
GO TO 50
25 VMIX=(0.5+SQRT (SQARG))/PORT
ZMIX=PORT*VMIX

```

```

C
C  CALCULATION OF FUGACITY COEFFICIENTS
C
  DO 40 I=1,NC
  SUMB=0.0
  DO 30 J=1,NC
  SUMB=SUMB + Y(J)*B(I,J)
30 CONTINUE
  SUMB=SUMB /SUMY
  PHI=EXP (2.0*SUMB/VMIX)/ZMIX
  PHIS = EXP((B(I,I)*VP(I))/RT)
  FACTOR=(P-VP(I))*VOL(I)
  PONTY= EXP(FACTOR/RT)
  FUG(I) = PHIS*PONTY/PHI
C  PRINT 300,I,PHIS,PONTY,PHI,FUG(I),Y(I),VP(I)
300  FORMAT(15,6G15.6)
  40 CONTINUE
  50  DHV=1.9872*T*(ZMIX-1.0)
C  PRINT 400,DHV,T,P
  400  FORMAT(5X,'DHV=',G15.6,'TEMP=',G15.6,'PRESS=',G15.6)
C  DO 51 I=1,NC
C  DO 51 J=1,NC
C  51  WRITE(6,401) I,J,B(I,J)
  401  FORMAT(5X,'B(',I2,',',I2,')=',G15.5)
  DO 52 I=1,NC
  52  VP(I)=VP(I)*760.0
  P=P*760.0
  RETURN
  END

```

1 SUBROUTINE TSONOP(T,P,MARK)

C THIS SUBROUTINE CALCULATES SECOND VIRIAL COEFFICIENTS FOR THE
C TRUNCATED VIRIAL EQUATION OF STATE BY USE OF TSONOPOULOS'
C CORRELATION; AICHE JOURNAL, VOL. 20, NO. 2, P. 263, MARCH 1974
C
C
C

C THE INPUT COMMONS PASS ALONG ALL THE INPUTTED INFORMATION
C

2 *COMMON /NJIT0/ NMEM(8M10)IDLK(10)IDLV, IDH, LDRUG, ISW, NDTM,
3 COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
* PC(10), OA(10), OR(10), AA(10,10), G(10,10)
4 COMMON /NJIT3/ TTC(10,10), PPC(10,10), VVC(10,10), ZZC(10,10),
* AK(10,10), AVALUE(10,10), BVALUE(10,10),
* HCRDSS(10,10), OMEGA(10), DTPOLE(10), B(10,10)

C THE OUTPT COMMONS PASS ALONG MOST OF THE CALCULATED INFORMATION
C
C
C

5 FTS0(TR)=0.1445-(0.330+0.1385/TR)/TR-(0.0121+0.000607
1 /TR**5)/TR**3
6 FTS1(TR)=0.0637+(0.331-0.423/TR-0.008/TR**6)/TR**2
7 FTS2(TR,AAA,BBB)=(AAA-BBB/TR**2)/TR**6

C CALL VIRMIX

MARK=-1

R=R2.055

DO 101 I=1,NC

TR=T/TTC(I,I)

101 R(I,I)=(FTS0(TR)+OMEGA(I)*FTS1(TR)+FTS2(TR,AVALUE(I,I),
1 BVALUE(I,I)))*R/PPC(I,I)*TTC(I,I)

DO 121 J=2,NC

KMAX=I-1

DO 122 K=1,KMAX

TR=T/TTC(J,K)

W=(OMEGA(J)+OMEGA(K))/2.0

R(J,K)=(FTS0(TR)+W*FTS1(TR)+FTS2(TR,AVALUE(J,K),BVALUE(J,K)))*R
1/PPC(J,K)*TTC(J,K)

R(K,J)=R(J,K)

122 CONTINUE

121 CONTINUE

RETURN

END

1 SUBROUTINE VIRMIK

2 COMMON /NJIT0/ NC, NCM1, IDLL, IDLV, IDH, LDRUG, ISW, NDIM,
* NAME(4,10),L(10)

3 COMMON /NJIT3/ TTC(10,10),PPC(10,10),VVC(10,10),ZC(10,10),
* AK(10,10), AVALUE(10,10), BVALUE(10,10),
* BCROSS(10,10), OMEGA(10), DIPOLE(10), B(10,10)
4 COMMON /NJIT2/ C(180), ALPHA(45), VC(10), VOL(10), TC(10),
* PC(10), OA(10), OB(10), AA(10,10), G(10,10)

C
C
C
C

THIS SUBROUTINE CALCULATES THE MIXING BINARY PROPERTIES

5 RCONST=82.057
6 EXPN = 1.0/3.0

C
C
C

CALCULATE CRITICAL PROPERTIES FOR BINARY INTERACTIONS

7 DO 100 J=2,NC
8 JMAX=J-1
9 DO 100 J=1,JMAX
0 TTC(I,I) = SQRT (TTC(I,I) * TTC(J,J))
1 TTC(J,J)=TTC(I,J)*(1.-AK(I,J))
2 PPC(I,J) = 4.0 * TTC(I,J) * (PPC(J,I) * VC(I) / TTC(I
1,I)+PPC(J,J) * VC(J) / TTC(J,I)) / (VC(I)**EXPN +
2 VC(J)**EXPN)**3
3 BVALUE(I,J) = (BVALUE(I,I) + BVALUE(J,J))/2.
4 AVALUE(I,J) = (AVALUE(I,I) +AVALUE(J,J))/2.
5 BVALUE(J,I) = BVALUE(I,J)
6 AVALUE(J,I) = AVALUE(I,J)
7 VVC(J,I)=VVC(I,J)
8 TTC(J,I)=TTC(I,J)
9 PPC(J,I)=PPC(I,J)
0 100 CONTINUE
1 RETURN
2 END

RAN IV G LEVEL 21

ONEDI

DATE = 81357

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```

1      SUBROUTINE ONEDI (P,ATEM,Y,AKA,ZDI,ZNODI)
      C
      C      ATEM IN DEGREE K
      C
      C      THIS SUBROUTINE CALCULATES THE VAPORPHASE DIMERIZATION CORRECTIONS
      C      WHEN NO MORE THAN ONE COMPONENT DIMERIZES.
      C      FOR THE DIMERIZATING COMPONENT, THE FUGACITY COEFFICIENT IS MODI-
      C      FIED BY THE FACTOR ZDI. FOR ALL THE OTHER COMPONENTS, THE FUGACITY
      C      COEFFICIENTS ARE MODIFIED BY THE FACTOR ZNODI.
      C
2      COMMON /DATA3/ BI1(10,10), BI2(10,10), T1, T2, A, RX
      C
3      IF (Y-0.0001)20,20,10
4      10 AKA=EXP((2.880814-A+RX/ATEM)*2.3025851)
5      AKOR=4.0*AKA*P*Y*(2.0-Y)
6      ZDI=(SQRT(1.0+AKOR)-1.0)*2.0/AKOR
7      ZNODI=2.0*(1.0-Y+SQRT(1.0+AKOR))/(2.0-Y)/(1.0+SQRT(1.0+AKOR))
8      GO TO 30
9      20 ZDI=1.0
0      ZNODI=1.0
1      30 CONTINUE
      C
2      RETURN
3      END

```


RAW IV G LEVEL 21

KFAC1

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```

1      SUBROUTINE KFAC1(I)
2      C
3      IMPLICIT REAL      (M)
4      INTEGER          COMPNT
5      REAL            XX(10), GAMX(10), YY(10), FIX(10), XFKX(10),
6      *              VSAVE(10)
7      C
8      COMMON /CTRL/    NCR, NPRT, NSTART, TITLE(20)
9      COMMON /NJIT0/   NC, NCM1, IDLL, IDLV, IDH, LDBG, ISW,
10     *              NAME(4,10), L(10), NDIM, COMPNT(10)
11     COMMON /NJIT1/   CPL(10,4), CPV(10,4), ENP(10,10), ANT(6,10),
12     *              K(10)
13     COMMON /INR0/    NST, NSTM1, NK1, NK2, NK11, NK21, NCASE, KUNITS,
14     *              NFEED, LF(9), ICODE, ITR, ITRMAX, ICODE
15     COMMON /INR1/    FSTR(9,11), FKV(9), HF(9), FLMAX(10), FVMAX(10),
16     *              DEST, DESTV, DESTL, BTMS, RFLX, BOIL, PP(50), T(50),
17     *              MU(50), SL(50), SV(50), FL(50), FLL(50,10),
18     *              FV(50), FVV(50,10), ZCONVT, ZCONVF
19     COMMON /INR2/    DFK(10,21), HL(50), HLL(50,10), CL(50), HHE(50),
20     *              DEXH(50,11), HV(50), HVV(50,10), CV(50)
21     COMMON /VLE/    XFK(10), GAM(10), X(10), Y(10), VP(10), FUG(10)
22     COMMON /KNST/   XFKNST(10)
23     C
24     DATA          SMALL /0.0/
25     C
26     HL(I)=0.0
27     HV(I)=0.0
28     CL(I)=0.0
29     CV(I)=0.0
30     DHV=0.0
31     HHE(I)=0.0
32     DO 1 J=1,NC
33     DO 1 K=1,NK2
34     1 DFK(J,K)=0.0
35     DO 2 J=1,NK1
36     2 DEXH(I,J)=0.0
37     C
38     TT=T(I)
39     TS=TT+5.0
40     P=PP(J)
41     XT=FL(I)
42     YT=FV(I)
43     DO 3 J=1,NC
44     Y(J)=FVV(J,J)/YT
45     3 X(J)=FLL(I,J)/XT
46     C
47     IF (I.NE.NST) GO TO 8
48     IF (ICODE.EQ.0) GO TO 8
49     VSAVE(NK1)=FV(NST)
50     YT=0.0
51     DO 5 J=1,NC
52     Y(J)=XFKNST(J)*X(J)
53     5 YT=YT+Y(J)
54     DO 6 J=1,NC
55     Y(J)=Y(J)/YT
56     VSAVE(J)=FVV(NST,J)

```

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```

6 FVV(NST,J)=Y(J)*DEST
  YI=DEST
  FV(NST)=DEST

```

```

8 DO 9 J=1,NC
  FIX(J)=1.0
  FUG(J)=1.0
  GAM(J)=1.0
  HLL(I,J)=(((ENP(J,5)*TT+ENP(J,4))*TT+ENP(J,3))*TT+ENP(J,2))*
  * TT+ENP(J,1)
  HVV(I,J)=(((ENP(J,10)*TT+ENP(J,9))*TT+ENP(J,8))*TT+
  * ENP(J,7))*TT+ENP(J,6)
  HL(I)=HL(I)+HLL(I,J)*FLL(I,J)
  HV(I)=HV(I)+HVV(I,J)*FVV(I,J)
  CL(I)=CL(I)+FLL(I,J)*(((CPL(J,4)*TT+CPL(J,3))*TT+CPL(J,2))*TT+
  * CPL(J,1))
  CV(I)=CV(I)+FVV(I,J)*(((CPV(J,4)*TT+CPV(J,3))*TT+CPV(J,2))*TT+
  * CPV(J,1))
  VP(J)=EXP(ANT(1,J)+ANT(2,J)/(ANT(3,J)+TT)+ANT(4,J)*TT+ANT(5,J))*
  * TT+ANT(6,J)*ALOG(TT)

```

```

9 CONTINUE
10 IF (NDIM.EQ.0) GO TO 11
  CALL ONEDI (P,TT,Y(NDIM),AKA,ZDIM,ZNODIM)
  VP(NDIM)=(SQRT(1.0+4.0*AKA*VP(NDIM))-1.0)/2.0/AKA

```

```

11 IF (IDLL.EQ.0) GO TO 20
  IS*=0
  CALL ACTVTY (TT,X,GAM)

```

```

20 IF (IDLV.EQ.0) GO TO 40
  CALL FUGCTY (TT,P,Y,VP,FUG,DHV)
  IF (IDH.EQ.2) GO TO 35
  HV(I)=HV(I)+DHV*FV(I)
  DO 30 J=1,NC
30 HVV(J,J)=HVV(J,J)+DHV
35 IF (IDLV.EQ.4) GO TO 60

```

```

40 DO 50 J=1,NC
50 XFK(J)=VP(J)/P*GAM(J)*FUG(J)

```

```

C
C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
C WITH RESPECT TO COMPONENT FLOW IN THE LIQUID PHASE.
C

```

```

60 IF (IDLL.EQ.0) GO TO 110
  IS*=1
  DO 105 K=1,NC
  IF (L(K).EQ.0 .OR. X(K).LT.SMALL) GO TO 105
  FLOWK=FLL(I,K)*0.01
  DO 70 J=1,NC
70 XX(J)=FLL(I,J)/(XT+FLOWK)
  XX(K)=(FLL(I,K)+FLOWK)/(XT+FLOWK)
91 CALL ACTVTY (TT,XX,GAMX)
99 DO 100 J=1,NC
100 DFK(J,K)=VP(J)/P*FUG(J)*(GAMX(J)-GAM(J))/FLOWK
105 CONTINUE

```

C
 C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
 C WITH RESPECT TO COMPONENT FLOW IN THE VAPOR PHASE.
 C

```

5 110 IF (IDLV.EQ.0.0) GO TO 140
6 DO 135 K=1,NC
7 IF (L(K).EQ.-1 .OR. Y(K).LT.SMALL) GO TO 135
8 FLOWK=FVV(I,K)*0.01
9 DO 120 J=1,NC
0 120 YY(J)=FVV(I,J)/(YT+FLOWK)
1 YY(K)=(FVV(I,K)+FLOWK)/(YT+FLOWK)
2 CALL FUGCTY (TT,P,YY,VP,FIX,DHVEK)
3 IF (IDH.EQ.2) GO TO 125
4 HVV(I,K)=HVV(I,K)+(DHVEK-DHV)/FLOWK*FV(J)
5 125 DO 130 J=1,NC
6 130 DFK(J,NK1)=VP(J)/P*GAM(J)*(FIX(J)-FUG(J))/FLOWK
7 135 CONTINUE

```

C
 C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
 C WITH RESPECT TO TEMPERATURE.
 C

```

8 140 DO 152 J=1,NC
9 152 DFK(J,NK1)=XFK(J)*(-ANT(2,J)/(TT+ANT(3,J))**2+ANT(4,J)+
* 2.0*ANT(5,J)*TT+ANT(6,J)/TT)

```

C
 C CALCULATION OF THE DERIVATIVE OF THE SEPARATION FACTOR
 C WITH RESPECT TO TEMPERATURE FOR A DIMERIZING COMPONENT.
 C

```

0 160 IF (NDIM.EQ.0) GO TO 161
1 CALL ONEDI (P,T5,Y(NDIM),AKAS,ZDIMM,ZND)
2 VPNDIM=EXP(ANT(1,NDIM)+ANT(2,NDIM)/(ANT(3,NDIM)+T5)+
* ANT(4,NDIM)*T5+ANT(5,NDIM)*T5*T5+ANT(6,NDIM)*ALOG(T5))
3 VPNDIM=(SQRT(1.0+4.0*AKAS*VPNDIM)-1.0)/2.0/AKAS
4 DFK(NDIM,NK1)=(VPNDIM-VP(NDIM))/5.0/P*GAM(NDIM)*FUG(NDIM)

```

C
 C 161 IF (IDLV.EQ.0) GO TO 180
 C CALL FUGCTY (T5,P,Y,VP,FIX,DHVT)
 C IF (IDH.EQ.2) GO TO 165
 C CV(I)=CV(I)+(DHVT-DHV)/5.0*FV(J)
 C 165 DO 170 J=1,NC
 C 170 DFK(J,NK1)=DFK(J,NK1)+VP(J)/P*GAM(J)*(FIX(J)-FUG(J))/5.0

C
 C 180 IF (IDLL.EQ.0) GO TO 200
 C ISW=0
 C CALL ACTVTY (T5,X,GAMX)
 C DO 190 J=1,NC
 C 190 DFK(J,NK1)=DFK(J,NK1)+VP(J)/P*FUG(J)*(GAMX(J)-GAM(J))/5.0

C
 C 200 IF (IDLL.EQ.0) GO TO 230
 C IF (IDH.GT.0) GO TO 230
 C ISV=0

C
 C CALL XSHLIN (T1,X,HHE(I))
 C ISW=1
 C DO 220 K=1,NC
 C IF (L(K).EQ.0 .OR. X(K).LT.SMALL) GO TO 220
 C FLOWK=FLL(I,K)*0.10

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```
      DO 210 J=1,NC
210  XX(J)=FLL(I,J)/(XT+FLOWK)
      XX(K)=(FLL(I,K)+FLOWK)/(XT+FLOWK)
      CALL XSHLIQ (TT,XX,XSH)
      DEXH(I,K)=(XSH-HHE(I))/FLOWK
220  CONTINUE
      ISW=0
      CALL XSHLIQ (TT+20.0,X,XSH)
      DEXH(I,NK1)=(XSH-HHE(I))/20.0
C
230  IF (I.NE.NST) GO TO 250
      IF (IDCODE.EQ.0) GO TO 250
      FV(NST)=VSAVE(NK1)
      DO 240 J=1,NC
      XFKNST(J)=XFK(J)
240  FVV(NST,J)=VSAVE(J)
C
250  RETURN
      END
```

SELECTED BIBLIOGRAPHY

- Abrams, D.S. and Prausnitz, J.M., Journal of the American Institute of Chemical Engineers, Vol. 21, No. 1, January 1975, pp 116-128.
- Anderson, T.F. and Prausnitz, J.M., Industrial and Engineering Chemistry Process Design and Development, Vol. 17, October 1978, pp 552-567.
- Benedict, M. and Rubin, L.C., Trans. American Institute of Chemical Engineers, Vol. 41, 1945, pp. 353-370.
- Benedict, M., Johnson, C.A., Solomon, E., and Rubin, L.C., Trans. American Institute of Chemical Engineers, Vol. 41, 1945, pp 371-392.
- Black, C., Chemical Engineering Progress, Sept. 1980, pp 78-85.
- Black, C. and Ditsler, D.E., "Extractive and Azeotropic Distillation," Advanced Chemical Series, Vol. 115, 1972, pp 1-15.
- Black, C., Golding, R.A., and Ditsler, D.E., "Extractive and Azeotropic Distillation," Adv. Chem. Ser., Vol. 115, 1972, pp 64-92.
- Block, U. and Hegner, B., Journal of the American Institute of Chemical Engineers, Vol. 22, No. 3, May 1976, pp 582-589.
- Boston, J.F. and Shah, V.B., "An Algorithm for Rigorous Distillation Calculations with Two Liquid Phases," paper presented at 86th AIChE National Meeting, Houston, April 1979.
- Deam, J.R. and Maddox, R.N., Hydrocarbon Processing, July 1969, pp 163-164.
- Ewell, R.H., Harrison, J.M., and Berg, L., Industrial and Engineering Chemistry, Vol. 36, No. 10, October 1944, pp 871.
- Colburn, A.P., Canadian Chemistry and Process Industries, April 1950, pp 286.
- Ellis, S.R.M. and Pearce, C.J., British Chemical Engineering, Dec. 1957, pp 648-653.
- Fredenslund, A., Gmehling, J., and Rasmussen, P., Vapor-Liquid Equilibria Using UNIFAC, Elsevier Scientific Publishing Company, Amsterdam, The Netherlands, 1977.

- Gasohol, "Ethanol Distillation," Chem Share Process Simulation, Houston, Texas, 1980.
- Gmehling, J. and Onken, U., Vapor-Liquid Equilibrium Data Collection, Vol. I, Part 1, DECHEMA, Frankfurt, W. Germany, 1977.
- Hachmuth, K.H., Chemical Engineering Progress, Vol. 48, No. 12, 1957, pp 617-626.
- Hayden, J.G. and O'Connell, J.P., Industrial and Engineering Chemistry, Process Design and Development, Vol. 14, No. 3, 1975, pp 209.
- Heidemann, R.A., Hydrocarbon Processing, Nov. 1974, pp 167-170.
- Holland, C.D., Gallun, S.E., and Lockett, M.J., Chemical Engineering, March 1981, pp 185-200.
- Holland, C.D., Multicomponent Distillation, Prentice-Hall, Inc., Englewood Cliffs, New Jersey, 1963.
- King, J.C., Separation Processes, McGraw-Hill Book Company, New York, 1971.
- Keister, W., "Computer Simulation of Ethanol Dehydration by Azeotropic Distillation with Benzene," Paper No. 21d, presented at the Winter AIChE Meeting, Feb. 28-March 3, 1982, Orlando, Florida.
- Keister, W., Private Communication, March 1982.
- Leach, M.J., Chemical Engineering, May 23, 1977, pp. 137-140.
- McDermott, C. and Stroud, S.E., Paper presented at CHISA Meeting, August 1978, in Prague.
- Naphtali, L.M. and Sandholm, D.P., Journal of the American Institute of Chemical Engineers, Vol. 17, No. 1, January 1971, pp 148-153.
- O'Connell, J.P. and Prausnitz, J.M., Industrial and Engineering Chemistry, Process Design and Development, Vol. 6, No. 2, April 1967, pp 246-250.
- Prausnitz, J.M., Private Communication, February 1981.
- Prausnitz, J.M., Molecular Thermodynamics of Fluid Phase Equilibria, Prentice-Hall, Inc., Englewood Cliffs, New Jersey, 1969.
- Prausnitz, J.M., Eckert, C.A., Orye, R.V., and O'Connell, J.P., Computer Calculations for Multicomponent Vapor-Liquid Equilibria, Prentice-Hall, Inc., Englewood Cliffs, N.J., 1967.

- Prausnitz, J.M., Anderson, T.F., Grem, E.A., Eckert, C.A., Hsieh, R., and O'Connell, J.P., Computer Calculations for Multicomponent Vapor-Liquid and Liquid-Liquid Equilibria, Prentice-Hall, Inc., Englewood Cliffs, New Jersey, 1980.
- Prokopakis, G.J., Seider, W.D., and Ross, B.A., "Azeotropic Distillation Tower with Two Liquid Phases," in Proceedings of Engineering Foundation Conference entitled Foundations of Computer-Aided Process Design, Hennike, N.H., July, 1980, Vol. 2, pp 239-269.
- Reid, R.C., Prausnitz, J.M., and Sherwood, T.K., The Properties of Gases and Liquids, McGraw-Hill Book Company, New York, N.Y., 3rd Edition, 1977.
- Renon, H., and Prausnitz, J.M., Journal of the American Institute of Chemical Engineers, Vol. 14, No. 1, January 1968, pp 135-143.
- Renon, H. and Prausnitz, J.M., Industrial and Engineering Chemistry, process Design and Development, Vol. 8, No. 3, July 1969, pp 413-419.
- Roche, E.C., Unpublished Work on Rigorous Multicomponent Distillation Calculation, 1978.
- Roche, E.C., Unpublished Work on Nonideal Liquid Flash Calculations, 1977.
- Ross, B.A. and Seider, W.D., "Simulation of Three-Phase Distillation Towers," Paper Presented at the 86th AIChE National Meeting, Houston, Texas, April, 1979.
- Smith, B.D., Design of Equilibrium Stage Processes, McGraw-Hill Book Company, New York, N.Y., 1963.
- Tsonopoulos, C., Journal of the American Institute of Chemical Engineers, Vol. 20, No. 2, March 1972, pp 263-272.
- Van Winkle, M., Distillation, McGraw-Hill Book Company, New York, N.Y., 1967.
- Yaws, C.L., Physical Properties, McGraw-Hill Book Company, New York, N.Y., 1977.