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NON-EQUILIBRIUM PARAMETRIC PUMPS FOR  
SEPARATING LIQUIDS OR GASES

By

ANIL KRISHNA RASTOGI

A THESIS

PRESENTED IN PARTIAL FULFILMENT OF  
THE REQUIREMENTS FOR THE DEGREE OF  
MASTER OF SCIENCE IN CHEMICAL ENGINEERING

AT

NEW-JERSEY INSTITUTE OF TECHNOLOGY

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NEWARK, NEW-JERSEY

1977

APPROVAL OF THESIS  
NON-EQUILIBRIUM PARAMETRIC PUMPS FOR  
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By

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FOR

DEPARTMENT OF CHEMICAL ENGINEERING  
NEWJERSEY INSTITUTE OF TECHNOLOGY

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ABSTRACT

An experimental and theoretical study has been done of Non-Equilibrium parametric pumps. Mathematical models and computer programmes are developed for both thermal and heatless parametric pumps based on non-equilibrium conditions and linear isotherms. A mathematical study was done for thermal parametric pumps whereas both experimental and theoretical studies were made for a heatless parametric pump. The effect of different operating conditions has been investigated on the separation of NaCl-H<sub>2</sub>O via a thermal parametric pump and He-CO<sub>2</sub> and He-C<sub>3</sub>H<sub>6</sub> gas mixtures via a heatless parametric pump.

The experimental study of heatless parametric pumps was extended to the separation of ternary gas mixtures. The effect of different operating conditions on the separation has been emphasized. The results based on binary gas model have been compared with ternary experimental results.

The mathematical study of thermal parametric pumps shows that separation of mixtures may be predicted. It also helps in predicting the effect of different operating conditions, such as  $\phi_B$ , half cycle time. Thus, mathematically optimum conditions can be found to attain a definite separation.

Similarly, a study of heatless parametric pumps shows that separation of gaseous mixtures is possible by this technique. The experimental studies shows that this technique is not only limited to binary mixtures but can be successfully applied to the multicomponent mixtures. The mathematically calculated results show the same effect as observed experimentally.

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TABLE OF CONTENTS

Page No.

|  |     |
|--|-----|
| Abstract   | ii  |
| Approval page  | iii |
| Acknowledgement  | i   |
| Table of contents  |     |
| List of figures  | 1   |
| List of tables   | 3   |
| Scope  | 4   |
| Background   | 6   |
| Theory   | 9   |
| A. Assumptions   | 9   |
| B. Development of characteristic equations for mathematical model. | 10  |
| C. Application of characteristic equations to                      |     |
| (i) thermal parametric pump  | 12  |
| (ii) heatless parametric pump                                      | 14  |
| D. Cyclic operation of thermal parametric pump                     | 17  |
| E. Computer calculation for thermal parametric pump                | 18  |
| F. Cyclic operation of heatless parametric pump                    | 20  |
| G. Computer calculation of heatless parametric pump                | 21  |
| Discussion of Results  | 26  |
| A. Thermal parametric pump - NaCl-H <sub>2</sub> O system          | 27  |
| B. Heatless parametric pump -                                      |     |
| (i) Binary gas mixture   | 28  |
| (ii) Ternary gas mixture   | 30  |

|  | Page No. |
|--|----------|
| Recommendations                              | 33       |
| Nomenclature                                 | 34       |
| References                                   | 37       |
| Appendices                                   |          |
| (i) Heatless parametric pump                 | 39       |
| I - Calculation of Mass transfer coefficient | 39       |
| II - Description of apparatus and operation  | 42       |
| (ii) Computer programmes for -               |          |
| III-A Thermal parametric pump                | 44       |
| III-B Heatless parametric pump               | 48       |
| (iii) Figures and Tables                     |          |
| IV- Figures                                  | 55       |
| V- Tables                                    | 56       |



LIST OF FIGURES

- Figure 1 Network of two families of characteristics.
- Figure 2 Adsorption isotherm for CO<sub>2</sub> and C<sub>3</sub>H<sub>6</sub> on silica gel at 30° C.
- Figure 3 Schematic drawing of continuous thermal parametric pump.
- Figure 4 Block diagram for thermal parametric pump computer calculation.
- Figure 5 Schematic drawing of heatless parametric pump.
- Figure 6 Stepwise outline of heatless parametric pumping.
- Figure 7 Repressurization and depressurization of columns.
- Figure 8 Block diagram for heatless parametric pump computer calculation.
- Figure 9 Adjustment of NNZ for He-CO<sub>2</sub> system.
- Figure 10 Concentration front for NaCl-H<sub>2</sub>O system.
- Figure 11 Effect of  $\phi_B$  on the separation of NaCl-H<sub>2</sub>O mixture.
- Figure 12 Effect of ' $\frac{\pi}{\omega}$ ' half cycle time on the separation of NaCl-H<sub>2</sub>O mixture.
- Figure 13 Steady state concentration vs. displacement for NaCl-H<sub>2</sub>O mixture.
- Figure 14 Adjustment of  $\alpha$  for He-CO<sub>2</sub> system.
- Figure 15,16 Calibration curve for He-CO<sub>2</sub> system.
- Figure 17 Comparison of experimental and calculated results for He-CO<sub>2</sub> mixture.
- Figure 18 Calibration curve for He-C<sub>3</sub>H<sub>6</sub> system.
- Figure 19 Comparison of experimental and calculated results for He-C<sub>3</sub>H<sub>6</sub> mixture.

- Figure 20 Effect of purge ratio ' $\gamma$ ' on the separation of He-C<sub>3</sub>H<sub>6</sub> mixture.
- Figure 21 Calibration curve for He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub> system.
- Figure 22 Separation of He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub> mixture at  $\gamma = 2.2$ .
- Figure 23 Saturation curve for He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub> mixture at 60 and 20 psia.
- Figure 24,25 Separation of He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub> mixture at  $\gamma = 1.5$  and  $\gamma = 1.0$  respectively.

| Table No.     | <u>Content of Tables</u>  |
|---------------|---|
| 1             | Mathematical result - Thermal Parametric Pump system NaCl-H <sub>2</sub> O                    |
| 2             | Experimental results - Heatless Parametric Pump   |
| 3             | Mathematical results- Heatless Parametric Pump  |
| 4,5           | Calibration for He-CO <sub>2</sub> system   |
| 6,7           | Cyclic run for He-CO <sub>2</sub> system.   |
| 8             | Calibration for He-C <sub>3</sub> H <sub>6</sub> system                                       |
| 9,10,11,12,13 | Cyclic run for He-C <sub>3</sub> H <sub>6</sub> system  |
| 14            | Calibration curve for He-CO <sub>2</sub> -C <sub>3</sub> H <sub>6</sub> system                |
| 15,16,17,18   | Cyclic run for He-CO <sub>2</sub> -C <sub>3</sub> H <sub>6</sub> system                       |
| 19,20         | Saturation of bed for CO <sub>2</sub> and C <sub>3</sub> H <sub>6</sub>                       |
| 21            | Thermal Parametric Pump parameters  |
| 22            | Heatless Parametric Pump parameters   |
| 23            | A typical case of computed concentration profiles for NaCl-H <sub>2</sub> O system            |
| 24            | A typical case of computed concentration profiles for He-C <sub>3</sub> H <sub>6</sub> system |

### SCOPE

Parametric pumping is a periodic separation technique which represents an alternative to the conventional adsorption process. This concept requires a solute/adsorbent system in which some controllable thermodynamic variable viz. temperature, pressure or pH affects the solute equilibrium distribution between the fluid and solid phases.

Parametric pumping has several advantage over conventional separation processes. In general, parametric cyclic technique is self generating. The operation is continuous. This technique increases the separating capacity for a given column. The capital investment for the process is relatively low for a given flow rates of products.

Parametric pumping technique can be categorized into two cases. In the first, solute and adsorbent(solid) are assumed to reach equilibrium instantaneously. The mass transfer coefficient between two phases is considered to be infinite. This is called ' Equilibrium parametric pumping.' In the second case which is closer to the real situation, the solute and solid do not reach equilibrium. The mass transfer rate is dependent upon the overall driving force available and the mass transfer coefficient is finite. This is known as NON-EQUILIBRIUM pumping.

Further, periodic separation caused by variable temperature is known as Thermal Parametric Pumping, whereas separation obtained using pressure as a thermodynamic variable is called Heatless Parametric Pumping. In principle, Thermal Parametric Pumping consists of changing the operating column temperature and reversing

the direction of flow during each half cycle. Two upflow and downflow half cycles constitute one cycle. During upflow or hot half cycle, solute is desorbed, while in reverse flow, cold half cycle solute is adsorbed. Thus, solute-rich and solute-lean products are obtained during respective half cycles. Thermal parametric pumps can be operated in batch, semicontinuously or continuously. On the other hand, heatless parametric pumping is based on changing of the total pressure during each half cycle, keeping the temperature of the system constant. During each half cycle, one column is maintained at high pressure and the other column at low pressure. At high pressure, solute is adsorbed and at low pressure, solute is desorbed, resulting in solute-lean product and solute-rich product from respective columns.

## BACKGROUND

Many workers have studied parametric pumping in one of its specific forms. Thermal parametric pumping was conceived by Wilhelm about 1962. Separation of NaCl-H<sub>2</sub>O liquid system using ion-exchange resins have been reported by Rice(1966), Wilhelm Rice and Bedelius(1966), Rolke(1967) and Wilhelm, Rice, Rolke and Sweed(1968). McAndrew(1967) studied recuperative thermal parametric pumping as applied to a gaseous methane-nitrogen system. Meyers(1968) showed the separation of equimolar feed of ethane and propane experimentally on a continuous flowrate basis. Meyers(1970) also developed a theoretical equilibrium plug flow model for the gaseous mixture separation.

Pigford, Baker and Blum(1969) also presented an equilibrium theory of the continuous parametric pumps. Sweed and Wilhelm(1969) computed the separation of Toluene-Heptane mixture using a numerical algorithm based on equilibrium operation and also determined the effect of displacement, cycle time, phase angle, and reservoir volume on separation. Chen and Hill(1971) investigated batch, semicontinuous, and continuous versions of the parametric pump in terms of an equilibrium theory. Patrick (1972) also showed the separation of the air-SO<sub>2</sub> mixture using a direct thermal mode and developed theoretical results based on instantaneous gas-solid equilibrium relationships. Patrick also studied the effect of the number of cycles, pumping cycle time and operating temperature range and level on the extent of separation over a wide range of operating conditions. Gregory(1974) made a

quantitative comparison of conventional adsorption with parametric pumping for a non-equilibrium system with nonlinear isotherms.

Some workers have investigated the separation of gaseous mixtures via heatless parametric pumps. A brief review is given below:

Alexis(1967) made a comparative study of heatless parametric pumping with other separation processes in upgrading hydrogen. Alexis showed that heatless parametric pumping is the most economical for treating moderate amounts of hydrogen. Kadlec (1971) separated nitrogen and methane using pressure as the driving force in a single column similar to a thermal parametric pump. Kadlec also developed a mathematical model based upon the assumption of instantaneous equilibrium between the gas phase and the adsorbed gas. Kadlec(1972) investigated the problem of optimum feed rate and optimum cycle time for heatless parametric pumping based on instantaneous equilibrium theory.

A different approach has been used by Shendalman(1972) in studying the separation of He-CO<sub>2</sub> mixture via a heatless parametric pump. Shendalman developed a mathematical model considering instantaneous equilibrium theory. Shendalman (1972) extended his experimental and theoretical investigation for non-equilibrium conditions.

The present work has been divided into three parts. In the first part, a mathematical model has been developed for the non-equilibrium thermal parametric pump. In the second part an experimental study has been done for non-equilibrium heatless-parametric pumps for separating binary gas mixtures and also a

mathematical model has been developed for the calculation of concentration profiles. In the third part, experimental study of heatless parametric pumps has been extended for the separation of ternary gas mixtures. The effect on the separation of different operating conditions such as half cycle time, operating pressure and purge ratio has been determined.



THEORY

Consider a differential element of thickness dx in the column. By material balance for the solute, one gets -

$$\epsilon \frac{\partial(cv)}{\partial x} + \rho_s (1-\epsilon) \frac{\partial w}{\partial t} + \epsilon \frac{\partial c}{\partial t} = 0 \quad \text{----(1)}$$

In the above equation the first term is due to bulk flow of fluid along the length of the column, the second and third terms are an accumulation in solid and fluid phases, respectively.

Rearranging equation (1) -

$$v \frac{\partial c}{\partial x} + \frac{\partial c}{\partial t} = - \frac{(1-\epsilon)}{\epsilon} \rho_s \frac{\partial w}{\partial t} - c \frac{\partial v}{\partial x} \quad \text{----(2)}$$

Similarly, for the total mass balance, at steady state-

$$\epsilon \frac{\partial \rho}{\partial t} + \frac{\partial(\rho v)}{\partial x} + (1-\epsilon) \rho_s \frac{\partial w}{\partial t} = 0 \quad \text{----(3)}$$

$$\frac{\partial \rho}{\partial t} + \epsilon \rho \frac{\partial v}{\partial x} + \epsilon v \frac{\partial \rho}{\partial x} + (1-\epsilon) \rho_s \frac{\partial w}{\partial t} = 0 \quad \text{----(4)}$$

Equations (2) and (4) are based on the following assumptions -

(A) ASSUMPTIONS :

1. Only one component is adsorbed in the flowing fluid.
2. Plug flow.
3. Negligible axial diffusion.
4. Molar density of flowing fluid has been considered constant.

Validity of this assumption for two different cases is explained below -

a) In liquid phase, the solute concentration is very small compared to that of the solvent. Hence, change in the concentration of solute has a negligible effect on the molar density of the fluid.

b) In the gas phase, the solute gas concentration is very small

and the density has been considered as that of the non adsorbing component helium only. If the pressure in the column is constant then the molar density of fluid will be constant.

5. Linear isotherms apply.

(B) Equations (2) and (4) are simplified further as below:

$$\frac{\partial \rho}{\partial t} = \frac{\partial \rho}{\partial x} = 0, \text{ for constant density fluid.}$$

Equation (4) then changes to :

$$\frac{\partial v}{\partial x} = - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{\rho} \frac{\partial w}{\partial t} \quad \text{----(5)}$$

$$\text{Also, } \frac{\partial w}{\partial t} = f(c, w, v) \quad \text{----(6)}$$

Equations (2), (5) and (6) are basic equations which are further used for generating characteristic equations. The steps are shown below :

$$c = f(x, t) \quad \text{----(7)}$$

$$\text{so, } dc = \frac{\partial c}{\partial x} dx + \frac{\partial c}{\partial t} dt,$$

$$\text{or } \frac{dc}{dx} = \frac{\partial c}{\partial x} + \frac{\partial c}{\partial t} \frac{dt}{dx} \quad \text{----(8)}$$

$$\text{If } \frac{dt}{dx} = \frac{1}{v},$$

Equation (8) changes to :

$$v \frac{dc}{dx} = v \frac{\partial c}{\partial x} + \frac{\partial c}{\partial t} \quad \text{----(9)}$$

According to the above condition, i.e., along a line of slope =  $\frac{1}{v}$ , the total derivative changes to the partial

derivative. Similarly,  $v = f(x, t)$ ,

$$\text{so, } dv = \frac{\partial v}{\partial x} dx + \frac{\partial v}{\partial t} dt,$$

or,  $\frac{dv}{dx} = \frac{\partial v}{\partial x} + \frac{\partial v}{\partial t} \frac{dt}{dx}$  . -----(10)

If  $\frac{dt}{dx} = 0$  ,

then  $\frac{dv}{dx} = \frac{\partial v}{\partial x}$  . -----(11)

This shows that along  $t =$  a constant, the total derivative changes to partial derivative. In other words, in a  $t-x$  axis at a fixed time, velocity is a function of  $x$  alone, which can be solved using Euler's iterative method or Runge-Kutta method.

Similarly,  $w = f(x,t)$  ,

$$dw = \frac{\partial w}{\partial x} dx + \frac{\partial w}{\partial t} dt$$
 . -----(12)

$$\frac{dw}{dt} = \frac{\partial w}{\partial x} \frac{dx}{dt} + \frac{\partial w}{\partial t}$$
 ,

If  $\frac{dx}{dt} = 0$  ,

Then equation (12) changes to -

$$\frac{dw}{dt} = \frac{\partial w}{\partial t}$$
 . -----(13)

Equation (13) shows that in a  $t-x$  axis, at a fixed distance in the column the solid concentration is a function of time only. Thus the solid phase concentration at any time can be solved using Euler's iterative method or Runge-Kutta method.

Combining equations (2), (5) and (6) with equations (9), (11) and (13) respectively, one obtains :

$$v \frac{dc}{dx} = - \frac{(1-\epsilon)}{\epsilon} \rho_s \frac{dw}{dt} - c \frac{dv}{dx}$$
 . -----(14)

Along,  $\frac{dt}{dx} = \frac{1}{v}$  . -----(14-a)

$$\frac{dv}{dx} = - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{v} \frac{dw}{dt}$$
 . -----(15)

Along,  $\frac{dt}{dx} = 0$  . -----(15-a)

and,  $\frac{dw}{dt} = f(c,w,v)$  . -----(16)

Along,  $\frac{dx}{dt} = 0$  . -----(16-a)

Equations (14) to (16-a) are used to generate concentration profiles for different parametric pumps. The application of the above equations in two types of pumps has been discussed separately.

(C)(i) THERMAL PARAMETRIC PUMPS

The following additional assumptions are used in simplifying equations (14) to (16-a) :

1. The concentration of the solute in the fluid mixture is very small. Hence, the velocity of fluid through the column is considered constant. Therefore,  $v = v_0$  ,
2. The fluid phase driving force is controlling for the calculation of mass-transfer coefficients.

If  $z = \frac{x}{v_0}$  , then the above equations become -

$\frac{dc}{dz} = - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{\rho} \frac{dw}{dt}$  . -----(17)

Along,  $\frac{dt}{dz} = 1$  . -----(17-a)

$\frac{dw}{dt} = \lambda \left( c - \frac{w}{M_T} \right)$  . -----(18)

Along,  $\frac{dz}{dt} = 0$  . -----(18-a)

Where,  $\lambda = \alpha v_0^{1-\beta}$  . -----(19)

and,  $M_T = a - b T$  . -----(20)

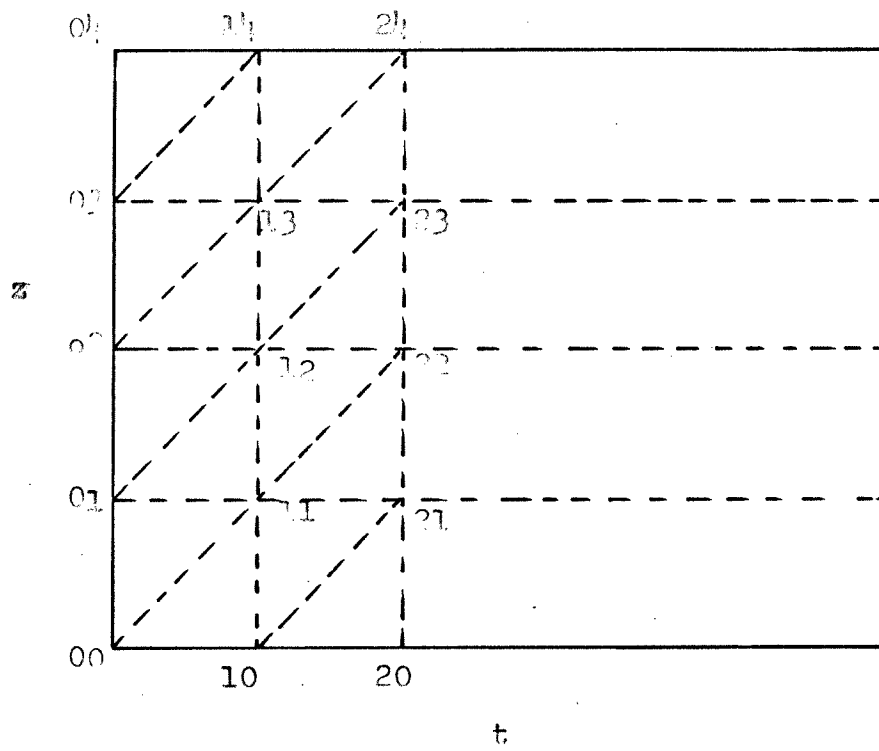


Figure 1

Network of Two Families of Characteristics

as given by Sweed (1971) values of constants  $\alpha, \beta, a$  and  $b$  are listed in Table-21. Equations (17) and (18) are solved in a  $t-z$  plane using Euler's iterative method as shown below :

In a  $t-z$  plane(Fig-1), the intersection of  $t = z + k \cdot \Delta t$  and  $z = l \cdot \Delta z$ , i.e., point  $(k, l)$  can be calculated using initial conditions along  $z = l \cdot \Delta z$  and  $\frac{dt}{dz} = 1$ . For example, if  $t$  is divided into  $k$  intervals and  $l = NNZ$ , then, considering  $t = 1 \cdot \Delta t$ ,  $c(1,1)$  and  $w(1,1)$  can be calculated using points  $(0,0)$  and  $(0,1)$ . The equations (17) to (20) can be written as below :

$$c(1,1) = c(0,0) - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{\rho} \lambda \int (c - \frac{w}{M_T}) dz \quad \text{-----(21)}$$

$$w(1,1) = w(0,1) + \lambda \int (c - \frac{w}{M_T}) dt \quad \text{-----(22)}$$

The first approximation for  $c(1,1)$  and  $w(1,1)$  will be :

$$c^1(1,1) = c(0,0) - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{\rho} \lambda (c(0,0) - \frac{w(0,0)}{M_T}) \Delta z \quad \text{-----(23)}$$

$$w^1(1,1) = w(0,0) + \lambda (c(0,1) - \frac{w(0,1)}{M_T}) \Delta z \quad \text{-----(24)}$$

A more exact result can be obtained by making use of a procedure analogous to the modified Euler's method used by Acrivos (1956) :

$$c^2(1,1) = c(0,0) - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{\rho} \lambda \frac{1}{2} (c^1(1,1) - \frac{w^1(1,1)}{M_T} + c(0,0) - \frac{w(0,0)}{M_T}) \Delta z \quad \text{-----(25)}$$

This iterative procedure is continued until the values of  $C(1,1)$

and  $w_{(1,1)}$  converge. In identical manner  $c_{(1,2)}$  and  $w_{(1,2)}$ ;  $c_{(1,3)}$  and  $w_{(1,3)}$ ;  $c_{(1,4)}$  and  $w_{(1,4)}$  can be calculated. The whole procedure is repeated for  $t = 2.\Delta t$ ;  $3.\Delta t$ ;  $k\Delta t$ . Thus, values of  $c$  and  $w$  can be generated at  $(i,j)$  point, once values at  $(i-1,j-1)$  and  $(i-1,j)$  are known.

(C) (ii) HEATLESS PARAMETRIC PUMPS

Equations (14) to (16-a) are again simplified for a heatless parametric pump with the following additional assumptions :

1. The velocity of gaseous mixture is constant through the column bed, i.e.,  $v = v_0$  ;
2. The gaseous mixture follows the simple ideal gas law. Hence,  $c = \rho y$  ;
3. The solid phase driving force is controlling for the calculation of mass-transfer coefficients.

Hence,

$$v_0 \frac{d(\rho y)}{dx} = - \frac{(1-\epsilon)}{\epsilon} \rho_s \frac{dw}{dt} ,$$

or, 
$$v_0 y \frac{d\rho}{dx} + v_0 \rho \frac{dy}{dx} = - \frac{(1-\epsilon)}{\epsilon} \rho_s \frac{dw}{dt} . \quad \text{----(27)}$$

Since,  $\rho$  is constant,  $\frac{d\rho}{dx} = 0$  , and equation (27)

changes to :

$$v_0 \frac{dy}{dx} = - \frac{(1-\epsilon)}{\epsilon} \frac{\rho_s}{\rho} \frac{dw}{dt} . \quad \text{----(28)}$$

Along, 
$$\frac{dt}{dx} = \frac{1}{v_0} . \quad \text{----(28-a)}$$

For the solid phase, concentration equations, (16) and (16-a) can be written as :

$$\frac{dw}{dt} = \lambda (\bar{w} - w) . \quad \text{----(29)}$$

and,  $\frac{dx}{dt} = 0$  . -----(29-a)

where  $\lambda$  is mass transfer coefficient that has been defined in Appendix-I.

Also,  $\ddot{w} = M_p \left( \frac{P y}{R T} \right)$  . -----(30)

where the value of equilibrium constant  $M_p$  is assumed to be constant calculated at  $P = 60$  psia from fig. - 2.

Let,  $z = \frac{x}{v_0}$ , equations (28) to (29-a) are transformed to the following form :

$\frac{dy}{dz} = - \frac{\rho_s}{\rho} \frac{(1-\epsilon)}{\epsilon} \left( \frac{y P M_p}{R T} - w \right)$  . -----(31)

Along,  $\frac{dt}{dz} = 1$  . -----(31-a)

$\frac{dw}{dt} = \lambda \left( \frac{y P M_p}{R T} - w \right)$  . -----(32)

Along,  $\frac{dz}{dt} = 0$  . -----(32-a)

Equations (31) and (32) are solved in a t-z plane using the Runge-Kutta method as discussed below .Consider in a t-z plane, two families of lines (31-a) and (32-a) such that the following conditions are satisfied :

$\frac{dt}{dz} = 1$  ----I ,

and,  $z = \text{constant}$  ----II ,

A network of I and II characteristics can be represented by the following equations :

$t = z + k \Delta t$  ,

$z = \ell \Delta z$  ,

where  $\Delta t$  and  $\Delta z$  are arbitrary infinitesimal increments and  $k, \ell$



SILICA GEL AT 30°C

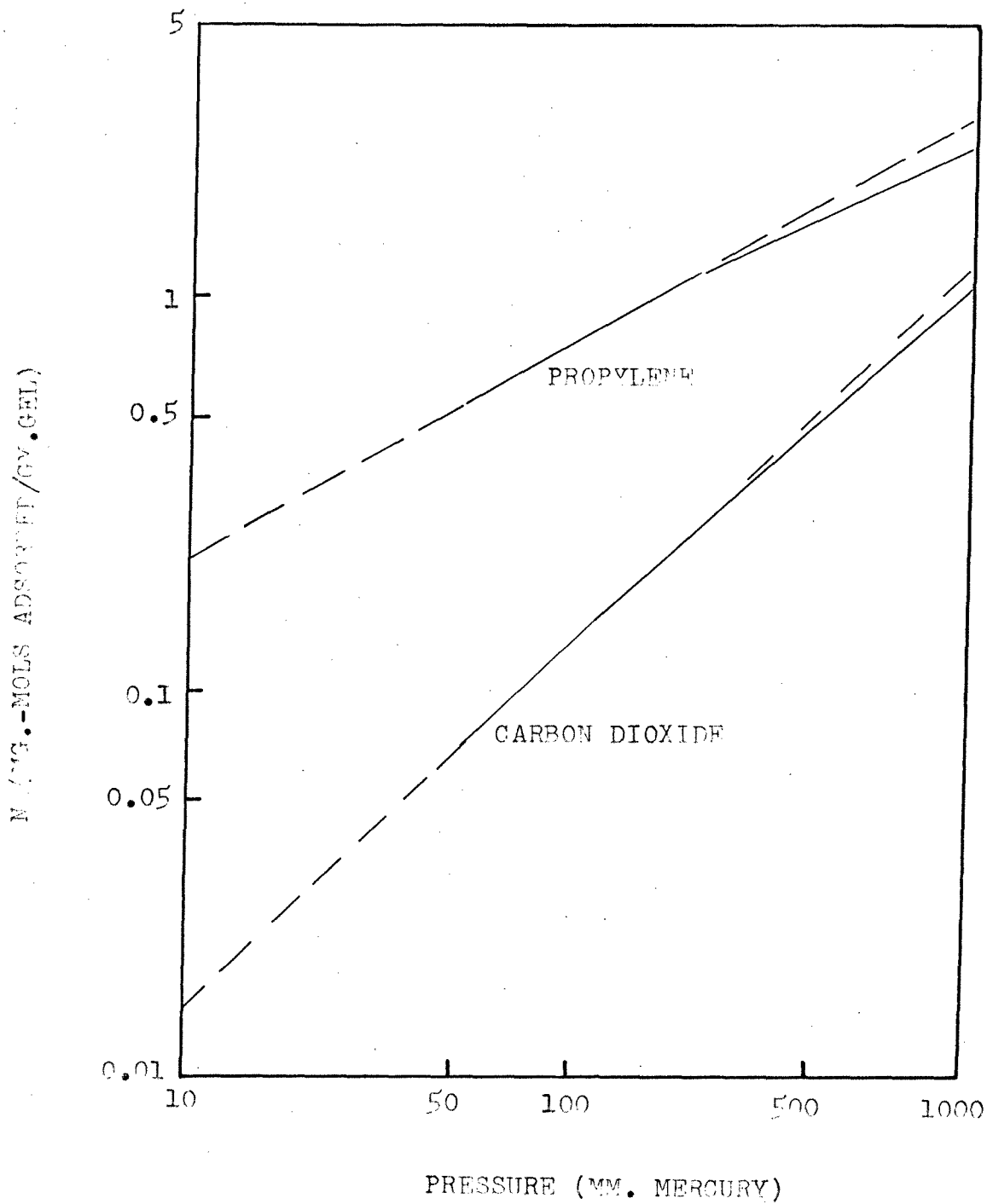


Figure 2

are integers. The point of intersection of two characteristics can be denoted by the symbol  $(k, \ell)$  and the values of  $y$  and  $w$  at these points will be denoted by  $y_{(k, \ell)}$  and  $w_{(k, \ell)}$  respectively. For example,  $(2, 1)$  refers to the intersection of the two characteristics :

$$t = z + 2 \Delta t \quad \text{and} \quad z = \Delta z \quad .$$

Evaluation of the quantities  $y_{(1,1)}$  and  $w_{(1,1)}$  proceeds in the following manner :

Equation (31) can be written as :

$$\frac{dy}{dz} = f(y, w) \quad ,$$

Using the Runge-Kutta method for  $\Delta z$  increment :

$$K_1 = f( y_{(0,0)} , w_{(0,0)} ) \Delta z \quad ,$$

$$y_{(1,1)}^1 = y_{(0,0)} + K_1/2 \quad ,$$

$$K_2 = f( y_{(1,1)}^1 , w_{(0,0)} ) \Delta z \quad ,$$

$$y_{(1,1)}^2 = y_{(0,0)} + K_2/2 \quad ,$$

$$K_3 = f( y_{(1,1)}^2 , w_{(0,0)} ) \Delta z \quad ,$$

$$y_{(1,1)}^3 = y_{(0,0)} + K_3/2 \quad ,$$

$$K_4 = f( y_{(1,1)}^3 , w_{(0,0)} ) \Delta z \quad ,$$

$$\Delta y = ( K_1 + 2 K_2 + 2 K_3 + K_4 ) / 6 \quad ,$$

$$y_{(1,1)} = y_{(0,0)} + \Delta y \quad ,$$

For the calculation of  $w_{(1,1)}$  :

$$\frac{dw}{dt} = f(y, w) \quad , \quad \text{using the Runge-Kutta method :}$$

$$K_1' = f ( y_{(0,1)} , w_{(0,1)} ) \Delta t ,$$

$$w_{(1,1)}^1 = w_{(0,1)} + K_1' / 2 ,$$

$$K_2' = f ( y_{(0,1)} , w_{(1,1)}^1 ) \Delta t ,$$

$$w_{(1,1)}^2 = w_{(0,1)} + K_2' / 2 ,$$

$$K_3' = f ( y_{(0,1)} , w_{(1,1)}^2 ) \Delta t ,$$

$$w_{(1,1)}^3 = w_{(0,1)} + K_3' / 2 ,$$

$$K_4' = f ( y_{(0,1)} , w_{(1,1)}^3 ) \Delta t ,$$

$$\Delta w = ( K_1' + 2 K_2' + 2 K_3' + K_4' ) / 6 ,$$

$$w_{(1,1)} = w_{(0,1)} + \Delta w$$

Similarly, for  $t = 1 \Delta t$ ,  $y_{(1,2)}$  and  $w_{(1,2)}$ ;  $y_{(1,3)}$  and  $w_{(1,3)}$ ;  $y_{(1,4)}$ ,  $w_{(1,4)}$  can be estimated. The same procedure is repeated for  $t = 2 \Delta t$ ,  $3 \Delta t$ ,  $k \Delta t$ . Therefore, in general,  $y_{(j,i)}$  and  $w_{(j,i)}$  can be calculated once  $y$  and  $w$  are known at points  $(j-1,i-1)$  and  $(j-1,i)$ .

#### (D) CYCLIC OPERATION OF THERMAL PARAMETRIC PUMPS

A thermal parametric pump consists of one jacketed column with two reservoirs, each at one end. Feed is introduced at the top. (Fig.-3) After each half cycle the column temperature is changed, keeping the pressure constant. At the beginning of cyclic operation the column is saturated at high temperature during the first half cycle. In general, one complete cycle can be divided into four parts as described herein :

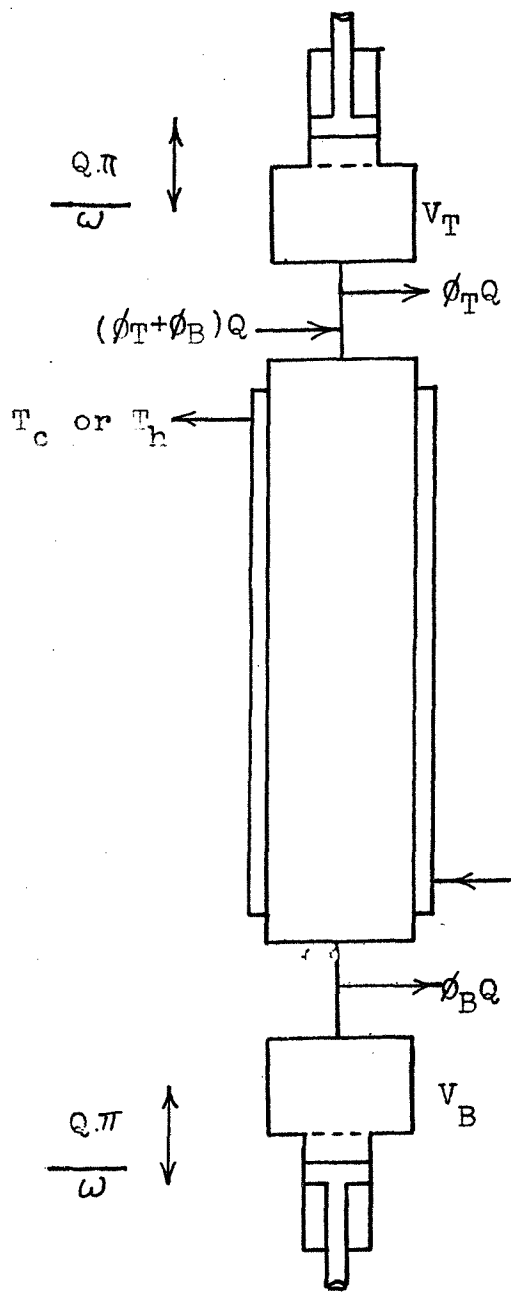


Figure 3

Continuous Thermal Parametric Pump

STEP 1 The direction of flow is upward and the column is maintained at a high temperature. Component A is desorbed; thus solute-enriched top product is obtained during this half cycle.

STEP 2 The temperature of the column changes from high temperature to low temperature and the direction of the flow is reversed.

STEP 3 The direction of flow is downward and the column is maintained at low temperature. Solute is adsorbed, resulting in a solute-lean product from the bottom end.

STEP 4 The temperature changes from high to low and the direction of flow changes from downward to upward. Steps 2 and 4 take place instantaneously and feed is introduced continuously at the top.

(E) COMPUTER CALCULATION FOR CYCLIC OPERATION (Fig. 4)

Total distance intervals = NNZ

STEP 1

$$v_1 = \frac{(1 - \phi_B) Q}{A \text{ TIME}} \quad \text{----(33)}$$

$$\Delta z_1 = \Delta t_1 = \frac{H}{v_1 \text{ NNZ}} \quad \text{----(34)}$$

Initial conditions :

(1) @ z = 0 for all t ≥ 0

$c_{(k,0)} = \langle \text{CBP1} \rangle_n$  , where  $\langle \text{CBP1} \rangle_n$  can be obtained by material balance at the bottom reservoir. (Step 4)

For n = 1 ,  $\langle \text{CBP1} \rangle = c_0$  ,

Also integrating equation (18) :

$$w_{(k,0)} = w_{(k-1,0)} M_T - (M_T c_{(k,0)} - w_{(k-1,0)}) \exp(-\Delta t) \quad \text{----(35)}$$

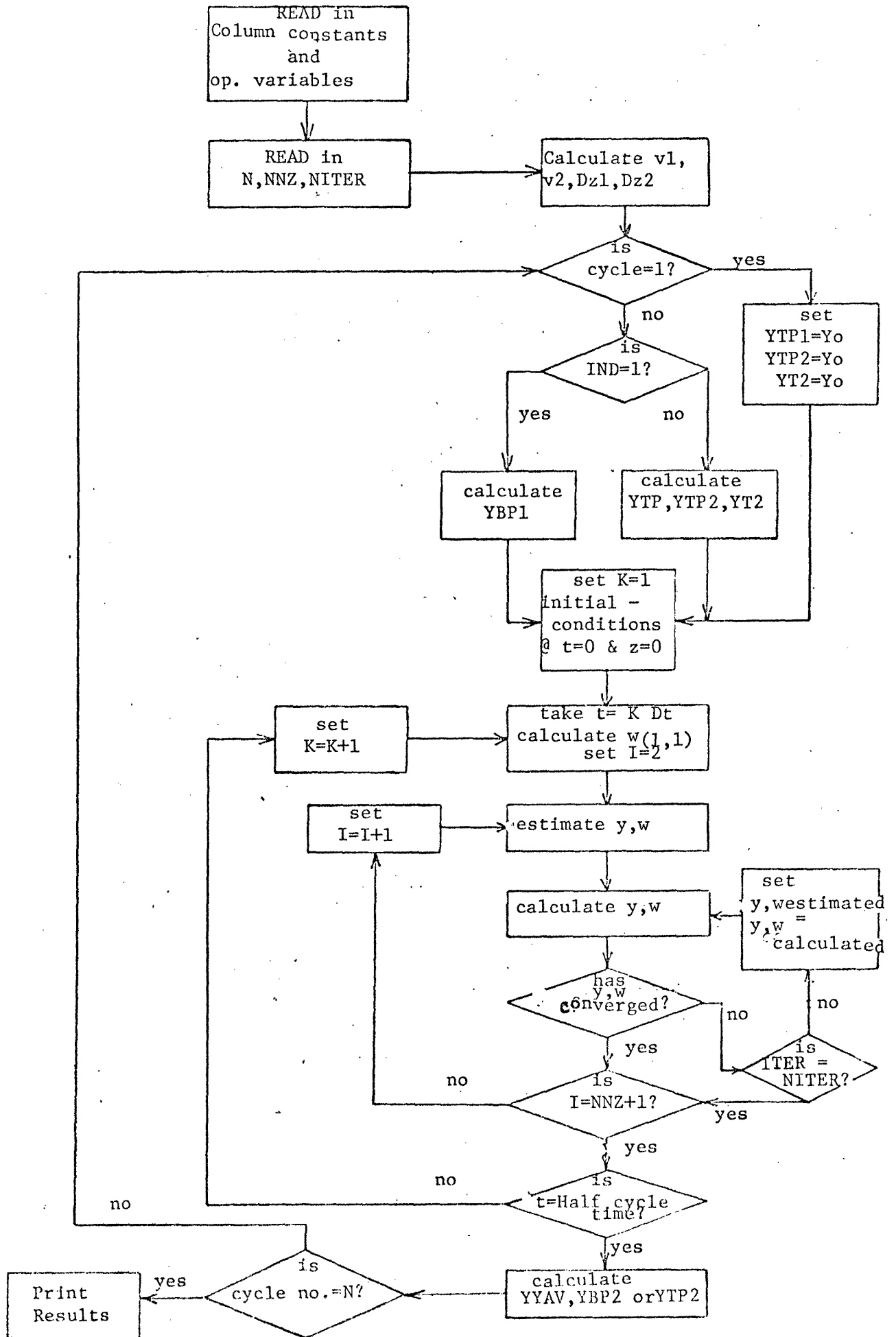


Figure 4: Block Diagram for Thermal Parametric Pumping

Where,  $t = k \Delta t_1$  ,

For  $n = 1$  ,  $w_{(k-1,0)} = w$  values obtained at the end of  $(n-1)^{th}$  cold half cycle.

(2) @  $t = 0$  ,  $x \geq 0$

$$\begin{matrix} c(0,i) \\ w(0,i) \end{matrix} \left| = \begin{matrix} c \\ w \end{matrix} \text{ , } w \text{ values obtained at the end of } (n-1)^{th} \text{ cold half cycle.}$$

Using the method of characteristics concentration profiles are calculated for  $t = \Delta t_1$  ;  $2 \Delta t_1$  ;  $k \Delta t_1$  .

At the end of  $n^{th}$  high temperature half cycle :

$\langle C_{T1} \rangle =$  average of concentrations obtained at the end of column at  $t = \Delta t_1$  ;  $2 \Delta t_1$  ;  $k \Delta t_1$

STEP 2

For  $n^{th}$  downflow half cycle, by making material balance at the top reservoir :

$$\langle C_{TP2} \rangle_n = \frac{Q \langle C_{TP1} \rangle_n + V_T \langle C_{TP2} \rangle_{n-1}}{V_T + Q} \quad \text{-----(36)}$$

$$\langle C_{T2} \rangle = ( \phi_T + \phi_B ) c_0 + ( 1 - \phi_T ) \langle C_{TP2} \rangle_{n-1} \quad \text{-----(37)}$$

There is instantaneous temperature change in the column.

STEP 3

Column at the low temperature

$$v_2 = \frac{( 1 + \phi_B ) Q}{A \text{ TIME}} \quad \text{-----(38)}$$

$$\Delta z_2 = \Delta t_2 = \frac{H}{v_2 \text{ NNZ}} \quad \text{-----(39)}$$

This step is similar to the step 1. At the end of half cycle:

$$\langle \text{CBP2} \rangle = \text{average of the concentration obtained at the end of } t = \Delta t_2 ; 2 \Delta t_2 ; k \Delta t_2 .$$

STEP 4

By material balance at the bottom reservoir

$$\langle \text{CBP1} \rangle_{n+1} = \langle \text{CBP2} \rangle_n Q + V_B \langle \text{CBP1} \rangle_n . \quad \text{----(40)}$$

During this step also, the solid and liquid phase concentrations stay at the value attained at the end of the  $n^{\text{th}}$  cold half cycle.

(F) CYCLIC OPERATION OF HEATLESS PARAMETRIC PUMPS

This technique uses two columns under operation. One column operates at high pressure and the other column operates at low pressure (Fig. 5). Before starting the cyclic operation, both columns are saturated with gas mixture by maintaining either both columns at the same pressure or both the columns at different pressures. In this work, both saturating conditions have been used. One complete cycle is divided into four steps as below:

( See fig. 6)

STEP 1 High pressure feed flows through the column 1 , solute is adsorbed, resulting in a solute-lean top product. A portion of solute-lean product is purged to low pressure column 2. In column 2 , there is decrease in pressure, resulting in desorption of solute. Hence, solute-enriched bottom product is obtained at the end of column 2.

STEP 2 Column 2 is repressurized to high pressure (repressurization) while column 1 is brought to low pressure ( blow-down). (Fig. 7)

STEP 3 Flow of feed is reversed from column 1 to column 2, now at high pressure.



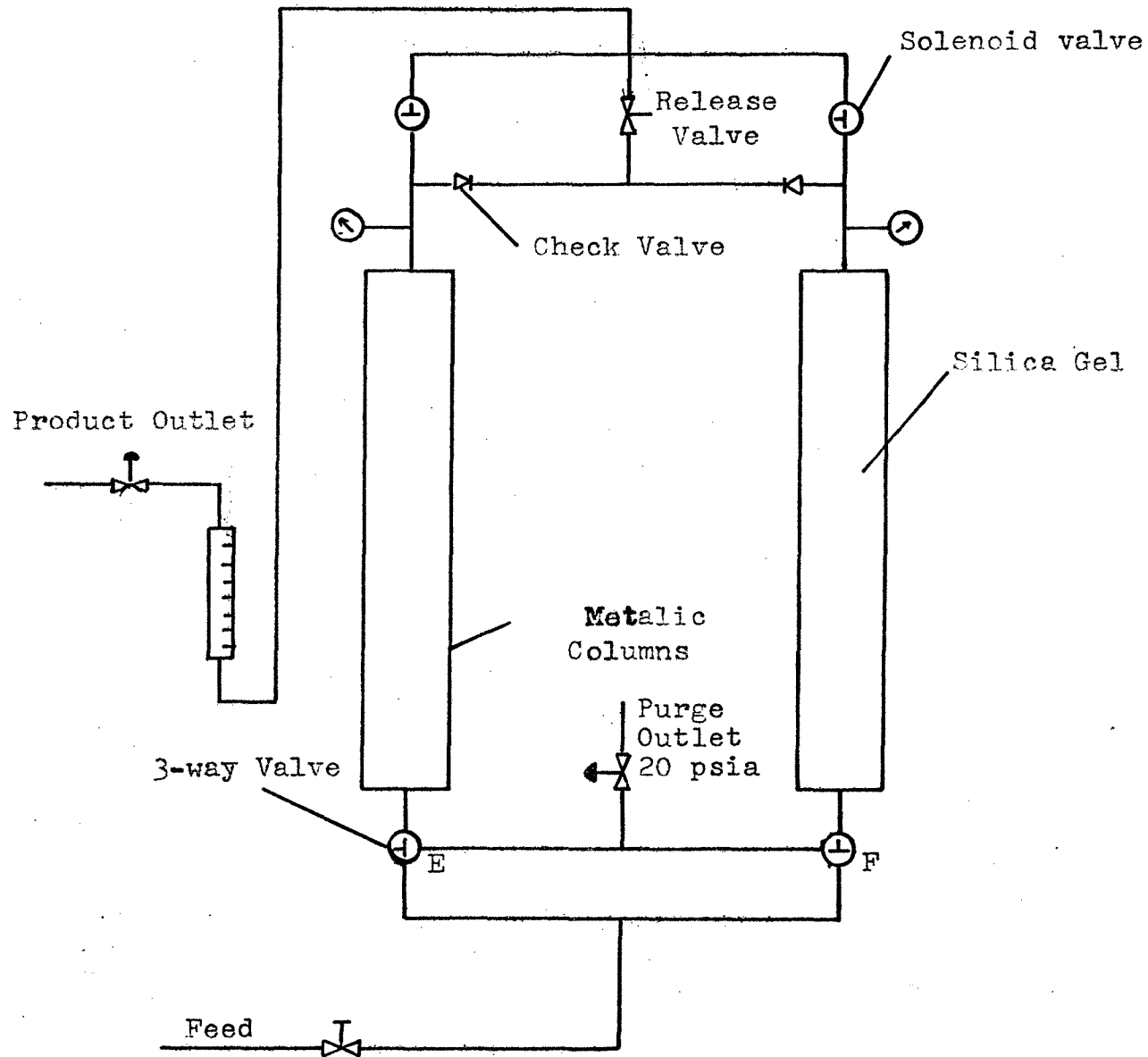


Figure 5

Schematic drawing of heatless Parametric Pump

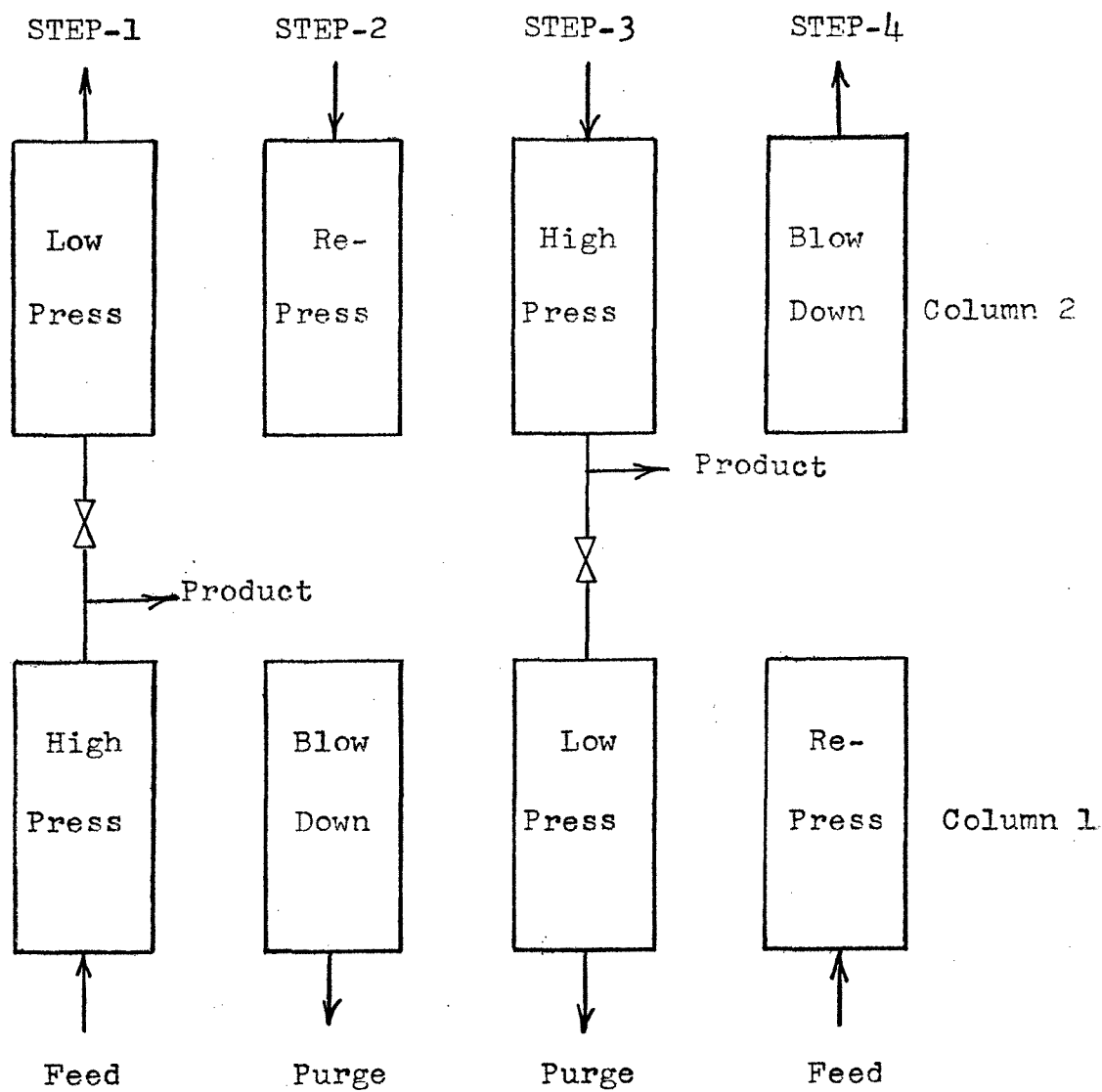


Figure 6

Stepwise Outline of Heatless Parametric Pumping

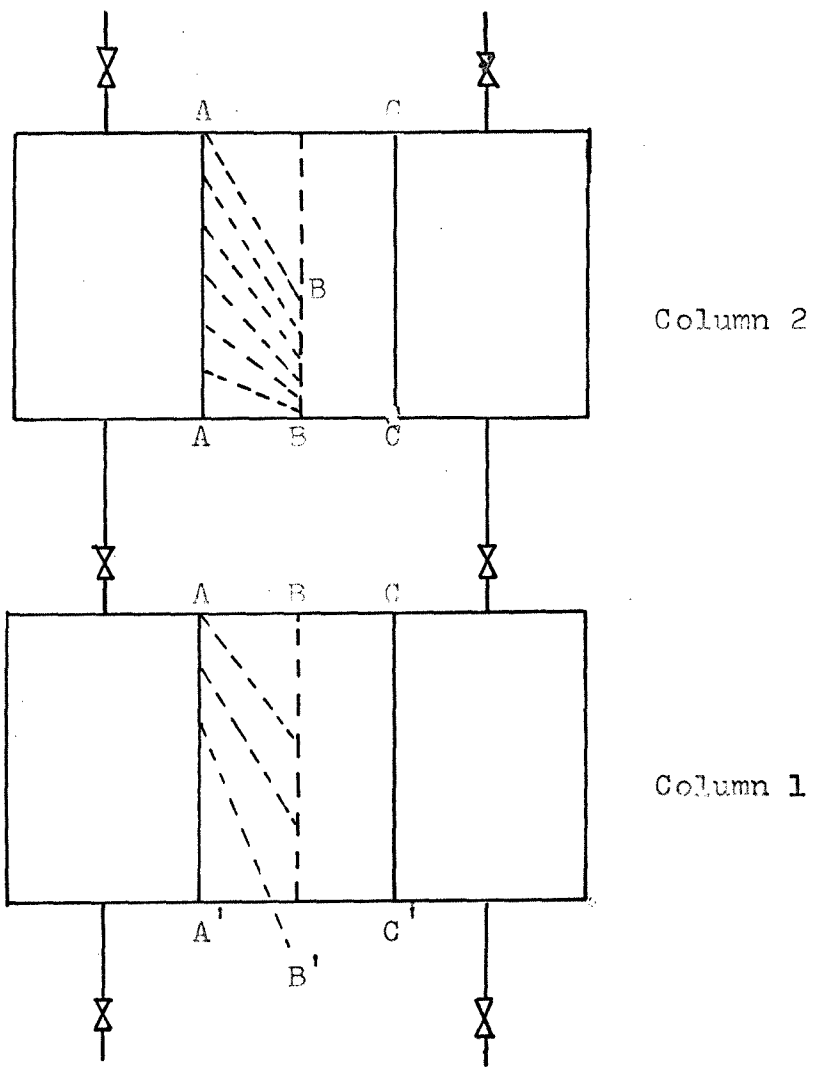


Figure 7

Repressurization and Depressurization of Columns

STEP 4 Column 1 is repressurized and column 2 is depressurized ( Fig. 7) .

During steps 2 and 4, solid concentration is considered to be frozen while fluid concentration ( in the bulk gas ) changes according to the pressure difference. Also, in these steps , gas is fed at high pressure but no gas is removed from the pressurized column whereas gas is blown off from the low pressure column but no gas is fed into it. Steps 2 and 4 take place instantaneously.

(G) COMPUTER CALCULATION OF CYCLIC OPERATION ( Fig. 8)

Let total number of intervals = NNZ

STEP 1 For high pressure column

$$\Delta z_1 = \Delta t_1 = \frac{H A}{NNZ Q} \quad \text{-----(41)}$$

$$v_1 = \frac{Q}{A} \quad \text{-----(42)}$$

For, n<sup>th</sup> half cycle, initial conditions are :

$$(1) \quad @ \quad z = 0 \quad \begin{aligned} y_{(k,0)} &= y_0 \quad , \\ w_{(k,0)} &= f ( w, y ) \quad , \end{aligned}$$

The solid phase concentration is calculated using Runge-Kutta method as below :

$$\frac{dw}{dt} = \lambda ( \overset{*}{w} - w ) \quad ,$$

Where, 
$$\overset{*}{w} = \frac{y_0 P M_p}{R T} \quad ,$$

Now 
$$K_1 = ( \overset{*}{w}_{(k,0)} - w_{(k-1,0)} ) \Delta z_1 \quad ,$$

$$w_{(k,0)}^1 = w_{(k-1,0)} + K_1/2 \quad ,$$

$$K_2 = \lambda \cdot ( \overset{*}{w}_{(k,0)} - w_{(k,0)}^1 ) \quad \Delta z_1 \quad ,$$

$$w_{(k,0)}^2 = w_{(k-1,0)} + K_3/2 \quad ,$$

$$K_3 = \lambda \cdot ( \overset{*}{w}_{(k,0)} - w_{(k,0)}^2 ) \quad \Delta z_1 \quad ,$$

$$w_{(k,0)}^3 = w_{(k,0)}^2 + K_4/2 \quad ,$$

$$K_4 = \lambda \cdot ( \overset{*}{w}_{(k,0)} - w_{(k,0)}^3 ) \quad \Delta z_1 \quad ,$$

$$Dw = ( K_1 + 2 K_2 + 2 K_3 + K_4 ) / 6 \quad ,$$

$$w_{(k,0)} = w_{(k-1,0)} + Dw \quad .$$

For  $k = 1$  ,  $w_{(k-1,0)} = w$  values obtained at the end of  $(n-1)^{th}$  half cycle.

(2) @  $t = 0$

$$y(0,i) = y \text{ and } w \text{ values obtained at the}$$

$$w(0,i) \text{ end of } (n-1)^{th} \text{ half cycle.}$$

Using the method of characteristics, the concentration profile is calculated for  $t = \Delta t_1 ; 2 \Delta t_1 , k \Delta t_1$ , until the half cycle time is complete. At the end of  $n^{th}$  half cycle :

<YTP> = average of the concentrations obtained at the end of column during  $t = \Delta t_1 , 2 \Delta t_1 , \dots k \Delta t_1$ .

For Low Pressure Column -

$$\Delta z_2 = \Delta t_2 = \frac{H \cdot A}{NNZ \cdot Q \cdot Y} \quad . \quad \text{----(43)}$$

$$v_2 = \frac{Q \cdot Y}{A} \quad . \quad \text{----(44)}$$

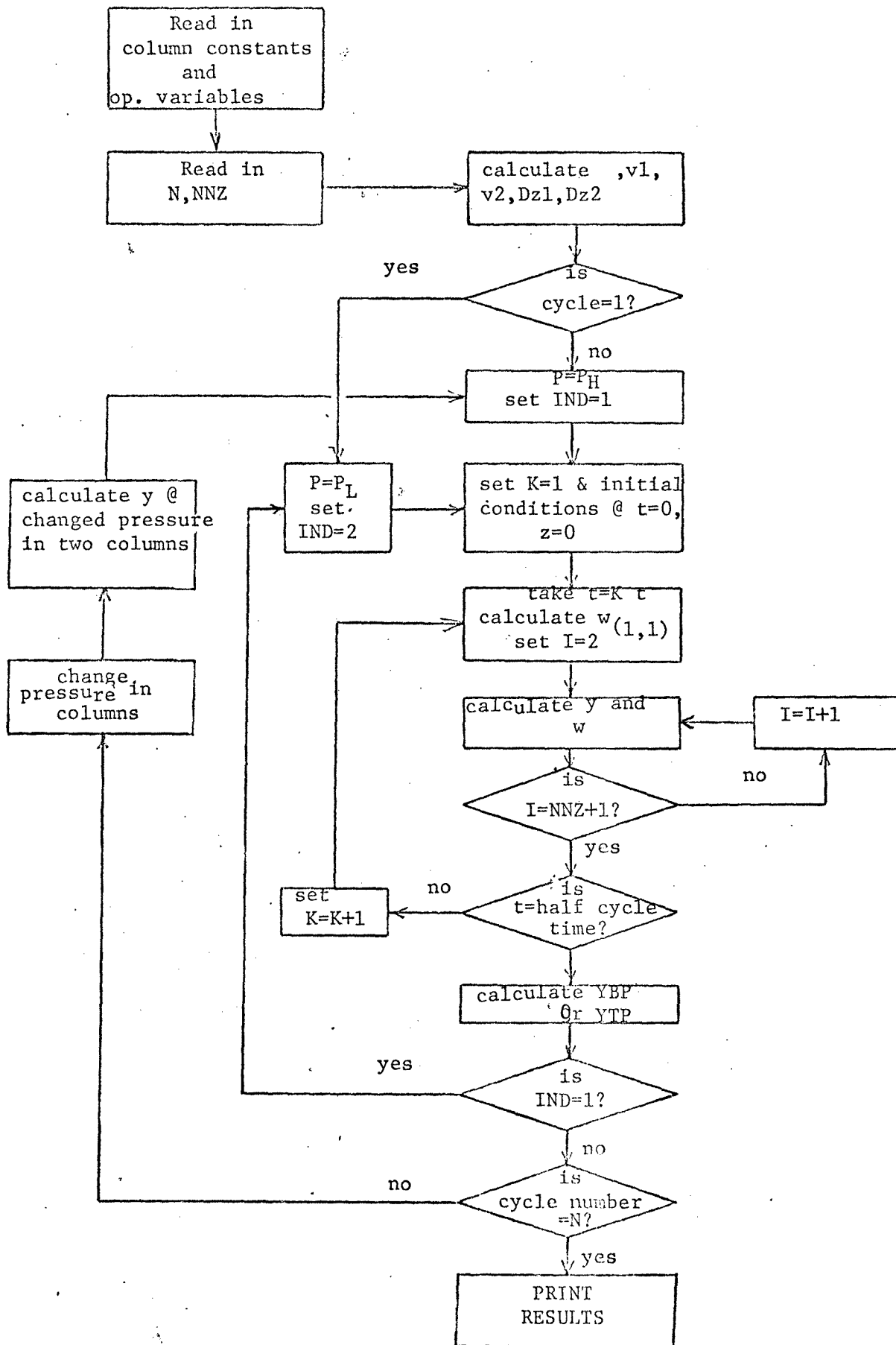


FIGURE 8 : Block Diagram for Heatless Parametric Pumping

Initial conditions :

(1) @  $z = 0$  ,  $y_{(k,0)}$  = y values evaluated by interpolation of the values of  $y_{(k,H)}$  obtained from high pressure column for  $n^{\text{th}}$  half cycle.

$w_{(k,0)}$  = w values calculated from Runge-Kutta method similar to the high pressure column calculation.

(2) @  $t = 0$

$y_{(0,i)}$   
 $w_{(0,i)}$  = y , w values obtained during (n-1)<sup>th</sup> half cycle.

Again using the method of characteristics, the concentration profiles are calculated. At the end of the half cycle :

<YBP> = average of the concentration obtained at the end of the column during  $t = \Delta t_2, 2 \Delta t_2, k \Delta t_2$ .

STEP 2 REPRESSURIZATION AND DEPRESSURIZATION OF COLUMNS

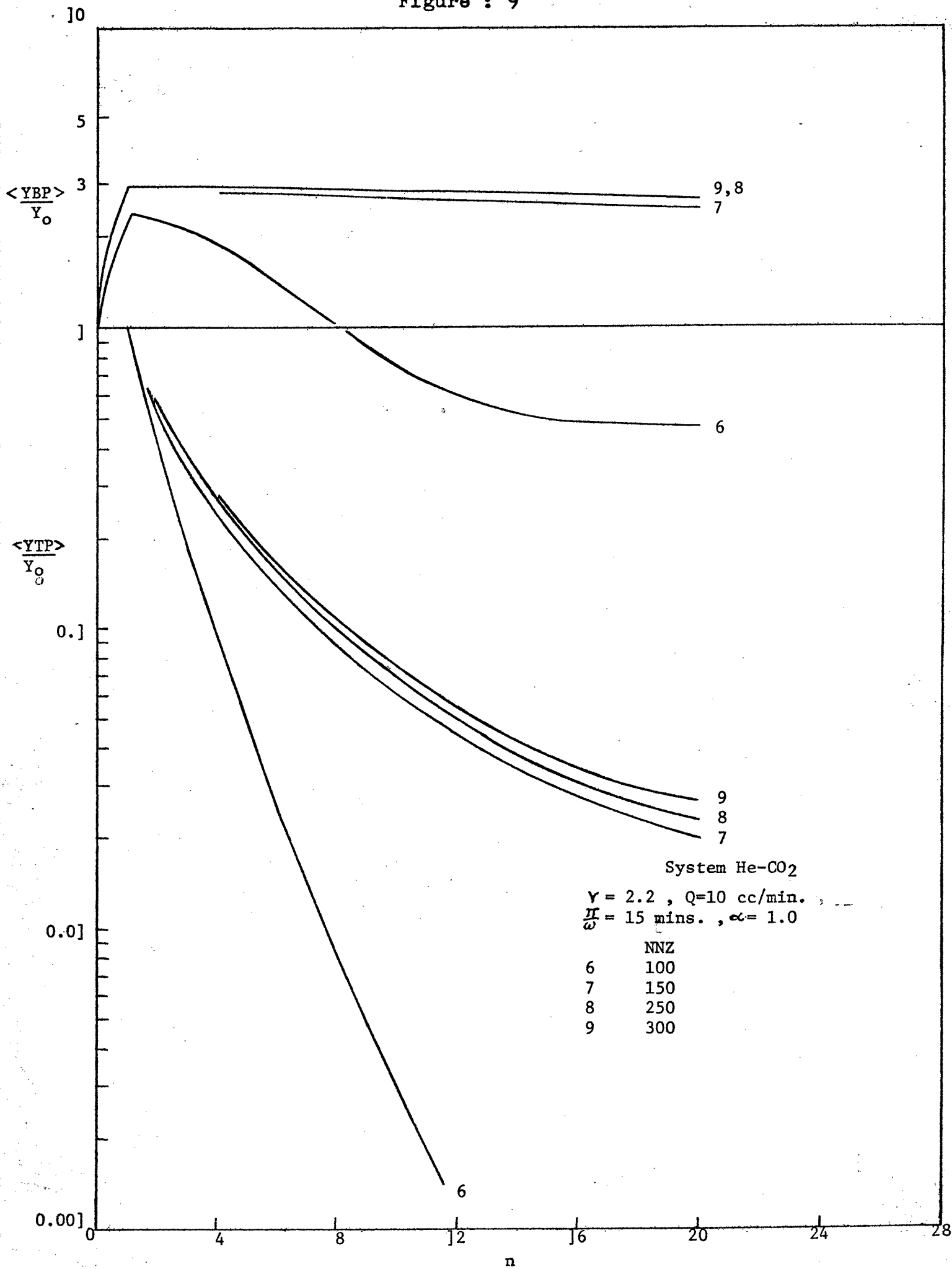
Before the start of  $(n+1)^{\text{th}}$  half cycle , the column which was at low pressure during the  $n^{\text{th}}$  half cycle , is repressurized to high pressure. Also, the solid phase concentration remains frozen to the value attained at the end of the  $n^{\text{th}}$  half cycle, but, because of the change of pressure, the gas phase is compressed and is given by the simple ideal gas law :

$$\begin{aligned} l_2 P_L &= l_1 P_H , \\ l_1 &= l_2 \frac{P_L}{P_H} . \end{aligned} \quad \text{-----(45)}$$

The void space formed by compression is filled by feed gas.(Fig. 7)

AA - position of the gas phase obtained at the end of the  $n^{\text{th}}$  cycle

Figure : 9





BB - position attained after the compression of gas phase.

CC - position of standard increments obtained by linear interpolation of position BB.

Similarly, the column which was at high pressure during the  $n^{\text{th}}$  half cycle is blown down to low pressure. Again the solid phase concentration remains frozen at the value obtained at the end of  $n^{\text{th}}$  half cycle, whereas fluid phase concentration is expanded according to the ideal gas law :

$$l_2 = l_1 \frac{P_H}{P_L} \quad \text{-----(46)}$$

In figure 7 :

AA' - position attained at the end of the  $n^{\text{th}}$  half cycle

BB' - position after compression

CC' - position obtained by linear interpolation of BB'

The calculations of steps 3 and 4 are similar to those of steps 1 and 2, respectively.

ADJUSTMENT OF OPTIMUM NUMBER OF INCREMENTS (NNZ)

In mathematical calculation, the selection of increments is important to achieve stability in concentration profiles. For the calculation of mass transfer coefficients in heatless parametric pumps, the following equation is used ( Appendix - I)

$$\lambda = \alpha \left[ \frac{D_p G}{\mu (1-\epsilon)} \right]^{-0.51} \frac{\left[ \frac{\mu}{\rho D} \right]^{G P} G P}{M_g P \rho_H \epsilon}$$

Here, value of  $\lambda$  depends on the two constants  $\alpha$ ,  $\beta$  and on the operating conditions. The value of  $\beta$  has been selected equal to -0.51.

For the He-CO<sub>2</sub> system assume that  $\alpha = 1.0$ . Computer calculations are carried out for the concentration profiles taking NNZ= 100,150, 200,250. It was observed that NNZ  $\geq$  150 gives an almost stable concentration profile for top and bottom products (Fig. 9). Thus, for  $\alpha = 1.0$ , NNZ = 150 is the value to attain stability.

Similarly, for the He-C<sub>3</sub>H<sub>6</sub> system :

$$\alpha = 0.25, \text{ NNZ} = 150 \text{ for stability.}$$

Other operating conditions are :

$$\Upsilon = 2.2, \quad Q = 10 \text{ cc/min.}$$

$$\frac{H}{\omega} = 15 \text{ mins.}, \quad y_0 = 1\% \text{ CO}_2 \text{ or } 1\% \text{ C}_3\text{H}_6.$$

DISCUSSION OF RESULTS

In this work a mathematical model has been developed for the NaCl-H<sub>2</sub>O system using a thermal parametric pump. Thirteen results have been mathematically calculated for different operating conditions and the effects of different operating conditions have been observed. In Table-1 different operating conditions for thirteen runs have been listed. A typical detailed display of concentration profiles is shown in Table-23 and Figure-10.

Both experimental and mathematical studies have been made for the separation of gaseous mixtures via a heatless parametric pump. Seventeen runs have been performed experimentally. Table-2 lists fifteen experimental runs with different systems and other operating conditions. Out of the fifteen runs listed, four runs are for the separation of the He-CO<sub>2</sub> mixture listed in Tables-4,5,6,7. Tables-4 and 5 give the calibration for peak height and pressure of CO<sub>2</sub> gas. Tables-6 and 7 present cyclic concentrations obtained after the each half cycle for CO<sub>2</sub> gas. In these tables concentration has been shown as the ratio of cyclic peak height and initial gas peak height.

It is shown in Appendix II that the peak height is proportional to y(concentration), so the peak height ratio is also the concentration ratio. Peak height ratio is plotted against half cycle time. Table-8 is calibration for He-C<sub>3</sub>H<sub>6</sub> and Tables-9,10, 11,12 and 13 are cyclic concentrations for He-C<sub>3</sub>H<sub>6</sub> separation. Table-14 is the calibration for ternary mixture He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub> and Tables 15 to 18 are for the cyclic separations. Tables-19 and 20 list degree of saturation of the bed with separation. Tables-21 and

22 list the operating conditions used in the mathematical models for two types of parametric pumps respectively. In Table-23, different operating conditions have been listed for the results calculated from the binary gas mathematical model. Table-24 presents a typical case in detail of the concentration profiles for the He-C<sub>3</sub>H<sub>6</sub> mixture.

(A) THERMAL PARAMETRIC PUMPS -

Figure - 10 is the presentation of characteristics concentration front movement for the NaCl-H<sub>2</sub>O system. It is obvious from the figure that the solute moves towards the upper end of the column with the change of cycle direction until the steady state is reached, finally, resulting in less solute at the bottom and more solute in the top product. Figures-11 and 12 show that the model developed predicts the separation of NaCl-H<sub>2</sub>O and also compares the effect of different operating conditions on separation.

Figure-11 indicates the effect of  $\phi_B$  with the ratio of bottom reservoir volume and the displacement with other conditions as-  $\frac{H}{\omega} = 35$  mins.,  $c_0 = 0.1$  M,  $Q(\frac{H}{\omega}) = 25$  cc. It shows that with the decrease in  $\phi_B$  from 0.16 to 0.04, fluid moves more from the bottom reservoir to the top reservoir through the column; this results in the increased mixing at the top reservoir and the better separation.

In separation, the rate of interphase mass transfer is dependent on the temperature, fluid velocity and cycle time. However, since velocity is inversely proportional to the cycle time for a given displacement, a study of cycle time alone includes implicitly the velocity effect. Figure-12 shows the top and bottom products concentration for different cycle time with  $\phi_B = 0.04$ ,  $Q(\frac{H}{\omega}) = 25$  cc.

and  $c_0 = 0.1$  M. Increase in the cycle time from 25 mins. to 100 mins. results in better separation because more time is allowed for the transfer to occur between two phases.

An optimization study for the displacement volume is shown in figure-13. It is a plot of the steady state bottom product concentration vs. volumetric displacement for  $\frac{T}{\omega} = 35$  mins.,  $\phi_B = 0.04$ ,  $c_0 = 0.1$ M and gives the following important information:

For the above mentioned conditions the minimum concentration is 0.018M, for which the optimum value of displacement  $Q(\frac{T}{\omega})$  is 5 cc. This curve is also helpful in determining the volumetric displacement required to obtain a desired bottom product concentration. For example, from figure-13 it can be shown that for obtaining 0.3 M bottom product ratio, the displacement  $Q(\frac{T}{\omega})$  should be 26 cc, while all other operating conditions are fixed.

Experimental study for the NaCl-H<sub>2</sub>O system has been done by Kim(25) and Rak(26).

## (B) HEATLESS PARAMETRIC PUMPS

### (i) Binary systems-

#### (a) Adjustment of ' $\alpha$ ' value -

For the He-CO<sub>2</sub> system the concentration curve with  $\alpha = 1.0$ ,  $\beta = -0.51$  and NNZ=150 differs from the experimental results, figure-14. Therefore, different values of  $\alpha$  have been tested to minimize the difference between the experimental and the computed results. A linear relation between  $\alpha$  and NNZ has been used as shown below:

$$\frac{\alpha_1}{\alpha_2} = \frac{NNZ_1}{NNZ_2} = \text{constant} ,$$

if

$$\frac{\alpha_1}{\alpha_2} = \frac{NNZ_1}{NNZ_2} = 1.5 ,$$

and  $\alpha_1 = 1.0$  ,  $NNZ_1 = 150$  ,  
 then  $\alpha_2 = 1.5$  ,  $NNZ_2 = 225$  .

This trial and error procedure is continued until two types of results are close; figure (14) shows the optimum value for the He-CO<sub>2</sub> system. Similarly, for the He-C<sub>3</sub>H<sub>6</sub> system the optimum values of  $\alpha$  and NNZ have been found. These values are tabulated as below - (  $\gamma = 2.2$ ,  $\frac{P}{\omega} = 15$  mins. ,  $Q = 10.0$  cc/min. )

| System                           | Optimum value of |     |
|----------------------------------|------------------|-----|
|                                  | $\alpha$         | NNZ |
| He-CO <sub>2</sub>               | 4.0              | 600 |
| He-C <sub>3</sub> H <sub>6</sub> | 1.25             | 600 |

(b) Effect of different operating conditions -

The separation of binary gaseous mixtures He-CO<sub>2</sub> and He-C<sub>3</sub>H<sub>6</sub> was observed experimentally, ( Figures (17) and (19) ). These two plots show clearly two curves for the bottom product. It is because both the columns were saturated at different pressures, i.e. , one column at 60 psia and the other at 20 psia initially. (  $\gamma = 2.2$ ,  $\frac{P}{\omega} = 15$  mins. and  $Q = 10$  cc/min. ). It is expected that these two curves will merge into one at steady state value. The same trend was observed with the computed results. Hence at this point it was decided that both the columns should be saturated at the same pressure, i.e. , at high initial pressure. Figures 19 and 20 are for the separation of He-C<sub>3</sub>H<sub>6</sub> mixture with a new saturation pressure of 60 psia and the other operating conditions -  $\gamma = 2.2$ ,  $\frac{P}{\omega} = 15$  mins. and  $Q = 10$  cc/min. These conditions result in only one curve for the bottom product.

Next, the effect of the purge ratio on the separation of He-C<sub>3</sub>H<sub>6</sub> mixtures was studied experimentally. It was found that

the decrease in the purge ratio or purge volume to the low pressure column increases separation of mixture while the other operating conditions are fixed at  $\frac{T}{\omega} = 15$  mins.,  $y_o = 1\% C_3H_6$  and  $Q = 10$  cc/min. Also, for the same operating conditions, the mathematically calculated results are close to the experimental results. Hence the mathematical model is in agreement with the experimental results.

(ii) Ternary gas mixtures

Figures-22, 24 and 25 are for the experimental separation of gas mixture He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub> using a heatless parametric pump. These figures also represent the effect of different operating variables such as the purge ratio ' $\gamma$ ' and half cycle time  $\frac{T}{\omega}$  on the separation. In figure - 23 it is observed that for the operating conditions  $\gamma = 2.2$ ,  $\frac{T}{\omega} = 15$  mins. and  $Q = 10$  cc/min. after the 15<sup>th</sup> half cycle, both the top and the bottom product concentrations for CO<sub>2</sub> drops below the feed concentration. This indicates accumulation of incoming CO<sub>2</sub> in the bed. It is considered that this occurs because of the following reasons:

1. half cycle time is too large ,
2. there is some interaction during the absorption of the gases. This causes the CO<sub>2</sub> disappearance from both the top and the bottom products.

To justify the above reasons the bed was saturated with CO<sub>2</sub> and C<sub>3</sub>H<sub>6</sub>. Figure-23 is the saturation curve for CO<sub>2</sub> and C<sub>3</sub>H<sub>6</sub> at 20 and 60 psia. From this figure for any gas, total amount absorbed is proportional to the area under the saturation curve. It was found that the net amount desorbed in changing the cycle from 60 psia to 20 psia is less for CO<sub>2</sub> than for C<sub>3</sub>H<sub>6</sub>. The numerical values are as given below:

| Pressure<br>psia | Amount adsorbed for        |                               |
|------------------|----------------------------|-------------------------------|
|                  | CO <sub>2</sub>            | C <sub>3</sub> H <sub>6</sub> |
| 20               | $\frac{2.084}{R T \rho_s}$ | $\frac{3.605}{R T \rho_s}$    |
| 60               | $\frac{4.200}{R T \rho_s}$ | $\frac{11.98}{R T \rho_s}$    |

Net amount desorbed in changing the pressure from 60 psia to 20 psia, for CO<sub>2</sub> =  $\frac{4.2 - 2.084}{R T \rho_s}$  ,

$$= \frac{2.116}{R T \rho_s} \text{ gm-moles / gm-solid .}$$

for, C<sub>3</sub>H<sub>6</sub> =  $\frac{11.98 - 3.605}{R T \rho_s}$  ,

$$= \frac{8.375}{R T \rho_s} \text{ gm-moles / gm-solid .}$$

Figure 23 shows less desorption of CO<sub>2</sub> than C<sub>3</sub>H<sub>6</sub>; hence, the half cycle time may affect the separation of CO<sub>2</sub> in a ternary system. Figures 24 and 25 ( $\gamma = 1.0$  and  $1.5$  respectively and  $\frac{H}{\omega} = 8$  mins.,  $Q = 10$  cc/min) support the reasoning for large half cycle time. Figures 24 and 25 also show the affect of purge ratio on the separation. As in the case of a binary mixture, a decrease in the purge ratio from 1.5 to 1.0 increases the separation.

The computed results based on the binary gas mixture model have been compared with the ternary experimental results. (Figures 22 and 25) The two results are quite close for the operating conditions  $Q = 10$  cc/min,  $\frac{H}{\omega} = 15$  mins., and  $\gamma = 2.2$  (Figure 22). However, for run 13, the deviation between the calculated and experimental results are higher than expected. (Figure 25). Thus, the above comparision indicates that for ternary mixtures, the



effect of interaction of gases on adsorption should be taken into account in the development of the mathematical model.

RECOMMENDATIONS

Further investigation in this field should be, determining the effect of different operating conditions such as feed flow rate, effect of half cycle time and purge ratio in more detail for both types of parametric pumps. For ternary or multicomponent system, a mathematical model can be developed similar to that for a binary system. In this mathematical model it would be important to consider the interaction of different gases in adsorption. Also, the experimental and mathematical studies of parametric pumping technique can be extended to the separation of different liquid and gaseous mixtures.

NOMENCLATURE

- a - constant
- A - cross sectional area of the column,  $\text{cm}^2$
- b - constant
- $c_0$  - molal concentration of solute in feed,  $\text{gmole/cc}$
- $\bar{c}$  - equilibrium molal concentration,  $\text{gmole/cc}$
- CBP1 - bottom product concentration during upflow,  $\text{gmole/cc}$
- CTP1 - top product concentration during upflow,  $\text{gmole/cc}$
- CT1 - outlet concentration in upflow cycle,  $\text{gmole/cc}$
- CT2 - inlet concentration in downflow,  $\text{gmole/cc}$
- CTP2 - top product concentration during downflow,  $\text{gmole/cc}$
- CBP2 - bottom product concentration in downflow cycle,  $\text{gmole/cc}$
- D - diffusivity,  $\text{cm}^2/\text{min}$ .
- $D_p$  - average particle diameter for 12-28 mesh size,  $\text{cm}$
- G - gas mass flow rate,  $\text{gm/cm}^2/\text{min}$ .
- H - height of the column,  $\text{cm}$
- $j_D$  - j - factor for mass transfer , dimensionless
- $\ell$  - length at any time,  $\text{cm}$
- M - molecular weight,  $\text{gm/gmole}$
- $M_p$  - equilibrium coefficient for gas phase
- $M_T$  - equilibrium coefficient for liquid phase
- NNZ - number of intervals
- N - number of moles
- $P_c$  - critical pressure,  $\text{atm}$ .
- Q - volumetric flow rate,  $\text{cc/min}$ .
- $K_1, K_2, K_3, K_4, K'_1, K'_2, K'_3, K'_4$  - constants in Runge-Kutta method

R - gas constant =  $82.06 \text{ atm-cm}^3/\text{gmole/ K}$

t - time, min.

T - temperature, K

$T_C$  - critical temperature, K

V - reservoir volume, cc

$v, v_0$  - velocity of fluid through the column, cm/min

w - solid phase concentration, gmole/gm of solid

x - linear distance, cm

y - mole fraction of solute, gmole/gmole

$z = x/v_0$ , min

x - increment in x, cm

z - increment in z, min

t - increment in t, min

YTP - top product concentration, gmole/gmole

YBP - bottom product concentration, gmole/gmole

Greek letters-

$\beta$  - ratio of reservoir volume to displacement, cc/cc

$\gamma$  - purge ratio

$\rho$  - density of bulk fluid, gm/cc

$\rho_s$  - solid density, gm/cc

$\epsilon$  - voidage of bed

$\mu$  - viscosity, gm/cm/min

$\sigma$  - collision diameter, cm

$\Omega$  - collision integral, dimensionless

$\lambda$  - mass transfer coefficient for, heatless parametric pump,  $\text{min}^{-1}$   
for thermal parametric pump,  
 $\text{cm}^3/\text{gm}$  of solid/min

Subscript-

$\langle \rangle$  - average

T - top product

B - bottom product

H - high pressure

L - low pressure

1 - hot cycle or upflow

2 - cold cycle or downflow

i, l - z

j, k - time

n - cycle number

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APPENDICES



APPENDIX - I

Calculation of mass transfer coefficient for heatless parametric pumps

Othmer (13) has given an expression for the calculation of mass transfer coefficient as below -

$$j_D = \frac{P}{G/M_g} \lambda \left[ \frac{\mu}{\rho \cdot D} \right]^{2/3} \quad \text{----(A)}$$

where,

$$j_D = \alpha \left[ \frac{D_p \cdot G}{\mu \cdot (1-\epsilon)} \right]^{-0.51} \quad \text{----(B)}$$

equations (A) and (B) are further simplified for ' $\lambda$ ' -

$$\lambda = \frac{\alpha \left[ \frac{D_p \cdot G}{\mu \cdot (1-\epsilon)} \right]^{-0.51} \left[ \frac{\mu}{\rho \cdot D} \right]^{-2/3} \cdot G \cdot P}{M_g \cdot P \cdot \rho \cdot H \cdot \epsilon} \quad (\text{min}^{-1}) \quad \text{----(C)}$$

Physical properties of gaseous mixture in the above equation have been considered as that of Helium only, because solute gas (CO<sub>2</sub> or C<sub>3</sub>H<sub>6</sub>) in the mixture is in very small amount (1% only). Viscosity of Helium and diffusivity of CO<sub>2</sub> and C<sub>3</sub>H<sub>6</sub> in helium have been estimated using the expression given by Bird (3).

- Calculation of Viscosity -

$$\mu = 2.6693 \cdot 10^{-5} \cdot \frac{M \cdot T}{\sigma^2 \cdot \Omega_\mu} \quad \text{----(D)}$$

for Helium -

$$\sigma = 2.576 \quad , \quad \frac{\epsilon}{K} = 10.2 \quad , \quad \Omega_\mu = 0.7005$$

$$\text{Also, } T = 25^\circ \text{C} = 298^\circ \text{K}$$

$$\begin{aligned} \mu_{\text{He}} &= \frac{2.6693 * 10^{-5} * 4 * 298 * 60}{(2.576)^2 * 0.7005} \\ &= 0.0018956 \\ &= 0.0019 \text{ gm / cm /min.} \end{aligned}$$

- Calculation of Diffusivity -

$$\frac{P \cdot D_{A-B}}{(P_{C-A} \cdot P_{C-B})^{1/3} \cdot (T_{C-A} \cdot T_{C-B})^{5/12} \cdot \left[ \frac{1}{M_A} + \frac{1}{M_B} \right]^{1/2}} = a \cdot \left[ \frac{T}{\sqrt{T_{C-A} \cdot T_{C-B}}} \right]^b$$

|                               | $T_C$<br>( K ) | $P_C$<br>( atm. ) | $M$  |
|-------------------------------|----------------|-------------------|------|
| He                            | 5.26           | 2.26              | 4.0  |
| CO <sub>2</sub>               | 304.2          | 72.9              | 44.0 |
| C <sub>3</sub> H <sub>6</sub> | 365.0          | 45.5              | 42.0 |

$$a = 2.745 * 10^{-4} \quad , \quad b = 1.823$$

(a) He - CO<sub>2</sub> system -

$$\left[ P_{C-CO_2} \cdot P_{C-He} \right]^{1/3} = 5.39$$

$$\left[ T_{C-CO_2} \cdot T_{C-He} \right]^{5/12} = 21.63$$

$$\sqrt{T_{C-CO_2} \cdot T_{C-He}} = 40.0$$

$$\frac{T}{\sqrt{T_{C-CO_2} \cdot T_{C-He}}} = 7.45$$

@ P = 4.0186 atm. (60 psia) -

$$D_{CO_2-He} = \frac{5.39 * 2.745 * 10^{-4} * 7.45^{1.823} * 21.63 * 0.5233}{4.0186}$$

or  $D_{\text{CO}_2\text{-He}} = 0.1618 \text{ cm}^2/\text{sec.}$   
 $= 9.708 \text{ cm}^2/\text{min.}$

@ P = 2.721 atm. ( 40 psia )

$D_{\text{CO}_2\text{-He}} = 14.34 \text{ cm}^2/\text{min.}$

@ P = 1.36 atm. ( 20 psia )

$D_{\text{CO}_2\text{-He}} = 29.14 \text{ cm}^2/\text{min.}$

Similarly for He - C<sub>3</sub>H<sub>6</sub> system -

@ P = 4.0186 atm. ( 60 psia )

$D_{\text{C}_3\text{H}_6\text{-He}} = 7.485 \text{ cm}^2/\text{min.}$

@ P = 2.721 atm. ( 40 psia )

$D_{\text{C}_3\text{H}_6\text{-He}} = 11.228 \text{ cm}^2/\text{min.}$

@ P = 1.36 atm. ( 20 psia )

$D_{\text{C}_3\text{H}_6\text{-He}} = 22.456 \text{ cm}^2/\text{min.}$

Apart from viscosity and diffusivity other properties are calculated as below -

Density -

$$\rho_H = \frac{P_H}{RT} \quad , \quad \rho_L = \frac{P_L}{RT}$$

Molecular weight -

He - CO<sub>2</sub> = 4.4

He - C<sub>3</sub>H<sub>6</sub> = 4.4

Other parameters such as D<sub>p</sub>, ε, H are given in table

## APPENDIX II

### Description of Apparatus -

The laboratory heatless continuous parametric pump, shown in fig. 5 , consists of two 3.175 cm \* 1.0 m metallic columns packed with 12-28 mesh PA-400, grade-408 silica gel. Feed is introduced alternatively to both columns. Feed rate is controlled by two electric actuated three way solenoid valves, E and F. At the top end of each column one pressure gauge is installed to measure the pressure in the column. A part of the gas coming out from the column under feed is purged to other column via a solenoid valve and pressure relief valve. Two check valves are used to check the direction of flow. Top product from the column under feed is withdrawn through a rotameter. Flow rate of top and bottom products is maintained by measuring flow of a soap bubble in a burette. Concentration of top and bottom product is measured by a Model 810, reserch chromatograph.

### Description of operation -

Initially two columns are saturated at the same pressure ( high-pressure ) or one column at high pressure and other column at low pressure. This is checked by measuring the product concentration. When product concentration is same as that of feed, the columns are considered to be saturated . After saturating the column, flow rate of top product and bottom product are adjusted to the operating values. After this, cyclic operation is started by actuating electronic

solenoid valves.

During the half cycle, at different intervals concentration of top or bottom product is analyzed. This concentration is measured as peak height which is represented as concentration by following relation :

$$\frac{y}{y_0} = \frac{\text{Peak height for top or bottom product}}{\text{Peak height for feed}}$$

At the end of half cycle direction of feed is reversed by switch system. This procedure is repeated for each cycle. Peak height can be adjusted by changing attenuation of column. Peak height can also be changed to concentration using calibration curve and ideal gas law. For any sample peak height corresponding pressure can be obtained from calibration curve, this pressure can be changed to concentration units using ideal gas law -

$$c = \frac{n}{v} = \frac{P}{RT} \propto y$$

APPENDIX- III - A

Computer programme for thermal parametric pump

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```

$JOB          RASTOGI
C *****
C  ANIL RASTOGI *****ANIL RASTOGI *****ANIL RASTOGI *****
C  THIS PROGRAMME HAS BEEN MADE BY ANIL K. RASTOGI ,NEW JERSEY INSTITUTE
C  OF TECHNOLOGY
C  SUBSCRIPT 1  -FOR HOT CYCLE - UPFLOW
C  SUBSCRIPT 2  FOR COLD CYCLE  DOWNFLOW
C  X = SOLID CONCENTRATION
C  Y = LIQUID CONCENTRATION
C  N=NUMBER OF CYCLE
C  NNZ = NUMBER OF INTERVALS
C  NCASE = NUMBER OF CASES FO DIFFERNT VALUES OF PHI BOTTOM & PHI TOP
C  NITER = NUMBER OF ITERATIONS
C  AM & BM = CONSTANTS FOR EQUILIBRIUM COEFFICIENTS
C  H = HEIGHT OF THE COLUMN
C  TIME = HALF CYCLE TIME
C  Y0 = FEED CONCENTRATION
C  Q = RESERVOIR VOLUME
C  VT & VB = DEAD VOLUME FOR TOP & BOTTOM RESPECTIVELY
C  VOID = VOIDAGE OF THE PACKING
C  AK1 , AK2 , BK = CONSTANTS FOR MASS-TRANSFER COEFFICIENTS
C  DENS = DENSITY OF THE PACKING
C  ERR = ITERATION ERROR
C  A = CROSS-SECTIONAL AREA OF THE COLUMN
C  PHOT = PHI TOP
C  PHOB = PHI BOTTOM
C  IND = NUMBER TO CHARACTERIZE THE DIRECTION OF FLOW  UP OR DOWN
C  M = NUMBER OF ANY CYCLE , M(MAXIMUM) = N
C  V1 & V2 = FLUID VELOCITY THROUGH THE COLUMN
C  DT1 & DT2 = INCREMENT FOR TIME
C  Y11 = COLUMN TOP CONCENTRATION FOR HOT CYCLE
C  Y1P1 = TOP PRODUCT CONCENTRATION FOR UPFLOW
C  Y1P22 = CONCENTRATION OF TOP DEAD VOLUME
C  Y1P2 = TOP PRODUCT CONCENTRATION
C  Y12 = INLET COLUMN CONCENTRATION
C  YBP22 = CONCENTRATION OF BOTTOM DEAD VOLUME
C  YBP2 = BOTTOM PRODUCT CONCENTRATION FOR DOWNFLOW
C  YBP1 = COLUMN INLET CONCENTRATION FOR HOT CYCLE-UPFLOW
C  XT = INITIAL SOLID CONCENTRATION AT THE COLUMN INLET
C  XXS & YYS = ESTIMATED VALUES AT ANY POINT
C  NTR = NUMBER OF ITERATIONS USED
C  K = NUMBER OF INCREMENT TAKEN HORIZONTALLY
C  TT = TIME AT ANY INCREMENT,TT(MAXIMUM) = TIME
C  X2 & Y2 = INITIAL SOLID & LIQUID CONCENTRATION ALONG HEIGHT OF THE COLUMN
C  XX & YY = SOLID & LIQUID CONCENTRATION ALONG HEIGHT OF THE COLUMN
C          FOR ANY INCREMENT TAKEN HORIZONTALLY
C  XX2 & YY2 = INITIALSOLID & LIQUID CONCENTRATION ALONG THE HEIGHT
C  OF THE COLUMN  FOR  SUBROUTINE
C  THIS PROGRAMME CALCULATES CONCENTRATION PROFILE FOR THERMAL PARAMETRIC
C  PUMP USING MODIFIED ITERATIVE EULERS METHOD
C *****
C  MAIN PROGRAMME
C *****
1  DIMENSION PHOT(10),PHOB(10),X2(100),Y1P2(100),Y2(100),YBP2(100)
   1,XX2(100),YY2(100),XX(100),YY(100),XT(100),XXS(100),YYS(100)
   2,NTR(100)
2  READ 10,N,NNZ,NCASE,NITER
3 10  FORMAT(4I10)

```

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|    |     |  |
|----|-----|--|
| 4  |     | READ 16,AM,BM  |
| 5  | 16  | FORMAT(4E20.5)   |
| 6  |     | READ 15,H,TIME,YO,Q,VT,VB,TEMP1,TEMP2,VOID,AK2,BK,AK1,DENS,ERR,A   |
| 7  | 15  | FORMAT(8F10.4)   |
| 8  |     | READ 20,(PHOT(I),PHOB(I),I=1,NCASE)  |
| 9  | 20  | FORMAT(2F10.4)   |
| 10 |     | DO 600 I=1,NCASE   |
| 11 |     | V1=(1.-PHOB(I))*Q/(A*VOID*TIME)  |
| 12 |     | V2=(1.+PHOB(I))*Q/(A*VOID*TIME)  |
| 13 |     | DT1=H/(NZ*V1)  |
| 14 |     | DT2=H/(NZ*V2)  |
| 15 |     | NZ=NZ+1  |
| 16 |     | YT1=YD   |
| 17 |     | DO 30 J=1,NZ   |
| 18 |     | Y2(J)=YD   |
| 19 | 30  | X2(J)=YD*(AM-BM*TEMP1)   |
| 20 |     | M=1  |
| 21 |     | IF(M-1)31,31,32  |
| 22 | 31  | YTP22=YD   |
| 23 |     | GO TO 33   |
| 24 | 32  | YTP22=YTP2(M-1)  |
| 25 | 33  | YTP1=((PHOT(I)+PHOB(I))*YD+(1.-PHOB(I))*YT1)/(1.+PHOT(I))  |
| 26 |     | IND=2  |
| 27 |     | YTP2(M)=(Q*YTP1+VT*YTP22)/(VT+Q)   |
| 28 |     | YT2=((PHOT(I)+PHOB(I))*YD+(1.-PHOT(I))*YTP2(M))/(1.+PHOB(I))   |
| 29 |     | YIN=YT2  |
|    | C   |  |
| 30 |     | CALL XYCAL(VOID,DENS,AK1,AK2,BK,AM,BM,TEMP1,TEMP2,V1,V2,IND,DT1,<br>DT2,NITER,X2,Y2,YIN,TIME,NZ,XX,YY,YB,XX2,YY2,M,ERR)  |
| 31 |     | YBP2(M)=YB   |
| 32 |     | IF(M-N)50,500,500  |
| 33 | 50  | IF(M-1)51,51,52  |
| 34 | 51  | YBP22=YD   |
| 35 |     | GO TO 53   |
| 36 | 52  | YBP22=YBP1   |
| 37 | 53  | YBP1=(Q*YBP2(M)+VB*YBP22)/(Q+VB)   |
| 38 |     | M=M+1  |
| 39 |     | YIN=YBP1   |
| 40 |     | DO 60 L=1,NZ   |
| 41 |     | LL=L-1   |
| 42 |     | X2(L)=XX(NZ-LL)  |
| 43 | 60  | Y2(L)=YY(NZ-LL)  |
| 44 |     | IND=1  |
|    | C   |  |
| 45 |     | CALL XYCAL(VOID,DENS,AK1,AK2,BK,AM,BM,TEMP1,TEMP2,V1,V2,IND,DT1,<br>DT2,NITER,X2,Y2,YIN,TIME,NZ,XX,YY,YT1,XX2,YY2,M,ERR) |
| 46 |     | DO 70 L=1,NZ   |
| 47 |     | LL=L-1   |
| 48 |     | X2(L)=XX(NZ-LL)  |
| 49 | 70  | Y2(L)=YY(NZ-LL)  |
| 50 |     | GO TO 32   |
| 51 | 500 | PRINT 510,(M,YTP2(M),YBP2(M),M=1,N)  |
| 52 | 510 | FORMAT('-',18X,'M',12X,'YTP2',16X,'YBP2'/(10X,I10,2E20.5))   |
| 53 | 600 | CONTINUE   |
| 54 |     | STOP   |
| 55 |     | END  |
|    | C   |  |
|    | C   |  |



```

56      SUBROUTINE XYCAL(VOID,DENS,AK1,AK2,BK,AM,BM,TEMP1,TEMP2,V1,V2,
      1IND,DT1,DT2,NITER,X2,Y2,YIN,TIME,NZ,XX,YY,YYAV,XX2,YY2,M,ERR)
C *****
C      THIS SUBROUTINE USES TRIAL AND ERROR METHOD FOR THE CALCULATION OF
C      LIQUID AND SOLID PHASE CONCENTRATIONS
C *****
C
57      DIMENSION X2(100),Y2(100),XX(100),YY(100),XT(100),YYS(100),XXS(100
      1),XX2(100),YY2(100),NTR(100)
58      SUM=0
59      IF((-1)*IND)20,10,10
60      10  CK=AK2*V2**BK
61      CM=AM-BM*TEMP2
62      DT=DT2
63      T=TEMP2
64      GO TO 30
65      20  CK=AK1*V1**BK
66      CM=AM-BM*TEMP1
67      DT=DT1
68      T=TEMP1
69      30  CNST=-(1.-VOID)*DENS*CK/VOID
70      XT(1)=X2(1)
71      DO 22 L=1,NZ
72      XX2(L)=X2(L)
73      22  YY2(L)=Y2(L)
74      K=1
75      155 CNK=K
76      TT=DT*CNK
77      YY(1)=YIN
78      XT(K+1)=CM*YY(1)-(CM*YIN-XT(1))*EXP(-CK*TT/CM)
79      XX(1)=XT(K+1)
80      I=2
C
81      85  ITER=1
82      NTR(1)=1
83      YYS(1)=YY(I-1)+CNST*DT*(YY(I-1)-XX(I-1)/CM)
84      XXS(1)=XX2(1)+CK*DT*(YY2(1)-XX2(1)/CM)
85      100 YY(I)=YY(I-1)+(DT/2.)*CNST*(YYS(I)-XXS(I)/CM+YY(I-1)-XX(I-1)/CM)
86      XX(I)=XX2(1)+(CK*DT/2.)*(YYS(I)-XXS(I)/CM+YY2(1)-XX2(1)/CM)
87      DEY=(YY(I)-YYS(1))/YYS(1)
88      IF(ABS(DEY)-ERR)50,50,60
89      50  DEVX=(XX(I)-XXS(1))/XXS(1)
90      IF(ABS(DEVX)-ERR)70,70,60
91      60  IF(ITER-NITER)80,70,70
92      80  YYS(1)=YY(I)
93      XXS(1)=XX(I)
94      ITER=ITER+1
95      GO TO 100
96      70  NTR(1)=ITER
97      IF(1-NZ)81,90,90
98      81  I=I+1
99      GO TO 85
C
100     90  SUM=SUM+YY(NZ)
101     IF(TT-TIME)150,200,200
102     150  DO 151 L=1,NZ
103     YY2(L)=YY(L)
104     151  XX2(L)=XX(L)

```

```

105      K=K+1
106      GO TO 155
107      200  TOTK=K
108      YYAV=SUM/TOTK
109      PRINT 525
110      525  FORMAT(' ')
111      PRINT 95,M,IND,TT
112      95   FORMAT('0',18X,'M',18X,'IND',15X,'TT'/(2120,F20.3))
113      PRINT 98
114      98   FORMAT(' ',18X,'I',12X,'YY',20X,'XX',23X,'NTR')
115      DO 97 I=1,NZ,1
116      97   PRINT 96,I,YY(I),XX(I),NTR(I)
117      96   FORMAT(I20,2E20.5,I20)
118      PRINT 210,YYAV
119      210  FORMAT(' ',20X,'YYAV'/10X,E20.5)
120      RETURN
121      END

```

C

\$ENTRY

| M | IND | TT     |
|---|-----|--------|
| 1 | 2   | 37.067 |

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.10000E-03 | 0.16384E-03 | 1   |
| 2  | 0.96375E-04 | 0.15780E-03 | 3   |
| 3  | 0.92913E-04 | 0.15264E-03 | 3   |
| 4  | 0.89685E-04 | 0.14826E-03 | 3   |
| 5  | 0.86734E-04 | 0.14457E-03 | 3   |
| 6  | 0.84075E-04 | 0.14148E-03 | 3   |
| 7  | 0.81710E-04 | 0.13889E-03 | 3   |
| 8  | 0.79627E-04 | 0.13674E-03 | 3   |
| 9  | 0.77609E-04 | 0.13496E-03 | 3   |
| 10 | 0.76234E-04 | 0.13348E-03 | 3   |
| 11 | 0.74879E-04 | 0.13226E-03 | 3   |
| 12 | 0.73719E-04 | 0.13125E-03 | 3   |
| 13 | 0.72731E-04 | 0.13043E-03 | 3   |
| 14 | 0.71893E-04 | 0.12975E-03 | 3   |
| 15 | 0.71187E-04 | 0.12920E-03 | 3   |
| 16 | 0.70591E-04 | 0.12874E-03 | 2   |
| 17 | 0.70093E-04 | 0.12837E-03 | 2   |
| 18 | 0.69676E-04 | 0.12807E-03 | 2   |
| 19 | 0.69329E-04 | 0.12783E-03 | 2   |
| 20 | 0.69041E-04 | 0.12763E-03 | 2   |
| 21 | 0.68803E-04 | 0.12747E-03 | 2   |
| 22 | 0.68606E-04 | 0.12734E-03 | 2   |
| 23 | 0.68443E-04 | 0.12723E-03 | 2   |
| 24 | 0.68309E-04 | 0.12715E-03 | 2   |
| 25 | 0.68199E-04 | 0.12708E-03 | 2   |
| 26 | 0.68109E-04 | 0.12702E-03 | 2   |
| 27 | 0.68035E-04 | 0.12698E-03 | 2   |
| 28 | 0.67976E-04 | 0.12694E-03 | 1   |

APPENDIX - III - B

Computer programme for heatless parametric pump

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```

$JOB          RASTOGI
C *****
C ***ANIL RASTOGI ***ANIL RASTOGI ***ANIL RASTOGI *****
C THIS PROGRAMME IS MADE BY ANIL K. RASTOGI , NEW JERSEY INSTITUTE OF
C TECHNOLOGY
C THIS PROGRAMME IS FOR THE CALCULATION OF CONCENTRATION PROFILE
C FOR THE HEATLESS PARAMETRIC PUMPING , USING RUNGE KUTTA METHOD
C NOMENCLATURE FOR THE PROGRAMME
C SUBSCRIPT 1-FOR HIGH PRESSURE COLUMN
C SUBSCRIPT 2 - FOR LOW PRESSURE COLUMN
C N - NUMBER OF CYCLES
C NNZ - NUMBER OF HEIGHT INTERVALS
C NCASE - NUMBER OF CASES FOR DIFFERENT PURGE RATIO
C H - HEIGHT OF COLUMN
C TEMP - OPERATING OF COLUMN
C TIME - HALF CYCLE TIME
C VOID - VOIDAGE OF THE BED
C A - CROSS SECTIONAL AREA OF THE COLUMN
C Q - VOLUMETRIC FLOW RATE TO THE COLUMN
C Y0 - INITIAL CONCENTRATION OF THE FEED
C PH- HIGH PRESSURE , PL - LOW PRESSURE
C PURG(1) - PURGE RATIO
C M - CYCLE NUMBER AT ANY TIME
C R - GAS CONSTANT
C DENS - DENSITY OF SOLID
C AK - EQUILIBRIUM CONSTANT
C DP - PACKING PARTICLE DIAMETER
C VISC - VISCOSITY OF GAS (HELIUM )
C DIFFH,DIFFL - DIFFUSIVITY AT HIGH PRESSURE AND LOW PRESSURE
C ALPHA , BETA - CONSTANTS FOR CALCULATION OF MASS-TRANSFER COEFFICIENTS
C AMAV - AVERAGE MOLECULAR WEIGHT OF THE GAS
C DH , DL - DENSITY OF GAS AT HIGH AND LOW PRESSURE RESPECTIVELY
C V01 , V02 - VELOCITY FOR HIGH AND LOW PRESSURE COLUMNS
C DZ1 , DZ2 -INCREMENTS IN Z FOR HIGH AND LOW PRESSURE COLUMNS
C DT - LINEAR INCREMENT IN HEIGHT
C Y - GAS PHASE CONCENTRATION , W - SOLID PHASE CONCENTRATION
C YY , WW - GAS AND SOLID PHASE CONCENTRATION OBTAINED AT THE END
C OF HALF CYCLE
C YYT - COLUMN INLET CONCENTRATION
C YP - PRODUCT CONCENTRATION
C CKG - MASS TRANSFER COEFFICIENT
C *****
C MAIN PROGRAMME
C *****
1 DIMENSION YMH(50),YML(50)
2 DIMENSION PURG(10),W1(800),Y1(800),W2(800),Y2(800),YIN(800),WIN(80
  10)
3 DIMENSION YTT(800),YYT(800)
4 DIMENSION YP1(40),YP2(40),YTPR(40),YBPR(40)
5 DIMENSION ZX(800),YY(800),WW(800),YN(800)
6 READ 1,N,NNZ,NCASE
7 READ 2,R,TEMP
8 READ 3,H,TIME,VOID,A,ERR,Y0,DENS
9 READ 4,Q,PH,PL
10 READ 6,(PURG(1),I=1,NCASE)
11 READ 5,AK
12 READ 7,DP,VISC,ALPHA,BETA,AMAV
13 7 FORMAT(8F10.5)
  
```

|    |      |   |
|----|------|---|
| 14 | 5    | FORMAT(F20.10)  |
| 15 | 2    | FORMAT(3F10.4)  |
| 16 | 1    | FORMAT(5I10)  |
| 17 | 3    | FORMAT(8F10.4)  |
| 18 | 2020 | FORMAT('1')   |
| 19 | 11   | FORMAT(///)   |
| 20 |      | PRINT 2020  |
| 21 | 6    | FORMAT(F10.4)   |
| 22 | 4    | FORMAT(3F10.4)  |
| 23 |      | PRINT 3030  |
| 24 | 3030 | FORMAT(9X,'N',8X,'NNZ',5X,'NCASE'//)  |
| 25 |      | PRINT 1,N,NNZ,NCASE   |
| 26 |      | PRINT 11  |
| 27 |      | PRINT 4040  |
| 28 | 4040 | FORMAT(5X,'R',5X,'TEMP'//)  |
| 29 |      | PRINT 2,R,TEMP  |
| 30 |      | PRINT 11  |
| 31 |      | PRINT 5050  |
| 32 | 5050 | FORMAT(5X,'H',7X,'TIME',7X,'VOID',9X,'A',8X,'ERR',8X,'YD',6X,<br>1'DENS'//)   |
| 33 |      | PRINT 3,H,TIME,VOID,A,ERR,YD,DENS   |
| 34 |      | PRINT 11  |
| 35 |      | PRINT 6060,AK   |
| 36 | 6060 | FORMAT('-',10X,'AK = ',F20.10)  |
| 37 |      | PRINT 11  |
| 38 |      | PRINT 7070  |
| 39 | 7070 | FORMAT(5X,'Q',8X,'PH',9X,'PL'//)  |
| 40 |      | PRINT 4,Q,PH,PL   |
| 41 |      | PRINT 11  |
| 42 |      | PRINT 8080  |
| 43 | 8080 | FORMAT(2X,'PURGE RATIO'//)  |
| 44 |      | PRINT 6,(PURG(I),I=1,NCASE)   |
| 45 |      | DH=PH/(R*TEMP)  |
| 46 |      | DL=PL/(R*TEMP)  |
| 47 |      | PRINT 6091  |
| 48 | 6091 | FORMAT('-',15X,'DH',15X,'DL')   |
| 49 |      | PRINT 8085,DH,DL  |
| 50 | 8085 | FORMAT('-',2E20.5)  |
| 51 |      | PRINT 9002  |
| 52 | 9002 | FORMAT('-',4X,'DP',7X,'VISC',7X,'DIFF',7X,'ALPHA',7X,'BETA',7X,'AM<br>1AV'//) |
| 53 |      | PRINT 7,DP,VISC,DIFF,ALPHA,BETA,AMAV  |
| 54 |      | YMH(1)=YO   |
| 55 |      | DU 60011=1,NCASE  |
| 56 |      | DT=H/NNZ  |
| 57 |      | V01=Q/(A*VOID)  |
| 58 |      | V02=Q*PURG(11)/(A*VOID)   |
| 59 |      | DZ1=H/(NNZ*V01)   |
| 60 |      | DZ2=H/(NNZ*V02)   |
| 61 |      | NNT1=TIME/DZ1   |
| 62 |      | PRINT 8086  |
| 63 | 8086 | FORMAT('-',15X,'DZ1',15X,'DZ2',10X,'NNT1',15X,'V01',15X,'V02')                |
| 64 |      | PRINT 6090,DZ1,DZ2,NNT1,V01,V02   |
| 65 | 6090 | FORMAT('-',2E20.5,110,2E20.5)   |
| 66 |      | GPH=Q*DH  |
| 67 |      | GPL=Q*DL*PURG(11)   |
| 68 |      | DIFFH=7.488   |
| 69 |      | DIFFL=22.476  |

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|     |      |  |
|-----|------|--|
| 70  |      | CKGH=CMASC(DP,VISC,DH,GPH,ALPHA,BETA,VOID,AMAV,H,A,DIFFH)  |
| 71  |      | CKGL=CMASC(DP,VISC,DL,GPL,ALPHA,BETA,VOID,AMAV,H,A,DIFFL)  |
| 72  |      | PRINT 9001,GPH,GPL,CKGH,CKGL   |
| 73  | 9001 | FORMAT(' ','GPH=',E12.5,10X,'GPL=',E12.5,10X,'CKGH=',E12.5,10X,'CKGL=',E12.5//)                              |
| 74  |      | NT1=NNT1+2   |
| 75  |      | NNT2=TIME/DZ2  |
| 76  |      | NT2=NNT2+2   |
| 77  |      | NZ=NNZ+1   |
| 78  |      | M=1  |
| 79  |      | DO 33 J=1,NZ   |
| 80  |      | AA=J-1   |
| 81  |      | ZX(J)=DT*AA  |
| 82  |      | Y1(J)=YD   |
| 83  |      | W1(J)=AK*PH*Y1(J)/(R*TEMP)   |
| 84  |      | YY(J)=YD   |
| 85  |      | WW(J)=W1(J)  |
| 86  |      | Y2(J)=YD   |
| 87  | 33   | W2(J)=AK*PH*Y2(J)/(R*TEMP)   |
| 88  |      | YP1(1)=YD  |
| 89  |      | NN=NT1   |
| 90  |      | DO 70 J=1,NT1  |
| 91  | 70   | YTT(J)=YD  |
| 92  |      | GO TO 71   |
| 93  | 8    | IND=1  |
| 94  |      | CALL CNEW(PH,PL,NNZ,H,IND,YD,Y2,YN,NZ,ZX)  |
| 95  |      | DO 20 K=1,NZ   |
| 96  |      | KK=K-1   |
| 97  |      | YIN(K)=YN(NZ-KK)   |
| 98  | 20   | WIN(K)=W2(NZ-KK)   |
| 99  |      | DO 34 J=1,NT1  |
| 100 | 34   | YTT(J)=YD  |
| 101 |      | CALL XYCAL(VOID,TIME,DENS,PH,DH,YIN,WIN,VO1,DZ1,YTT,M,IND,NTT,YY,<br>1WW,YYAV1,YTT,NZ,R,TEMP,CKGH,AK,H,YEND) |
| 102 |      | YMH(M)=YEND  |
| 103 |      | YP1(M)=YYAV1   |
| 104 |      | NN=NTT+1   |
| 105 | 71   | CALL TOPCAL(NN,NT2,DZ1,DZ2,YTT,YTT)  |
| 106 |      | IND=2  |
| 107 |      | DO 99 J=1,NZ   |
| 108 |      | JJ=J-1   |
| 109 | 99   | Y2(J)=Y1(NZ-JJ)  |
| 110 |      | CALL CNEW(PH,PL,NNZ,H,IND,YD,Y2,YN,NZ,ZX)  |
| 111 |      | DO 97 K=1,NZ   |
| 112 |      | KK=K-1   |
| 113 |      | YIN(K)=YN(K)   |
| 114 | 97   | WIN(K)=W1(NZ-KK)   |
| 115 |      | DO 66 L=1,NZ   |
| 116 |      | YI(L)=YY(L)  |
| 117 | 66   | WI(L)=WW(L)  |
| 118 |      | CALL XYCAL(VOID,TIME,DENS,PL,DL,YIN,WIN,VO2,DZ2,YTT,M,IND,NTT,Y2,<br>1W2,YYAV2,YTT,NZ,R,TEMP,CKGL,AK,H,YEND) |
| 119 |      | YML(M)=YEND  |
| 120 |      | YP2(M)=YYAV2   |
| 121 |      | IF (M.EQ.N)GO TO 104   |
| 122 |      | M=M+1  |
| 123 |      | GO TO 8  |
| 124 | 600  | CONTINUE   |

```

125 333 FORMAT(I5,10X,2(E20.5,10X),2E20.5)
126 104 PRINT 105
127 105 FORMAT('-',3X,'N',21X,'<YTP>',24X,'<YBP>',24X,'YPT/YO',14X,'YBP/YO
1'///)
128 DO 101 J=1,N
129 YTPR(J)=YP1(J)/YO
130 YBPR(J)=YP2(J)/YO
131 101 PRINT 333,J,YP1(J),YP2(J),YTPR(J),YBPR(J)
132 PRINT 446
133 446 FORMAT('1',15X,'MINIMUM AND MAXIMUM CONCENTRATION AT THE END OF
1HALF CYCLE'///)
134 PRINT 447
135 447 FORMAT('-',12X,'YMIN',16X,'YMAX'///)
136 445 FORMAT(2E20.5)
137 DO 444 JJ=1,N
138 444 PRINT 445,YMH(JJ),YML(JJ)
139 STOP
140 END

```

```

141 FUNCTION CMASC(DP,VISC,DS,Q,ALPHA,BETA,VOID,AMAV,H,A,DIFF)

```

```

C *****
C THIS SUBFUNCTION CALCULATES MASS TRANSFER COEFFICIENT
C *****

```

```

142 BF=(VISC/(DS*DIFF*AMAV))**(2./3.)
143 AJD=ALPHA*((Q*DP*AMAV/(VISC*(1.-VOID)*A*VOID))**(BETA))
144 CMASC=AJD*Q/(BF*H*DS*A*VOID)
145 RETURN
146 END

```

C

```

147 FUNCTION RK(WINE,WI,CKG,DZ)

```

```

C *****
C RUNGE - KUTTA METHOD FOR THE CALCULATION OF SOLID PHASE CONCENTRATION
C SOLID PHASE EQUILIBRIUM CONCENTRATION HAS BEEN TAKEN CONSTANT
C *****

```

```

148 AK1=CKG*(WINE-WI)*DZ
149 W=WI+AK1/2.
150 AK2=CKG*(WINE-W)*DZ
151 W=WI+AK2/2.
152 AK3=CKG*(WINE-W)*DZ
153 W=WI+AK3
154 AK4=CKG*(WINE-W)*DZ
155 DW=(AK1+2.*AK2+2.*AK3+AK4)/6.
156 RK=WI+DW
157 RETURN
158 END

```

C

```

159 SUBROUTINE CNEW(PH,PL,NNZ,H,IND,YO,Y,YN,NP,DHX)

```

```

C *****
C THIS SUBROUTINE IS FOR DEPRESSURIZATION AND REPRESSURIZATION STEP
C THIS IS BASED ON IDEAL GAS LAW  $P1*V1 = P2*V2$ 
C *****

```

```

160 DIMENSION Y(800),YN(800),DIST(800),DHX(800)
161 DIST(1)=0.0

```

```

162      NZ=NNZ+1
163      DO 40 I=2,NP
164      IF((-1)**IND.LT.0.0)GO TO 29
165      DIST(I)=DHX(I)*PH/PL
166      GO TO 40
167      29  DIST(I)=DHX(I)*PL/PH
168      40  CONTINUE
169      A=10.**(-10.)
170      YN(1)=Y(1)
171      DO 24 J=2,NZ
172      DO 25 M=2,NP
173      IF (Y(J).LT.A)GO TO 10
174      GO TO 18
175      10  PRINT15,J,Y(J)
176      15  FORMAT(' ',Y(' ',I2,' ')=' ',5X,E20.5)
177      18  IF(DHX(J)-DIST(M))26,26,22
178      22  IF(M.EQ.NZ)GO TO 27
179      25  CONTINUE
180      26  YN(J)=Y(M-1)+(Y(M)-Y(M-1))*(DHX(J)-DIST(M-1))/(DIST(M)-DIST(M-1))
181      GO TO 24
182      27  YN(J)=Y0
183      24  CONTINUE
184      RETURN
185      END

```

C

```

186      SUBROUTINE TOPCAL(NN,NT2,DZ1,DZ2,YYT,YYT2)
C *****
C THIS SUBROUTINE IS FOR LINEAR INTERPOLATION OF TOP PRODUCT OR INLET
C CONCENTRATION FOR SECOND COLUMN
C *****
187      DIMENSION YYT(800),YYT2(800)
188      YYT2(1)=YYT(1)
189      DO 40 J=2,NT2
190      CNK =J-1
191      TT2=CNK*DZ2
192      M=1
193      20  CN=M
194      TT1=CN*DZ1
195      IF(TT2-TT1)50,55,60
196      60  IF ((M+1).EQ.NN)GO TO 55
197      M=M+1
198      GO TO 20
199      55  YYT2(J)=YYT(M+1)
200      GO TO 40
201      50  YYT2(J)=YYT(M) +(YYT(M+1)-YYT(M))*DZ2/DZ1
202      40  CONTINUE
203      RETURN
204      END

```

```

205      FUNCTION RK1(WE,WA,CNST,DZ,CNST1,YI)
C *****
C RUNGE - KUTTA METHOD FOR THE CALCULATION OF GAS PHASE CONCENTRATION
C SOLID PHASE CONCENTRATION HAS BEEN TAKEN CONSTANT
206      AKK1=CNST*(WE-WA)*DZ
207      Y=YI+AKK1/2.

```

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```

208 WE=Y*CNST1
209 AKK2=CNST*(WE-WA)*DZ
210 Y=YI+AKK2/2.
211 WE=Y*CNST1
212 AKK3=CNST*(WE-WA)*DZ
213 Y=YI+AKK3
214 WE=Y*CNST1
215 AKK4=CNST*(WE-WA)*DZ
216 DY=(AKK1+2.*AKK2+2.*AKK3+AKK4)/6.
217 RK1=YI+DY
218 RETURN
219 END

```

C

```

220 SUBROUTINE XYCAL(VOID,TIME,DENS,PS,DF,YY2,WW2,VO,DZ,YTT,M,IND,K,YY
1,WW,YYAV,YBT,NZ,R,TEMP,CKG,AK,H,YYAV1)

```

C \*\*\*\*\*

C THIS SUBROUTINE CALCULATES CONCENTRATION PROFILE

C \*\*\*\*\*

```

221 DIMENSION YTT(800),YBT(800)
222 DIMENSION YY2(800),WW2(800),YY(800),WW(800)
223 CNST1=AK*PS/(R*TEMP)
224 CNST=- (1.-VOID)*DENS*CKG/(VOID*DF)
225 NNZ=NZ-1
226 SUM=YY2(NZ)
227 YBT(1)=YY2(NZ)
228 K=1
229 155 CNK=K
230 WI=WW2(1)
231 TT=DZ*CNK
232 YY(1)=YTT(K+1)
233 YIN=YTT(K+1)
234 WINE=AK*PS*YIN/(R*TEMP)
235 WW(1)=RK(WINE,WI,CKG,DZ)
236 WWIE=AK*PS*YY2(1)/(R*TEMP)
237 I=2
238 85 WWZE=AK*PS*YY2(I)/(R*TEMP)
239 YY(I)=RK1(WWIE,WW2(I-1),CNST,DZ,CNST1,YY2(I-1))
240 WW(I)=RK(WWZE,WW2(I),CKG,DZ)
241 IF(I-NZ)81,90,90
242 81 I=I+1
243 WWIE=WWZE
244 GO TO 85
245 90 SUM=SUM+YY(NZ)
246 YBT(K+1)=YY(NZ)
247 IF(TT-TIME)150,200,200
248 150 DO 151 L=1,NZ
249 YY2(L)=YY(L)
250 151 WW2(L)=WW(L)
251 K=K+1
252 GO TO 155
253 200 TOTK=K+1
254 YYAV=SUM/TOTK
255 YYAV1=YY(NZ)
256 IF(IND.EQ.2)GO TO 1
257 PRINT 995,M
258 995 FORMAT('-',10X,' END VALUES OF HIGH PRESSURE COLUMN WHEN M=',I2//)

```

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```
259      GO TO 2
260      1      PRINT 996,M
261      996    FORMAT('-',10X,' END VALUES OF LOW PRESSURE COLUMN WHEN M=',I2//)
262      2      PRINT 998
263      998    FORMAT('-', 'END VALUES',7X,'I',8X,'WW(I)',15X,'YY(I)')
264      999    FORMAT('-',10X,I10,3E20.8)
265      DO 153 L=1,NZ,20
266      153    PRINT 999,L      ,WW(L),YY(L)
267      MC=M
268      PRINT 800,MC,YYAV,YYAV1
269      800    FORMAT('-',20X,'M=',I2,10X,'YYAV=',E12.5,5X,'YYAV1=',E12.5)
270      RETURN
271      END
```

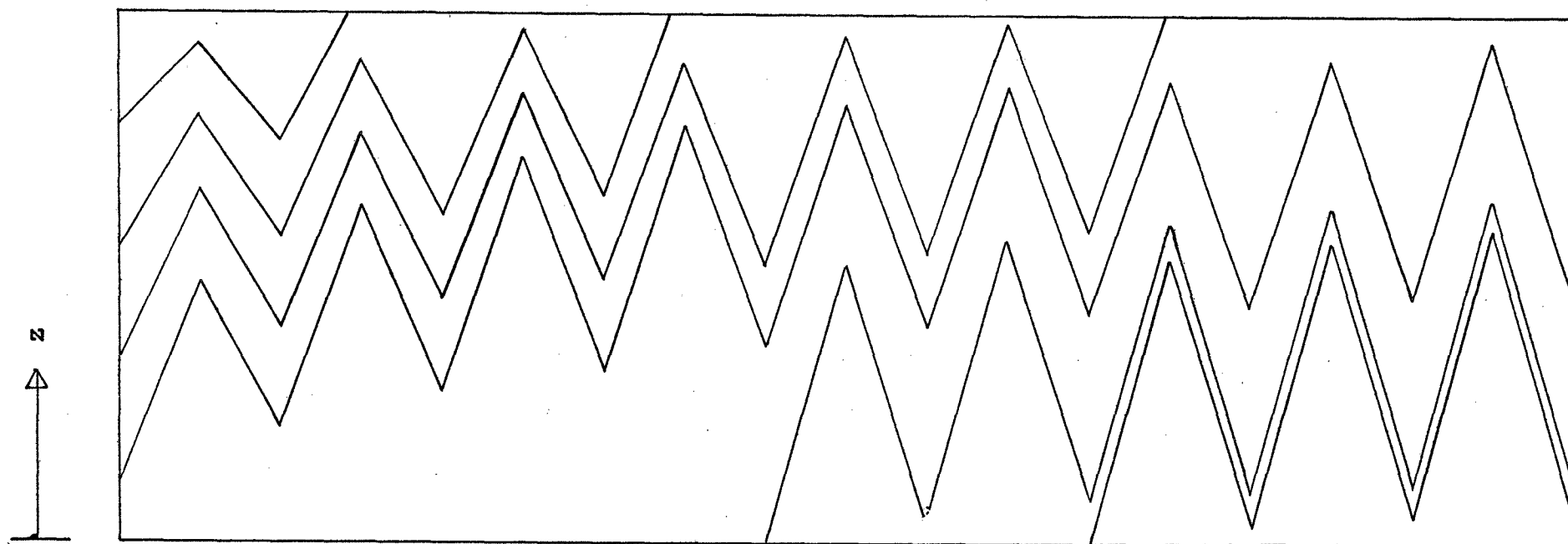
APPENDIX - IV

Figures

Figure :10

$$\frac{\bar{t}}{\omega} = 60 \text{ mins.}, Q\left(\frac{\pi}{\omega}\right) = 25 \text{ cc}$$

$$\phi_B = 0.01, C_0 = 0.1 \text{ M}$$



→ t

Concentration Front

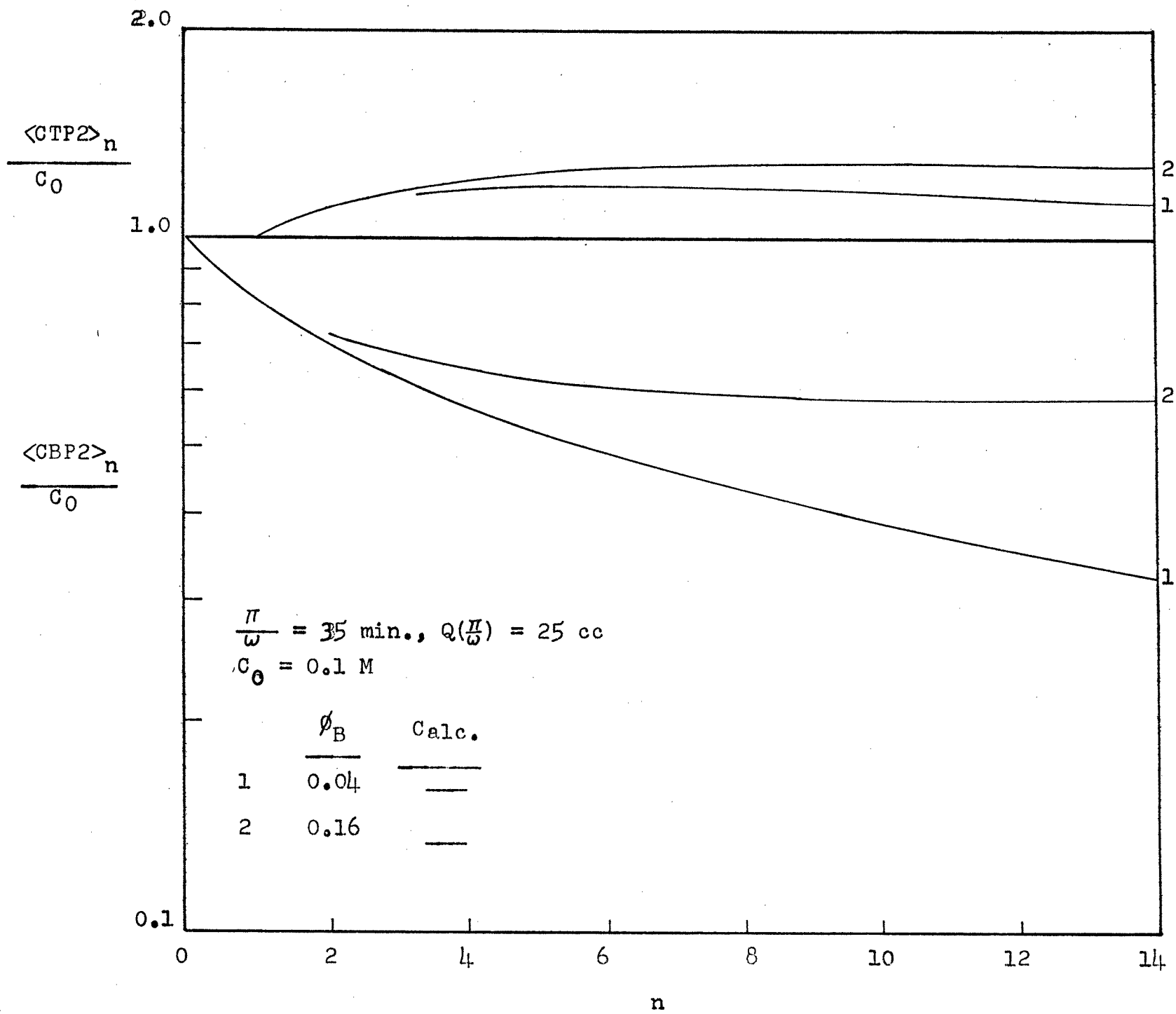
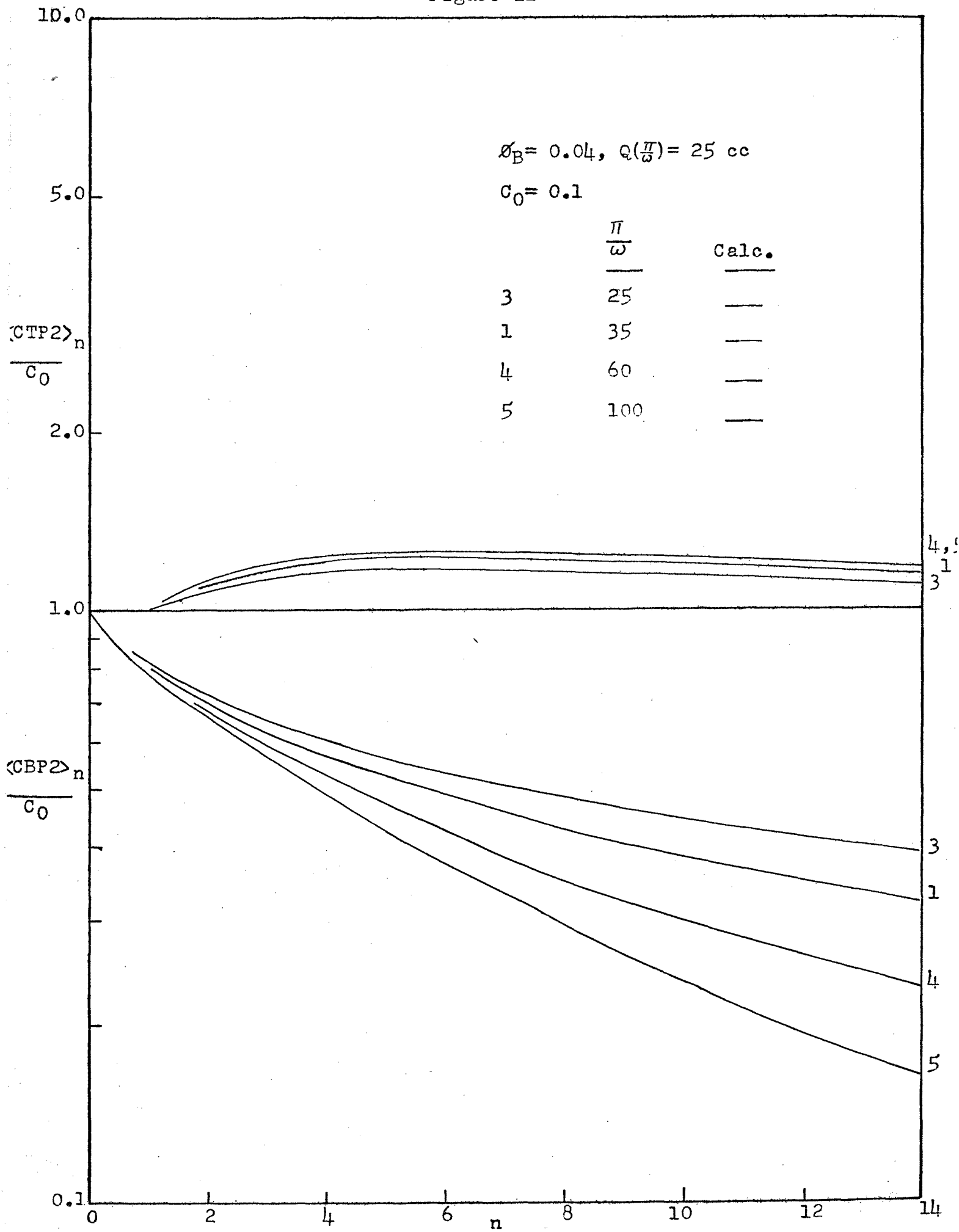


Figure 12



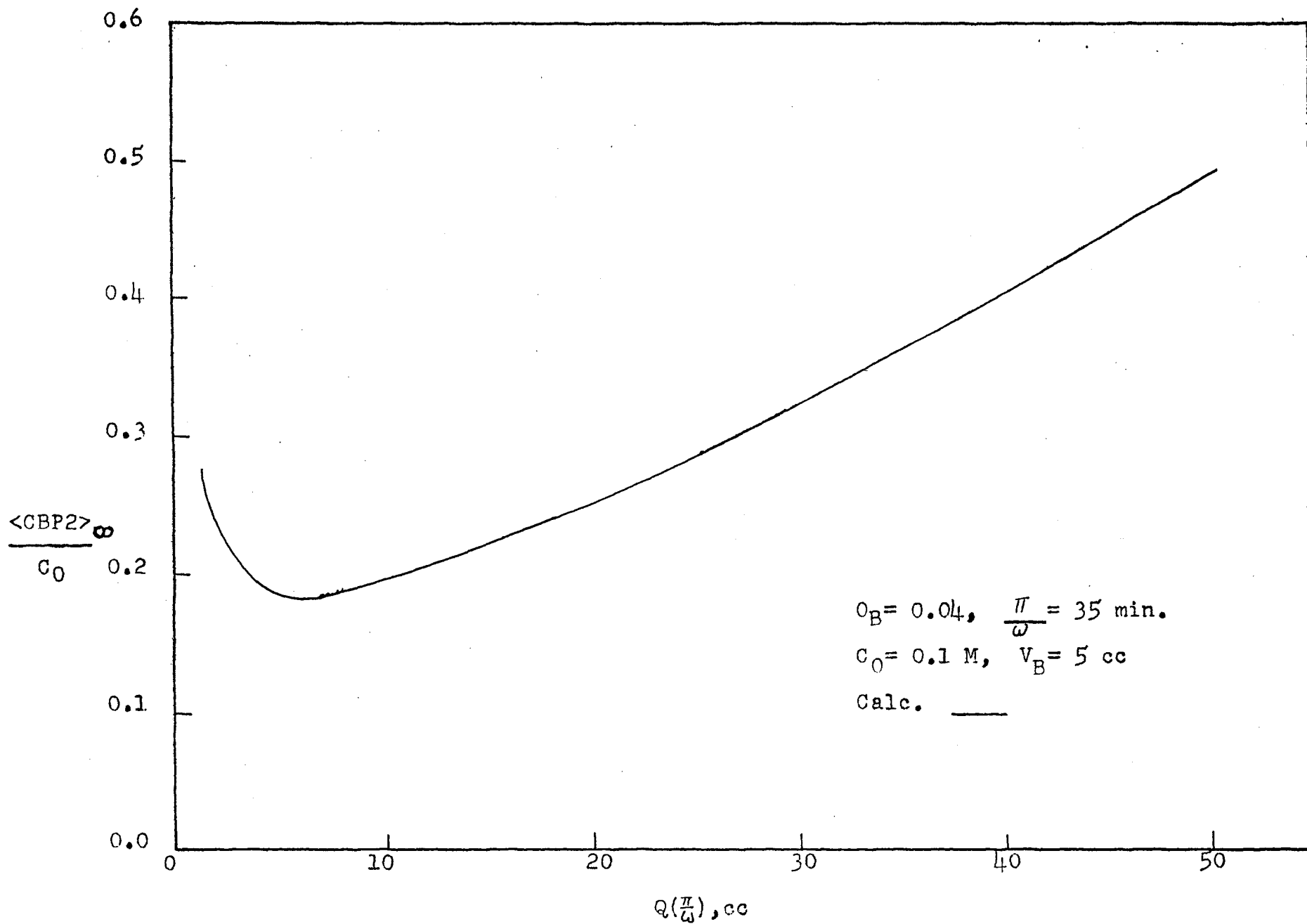


Figure 14

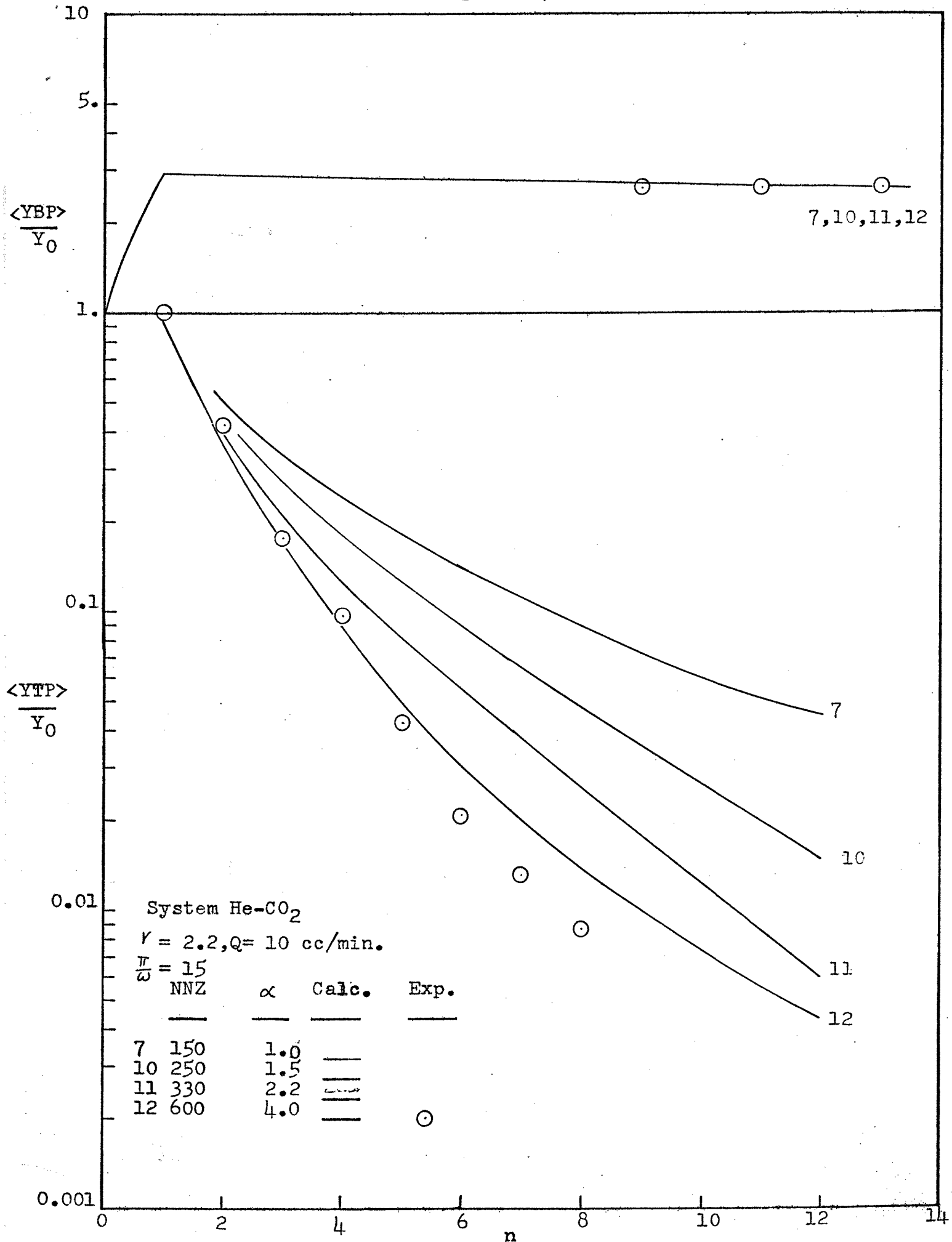




Figure 15

Calibration for He-CO<sub>2</sub> system

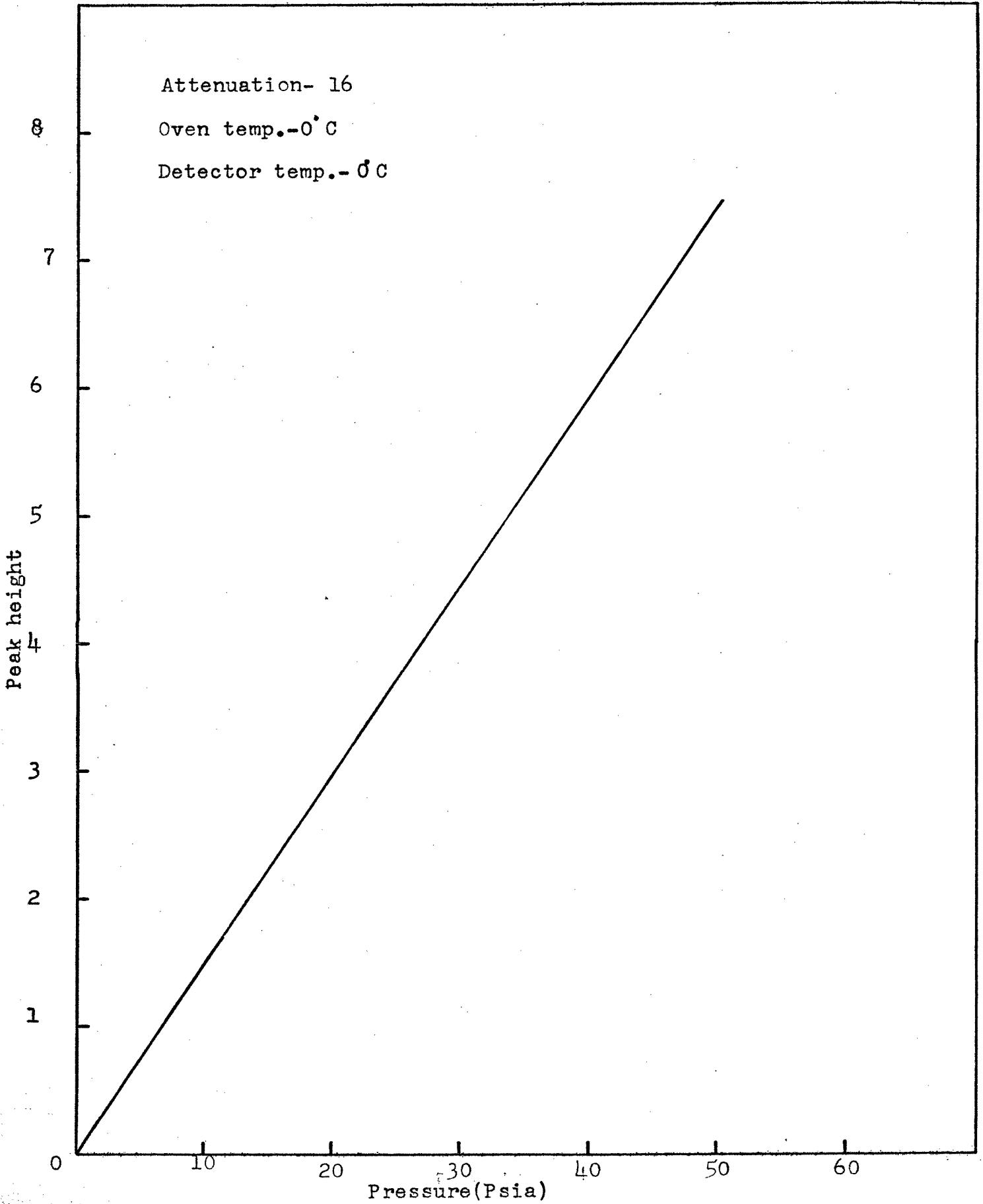


Figure 16

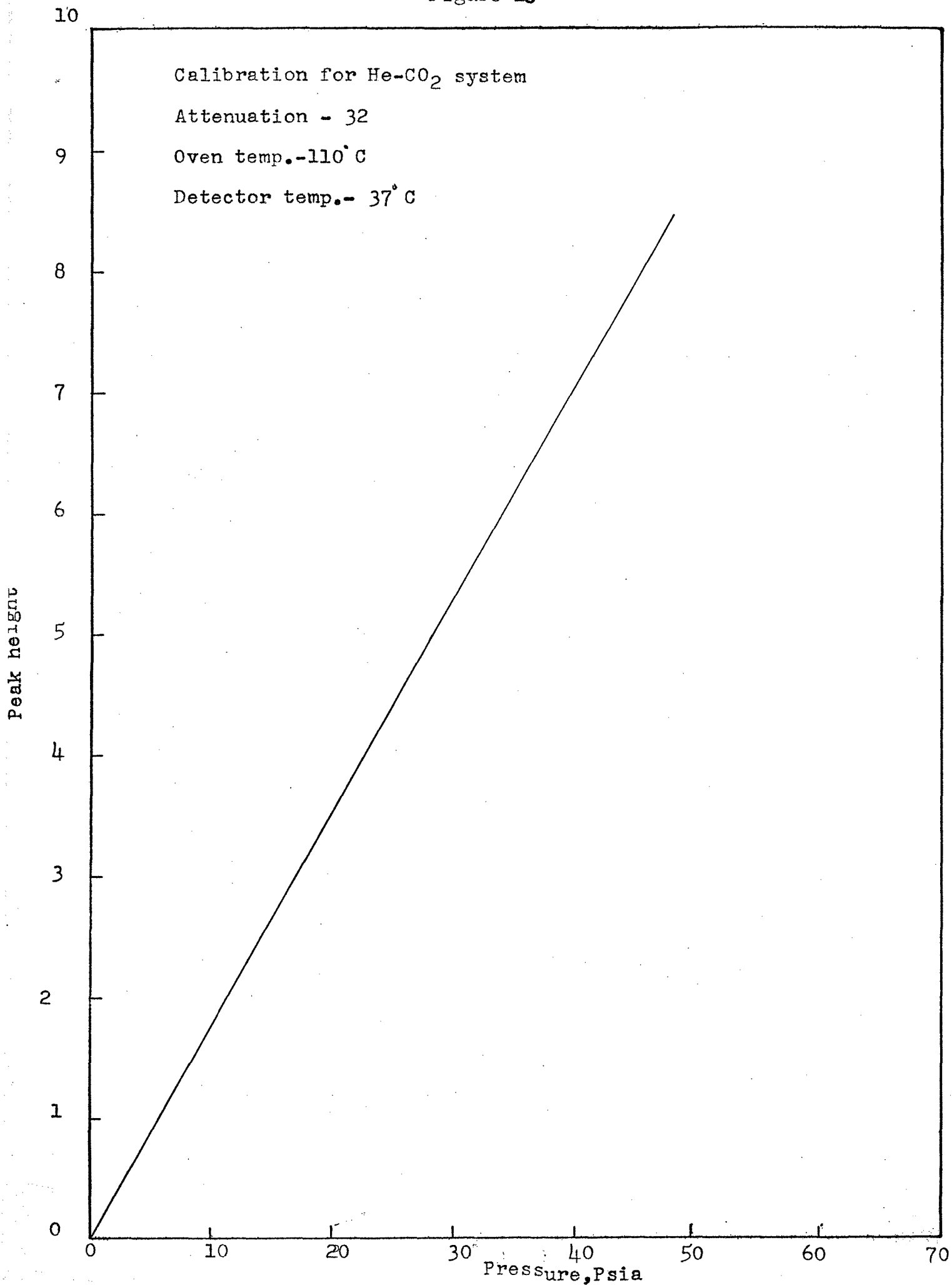


Figure 17

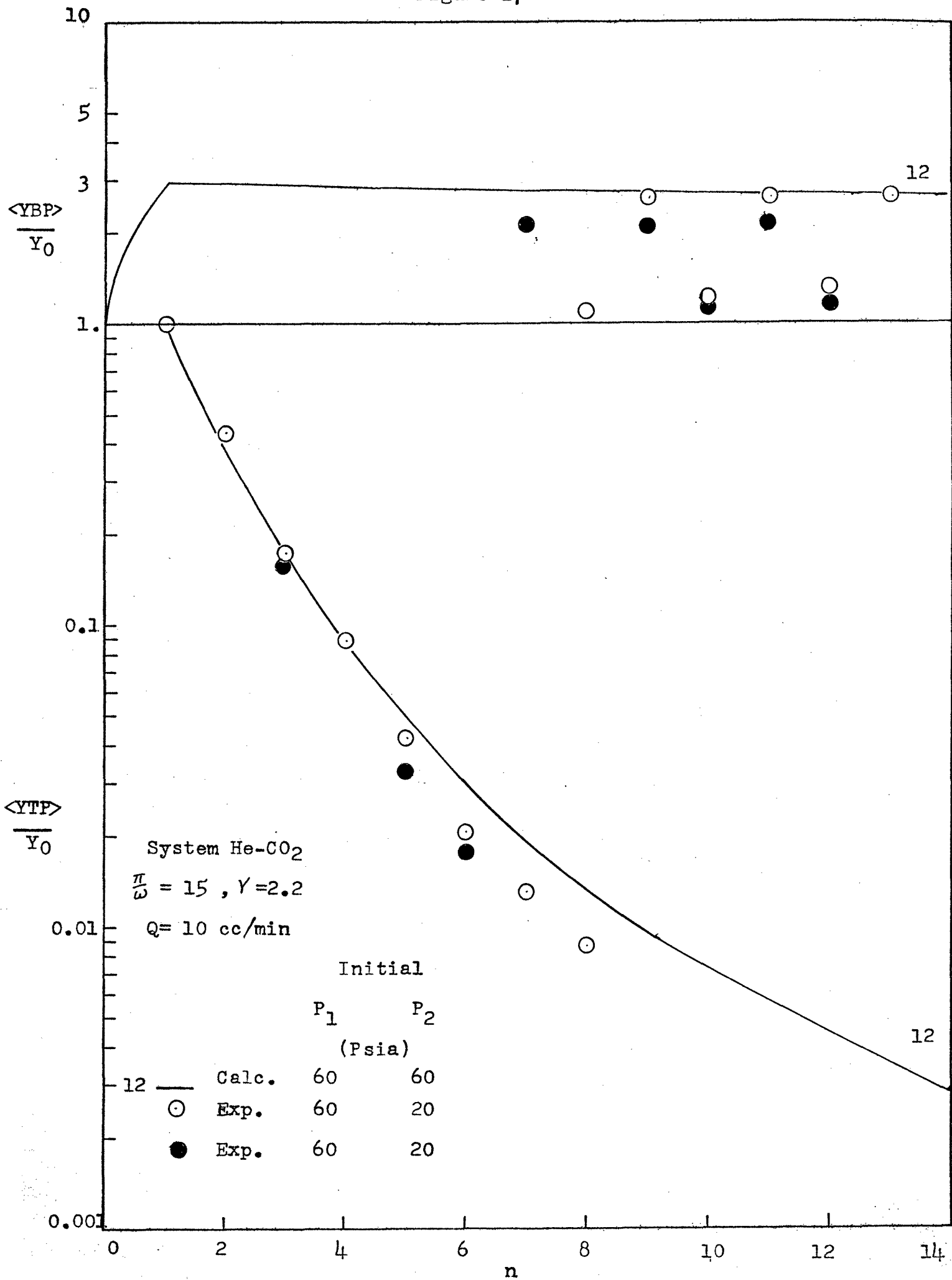


Figure 18

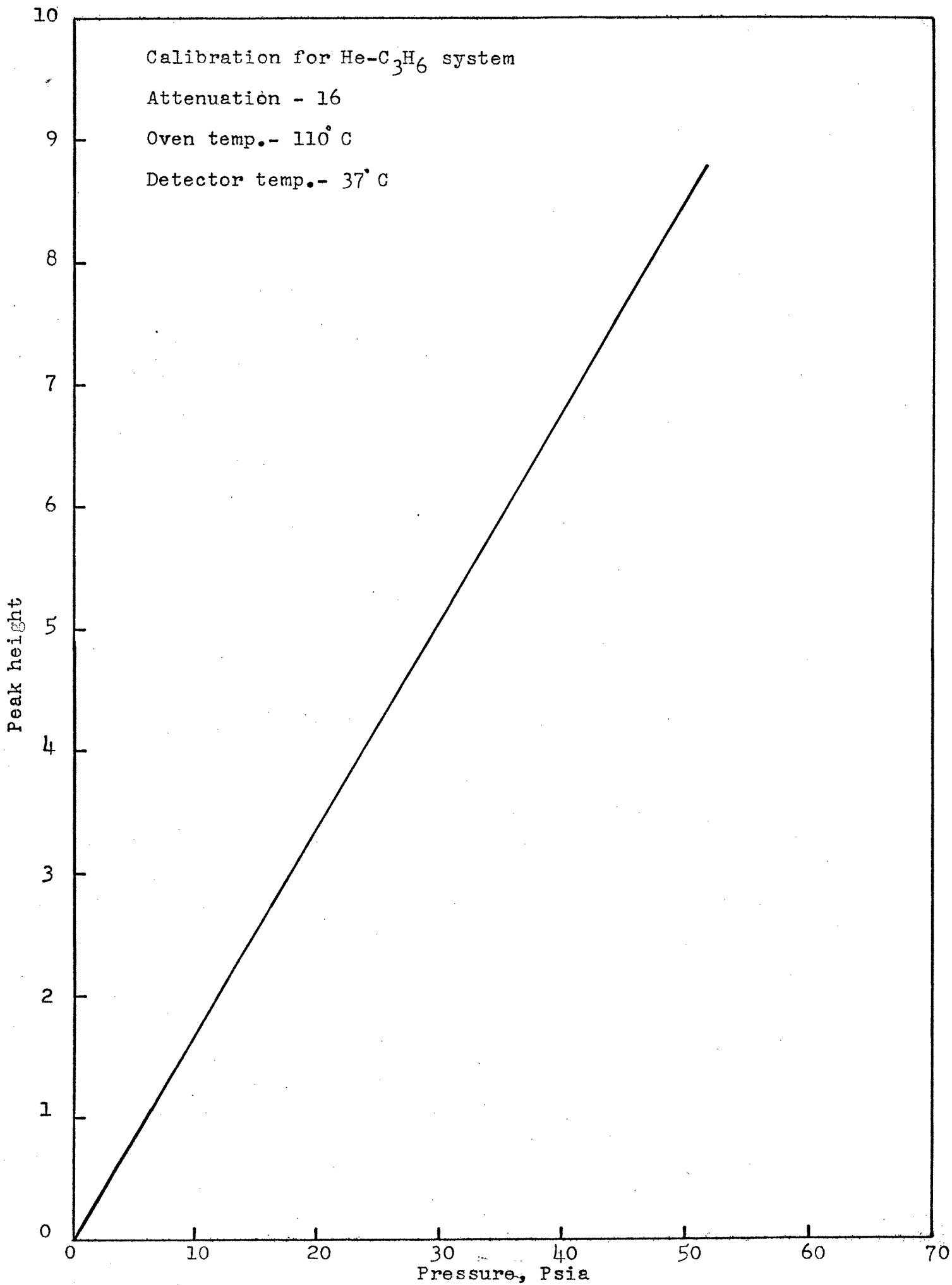


Figure 19

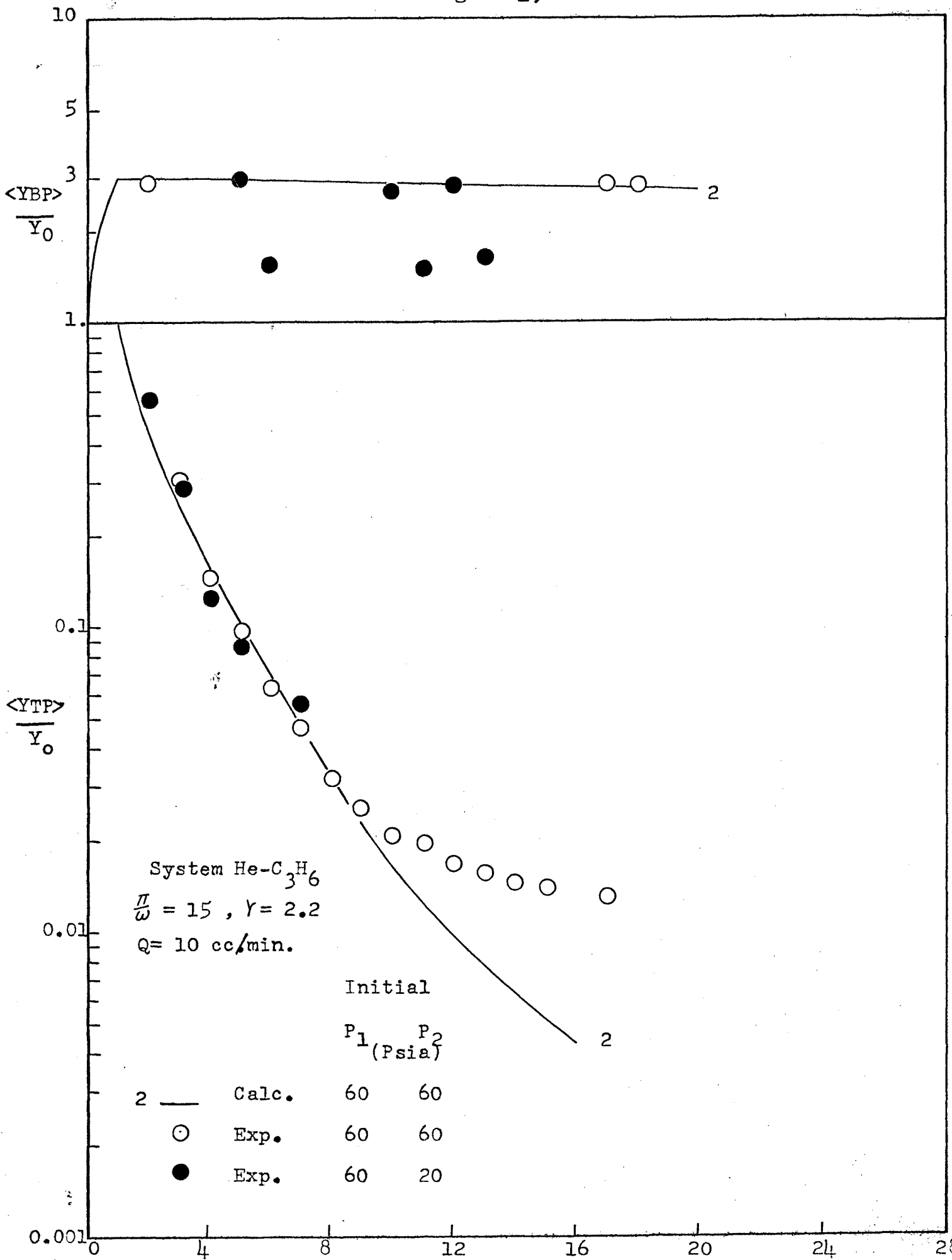


Figure 20

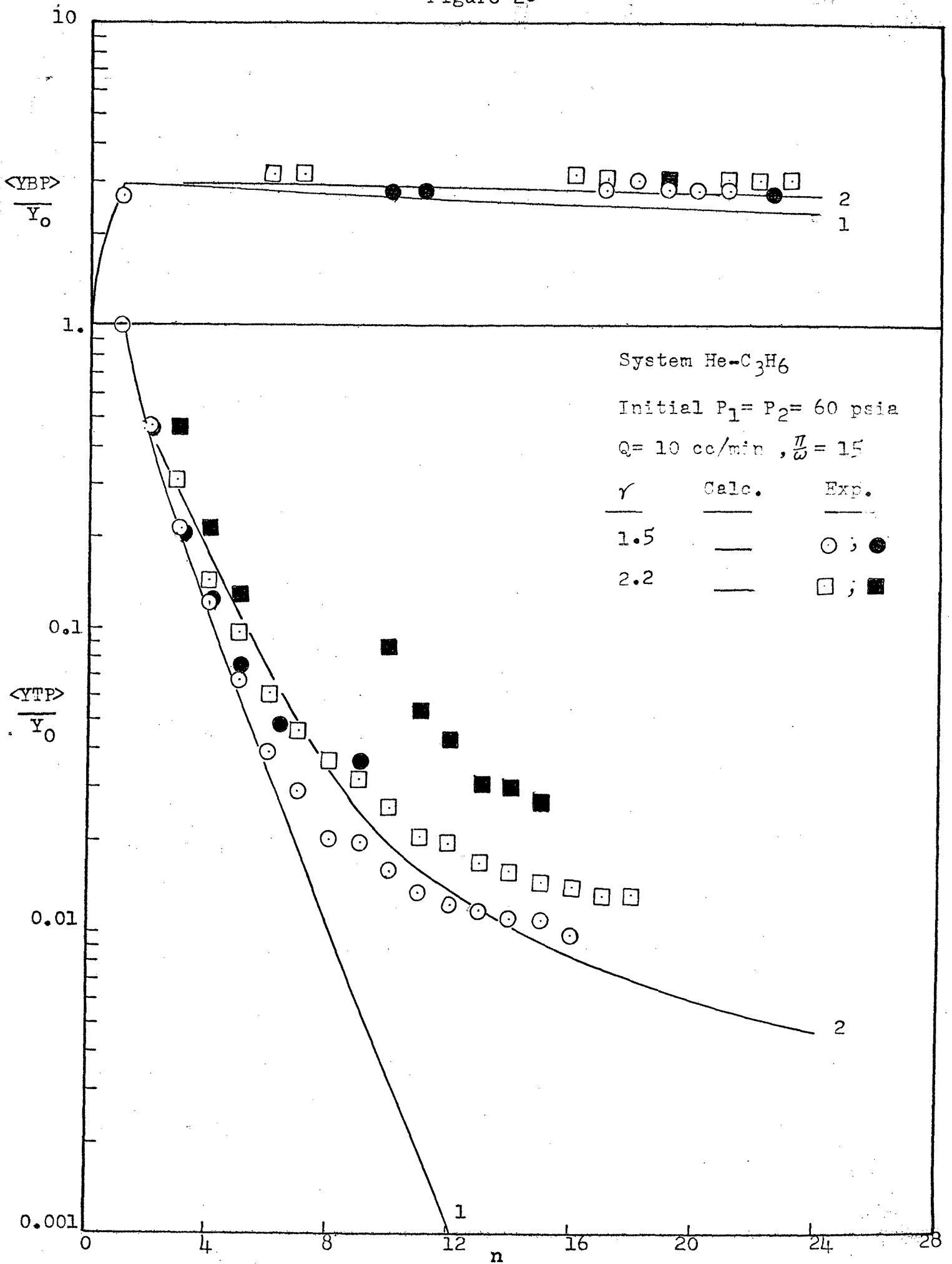


Figure 21

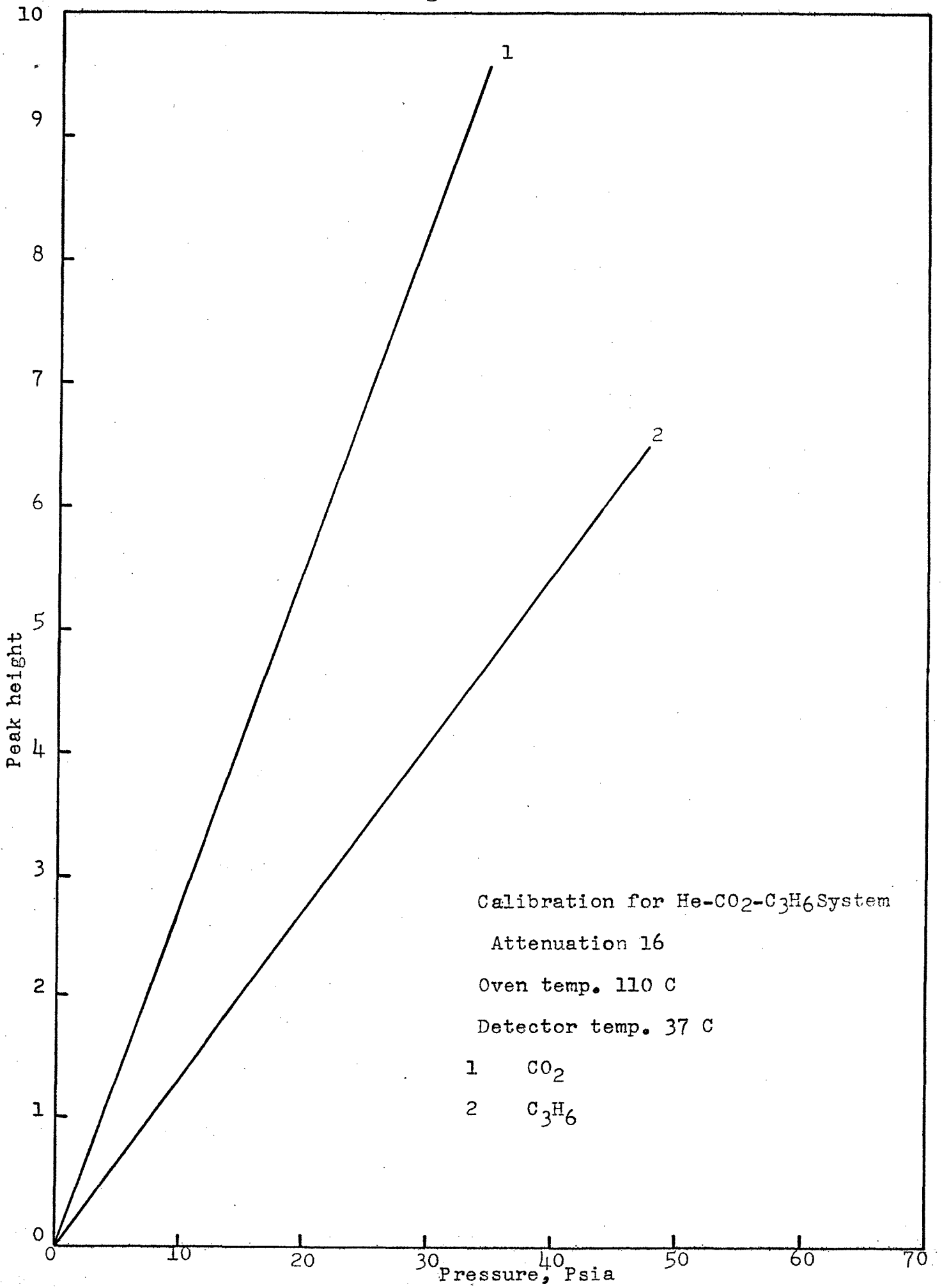


Figure : 22

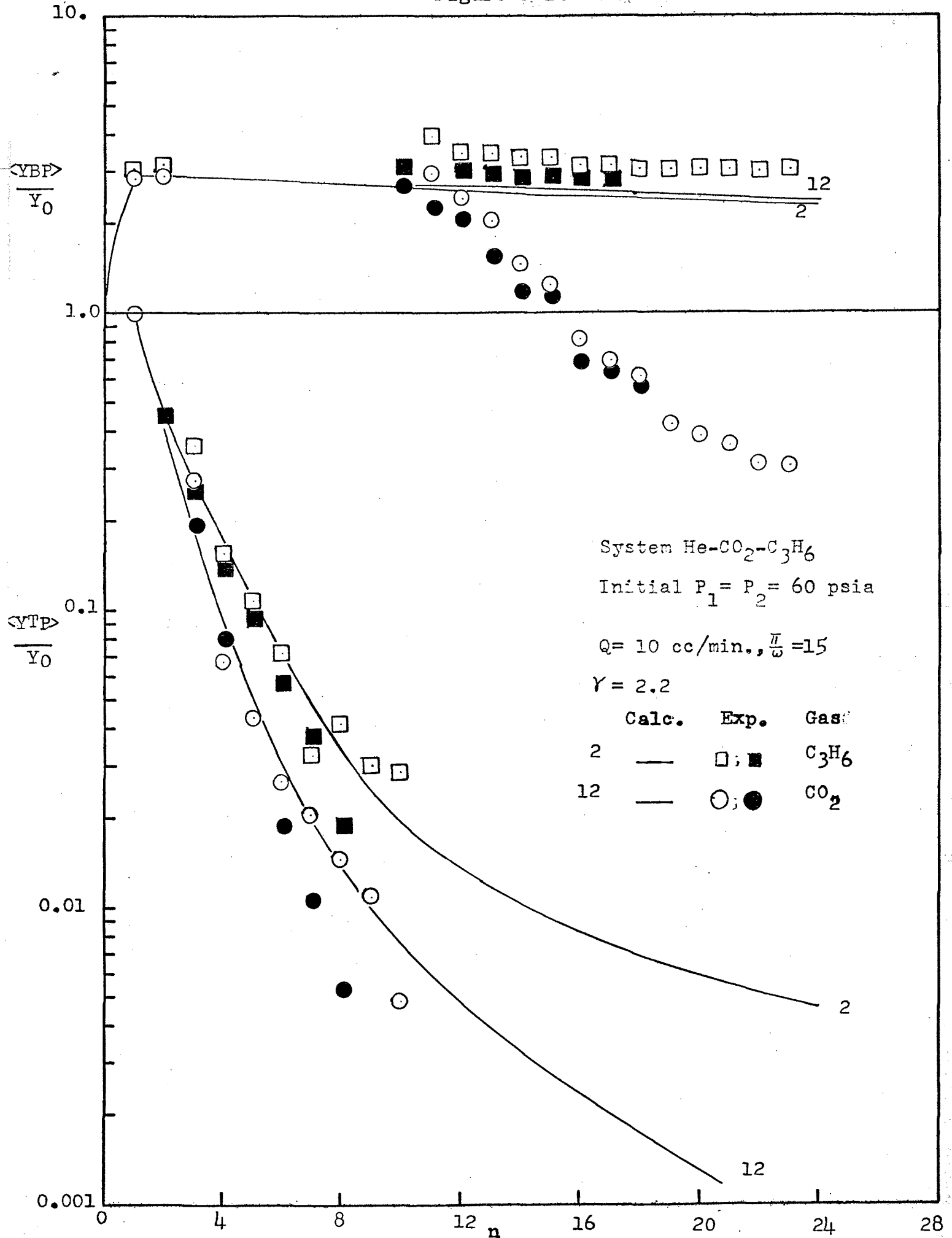




Figure : 23

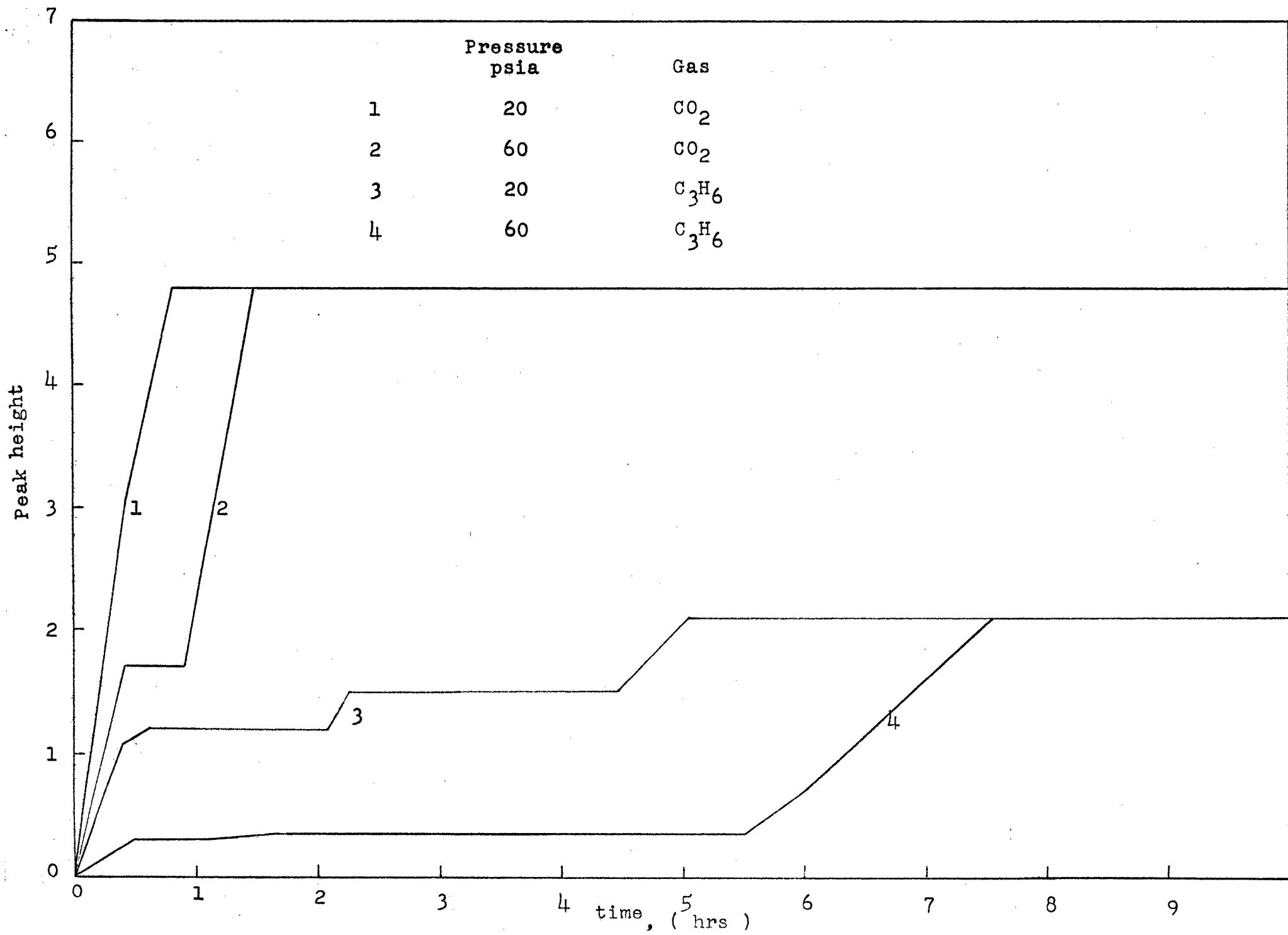


Figure 24

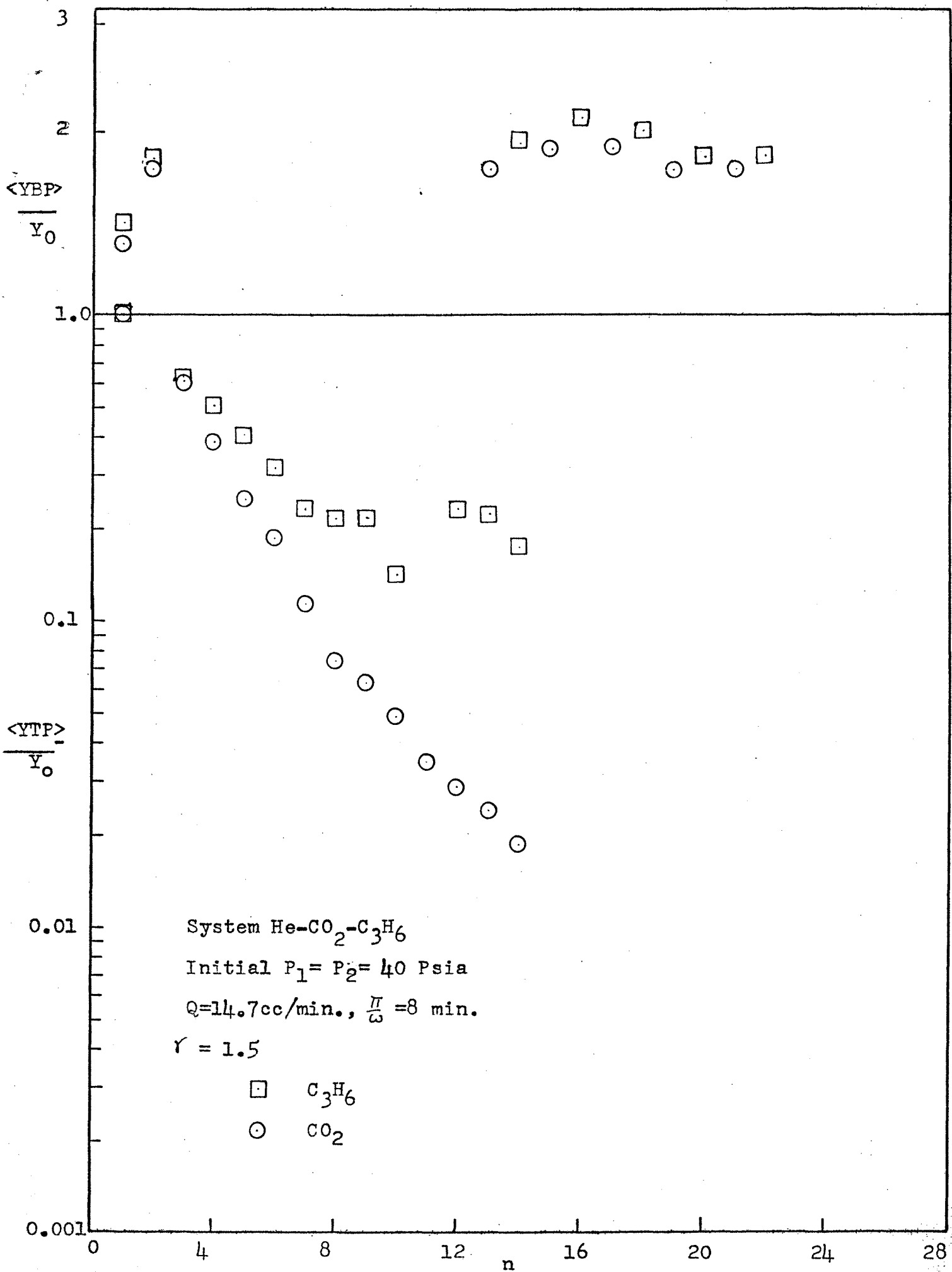
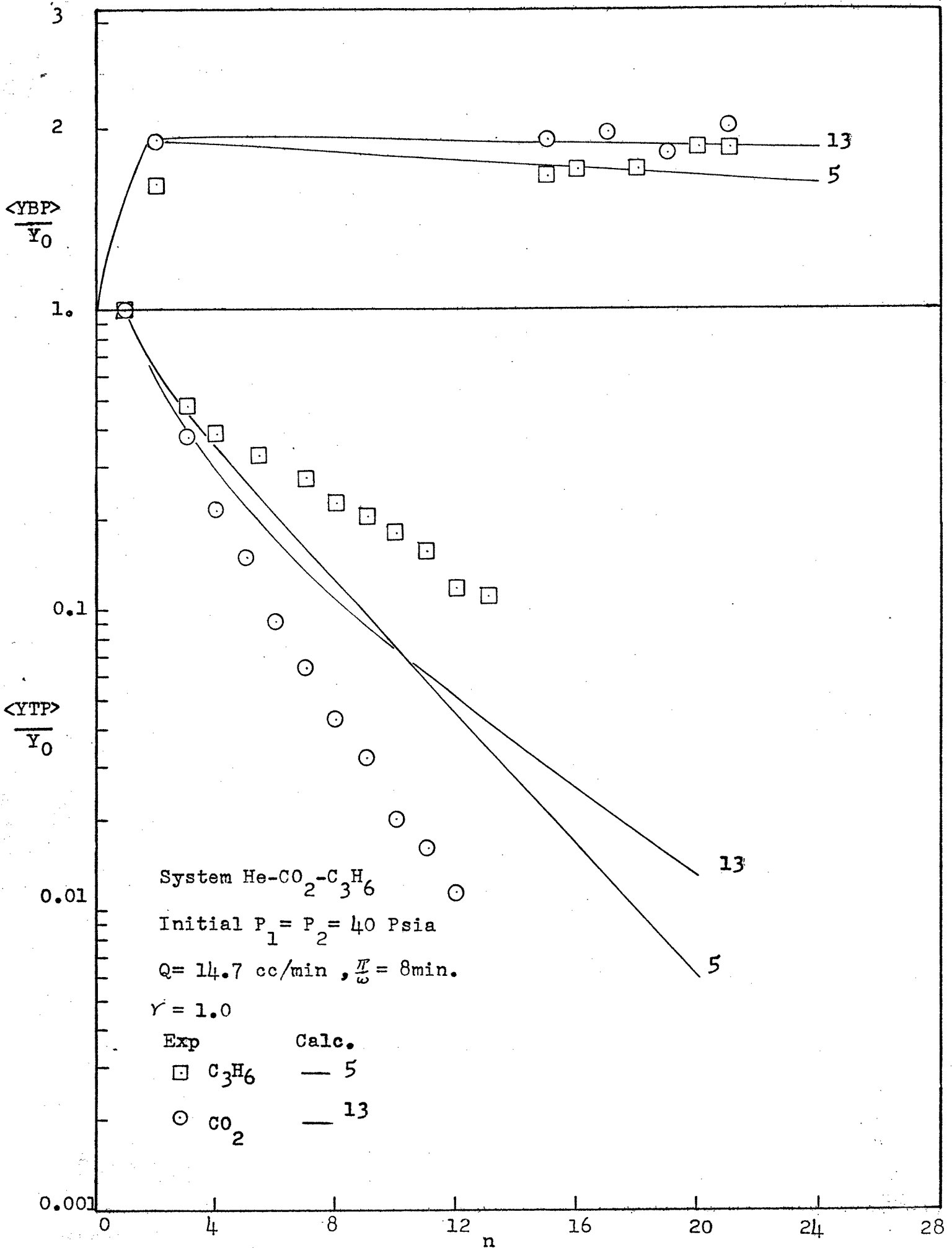


Figure 25



APPENDIX - V

Tables

Table no. 1

MATHEMATICAL RESULT - THERMAL PARAMETRIC PUMPING - SYSTEM : NaCl - H<sub>2</sub>O

| Run No. | $C_0$<br>(M) | $\frac{\pi}{\omega}$<br>(min.) | $Q(\frac{\pi}{\omega})$<br>(cc.) | $\phi_B$ | $\phi_B + \phi_T$ | $C_1 = V_T/Q$ | $C_2 = V_B/Q$ |
|---------|--------------|--------------------------------|----------------------------------|----------|-------------------|---------------|---------------|
| 1       | 0.1          | 35                             | 25                               | 0.04     | 0.4               | 5/25          | 5/25          |
| 2       | 0.1          | 35                             | 25                               | 0.16     | 0.4               | 5/25          | 5/25          |
| 3       | 0.1          | 25                             | 25                               | 0.04     | 0.4               | 5/25          | 5/25          |
| 4       | 0.1          | 60                             | 25                               | 0.04     | 0.4               | 5/25          | 5/25          |
| 5       | 0.1          | 100                            | 25                               | 0.04     | 0.4               | 5/25          | 5/25          |
| 6       | 0.1          | 35                             | 35                               | 0.04     | 0.4               | 5/35          | 5/35          |
| 7       | 0.1          | 100                            | 25                               | 0.16     | 0.4               | 5/25          | 5/25          |
| 8       | 0.1          | 35                             | 25                               | 0.01     | 0.4               | 5/25          | 5/25          |
| 9       | 0.1          | 60                             | 25                               | 0.01     | 0.4               | 5/25          | 5/25          |
| 10      | 0.1          | 60                             | 25                               | 0.16     | 0.4               | 5/25          | 5/25          |
| 11      | 0.1          | 100                            | 25                               | 0.01     | 0.4               | 5/25          | 5/25          |
| 12      | 0.05         | 35                             | 25                               | 0.04     | 0.4               | 5/25          | 5/25          |
| 13      | 0.5          | 35                             | 25                               | 0.04     | 0.4               | 5/25          | 5/25          |

Table No.2

## EXPERIMENTAL RESULTS - HEATLESS PARAMETRIC PUMP

| Run No. | Y <sub>0</sub>   | Initial Pressure (Psia) |                | Operating Pressure (Psia) |                | $\frac{\pi}{\omega}$<br>(min.) | Q<br>( $\frac{cc}{min.}$ ) | Y   | figure No. | System   |
|---------|--|-------------------------|----------------|---------------------------|----------------|--------------------------------|----------------------------|-----|------------|--|
|         |  | P <sub>1</sub>          | P <sub>2</sub> | P <sub>1</sub>            | P <sub>2</sub> |                                |                            |     |            |  |
| 1       | 1 % CO <sub>2</sub>                                      | -                       | -              | -                         | -              | --                             | --                         | --  | 16         | He - CO <sub>2</sub><br>(calibration)                                    |
| 2       | 1 % CO <sub>2</sub>                                      | 60                      | 20             | 60                        | 20             | 15                             | 10                         | 2.2 | 19         | He - CO <sub>2</sub>   |
| 3       | 1 % CO <sub>2</sub>                                      | -                       | -              | -                         | -              | --                             | --                         | --  | 1          | He - CO <sub>2</sub><br>(calibration)                                    |
| 4       | 1 % CO <sub>2</sub>                                      | 60                      | 20             | 60                        | 20             | 15                             | 10                         | 2.2 |            | He - CO <sub>2</sub>   |
| 5       | 1 % C <sub>3</sub> H <sub>6</sub>                        | -                       | -              | -                         | -              | -                              | -                          | -   |            | He - C <sub>3</sub> H <sub>6</sub><br>(calibration)                      |
| 6       | 1 % C <sub>3</sub> H <sub>6</sub>                        | 60                      | 20             | 60                        | 20             | 15                             | 10                         | 2.2 |            | He - C <sub>3</sub> H <sub>6</sub>                                       |
| 7       | 1 % C <sub>3</sub> H <sub>6</sub>                        | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2 |            | He - C <sub>3</sub> H <sub>6</sub>                                       |
| 8       | 1 % C <sub>3</sub> H <sub>6</sub>                        | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 1.5 |            | He - C <sub>3</sub> H <sub>6</sub>                                       |
| 9       | 1 % C <sub>3</sub> H <sub>6</sub>                        | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 1.5 |            | He - C <sub>3</sub> H <sub>6</sub>                                       |
| 10      | 1 % C <sub>3</sub> H <sub>6</sub>                        | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2 |            | He - C <sub>3</sub> H <sub>6</sub>                                       |
| 11      | 1 % C <sub>3</sub> H <sub>6</sub><br>1 % CO <sub>2</sub> | -                       | -              | -                         | -              | -                              | -                          | -   |            | He - CO <sub>2</sub> -<br>C <sub>3</sub> H <sub>6</sub><br>(calibration) |
| 12      | 1 % C <sub>3</sub> H <sub>6</sub><br>1 % CO <sub>2</sub> | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2 |            | He - CO <sub>2</sub> -<br>C <sub>3</sub> H <sub>6</sub>                  |
| 13      | 1 % C <sub>3</sub> H <sub>6</sub><br>1 % CO <sub>2</sub> | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2 |            | He - CO <sub>2</sub> -<br>C <sub>3</sub> H <sub>6</sub>                  |
| 14      | 1 % C <sub>3</sub> H <sub>6</sub><br>1 % CO <sub>2</sub> | 40                      | 40             | 40                        | 20             | 8                              | 14.7                       | 1.5 |            | He - CO <sub>2</sub> -<br>C <sub>3</sub> H <sub>6</sub>                  |
| 15      | 1 % C <sub>3</sub> H <sub>6</sub><br>1 % CO <sub>2</sub> | 40                      | 40             | 40                        | 20             | 8                              | 14.7                       | 1.0 |            | He - CO <sub>2</sub> -<br>C <sub>3</sub> H <sub>6</sub>                  |

Table no. 3

## MATHEMATICAL RESULTS - HEATLESS PARAMETRIC PUMP

| Run No. | Y <sub>0</sub>                    | Initial Pressure (Psia) |                | Operating Pressure (Psia) |                | $\frac{\pi}{\omega}$<br>(min.) | Q<br>( $\frac{cc}{min.}$ ) | $\gamma$ | $\alpha$ | NNZ | System                             |
|---------|-----------------------------------|-------------------------|----------------|---------------------------|----------------|--------------------------------|----------------------------|----------|----------|-----|------------------------------------|
|         |                                   | P <sub>1</sub>          | P <sub>2</sub> | P <sub>1</sub>            | P <sub>2</sub> |                                |                            |          |          |     |                                    |
| 1       | 1 % C <sub>3</sub> H <sub>6</sub> | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 1.5      | 1.25     | 600 | He - C <sub>3</sub> H <sub>6</sub> |
| 2       | 1 % C <sub>3</sub> H <sub>6</sub> | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 1.25     | 600 | He - C <sub>3</sub> H <sub>6</sub> |
| 3       | 1 % C <sub>3</sub> H <sub>6</sub> | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 0.25     | 100 | He - C <sub>3</sub> H <sub>6</sub> |
| 4       | 1 % C <sub>3</sub> H <sub>6</sub> | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 0.25     | 150 | He - C <sub>3</sub> H <sub>6</sub> |
| 5       | 1 % C <sub>3</sub> H <sub>6</sub> | 40                      | 40             | 40                        | 20             | 8                              | 14.7                       | 1.0      | 1.25     | 600 | He - C <sub>3</sub> H <sub>6</sub> |
| 6       | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 1.00     | 100 | He - CO <sub>2</sub>               |
| 7       | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 1.00     | 150 | He - CO <sub>2</sub>               |
| 8       | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 1.00     | 250 | He - CO <sub>2</sub>               |
| 9       | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 1.00     | 300 | He - CO <sub>2</sub>               |
| 10      | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 1.50     | 250 | He - CO <sub>2</sub>               |
| 11      | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 2.20     | 330 | He - CO <sub>2</sub>               |
| 12      | 1 % CO <sub>2</sub>               | 60                      | 60             | 60                        | 20             | 15                             | 10                         | 2.2      | 4.00     | 600 | He - CO <sub>2</sub>               |
| 13      | 1 % CO <sub>2</sub>               | 40                      | 40             | 40                        | 20             | 8                              | 14.7                       | 1.0      | 4.00     | 600 | He - CO <sub>2</sub>               |

Table no.- 4

|                                   |                            |
|-----------------------------------|----------------------------|
| Run No.- 1                        | System- He-CO <sub>2</sub> |
| Oven temp.: 0° C                  | Cell current: 185 mA.      |
| Pressure of carrier gas = 30 psig |                            |
| Detector temp.: 0° C              | Carrier gas: Helium        |
| Attenuation = 8                   |                            |

| Pressure absolute<br>psia | Peak height |
|---------------------------|-------------|
| 19.7                      | 3.08        |
| 24.7                      | 3.70        |
| 34.7                      | 5.20        |
| 39.7                      | 5.95        |
| 44.7                      | 6.63        |



Table no.- 5

Run No.- 3

System- He-CO<sub>2</sub>

Oven temp.: 110° C

Cell current: 185 mA.

Pressure of carrier gas = 30 psig

Detector temp.: 37° C

Carrier gas: Helium

Attenuation = 32 for CO<sub>2</sub>

Pressure absolute

Peak height

psia

14.7

2.9

19.7

3.55

24.7

4.15

29.7

4.65

Table no.- 6

Run No.- 2

System- He-CO<sub>2</sub>

Oven temp.: 0° C  
 Detector temp: 0° C  
 Feed concn.: 1 % CO<sub>2</sub>  
 Half cycle time: 15.0 mints.  
 Initial saturation pressure for:-  
 Column 1= 60. psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia= 10.0cc/min.  
 Purge @ 20.0 psia= 22.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 5.0 @ 8

Cell current: 185 mA.  
 Carrier gas : Helium  
 Pressure of Carrier gas: 30 psig  
 Purge ratio: 2.2  
 Column 2= 20.0 psia  
 Attenuation for CO<sub>2</sub>

| End of half cycle no. | Attn. | Peak height Top Product | Bottom Product | h/h <sub>0</sub> |
|-----------------------|-------|-------------------------|----------------|------------------|
| 0                     | 8     | 5.0                     | 5.0            | 1.00             |
| 1                     | 8     | 2.05                    | ---            | 0.41             |
| 2                     | 1     | 6.30                    | ---            | 0.1575           |
| 3                     | 1     | 3.10                    | ---            | 0.0775           |
| 4                     | 1     | 1.30                    | ---            | 0.0325           |
| 5                     | 1     | 0.70                    | ---            | 0.0175           |
| 6                     | -     | -----                   | ---            | -----            |
| 7                     | 16    | -----                   | 6.5            | 2.6000           |
| 8                     | 16    | -----                   | 2.8            | 1.1200           |
| 9                     | 16    | -----                   | 6.4            | 2.6000           |
| 10                    | 16    | -----                   | 2.8            | 1.1200           |

Table no.- 7

Run No.- 4

System- He-CO<sub>2</sub>

Oven temp.: 110° C  
 Detector temp: 37° C  
 Feed concn.: 1 % CO<sub>2</sub>  
 Half cycle time: 15.0 mints. Purge ratio: 2.2  
 Initial saturation pressure for:-  
 Column 1= 60.0 psia      Column 2= 20.0 psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia= 10.0 cc/min.  
 Purge @ 20.0 psia= 22.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 5.8 @ 16 Attenuation for CO<sub>2</sub>

| End of half cycle no. | Attn. | Peak height Top Product | Bottom Product | h/h <sub>0</sub> |
|-----------------------|-------|-------------------------|----------------|------------------|
| 0                     | 16    | 5.8                     | 5.8            | 1.00             |
| 1                     | 16    | 2.5                     | ---            | 0.431            |
| 2                     | 4     | 4.0                     | ---            | 0.1724           |
| 3                     | 2     | 4.1                     | ---            | 0.0884           |
| 4                     | 1     | 3.9                     | ---            | 0.0420           |
| 5                     | 1     | 1.9                     | ---            | 0.0205           |
| 6                     | 1     | 1.2                     | ---            | 0.0129           |
| 7                     | 1     | 0.8                     | ---            | 0.0086           |
| 8                     | 16    | ---                     | 6.3            | 1.0862           |
| 9                     | 32    | ---                     | 8.4            | 2.8960           |
| 10                    | 16    | ---                     | 6.9            | 1.1896           |
| 11                    | 32    | ---                     | 8.1            | 2.7460           |
| 12                    | 16    | ---                     | 7.5            | 1.2931           |

Table no.- 8

Run No.- 5

System-He-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C

Cell current: 185 mA.

Pressure of carrier gas = 30 psig

Detector temp.: 37° C

Carrier gas: Helium

Attenuation = 16 for C<sub>3</sub>H<sub>6</sub>

Pressure absolute

Peak height

psia

14.7

2.7

19.7

3.4

24.7

4.2

29.7

5.0

34.7

5.8

39.7

6.5

Table no.- 9

Run No.- 6

System-He-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C  
 Detector temp: 37° C  
 Feed concn.: 1 % C<sub>3</sub>H<sub>6</sub>  
 Half cycle time: 15.0 mints. Purge ratio: 2.2  
 Initial saturation pressure for:-  
 Column 1= 60. psia      Column 2= 20.0 psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia= 10.0 cc/min.  
 Purge @ 20.0 psia= 22.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 2.5 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

| End of half cycle no. | Attn. | Peak height Top Product | Bottom Product | h/h <sub>0</sub> |
|-----------------------|-------|-------------------------|----------------|------------------|
| 0                     | 16    | 2.5                     | 2.5            | 1.0              |
| 1                     | 8     | 2.95                    | ---            | 0.5673           |
| 2                     | 8     | 1.50                    | ---            | 0.2885           |
| 3                     | 2     | 2.60                    | ---            | 0.1250           |
| 4                     | 2     | 1.80                    | ---            | 0.0865           |
| 5                     | 16    | ----                    | 7.7            | 2.9615           |
| 6                     | 16    | ----                    | 4.05           | 1.5570           |
| 7                     | --    | ----                    | ---            | -----            |
| 8                     | 2     | 1.20                    | ---            | 0.0577           |
| 9                     | 16    | ----                    | 7.5            | 2.8800           |
| 10                    | 16    | ----                    | 3.9            | 1.5000           |
| 11                    | 16    | ----                    | 7.6            | 2.9230           |
| 12                    | 16    | ----                    | 4.2            | 1.6150           |

Table no.- 10

Run No.- 7

System- He-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110°C  
 Detector temp: 37°C  
 Feed concn.: 1 % C<sub>3</sub>H<sub>6</sub>  
 Half cycle time: 15.0 mints.  
 Initial saturation pressure for:-  
 Column 1= 60.0 psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia= 10.0 cc/min.  
 Purge @ 20.0 psia= 22.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 2.6 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

Cell current: 185 mA.  
 Carrier gas : Helium  
 Pressure of Carrier gas: 30 psig  
 Purge ratio: 2.2  
 Column 2= 60.0 psia

| End of half<br>cycle no. | Attn. | Peak height<br>Top<br>Product | Bottom<br>Product | h/h <sub>0</sub> |
|--------------------------|-------|-------------------------------|-------------------|------------------|
| 0                        | 16    | 2.6                           | 2.6               | 1.00             |
| 1                        | 16    | 2.6                           | ---               | 1.00             |
| 2                        | --    | ---                           | ---               | ---              |
| 3                        | 8     | 2.4                           | ---               | 0.4615           |
| 4                        | 8     | 1.1                           | ---               | 0.2115           |
| 5                        | 2     | 2.7                           | ---               | 0.1298           |
| 6                        | 16    | ---                           | 8.2               | 3.1538           |
| 7                        | 16    | ---                           | 8.3               | 3.1923           |
| 8                        | --    | ---                           | ---               | ---              |
| 9                        | --    | ---                           | ---               | ---              |
| 10                       | 2     | 1.8                           | ---               | 0.0865           |
| 11                       | 2     | 1.1                           | ---               | 0.0529           |
| 12                       | 1     | 1.8                           | ---               | 0.0433           |
| 13                       | 1     | 1.25                          | ---               | 0.0301           |
| 14                       | 1     | 1.25                          | ---               | 0.0301           |
| 15                       | 1     | 1.00                          | ---               | 0.0264           |
| 16                       | 16    | ---                           | 8.2               | 3.1538           |
| 17                       | 16    | ---                           | 7.9               | 3.0380           |
| 18                       | --    | ---                           | ---               | ---              |
| 19                       | 16    | ---                           | 7.9               | 3.0380           |
| 20                       | --    | ---                           | ---               | ---              |
| 21                       | 16    | ---                           | 8.0               | 3.0770           |

Table no.- 11

Run No.- 8

System- He-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C  
 Detector temp: 37° C  
 Feed concn.: 1 % C<sub>3</sub>H<sub>6</sub>  
 Half cycle time: 15.0 mints.  
 Initial saturation pressure for:-  
 Column 1= 60.0 psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia= 10.0 cc/min.  
 Purge @ 20.0 psia= 15.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 2.6 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

Cell current: 185 mA.  
 Carrier gas : Helium  
 Pressure of Carrier gas: 30 psig  
 Purge ratio: 1.5  
 Column 2= 60.0 psia

| End of half cycle no. | Attn. | Peak height Top Product | Bottom Product | h/h <sub>0</sub> |
|-----------------------|-------|-------------------------|----------------|------------------|
| 0                     | 16    | 2.6                     | 2.6            | 1.00             |
| 1                     | 16    | 2.6                     | ---            | 1.00             |
| 2                     | 16    | 1.2                     | ---            | 0.4615           |
| 3                     | 8     | 1.1                     | ---            | 0.2115           |
| 4                     | 4     | 1.3                     | ---            | 0.1250           |
| 5                     | 2     | 1.55                    | ---            | 0.0745           |
| 6                     | 2     | 1.00                    | ---            | 0.04801          |
| 7                     | --    | ----                    | ----           | -----            |
| 8                     | -     | -----                   | -----          | -----            |
| 9                     | 2     | 0.75                    | ---            | 0.0306           |
| 10                    | 16    | ---                     | 7.3            | 2.8070           |
| 11                    | 16    | ---                     | 7.5            | 2.8850           |
| 12                    | ---   | -----                   | ---            | -----            |
| 13                    | ---   | -----                   | ---            | -----            |
| 14                    | 16    | ---                     | 7.1            | 2.7310           |
| 15                    | 16    | ---                     | 7.4            | 2.8460           |

Table no.-12

Run No.- 9

System-He-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110°C  
 Detector temp: 37°C  
 Feed concn.: 1 % C<sub>3</sub>H<sub>6</sub>  
 Half cycle time: 15.0 mints.  
 Initial saturation pressure for:-  
 Column 1= 60.0 psia  
 Volumetric flow rate of:-  
 Feed @ 60 psia= 10.0 cc/min.  
 Purge @ 20.0 psia= 15.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 2.6 @ 16  
 Cell current: 185 mA.  
 Carrier gas : Helium  
 Pressure of Carrier gas: 30 psig  
 Purge ratio: 1.5  
 Column 2= 60.0 psia  
 Attenuation for C<sub>3</sub>H<sub>6</sub>

| End of half cycle no. | Attn. | Peak height Top Product | Bottom Product | h/h <sub>0</sub> |
|-----------------------|-------|-------------------------|----------------|------------------|
| 0                     | 16    | 2.6                     | 2.6            | 1.00             |
| 1                     | 16    | ---                     | 7.2            | 2.769            |
| 2                     | 16    | 1.2                     | ---            | 0.4615           |
| 3                     | 8     | 1.1                     | ---            | 0.2115           |
| 4                     | 4     | 1.25                    | ---            | 0.1202           |
| 5                     | 2     | 1.40                    | ---            | 0.0673           |
| 6                     | 2     | 0.80                    | ---            | 0.0385           |
| 7                     | 1     | 1.20                    | ---            | 0.0289           |
| 8                     | 1     | 0.85                    | ---            | 0.0204           |
| 9                     | 1     | 0.80                    | ---            | 0.0192           |
| 10                    | 1     | 0.65                    | ---            | 0.0156           |
| 11                    | 1     | 0.55                    | ---            | 0.0132           |
| 12                    | 1     | 0.50                    | ---            | 0.0120           |
| 13                    | 1     | 0.48                    | ---            | 0.0115           |
| 14                    | 1     | 0.45                    | ---            | 0.0108           |
| 15                    | 1     | 0.45                    | ---            | 0.0108           |
| 16                    | 1     | 0.40                    | ---            | 0.0096           |
| 17                    | 16    | ---                     | 7.5            | 2.8850           |
| 18                    | 16    | ---                     | 7.8            | 3.0000           |



Table no.- 13

Run No.- 10

System-He-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C      Cell current: 185 mA.  
 Detector temp: 37° C      Carrier gas : Helium  
 Feed concn.: 1 % C<sub>3</sub>H<sub>6</sub>      Pressure of Carrier gas: 30 psig  
 Half cycle time: 15.0 mints. Purge ratio: 2.2  
 Initial saturation pressure for:-  
 Column 1= 60 psia      Column 2= 60 psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia= 10.0cc/min.  
 Purge @ 20.0 psia= 22.0 cc/min.  
 Feed peak height, h<sub>0</sub>= 2.6 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

| End of half<br>cycle no. | Attn. | Peak height<br>Top<br>Product | Bottom<br>Product | h/h <sub>0</sub> |
|--------------------------|-------|-------------------------------|-------------------|------------------|
| 0                        | 16    | 2.6                           | 2.6               | 1.0              |
| 1                        | 16    | ---                           | 7.4               | 2.846            |
| 2                        | ---   | ---                           | ---               | -----            |
| 3                        | 16    | 0.8                           | ---               | 0.3077           |
| 4                        | 4     | 1.5                           | ---               | 0.1442           |
| 5                        | 2     | 2.0                           | ---               | 0.0962           |
| 6                        | 1     | 2.5                           | ---               | 0.0601           |
| 7                        | 1     | 1.9                           | ---               | 0.0457           |
| 8                        | 1     | 1.5                           | ---               | 0.0361           |
| 9                        | 1     | 1.3                           | ---               | 0.0313           |
| 10                       | 1     | 1.05                          | ---               | 0.0252           |
| 11                       | 1     | 0.85                          | ---               | 0.0204           |
| 12                       | 1     | 0.80                          | ---               | 0.0192           |
| 13                       | 1     | 0.70                          | ---               | 0.0168           |
| 14                       | 1     | 0.65                          | ---               | 0.0156           |
| 15                       | 1     | 0.60                          | ---               | 0.0144           |
| 16                       | 1     | 0.58                          | ---               | 0.0139           |
| 17                       | 1     | 0.54                          | ---               | 0.0130           |
| 18                       | 1     | 0.55                          | ---               | 0.0132           |

Table no.- 14

Run No.- 11

System- He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110°C

Cell current: 185 mA.

Pressure of carrier gas = 30 psig

Detector temp.: 37°C

Carrier gas: Helium

Attenuation = 16

| Pressure absolute<br>psia | Peak height     |                               |
|---------------------------|-----------------|-------------------------------|
|                           | CO <sub>2</sub> | C <sub>3</sub> H <sub>6</sub> |
| 14.7                      | 4.8             | 2.1                           |
| 19.7                      | 5.8             | 2.7                           |
| 24.7                      | 7.1             | 3.4                           |
| 29.7                      | 8.1             | 4.0                           |
| 34.7                      | 8.9             | 4.7                           |

Table no.- 15

Run No.-12

System-He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C                      Cell current: 185 mA.  
 Detector temp.: 37° C                    Carrier gas : Helium  
 Half cycle time: 15.0 mints. Pressure of carrier gas: 30 psig  
 Feed concn.: 1% CO<sub>2</sub>, 1% C<sub>3</sub>H<sub>6</sub>            Purge ratio: 2.2  
 Initial saturation pressure for -  
 Column 1 = 60.0 psia                      Column 2 = 60.0 psia  
 Volumetric flow rate of:-  
 Feed @ 60.0 psia = 10.0 cc/min.  
 Purge @ 20.0 psia = 22.0 cc/min.  
 Feed peak height, h<sub>0</sub> = 4.7 @ 16 Attenuation for CO<sub>2</sub>

= 2.0 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

| End of half<br>cycle no. | Attn. | Peak height<br>Top<br>Product |                               | Bottom<br>Product |                               | h/h <sub>0</sub> |                               |
|--------------------------|-------|-------------------------------|-------------------------------|-------------------|-------------------------------|------------------|-------------------------------|
|                          |       | CO <sub>2</sub>               | C <sub>3</sub> H <sub>6</sub> | CO <sub>2</sub>   | C <sub>3</sub> H <sub>6</sub> | CO <sub>2</sub>  | C <sub>3</sub> H <sub>6</sub> |
| 0                        | 16    | 4.7                           | 2.0                           | 4.7               | 2.0                           | 1.0              | 1.0                           |
| 1                        | 16    | 4.7                           | 2.0                           | ---               | ---                           | 1.0              | 1.0                           |
| 2                        | 16    | 2.0                           | 0.9                           | ---               | ---                           | 0.4255           | 0.450                         |
| 3                        | 4     | 3.6                           | 2.0                           | ---               | ---                           | 0.1915           | 0.250                         |
| 4                        | 4     | 1.5                           | 1.10                          | ---               | ---                           | 0.0798           | 0.1375                        |
| 5                        | 2     | 1.6                           | 1.5                           | ---               | ---                           | 0.0426           | 0.0938                        |
| 6                        | 2     | 0.7                           | 0.9                           | ---               | ---                           | 0.0186           | 0.0563                        |
| 7                        | 1     | 0.8                           | 1.2                           | ---               | ---                           | 0.0106           | 0.0375                        |
| 8                        | 1     | 0.4                           | 0.6                           | ---               | ---                           | 0.0053           | 0.0186                        |
| 9                        | 1     | 0.4                           | 0.6                           | ---               | ---                           | 0.0053           | 0.0186                        |
| 10                       | 32    | ---                           | ---                           | 6.3               | 3.1                           | 2.681            | 3.100                         |
| 11                       | 32    | ---                           | ---                           | 5.3               | 3.0                           | 2.255            | 3.000                         |
| 12                       | 32    | ---                           | ---                           | 4.9               | 3.0                           | 2.085            | 3.000                         |
| 13                       | 32    | ---                           | ---                           | 3.6               | 2.9                           | 1.532            | 2.900                         |
| 14                       | 32    | ---                           | ---                           | 2.8               | 2.8                           | 1.122            | 2.800                         |
| 15                       | 32    | ---                           | ---                           | 2.7               | 2.9                           | 1.119            | 2.900                         |
| 16                       | 32    | ---                           | ---                           | 1.6               | 2.8                           | 0.681            | 2.800                         |
| 17                       | 32    | ---                           | ---                           | 1.5               | 2.8                           | 0.638            | 2.800                         |
| 18                       | 32    | ---                           | ---                           | 1.3               | 2.8                           | 0.565            | 2.800                         |



Table no.- 17

Run No.-14

System-He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C                      Cell current: 185 mA.  
 Detector temp.: 37° C                    Carrier gas : Helium  
 Half cycle time: 8.0 mints. Pressure of carrier gas: 30 psig  
 Feed concn.: 1% CO<sub>2</sub>, 1% C<sub>3</sub>H<sub>6</sub>            Purge ratio: 1.5  
 Initial saturation pressure for -  
 Column 1 = 40.0 psia                      Column 2 = 40.0 psia  
 Volumetric flow rate of:-  
 Feed @ 40.0 psia = 14.7 cc/min.  
 Purge @ 20.0 psia = 22.0 cc/min.  
 Feed peak height, h<sub>0</sub> = 4.4 @ 16 Attenuation for CO<sub>2</sub>

= 2.05 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

| End of half<br>cycle no. | Attn. | Peak height<br>Top<br>Product |                               | Bottom<br>Product |                               | h/h <sub>0</sub> |                               |
|--------------------------|-------|-------------------------------|-------------------------------|-------------------|-------------------------------|------------------|-------------------------------|
|                          |       | CO <sub>2</sub>               | C <sub>3</sub> H <sub>6</sub> | CO <sub>2</sub>   | C <sub>3</sub> H <sub>6</sub> | CO <sub>2</sub>  | C <sub>3</sub> H <sub>6</sub> |
| 0                        | 16    | 4.4                           | 2.05                          | ---               | ---                           | 1.0              | 1.0                           |
| 1                        | 16    | 4.4                           | 2.05                          | ---               | ---                           | 1.0              | 1.0                           |
| 2                        | 16    | 4.4                           | 2.05                          | ---               | ---                           | 1.0              | 1.0                           |
| 3                        | 16    | 2.7                           | 1.30                          | ---               | ---                           | 0.6136           | 0.634                         |
| 4                        | 16    | 1.7                           | 1.05                          | ---               | ---                           | 0.3864           | 0.5122                        |
| 5                        | 16    | 1.1                           | 0.84                          | ---               | ---                           | 0.2500           | 0.4097                        |
| 6                        | 4     | 3.3                           | 2.60                          | ---               | ---                           | 0.1875           | 0.3170                        |
| 7                        | 4     | 2.0                           | 1.90                          | ---               | ---                           | 0.1136           | 0.2317                        |
| 8                        | 4     | 1.3                           | 1.8                           | ---               | ---                           | 0.0739           | 0.2195                        |
| 9                        | 2     | 2.2                           | 3.60                          | ---               | ---                           | 0.0625           | 0.2195                        |
| 10                       | 2     | 1.7                           | 2.30                          | ---               | ---                           | 0.0483           | 0.1402                        |
| 11                       | 2     | 1.2                           | 4.70                          | ---               | ---                           | 0.0341           | 0.2866                        |
| 12                       | 2     | 1.0                           | 3.80                          | ---               | ---                           | 0.0284           | 0.2317                        |
| 13                       | 1     | 1.7                           | 7.30                          | ---               | ---                           | 0.0240           | 0.2226                        |
| 14                       | 1     | 1.3                           | 5.70                          | ---               | ---                           | 0.0195           | 0.2738                        |
| 15                       | 32    | ---                           | ---                           | 4.2               | 1.7                           | 1.9100           | 1.6550                        |
| 16                       | 32    | ---                           | ---                           | 4.2               | 1.75                          | 1.9100           | 1.7070                        |
| 17                       | 32    | ---                           | ---                           | 4.35              | 1.75                          | 1.9770           | 1.7070                        |
| 18                       | 32    | ---                           | ---                           | 4.30              | 1.75                          | 1.9540           | 1.7070                        |
| 19                       | 32    | ---                           | ---                           | 4.00              | 1.30                          | 1.8180           | 1.2680                        |
| 20                       | 32    | ---                           | ---                           | 3.90              | 1.90                          | 1.7730           | 1.8540                        |
| 21                       | 32    | ---                           | ---                           | 4.40              | 1.90                          | 2.0000           | 1.8540                        |

Table no.-18

Run No.- 15

System-He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub>

Oven temp.: 110° C      Cell current: 185 mA.  
 Detector temp.: 37° C      Carrier gas : Helium  
 Half cycle time: 8.0 mints. Pressure of carrier gas: 30 psig  
 Feed concn.: 1% CO<sub>2</sub>, 1% C<sub>3</sub>H<sub>6</sub>      Purge ratio: 1.0  
 Initial saturation pressure for -  
 Column 1 = 40.0 psia      Column 2 = 40.0 psia  
 Volumetric flow rate of:-  
 Feed @ 40.0 psia = 14.7 cc/min.  
 Purge @ 20.0 psia = 22.0 cc/min.  
 Feed peak height, h<sub>0</sub> = 4.7 @ 16 Attenuation for CO<sub>2</sub>

= 2.1 @ 16 Attenuation for C<sub>3</sub>H<sub>6</sub>

| End of half<br>cycle no. | Attn. | Peak height<br>Top<br>Product |                               | Bottom<br>Product |                               | h/h <sub>0</sub> |                               |
|--------------------------|-------|-------------------------------|-------------------------------|-------------------|-------------------------------|------------------|-------------------------------|
|                          |       | CO <sub>2</sub>               | C <sub>3</sub> H <sub>6</sub> | CO <sub>2</sub>   | C <sub>3</sub> H <sub>6</sub> | CO <sub>2</sub>  | C <sub>3</sub> H <sub>6</sub> |
| 0                        | 16    | 4.7                           | 2.1                           | ---               | ---                           | 1.0              | 1.0                           |
| 1                        | 32    | ---                           | ---                           | 3.1               | 1.5                           | 1.319            | 1.428                         |
| 2                        | 32    | ---                           | ---                           | 4.1               | 1.9                           | 1.745            | 1.810                         |
| 3                        | 16    | 1.7                           | 1.0                           | ---               | ---                           | 0.362            | 0.476                         |
| 4                        | 16    | 1.0                           | 0.8                           | ---               | ---                           | 0.213            | 0.381                         |
| 5                        | 4     | 2.8                           | 2.9                           | ---               | ---                           | 0.119            | 0.321                         |
| 6                        | 4     | 1.7                           | 2.3                           | ---               | ---                           | 0.090            | 0.271                         |
| 7                        | 4     | 1.2                           | 1.9                           | ---               | ---                           | 0.064            | 0.226                         |
| 8                        | 2     | 1.6                           | 3.4                           | ---               | ---                           | 0.043            | 0.202                         |
| 9                        | 2     | 1.2                           | 3.0                           | ---               | ---                           | 0.031            | 0.179                         |
| 10                       | 1     | 1.5                           | 5.2                           | ---               | ---                           | 0.019            | 0.159                         |
| 11                       | 1     | 1.2                           | 3.9                           | ---               | ---                           | 0.015            | 0.116                         |
| 12                       | 1     | 0.9                           | 3.7                           | ---               | ---                           | 0.012            | 0.110                         |
| 13                       | 32    | ---                           | ---                           | 3.2               | 1.5                           | 1.715            | 1.710                         |
| 14                       | 32    | ---                           | ---                           | 4.4               | 1.8                           | 1.872            | 1.990                         |
| 15                       | 32    | ---                           | ---                           | 4.4               | 2.0                           | 1.872            | 1.990                         |
| 16                       | 32    | ---                           | ---                           | 4.5               | 2.2                           | 1.915            | 2.090                         |
| 17                       | 32    | ---                           | ---                           | 4.4               | 2.0                           | 1.872            | 1.990                         |
| 18                       | 32    | ---                           | ---                           | 4.4               | 2.1                           | 1.872            | 2.000                         |
| 19                       | 32    | ---                           | ---                           | 4.05              | 1.85                          | 1.723            | 1.752                         |
| 20                       | 32    | ---                           | ---                           | 4.1               | 1.9                           | 1.715            | 1.810                         |
| 21                       | 32    | ---                           | ---                           | 4.2               | 1.9                           | 1.715            | 1.810                         |
| 22                       | 32    | ---                           | ---                           | 4.1               | 1.9                           | 1.715            | 1.810                         |

Table No: 19

Run no: 16

System: He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub>

Saturation pressure - 20 psia

Detector temp. - 37° C

Oven temp. - 110° C

Cell current - 185 mA

Carrier gas - Helium

Attenuation - 16

| Time<br>hr - min. | Peak height     |                               |
|-------------------|-----------------|-------------------------------|
|                   | CO <sub>2</sub> | C <sub>3</sub> H <sub>6</sub> |
| 0 - 15            | 1.8             | 0.65                          |
| 0 - 25            | 3.0             | 1.05                          |
| 0 - 50            | 4.8             | 1.20                          |
| 1 - 15            | 4.8             | 1.20                          |
| 1 - 30            | 4.8             | 1.20                          |
| 2 - 05            | 4.8             | 1.20                          |
| 2 - 15            | 4.8             | 1.50                          |
| 2 - 30            | 4.8             | 1.50                          |
| 3 - 00            | 4.8             | 1.50                          |
| 3 - 30            | 4.8             | 1.50                          |
| 4 - 00            | 4.8             | 1.50                          |
| 4 - 30            | 4.8             | 1.50                          |
| 5 - 00            | 4.8             | 2.10                          |
| 5 - 30            | 4.8             | 2.10                          |
| 6 - 00            | 4.8             | 2.10                          |

Table No: 20

Run no: 17

System: He-CO<sub>2</sub>-C<sub>3</sub>H<sub>6</sub>

Saturation pressure - 60 psia

Detector temp. - 37°C

Oven temp. - 110°C

Cell Current - 185 mA

Carrier gas - Helium

Attenuation - 16

| Time<br>hr - min. | Peak height     |                               |
|-------------------|-----------------|-------------------------------|
|                   | CO <sub>2</sub> | C <sub>3</sub> H <sub>6</sub> |
| 0 - 15            | 1.0             | 0.15                          |
| 0 - 25            | 1.7             | 0.25                          |
| 0 - 45            | 1.7             | 0.30                          |
| 0 - 55            | 1.7             | 0.30                          |
| 1 - 10            | 3.05            | 0.30                          |
| 1 - 30            | 4.8             | 0.35                          |
| 2 - 00            | 4.8             | 0.35                          |
| 2 - 30            | 4.8             | 0.35                          |
| 3 - 00            | 4.8             | 0.35                          |
| 3 - 30            | 4.8             | 0.35                          |
| 4 - 00            | 4.8             | 0.35                          |
| 4 - 30            | 4.8             | 0.35                          |
| 5 - 00            | 4.8             | 0.35                          |
| 5 - 30            | 4.8             | 0.35                          |
| 6 - 00            | 4.8             | 0.75                          |
| 6 - 30            | 4.8             | 1.15                          |
| 7 - 00            | 4.8             | 1.60                          |
| 7 - 30            | 4.8             | 2.10                          |
| 8 - 00            | 4.8             | 2.10                          |



Table No:21

THERMAL PARAMETRIC PUMP

- Physical parameters for use in computations -

Column diameter = 1.0 cm

Column height = 90.0 cm.

Column cross-sectional area = 0.785 cm<sup>2</sup>

Packing - Bio-Rad AG 11 AS-Resin

Density for silica gel '  $\rho_s$  ' = 0.761 gm/cm<sup>3</sup>

Bed voidage '  $\epsilon$  ' = 0.38

bed height = 90.0 cms.

Top reservoir dead volume '  $V_T$  ' = 5.0 cc.

bottom reservoir dead volume '  $V_B$  ' = 5.0 cc.

Operating temperature -

Hot temperature = 70.0°C

Cold temperature = 5.0°C

Equilibrium constants for system NaCl-H<sub>2</sub>O-

An expression given by Swwed (1971) has been used:

$$M_T = a_m - b_m \cdot T, \text{ where } a_m = 4.6347$$
$$b_m = 0.00993$$

Mass transfer coefficient for NaCl-H<sub>2</sub>O system-

Following expression has been used as given by Sweed

(1971) -

$$\lambda = \alpha \cdot v^{1-\beta}$$

where,  $\beta = 0.7$

$$\alpha = 0.328 \text{ @ } 70^\circ\text{C}$$

$$\alpha = 0.0736 \text{ @ } 5^\circ\text{C}$$

Table No: 22

HEATLESS PARAMETRIC PUMP

- Physical parameters for use in computations -

Column diameter = 3.175 cm.

column diameter = 100.0 cm.

Column cross-sectional area = 7.913 cm<sup>2</sup>

Packing - Silica gel type

density of silica gel '  $\rho_s$  ' = 0.73 gm./cm<sup>3</sup>

Particle diameter for 12-28 taylor mesh size = 0.1 cm

bed voidage '  $\epsilon$  ' = 0.42

bed height = 100 cm.

A constant value of equilibrium coefficient '  $K_p$  ' calculated at 60 psia has been used for both systems.

| System                             | $K_p$ |
|------------------------------------|-------|
| He - CO <sub>2</sub>               | 52.7  |
| He - C <sub>3</sub> H <sub>6</sub> | 250.0 |

Operating temperature = 25.0°C

Mas transfer coefficient calculated from equation (C)-

Appendix - I

| System                           | flow rate<br>(cc/min) |      | (min <sup>-1</sup> ) @ Pressure |     |         |         |         |
|----------------------------------|-----------------------|------|---------------------------------|-----|---------|---------|---------|
|                                  |                       |      | 60                              | 40  | 20      |         |         |
| He-CO <sub>2</sub>               | 10.0                  | 4.0  | -0.51                           | 2.2 | 0.49465 | --      | 1.2746  |
| He-CO <sub>2</sub>               | 14.7                  | 4.0  | -0.51                           | 1.0 | --      | 0.72717 | 1.0461  |
| He-C <sub>3</sub> H <sub>6</sub> | 10.0                  | 1.25 | -0.51                           | 2.2 | 0.13001 | --      | 0.33515 |
| He-C <sub>3</sub> H <sub>6</sub> | 10.0                  | 1.25 | -0.51                           | 1.5 | 0.13001 | --      | 0.27780 |
| He-C <sub>3</sub> H <sub>6</sub> | 14.7                  | 1.25 | -0.51                           | 1.0 | --      | 0.19304 | 0.27507 |

Table : 23

$$\phi_B = 0.01, \frac{T}{\omega} = 60 \text{ mins.}, c_0 = 0.1 \text{ M}$$

| M  | IND         | TT          |     |
|----|-------------|-------------|-----|
| 1  | 2           | 60,851      |     |
| I  | YY          | XX          | NTR |
| 1  | 0.10000E-03 | 0.18271E-03 | 1   |
| 2  | 0.99735E-04 | 0.18150E-03 | 2   |
| 3  | 0.99441E-04 | 0.18026E-03 | 2   |
| 4  | 0.99108E-04 | 0.17899E-03 | 2   |
| 5  | 0.98742E-04 | 0.17770E-03 | 2   |
| 6  | 0.98345E-04 | 0.17640E-03 | 2   |
| 7  | 0.97919E-04 | 0.17508E-03 | 2   |
| 8  | 0.97467E-04 | 0.17375E-03 | 2   |
| 9  | 0.96990E-04 | 0.17242E-03 | 2   |
| 10 | 0.96491E-04 | 0.17109E-03 | 2   |
| 11 | 0.95972E-04 | 0.16976E-03 | 2   |
| 12 | 0.95434E-04 | 0.16843E-03 | 1   |
| 13 | 0.94880E-04 | 0.16711E-03 | 1   |
| 14 | 0.94313E-04 | 0.16580E-03 | 1   |
| 15 | 0.93733E-04 | 0.16450E-03 | 1   |
| 16 | 0.93143E-04 | 0.16322E-03 | 1   |
| 17 | 0.92544E-04 | 0.16195E-03 | 1   |
| 18 | 0.91935E-04 | 0.16070E-03 | 1   |
| 19 | 0.91328E-04 | 0.15948E-03 | 1   |
| 20 | 0.90713E-04 | 0.15827E-03 | 1   |
| 21 | 0.90097E-04 | 0.15708E-03 | 1   |
| 22 | 0.89479E-04 | 0.15592E-03 | 1   |
| 23 | 0.88862E-04 | 0.15478E-03 | 1   |
| 24 | 0.88246E-04 | 0.15367E-03 | 1   |
| 25 | 0.87633E-04 | 0.15258E-03 | 1   |
| 26 | 0.87024E-04 | 0.15152E-03 | 1   |
| 27 | 0.86415E-04 | 0.15049E-03 | 1   |
| 28 | 0.85820E-04 | 0.14948E-03 | 1   |
| 29 | 0.85228E-04 | 0.14850E-03 | 1   |
| 30 | 0.84643E-04 | 0.14755E-03 | 1   |
| 31 | 0.84065E-04 | 0.14662E-03 | 1   |
| 32 | 0.83496E-04 | 0.14572E-03 | 1   |
| 33 | 0.82937E-04 | 0.14485E-03 | 1   |
| 34 | 0.82387E-04 | 0.14400E-03 | 1   |
| 35 | 0.81846E-04 | 0.14318E-03 | 1   |
| 36 | 0.81317E-04 | 0.14239E-03 | 1   |
| 37 | 0.80798E-04 | 0.14162E-03 | 1   |
| 38 | 0.80290E-04 | 0.14088E-03 | 1   |
| 39 | 0.79793E-04 | 0.14017E-03 | 1   |
| 40 | 0.79308E-04 | 0.13948E-03 | 1   |
| 41 | 0.78834E-04 | 0.13881E-03 | 1   |
| 42 | 0.78372E-04 | 0.13817E-03 | 1   |
| 43 | 0.77922E-04 | 0.13755E-03 | 1   |
| 44 | 0.77484E-04 | 0.13695E-03 | 1   |
| 45 | 0.77058E-04 | 0.13637E-03 | 1   |
| 46 | 0.76643E-04 | 0.13582E-03 | 1   |
| 47 | 0.76241E-04 | 0.13529E-03 | 1   |
| 48 | 0.75850E-04 | 0.13478E-03 | 1   |
| 49 | 0.75471E-04 | 0.13429E-03 | 1   |
| 50 | 0.75103E-04 | 0.13382E-03 | 1   |
| 51 | 0.74747E-04 | 0.13337E-03 | 1   |
| 52 | 0.74402E-04 | 0.13293E-03 | 1   |
| 53 | 0.74069E-04 | 0.13252E-03 | 1   |
| 54 | 0.73746E-04 | 0.13212E-03 | 1   |
| 55 | 0.73434E-04 | 0.13174E-03 | 1   |
| 56 | 0.73133E-04 | 0.13137E-03 | 1   |
| 57 | 0.72843E-04 | 0.13102E-03 | 1   |
| 58 | 0.72562E-04 | 0.13069E-03 | 1   |
| 59 | 0.72292E-04 | 0.13037E-03 | 1   |
| 60 | 0.72031E-04 | 0.13006E-03 | 1   |
| 61 | 0.71780E-04 | 0.12977E-03 | 1   |
| 62 | 0.71538E-04 | 0.12949E-03 | 1   |
| 63 | 0.71305E-04 | 0.12922E-03 | 1   |
| 64 | 0.71081E-04 | 0.12897E-03 | 1   |
| 65 | 0.70866E-04 | 0.12872E-03 | 1   |
| 66 | 0.70659E-04 | 0.12849E-03 | 1   |

M  
2

IND  
1

TY  
60.077

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.73255E-04 | 0.89893E-04 | 1   |
| 2  | 0.73255E-04 | 0.89893E-04 | 1   |
| 3  | 0.73255E-04 | 0.89893E-04 | 1   |
| 4  | 0.73255E-04 | 0.89894E-04 | 1   |
| 5  | 0.73256E-04 | 0.89895E-04 | 1   |
| 6  | 0.73256E-04 | 0.89897E-04 | 1   |
| 7  | 0.73256E-04 | 0.89902E-04 | 1   |
| 8  | 0.73261E-04 | 0.89910E-04 | 1   |
| 9  | 0.73266E-04 | 0.89923E-04 | 1   |
| 10 | 0.73275E-04 | 0.89945E-04 | 1   |
| 11 | 0.73288E-04 | 0.89977E-04 | 1   |
| 12 | 0.73305E-04 | 0.90023E-04 | 1   |
| 13 | 0.73339E-04 | 0.90087E-04 | 1   |
| 14 | 0.73379E-04 | 0.90174E-04 | 2   |
| 15 | 0.73434E-04 | 0.90286E-04 | 2   |
| 16 | 0.73507E-04 | 0.90430E-04 | 2   |
| 17 | 0.73601E-04 | 0.90610E-04 | 2   |
| 18 | 0.73721E-04 | 0.90832E-04 | 2   |
| 19 | 0.73870E-04 | 0.91099E-04 | 2   |
| 20 | 0.74052E-04 | 0.91418E-04 | 2   |
| 21 | 0.74270E-04 | 0.91791E-04 | 2   |
| 22 | 0.74528E-04 | 0.92222E-04 | 2   |
| 23 | 0.74830E-04 | 0.92713E-04 | 2   |
| 24 | 0.75177E-04 | 0.93266E-04 | 2   |
| 25 | 0.75571E-04 | 0.93882E-04 | 2   |
| 26 | 0.76014E-04 | 0.94561E-04 | 2   |
| 27 | 0.76507E-04 | 0.95302E-04 | 2   |
| 28 | 0.77050E-04 | 0.96102E-04 | 2   |
| 29 | 0.77641E-04 | 0.96960E-04 | 2   |
| 30 | 0.78280E-04 | 0.97872E-04 | 2   |
| 31 | 0.78965E-04 | 0.98833E-04 | 2   |
| 32 | 0.79692E-04 | 0.99839E-04 | 2   |
| 33 | 0.80460E-04 | 0.10088E-03 | 2   |
| 34 | 0.81264E-04 | 0.10196E-03 | 2   |
| 35 | 0.82100E-04 | 0.10307E-03 | 2   |
| 36 | 0.82964E-04 | 0.10420E-03 | 2   |
| 37 | 0.83853E-04 | 0.10535E-03 | 2   |
| 38 | 0.84760E-04 | 0.10651E-03 | 2   |
| 39 | 0.85684E-04 | 0.10768E-03 | 1   |
| 40 | 0.86618E-04 | 0.10885E-03 | 1   |
| 41 | 0.87558E-04 | 0.11002E-03 | 1   |
| 42 | 0.88500E-04 | 0.11118E-03 | 1   |
| 43 | 0.89441E-04 | 0.11232E-03 | 1   |
| 44 | 0.90377E-04 | 0.11346E-03 | 1   |
| 45 | 0.91306E-04 | 0.11457E-03 | 1   |
| 46 | 0.92225E-04 | 0.11567E-03 | 1   |
| 47 | 0.93131E-04 | 0.11674E-03 | 1   |
| 48 | 0.94024E-04 | 0.11780E-03 | 1   |
| 49 | 0.94901E-04 | 0.11883E-03 | 1   |
| 50 | 0.95764E-04 | 0.11984E-03 | 2   |
| 51 | 0.96608E-04 | 0.12084E-03 | 1   |
| 52 | 0.97437E-04 | 0.12180E-03 | 2   |
| 53 | 0.98248E-04 | 0.12275E-03 | 2   |
| 54 | 0.99040E-04 | 0.12368E-03 | 1   |
| 55 | 0.99817E-04 | 0.12459E-03 | 1   |
| 56 | 0.10058E-03 | 0.12548E-03 | 1   |
| 57 | 0.10132E-03 | 0.12636E-03 | 1   |
| 58 | 0.10206E-03 | 0.12722E-03 | 1   |
| 59 | 0.10276E-03 | 0.12808E-03 | 1   |
| 60 | 0.10349E-03 | 0.12892E-03 | 1   |
| 61 | 0.10419E-03 | 0.12976E-03 | 1   |
| 62 | 0.10489E-03 | 0.13060E-03 | 1   |
| 63 | 0.10557E-03 | 0.13143E-03 | 1   |
| 64 | 0.10626E-03 | 0.13226E-03 | 1   |
| 65 | 0.10694E-03 | 0.13309E-03 | 1   |
| 66 | 0.10762E-03 | 0.13392E-03 | 1   |

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IND  
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| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.10798E-03 | 0.19739E-03 | 1   |
| 2  | 0.10770E-03 | 0.19604E-03 | 2   |
| 3  | 0.10738E-03 | 0.19464E-03 | 2   |
| 4  | 0.10702E-03 | 0.19320E-03 | 2   |
| 5  | 0.10662E-03 | 0.19172E-03 | 2   |
| 6  | 0.10618E-03 | 0.19020E-03 | 2   |
| 7  | 0.10570E-03 | 0.18865E-03 | 2   |
| 8  | 0.10519E-03 | 0.18708E-03 | 2   |
| 9  | 0.10465E-03 | 0.18548E-03 | 2   |
| 10 | 0.10408E-03 | 0.18385E-03 | 2   |
| 11 | 0.10348E-03 | 0.18222E-03 | 2   |
| 12 | 0.10285E-03 | 0.18056E-03 | 2   |
| 13 | 0.10220E-03 | 0.17890E-03 | 2   |
| 14 | 0.10153E-03 | 0.17722E-03 | 2   |
| 15 | 0.10083E-03 | 0.17554E-03 | 2   |
| 16 | 0.10012E-03 | 0.17385E-03 | 1   |
| 17 | 0.99385E-04 | 0.17215E-03 | 1   |
| 18 | 0.98636E-04 | 0.17045E-03 | 1   |
| 19 | 0.97872E-04 | 0.16876E-03 | 1   |
| 20 | 0.97095E-04 | 0.16706E-03 | 1   |
| 21 | 0.96305E-04 | 0.16536E-03 | 1   |
| 22 | 0.95504E-04 | 0.16367E-03 | 1   |
| 23 | 0.94694E-04 | 0.16197E-03 | 1   |
| 24 | 0.93874E-04 | 0.16029E-03 | 1   |
| 25 | 0.93048E-04 | 0.15860E-03 | 1   |
| 26 | 0.92214E-04 | 0.15693E-03 | 1   |
| 27 | 0.91375E-04 | 0.15526E-03 | 1   |
| 28 | 0.90531E-04 | 0.15360E-03 | 1   |
| 29 | 0.89683E-04 | 0.15194E-03 | 1   |
| 30 | 0.88833E-04 | 0.15030E-03 | 1   |
| 31 | 0.87979E-04 | 0.14866E-03 | 1   |
| 32 | 0.87125E-04 | 0.14704E-03 | 1   |
| 33 | 0.86269E-04 | 0.14543E-03 | 1   |
| 34 | 0.85414E-04 | 0.14384E-03 | 1   |
| 35 | 0.84559E-04 | 0.14226E-03 | 1   |
| 36 | 0.83706E-04 | 0.14069E-03 | 1   |
| 37 | 0.82855E-04 | 0.13915E-03 | 1   |
| 38 | 0.82007E-04 | 0.13762E-03 | 1   |
| 39 | 0.81162E-04 | 0.13612E-03 | 1   |
| 40 | 0.80322E-04 | 0.13463E-03 | 1   |
| 41 | 0.79487E-04 | 0.13317E-03 | 1   |
| 42 | 0.78658E-04 | 0.13174E-03 | 1   |
| 43 | 0.77836E-04 | 0.13033E-03 | 1   |
| 44 | 0.77021E-04 | 0.12895E-03 | 1   |
| 45 | 0.76214E-04 | 0.12760E-03 | 1   |
| 46 | 0.75416E-04 | 0.12627E-03 | 1   |
| 47 | 0.74627E-04 | 0.12498E-03 | 1   |
| 48 | 0.73849E-04 | 0.12372E-03 | 1   |
| 49 | 0.73081E-04 | 0.12249E-03 | 1   |
| 50 | 0.72324E-04 | 0.12129E-03 | 1   |
| 51 | 0.71579E-04 | 0.12012E-03 | 1   |
| 52 | 0.70847E-04 | 0.11899E-03 | 1   |
| 53 | 0.70128E-04 | 0.11789E-03 | 1   |
| 54 | 0.69422E-04 | 0.11682E-03 | 2   |
| 55 | 0.68729E-04 | 0.11579E-03 | 2   |
| 56 | 0.68050E-04 | 0.11479E-03 | 2   |
| 57 | 0.67386E-04 | 0.11381E-03 | 2   |
| 58 | 0.66736E-04 | 0.11287E-03 | 2   |
| 59 | 0.66101E-04 | 0.11196E-03 | 2   |
| 60 | 0.65481E-04 | 0.11109E-03 | 2   |
| 61 | 0.64876E-04 | 0.11024E-03 | 2   |
| 62 | 0.64287E-04 | 0.10942E-03 | 2   |
| 63 | 0.63712E-04 | 0.10863E-03 | 2   |
| 64 | 0.63152E-04 | 0.10786E-03 | 2   |
| 65 | 0.62608E-04 | 0.10713E-03 | 2   |
| 66 | 0.62078E-04 | 0.10642E-03 | 2   |

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✓ 0.54911E-04

✓ M  
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IND  
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| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.41034E-04 | 0.50354E-04 | 1   |
| 2  | 0.41034E-04 | 0.50354E-04 | 1   |
| 3  | 0.41034E-04 | 0.50354E-04 | 1   |
| 4  | 0.41035E-04 | 0.50355E-04 | 1   |
| 5  | 0.41035E-04 | 0.50356E-04 | 1   |
| 6  | 0.41035E-04 | 0.50357E-04 | 1   |
| 7  | 0.41037E-04 | 0.50361E-04 | 1   |
| 8  | 0.41039E-04 | 0.50368E-04 | 1   |
| 9  | 0.41043E-04 | 0.50379E-04 | 1   |
| 10 | 0.41050E-04 | 0.50397E-04 | 1   |
| 11 | 0.41062E-04 | 0.50424E-04 | 1   |
| 12 | 0.41079E-04 | 0.50463E-04 | 1   |
| 13 | 0.41103E-04 | 0.50517E-04 | 2   |
| 14 | 0.41137E-04 | 0.50589E-04 | 2   |
| 15 | 0.41183E-04 | 0.50685E-04 | 2   |
| 16 | 0.41245E-04 | 0.50808E-04 | 2   |
| 17 | 0.41327E-04 | 0.50964E-04 | 2   |
| 18 | 0.41430E-04 | 0.51158E-04 | 2   |
| 19 | 0.41560E-04 | 0.51394E-04 | 2   |
| 20 | 0.41719E-04 | 0.51677E-04 | 2   |
| 21 | 0.41912E-04 | 0.52011E-04 | 2   |
| 22 | 0.42143E-04 | 0.52401E-04 | 2   |
| 23 | 0.42414E-04 | 0.52850E-04 | 2   |
| 24 | 0.42729E-04 | 0.53362E-04 | 2   |
| 25 | 0.43093E-04 | 0.53938E-04 | 3   |
| 26 | 0.43506E-04 | 0.54581E-04 | 3   |
| 27 | 0.43970E-04 | 0.55292E-04 | 3   |
| 28 | 0.44486E-04 | 0.56071E-04 | 3   |
| 29 | 0.45056E-04 | 0.56916E-04 | 3   |
| 30 | 0.45679E-04 | 0.57827E-04 | 3   |
| 31 | 0.46356E-04 | 0.58804E-04 | 3   |
| 32 | 0.47084E-04 | 0.59842E-04 | 2   |
| 33 | 0.47864E-04 | 0.60937E-04 | 2   |
| 34 | 0.48693E-04 | 0.62089E-04 | 2   |
| 35 | 0.49571E-04 | 0.63293E-04 | 2   |
| 36 | 0.50494E-04 | 0.64547E-04 | 2   |
| 37 | 0.51460E-04 | 0.65848E-04 | 2   |
| 38 | 0.52468E-04 | 0.67191E-04 | 2   |
| 39 | 0.53513E-04 | 0.68572E-04 | 2   |
| 40 | 0.54593E-04 | 0.69990E-04 | 2   |
| 41 | 0.55705E-04 | 0.71440E-04 | 2   |
| 42 | 0.56848E-04 | 0.72919E-04 | 2   |
| 43 | 0.58017E-04 | 0.74424E-04 | 2   |
| 44 | 0.59211E-04 | 0.75954E-04 | 2   |
| 45 | 0.60428E-04 | 0.77505E-04 | 2   |
| 46 | 0.61665E-04 | 0.79077E-04 | 2   |
| 47 | 0.62921E-04 | 0.80666E-04 | 2   |
| 48 | 0.64194E-04 | 0.82273E-04 | 2   |
| 49 | 0.65482E-04 | 0.83895E-04 | 2   |
| 50 | 0.66785E-04 | 0.85532E-04 | 2   |
| 51 | 0.68102E-04 | 0.87184E-04 | 2   |
| 52 | 0.69433E-04 | 0.88849E-04 | 1   |
| 53 | 0.70774E-04 | 0.90530E-04 | 2   |
| 54 | 0.72129E-04 | 0.92224E-04 | 1   |
| 55 | 0.73496E-04 | 0.93932E-04 | 1   |
| 56 | 0.74874E-04 | 0.95654E-04 | 1   |
| 57 | 0.76263E-04 | 0.97390E-04 | 1   |
| 58 | 0.77664E-04 | 0.99142E-04 | 1   |
| 59 | 0.79077E-04 | 0.10091E-03 | 1   |
| 60 | 0.80502E-04 | 0.10269E-03 | 1   |
| 61 | 0.81939E-04 | 0.10449E-03 | 1   |
| 62 | 0.83389E-04 | 0.10630E-03 | 1   |
| 63 | 0.84851E-04 | 0.10813E-03 | 1   |
| 64 | 0.86326E-04 | 0.10997E-03 | 1   |
| 65 | 0.87814E-04 | 0.11184E-03 | 1   |
| 66 | 0.89316E-04 | 0.11371E-03 | 1   |

✓ YYAV  
0.12067E-03

M  
5

IND  
2

TT  
60,051

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.10920E-03 | 0.19808E-03 | 1   |
| 2  | 0.10880E-03 | 0.19616E-03 | 2   |
| 3  | 0.10839E-03 | 0.19421E-03 | 2   |
| 4  | 0.10790E-03 | 0.19219E-03 | 2   |
| 5  | 0.10735E-03 | 0.19010E-03 | 2   |
| 6  | 0.10674E-03 | 0.18795E-03 | 2   |
| 7  | 0.10608E-03 | 0.18574E-03 | 2   |
| 8  | 0.10537E-03 | 0.18349E-03 | 2   |
| 9  | 0.10461E-03 | 0.18119E-03 | 2   |
| 10 | 0.10381E-03 | 0.17885E-03 | 2   |
| 11 | 0.10296E-03 | 0.17648E-03 | 2   |
| 12 | 0.10208E-03 | 0.17407E-03 | 2   |
| 13 | 0.10115E-03 | 0.17165E-03 | 2   |
| 14 | 0.10019E-03 | 0.16920E-03 | 2   |
| 15 | 0.99197E-04 | 0.16674E-03 | 2   |
| 16 | 0.98173E-04 | 0.16426E-03 | 2   |
| 17 | 0.97120E-04 | 0.16178E-03 | 2   |
| 18 | 0.96040E-04 | 0.15929E-03 | 2   |
| 19 | 0.94937E-04 | 0.15679E-03 | 2   |
| 20 | 0.93813E-04 | 0.15430E-03 | 2   |
| 21 | 0.92668E-04 | 0.15181E-03 | 2   |
| 22 | 0.91506E-04 | 0.14932E-03 | 1   |
| 23 | 0.90329E-04 | 0.14684E-03 | 1   |
| 24 | 0.89138E-04 | 0.14437E-03 | 1   |
| 25 | 0.87935E-04 | 0.14191E-03 | 1   |
| 26 | 0.86722E-04 | 0.13947E-03 | 1   |
| 27 | 0.85501E-04 | 0.13705E-03 | 1   |
| 28 | 0.84273E-04 | 0.13464E-03 | 1   |
| 29 | 0.83041E-04 | 0.13225E-03 | 1   |
| 30 | 0.81805E-04 | 0.12989E-03 | 1   |
| 31 | 0.80567E-04 | 0.12755E-03 | 1   |
| 32 | 0.79329E-04 | 0.12523E-03 | 1   |
| 33 | 0.78091E-04 | 0.12295E-03 | 1   |
| 34 | 0.76857E-04 | 0.12069E-03 | 1   |
| 35 | 0.75625E-04 | 0.11847E-03 | 1   |
| 36 | 0.74395E-04 | 0.11628E-03 | 1   |
| 37 | 0.73175E-04 | 0.11412E-03 | 1   |
| 38 | 0.71967E-04 | 0.11200E-03 | 1   |
| 39 | 0.70763E-04 | 0.10992E-03 | 1   |
| 40 | 0.69565E-04 | 0.10788E-03 | 1   |
| 41 | 0.68385E-04 | 0.10588E-03 | 1   |
| 42 | 0.67214E-04 | 0.10392E-03 | 1   |
| 43 | 0.66056E-04 | 0.10200E-03 | 2   |
| 44 | 0.64911E-04 | 0.10013E-03 | 2   |
| 45 | 0.63781E-04 | 0.98305E-04 | 2   |
| 46 | 0.62667E-04 | 0.96525E-04 | 2   |
| 47 | 0.61570E-04 | 0.94791E-04 | 2   |
| 48 | 0.60490E-04 | 0.93104E-04 | 2   |
| 49 | 0.59428E-04 | 0.91464E-04 | 2   |
| 50 | 0.58385E-04 | 0.89871E-04 | 2   |
| 51 | 0.57351E-04 | 0.88325E-04 | 2   |
| 52 | 0.56338E-04 | 0.86827E-04 | 2   |
| 53 | 0.55337E-04 | 0.85374E-04 | 2   |
| 54 | 0.54412E-04 | 0.83968E-04 | 2   |
| 55 | 0.53471E-04 | 0.82608E-04 | 2   |
| 56 | 0.52551E-04 | 0.81292E-04 | 2   |
| 57 | 0.51653E-04 | 0.80021E-04 | 2   |
| 58 | 0.50777E-04 | 0.78793E-04 | 2   |
| 59 | 0.49923E-04 | 0.77607E-04 | 2   |
| 60 | 0.49091E-04 | 0.76463E-04 | 2   |
| 61 | 0.48281E-04 | 0.75360E-04 | 2   |
| 62 | 0.47492E-04 | 0.74297E-04 | 2   |
| 63 | 0.46726E-04 | 0.73273E-04 | 2   |
| 64 | 0.45981E-04 | 0.72287E-04 | 2   |
| 65 | 0.45258E-04 | 0.71338E-04 | 2   |
| 66 | 0.44557E-04 | 0.70425E-04 | 2   |

✓ YYAV  
0.35076E-04

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IND  
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| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.29738E-04 | 0.36492E-04 | 1   |
| 2  | 0.29738E-04 | 0.36492E-04 | 1   |
| 3  | 0.29738E-04 | 0.36492E-04 | 1   |
| 4  | 0.29738E-04 | 0.36493E-04 | 1   |
| 5  | 0.29738E-04 | 0.36493E-04 | 1   |
| 6  | 0.29739E-04 | 0.36495E-04 | 1   |
| 7  | 0.29740E-04 | 0.36498E-04 | 1   |
| 8  | 0.29742E-04 | 0.36503E-04 | 1   |
| 9  | 0.29745E-04 | 0.36513E-04 | 1   |
| 10 | 0.29751E-04 | 0.36528E-04 | 1   |
| 11 | 0.29761E-04 | 0.36550E-04 | 1   |
| 12 | 0.29774E-04 | 0.36583E-04 | 2   |
| 13 | 0.29795E-04 | 0.36627E-04 | 2   |
| 14 | 0.29823E-04 | 0.36688E-04 | 2   |
| 15 | 0.29862E-04 | 0.36769E-04 | 2   |
| 16 | 0.29914E-04 | 0.36872E-04 | 2   |
| 17 | 0.29982E-04 | 0.37003E-04 | 2   |
| 18 | 0.30069E-04 | 0.37166E-04 | 2   |
| 19 | 0.30178E-04 | 0.37365E-04 | 2   |
| 20 | 0.30313E-04 | 0.37604E-04 | 2   |
| 21 | 0.30476E-04 | 0.37888E-04 | 2   |
| 22 | 0.30672E-04 | 0.38220E-04 | 3   |
| 23 | 0.30904E-04 | 0.38604E-04 | 3   |
| 24 | 0.31174E-04 | 0.39042E-04 | 3   |
| 25 | 0.31494E-04 | 0.39537E-04 | 3   |
| 26 | 0.31837E-04 | 0.40090E-04 | 3   |
| 27 | 0.32235E-04 | 0.40704E-04 | 3   |
| 28 | 0.32675E-04 | 0.41377E-04 | 3   |
| 29 | 0.33171E-04 | 0.42111E-04 | 3   |
| 30 | 0.33710E-04 | 0.42905E-04 | 3   |
| 31 | 0.34298E-04 | 0.43757E-04 | 3   |
| 32 | 0.34934E-04 | 0.44668E-04 | 3   |
| 33 | 0.35617E-04 | 0.45635E-04 | 3   |
| 34 | 0.36347E-04 | 0.46656E-04 | 3   |
| 35 | 0.37122E-04 | 0.47728E-04 | 3   |
| 36 | 0.37939E-04 | 0.48849E-04 | 2   |
| 37 | 0.38798E-04 | 0.50016E-04 | 2   |
| 38 | 0.39698E-04 | 0.51226E-04 | 2   |
| 39 | 0.40635E-04 | 0.52478E-04 | 2   |
| 40 | 0.41609E-04 | 0.53769E-04 | 2   |
| 41 | 0.42617E-04 | 0.55096E-04 | 2   |
| 42 | 0.43657E-04 | 0.56457E-04 | 2   |
| 43 | 0.44727E-04 | 0.57851E-04 | 2   |
| 44 | 0.45826E-04 | 0.59275E-04 | 2   |
| 45 | 0.46952E-04 | 0.60727E-04 | 2   |
| 46 | 0.48103E-04 | 0.62206E-04 | 2   |
| 47 | 0.49278E-04 | 0.63712E-04 | 2   |
| 48 | 0.50475E-04 | 0.65242E-04 | 2   |
| 49 | 0.51695E-04 | 0.66796E-04 | 2   |
| 50 | 0.52935E-04 | 0.68374E-04 | 2   |
| 51 | 0.54196E-04 | 0.69975E-04 | 2   |
| 52 | 0.55476E-04 | 0.71599E-04 | 2   |
| 53 | 0.56775E-04 | 0.73246E-04 | 2   |
| 54 | 0.58094E-04 | 0.74916E-04 | 2   |
| 55 | 0.59431E-04 | 0.76608E-04 | 2   |
| 56 | 0.60787E-04 | 0.78324E-04 | 2   |
| 57 | 0.62163E-04 | 0.80063E-04 | 2   |
| 58 | 0.63557E-04 | 0.81826E-04 | 2   |
| 59 | 0.64970E-04 | 0.83612E-04 | 2   |
| 60 | 0.66402E-04 | 0.85423E-04 | 2   |
| 61 | 0.67854E-04 | 0.87257E-04 | 2   |
| 62 | 0.69325E-04 | 0.89116E-04 | 2   |
| 63 | 0.70816E-04 | 0.90999E-04 | 2   |
| 64 | 0.72327E-04 | 0.92907E-04 | 2   |
| 65 | 0.73857E-04 | 0.94839E-04 | 2   |
| 66 | 0.75407E-04 | 0.96796E-04 | 2   |



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IND  
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TT  
60,851

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.10415E-03 | 0.18811E-03 | 1   |
| 2  | 0.10375E-03 | 0.18601E-03 | 2   |
| 3  | 0.10326E-03 | 0.18391E-03 | 2   |
| 4  | 0.10274E-03 | 0.18172E-03 | 2   |
| 5  | 0.10215E-03 | 0.17947E-03 | 2   |
| 6  | 0.10149E-03 | 0.17715E-03 | 2   |
| 7  | 0.10078E-03 | 0.17477E-03 | 2   |
| 8  | 0.10001E-03 | 0.17235E-03 | 2   |
| 9  | 0.99194E-04 | 0.16988E-03 | 2   |
| 10 | 0.98328E-04 | 0.16737E-03 | 2   |
| 11 | 0.97415E-04 | 0.16482E-03 | 2   |
| 12 | 0.96460E-04 | 0.16225E-03 | 2   |
| 13 | 0.95465E-04 | 0.15966E-03 | 2   |
| 14 | 0.94433E-04 | 0.15705E-03 | 2   |
| 15 | 0.93365E-04 | 0.15442E-03 | 2   |
| 16 | 0.92266E-04 | 0.15179E-03 | 2   |
| 17 | 0.91138E-04 | 0.14915E-03 | 2   |
| 18 | 0.89983E-04 | 0.14650E-03 | 2   |
| 19 | 0.88803E-04 | 0.14386E-03 | 2   |
| 20 | 0.87602E-04 | 0.14123E-03 | 2   |
| 21 | 0.86381E-04 | 0.13860E-03 | 2   |
| 22 | 0.85143E-04 | 0.13598E-03 | 1   |
| 23 | 0.83890E-04 | 0.13337E-03 | 1   |
| 24 | 0.82625E-04 | 0.13078E-03 | 1   |
| 25 | 0.81349E-04 | 0.12821E-03 | 1   |
| 26 | 0.80064E-04 | 0.12566E-03 | 1   |
| 27 | 0.78772E-04 | 0.12313E-03 | 1   |
| 28 | 0.77476E-04 | 0.12063E-03 | 1   |
| 29 | 0.76176E-04 | 0.11815E-03 | 1   |
| 30 | 0.74876E-04 | 0.11571E-03 | 1   |
| 31 | 0.73575E-04 | 0.11329E-03 | 1   |
| 32 | 0.72276E-04 | 0.11091E-03 | 1   |
| 33 | 0.70980E-04 | 0.10856E-03 | 1   |
| 34 | 0.69690E-04 | 0.10624E-03 | 1   |
| 35 | 0.68405E-04 | 0.10396E-03 | 1   |
| 36 | 0.67128E-04 | 0.10173E-03 | 1   |
| 37 | 0.65860E-04 | 0.99529E-04 | 1   |
| 38 | 0.64602E-04 | 0.97373E-04 | 1   |
| 39 | 0.63355E-04 | 0.95260E-04 | 1   |
| 40 | 0.62120E-04 | 0.93191E-04 | 2   |
| 41 | 0.60898E-04 | 0.91167E-04 | 2   |
| 42 | 0.59691E-04 | 0.89190E-04 | 2   |
| 43 | 0.58500E-04 | 0.87259E-04 | 2   |
| 44 | 0.57325E-04 | 0.85376E-04 | 2   |
| 45 | 0.56167E-04 | 0.83542E-04 | 2   |
| 46 | 0.55027E-04 | 0.81756E-04 | 2   |
| 47 | 0.53906E-04 | 0.80019E-04 | 2   |
| 48 | 0.52805E-04 | 0.78332E-04 | 2   |
| 49 | 0.51724E-04 | 0.76694E-04 | 2   |
| 50 | 0.50664E-04 | 0.75105E-04 | 2   |
| 51 | 0.49625E-04 | 0.73565E-04 | 2   |
| 52 | 0.48607E-04 | 0.72073E-04 | 2   |
| 53 | 0.47612E-04 | 0.70629E-04 | 2   |
| 54 | 0.46639E-04 | 0.69233E-04 | 2   |
| 55 | 0.45689E-04 | 0.67883E-04 | 2   |
| 56 | 0.44762E-04 | 0.66578E-04 | 2   |
| 57 | 0.43858E-04 | 0.65319E-04 | 2   |
| 58 | 0.42977E-04 | 0.64104E-04 | 2   |
| 59 | 0.42119E-04 | 0.62932E-04 | 2   |
| 60 | 0.41284E-04 | 0.61803E-04 | 2   |
| 61 | 0.40472E-04 | 0.60714E-04 | 2   |
| 62 | 0.39683E-04 | 0.59666E-04 | 2   |
| 63 | 0.38917E-04 | 0.58657E-04 | 2   |
| 64 | 0.38173E-04 | 0.57687E-04 | 2   |
| 65 | 0.37451E-04 | 0.56753E-04 | 2   |
| 66 | 0.36752E-04 | 0.55856E-04 | 2   |

Wellington Business Forms, LTD.

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| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.25045E-04 | 0.30733E-04 | 1   |
| 2  | 0.25045E-04 | 0.30733E-04 | 1   |
| 3  | 0.25045E-04 | 0.30734E-04 | 1   |
| 4  | 0.25045E-04 | 0.30734E-04 | 1   |
| 5  | 0.25045E-04 | 0.30734E-04 | 1   |
| 6  | 0.25046E-04 | 0.30736E-04 | 1   |
| 7  | 0.25047E-04 | 0.30739E-04 | 1   |
| 8  | 0.25049E-04 | 0.30744E-04 | 1   |
| 9  | 0.25052E-04 | 0.30752E-04 | 1   |
| 10 | 0.25057E-04 | 0.30765E-04 | 1   |
| 11 | 0.25066E-04 | 0.30786E-04 | 1   |
| 12 | 0.25078E-04 | 0.30815E-04 | 2   |
| 13 | 0.25096E-04 | 0.30856E-04 | 2   |
| 14 | 0.25122E-04 | 0.30911E-04 | 2   |
| 15 | 0.25157E-04 | 0.30983E-04 | 2   |
| 16 | 0.25205E-04 | 0.31078E-04 | 2   |
| 17 | 0.25266E-04 | 0.31197E-04 | 2   |
| 18 | 0.25345E-04 | 0.31346E-04 | 2   |
| 19 | 0.25445E-04 | 0.31527E-04 | 2   |
| 20 | 0.25567E-04 | 0.31745E-04 | 2   |
| 21 | 0.25717E-04 | 0.32005E-04 | 3   |
| 22 | 0.25897E-04 | 0.32309E-04 | 3   |
| 23 | 0.26108E-04 | 0.32661E-04 | 3   |
| 24 | 0.26355E-04 | 0.33063E-04 | 3   |
| 25 | 0.26635E-04 | 0.33518E-04 | 3   |
| 26 | 0.26964E-04 | 0.34026E-04 | 3   |
| 27 | 0.27329E-04 | 0.34591E-04 | 3   |
| 28 | 0.27738E-04 | 0.35212E-04 | 3   |
| 29 | 0.28191E-04 | 0.35890E-04 | 3   |
| 30 | 0.28689E-04 | 0.36624E-04 | 3   |
| 31 | 0.29232E-04 | 0.37414E-04 | 3   |
| 32 | 0.29820E-04 | 0.38259E-04 | 3   |
| 33 | 0.30453E-04 | 0.39158E-04 | 3   |
| 34 | 0.31130E-04 | 0.40108E-04 | 3   |
| 35 | 0.31851E-04 | 0.41109E-04 | 3   |
| 36 | 0.32613E-04 | 0.42157E-04 | 3   |
| 37 | 0.33416E-04 | 0.43252E-04 | 3   |
| 38 | 0.34257E-04 | 0.44390E-04 | 2   |
| 39 | 0.35136E-04 | 0.45567E-04 | 2   |
| 40 | 0.36050E-04 | 0.46785E-04 | 2   |
| 41 | 0.36998E-04 | 0.48039E-04 | 2   |
| 42 | 0.37979E-04 | 0.49329E-04 | 2   |
| 43 | 0.38991E-04 | 0.50652E-04 | 2   |
| 44 | 0.40032E-04 | 0.52007E-04 | 2   |
| 45 | 0.41100E-04 | 0.53392E-04 | 2   |
| 46 | 0.42195E-04 | 0.54807E-04 | 2   |
| 47 | 0.43316E-04 | 0.56250E-04 | 2   |
| 48 | 0.44461E-04 | 0.57720E-04 | 2   |
| 49 | 0.45630E-04 | 0.59218E-04 | 2   |
| 50 | 0.46821E-04 | 0.60741E-04 | 2   |
| 51 | 0.48035E-04 | 0.62291E-04 | 2   |
| 52 | 0.49271E-04 | 0.63866E-04 | 2   |
| 53 | 0.50528E-04 | 0.65467E-04 | 2   |
| 54 | 0.51806E-04 | 0.67094E-04 | 2   |
| 55 | 0.53106E-04 | 0.68747E-04 | 2   |
| 56 | 0.54427E-04 | 0.70426E-04 | 2   |
| 57 | 0.55769E-04 | 0.72132E-04 | 2   |
| 58 | 0.57133E-04 | 0.73864E-04 | 2   |
| 59 | 0.58518E-04 | 0.75623E-04 | 2   |
| 60 | 0.59925E-04 | 0.77408E-04 | 2   |
| 61 | 0.61354E-04 | 0.79222E-04 | 2   |
| 62 | 0.62805E-04 | 0.81062E-04 | 2   |
| 63 | 0.64277E-04 | 0.82930E-04 | 2   |
| 64 | 0.65772E-04 | 0.84825E-04 | 2   |
| 65 | 0.67295E-04 | 0.86747E-04 | 2   |
| 66 | 0.68829E-04 | 0.88696E-04 | 2   |

Willingon Business Forms, LTD.

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Wellington Business Farms, LTD

| 1  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.23008E-04 | 0.28234E-04 | 1   |
| 2  | 0.23008E-04 | 0.28234E-04 | 1   |
| 3  | 0.23008E-04 | 0.28234E-04 | 1   |
| 4  | 0.23008E-04 | 0.28234E-04 | 1   |
| 5  | 0.23008E-04 | 0.28235E-04 | 1   |
| 6  | 0.23009E-04 | 0.28236E-04 | 1   |
| 7  | 0.23010E-04 | 0.28239E-04 | 1   |
| 8  | 0.23011E-04 | 0.28243E-04 | 1   |
| 9  | 0.23014E-04 | 0.28252E-04 | 1   |
| 10 | 0.23020E-04 | 0.28264E-04 | 1   |
| 11 | 0.23028E-04 | 0.28284E-04 | 1   |
| 12 | 0.23040E-04 | 0.28312E-04 | 2   |
| 13 | 0.23057E-04 | 0.28350E-04 | 2   |
| 14 | 0.23081E-04 | 0.28403E-04 | 2   |
| 15 | 0.23115E-04 | 0.28472E-04 | 2   |
| 16 | 0.23160E-04 | 0.28562E-04 | 2   |
| 17 | 0.23219E-04 | 0.28676E-04 | 2   |
| 18 | 0.23295E-04 | 0.28818E-04 | 2   |
| 19 | 0.23320E-04 | 0.28922E-04 | 2   |
| 20 | 0.23507E-04 | 0.29201E-04 | 2   |
| 21 | 0.23651E-04 | 0.29450E-04 | 3   |
| 22 | 0.23823E-04 | 0.29741E-04 | 3   |
| 23 | 0.24025E-04 | 0.30079E-04 | 3   |
| 24 | 0.24262E-04 | 0.30464E-04 | 3   |
| 25 | 0.24534E-04 | 0.30901E-04 | 3   |
| 26 | 0.24845E-04 | 0.31390E-04 | 3   |
| 27 | 0.25196E-04 | 0.31932E-04 | 3   |
| 28 | 0.25589E-04 | 0.32529E-04 | 3   |
| 29 | 0.26024E-04 | 0.33182E-04 | 3   |
| 30 | 0.26503E-04 | 0.33889E-04 | 3   |
| 31 | 0.27026E-04 | 0.34650E-04 | 3   |
| 32 | 0.27593E-04 | 0.35466E-04 | 3   |
| 33 | 0.28203E-04 | 0.36333E-04 | 3   |
| 34 | 0.28857E-04 | 0.37252E-04 | 3   |
| 35 | 0.29552E-04 | 0.38220E-04 | 3   |
| 36 | 0.30289E-04 | 0.39235E-04 | 3   |
| 37 | 0.31066E-04 | 0.40295E-04 | 3   |
| 38 | 0.31880E-04 | 0.41399E-04 | 2   |
| 39 | 0.32732E-04 | 0.42543E-04 | 2   |
| 40 | 0.33619E-04 | 0.43726E-04 | 2   |
| 41 | 0.34535E-04 | 0.44946E-04 | 2   |
| 42 | 0.35492E-04 | 0.46202E-04 | 2   |
| 43 | 0.36477E-04 | 0.47492E-04 | 2   |
| 44 | 0.37490E-04 | 0.48815E-04 | 2   |
| 45 | 0.38532E-04 | 0.50168E-04 | 2   |
| 46 | 0.39601E-04 | 0.51552E-04 | 2   |
| 47 | 0.40698E-04 | 0.52965E-04 | 2   |
| 48 | 0.41816E-04 | 0.54407E-04 | 2   |
| 49 | 0.42960E-04 | 0.55876E-04 | 2   |
| 50 | 0.44128E-04 | 0.57372E-04 | 2   |
| 51 | 0.45319E-04 | 0.58896E-04 | 2   |
| 52 | 0.46532E-04 | 0.60447E-04 | 2   |
| 53 | 0.47768E-04 | 0.62025E-04 | 2   |
| 54 | 0.49027E-04 | 0.63630E-04 | 2   |
| 55 | 0.50307E-04 | 0.65262E-04 | 2   |
| 56 | 0.51610E-04 | 0.66921E-04 | 2   |
| 57 | 0.52935E-04 | 0.68608E-04 | 2   |
| 58 | 0.54283E-04 | 0.70323E-04 | 2   |
| 59 | 0.55653E-04 | 0.72066E-04 | 2   |
| 60 | 0.57045E-04 | 0.73837E-04 | 2   |
| 61 | 0.58461E-04 | 0.75637E-04 | 2   |
| 62 | 0.59899E-04 | 0.77465E-04 | 2   |
| 63 | 0.61361E-04 | 0.79322E-04 | 2   |
| 64 | 0.62846E-04 | 0.81207E-04 | 2   |
| 65 | 0.64354E-04 | 0.83121E-04 | 2   |
| 66 | 0.65885E-04 | 0.85063E-04 | 2   |

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Wellington Business Farms, LTD.

M 14 IND 2 TT 60,851

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.99869E-04 | 0.18000E-03 | 1   |
| 2  | 0.99550E-04 | 0.17781E-03 | 2   |
| 3  | 0.99159E-04 | 0.17563E-03 | 2   |
| 4  | 0.98503E-04 | 0.17337E-03 | 2   |
| 5  | 0.97803E-04 | 0.17103E-03 | 2   |
| 6  | 0.97203E-04 | 0.16864E-03 | 2   |
| 7  | 0.96464E-04 | 0.16618E-03 | 2   |
| 8  | 0.95669E-04 | 0.16368E-03 | 2   |
| 9  | 0.94822E-04 | 0.16113E-03 | 2   |
| 10 | 0.93825E-04 | 0.15854E-03 | 2   |
| 11 | 0.92902E-04 | 0.15593E-03 | 2   |
| 12 | 0.91995E-04 | 0.15328E-03 | 2   |
| 13 | 0.91087E-04 | 0.15062E-03 | 2   |
| 14 | 0.89901E-04 | 0.14794E-03 | 2   |
| 15 | 0.88601E-04 | 0.14525E-03 | 2   |
| 16 | 0.87288E-04 | 0.14255E-03 | 2   |
| 17 | 0.85906E-04 | 0.13985E-03 | 2   |
| 18 | 0.85318E-04 | 0.13714E-03 | 2   |
| 19 | 0.84105E-04 | 0.13445E-03 | 2   |
| 20 | 0.82871E-04 | 0.13176E-03 | 2   |
| 21 | 0.81619E-04 | 0.12908E-03 | 2   |
| 22 | 0.80350E-04 | 0.12642E-03 | 1   |
| 23 | 0.79066E-04 | 0.12378E-03 | 1   |
| 24 | 0.77771E-04 | 0.12115E-03 | 1   |
| 25 | 0.76467E-04 | 0.11855E-03 | 1   |
| 26 | 0.75155E-04 | 0.11597E-03 | 1   |
| 27 | 0.73837E-04 | 0.11341E-03 | 1   |
| 28 | 0.72516E-04 | 0.11089E-03 | 1   |
| 29 | 0.71194E-04 | 0.10840E-03 | 1   |
| 30 | 0.69871E-04 | 0.10594E-03 | 1   |
| 31 | 0.68550E-04 | 0.10351E-03 | 1   |
| 32 | 0.67232E-04 | 0.10112E-03 | 1   |
| 33 | 0.65919E-04 | 0.98767E-04 | 1   |
| 34 | 0.64613E-04 | 0.96454E-04 | 1   |
| 35 | 0.63314E-04 | 0.94181E-04 | 1   |
| 36 | 0.62025E-04 | 0.91950E-04 | 1   |
| 37 | 0.60746E-04 | 0.89762E-04 | 1   |
| 38 | 0.59478E-04 | 0.87619E-04 | 2   |
| 39 | 0.58223E-04 | 0.85520E-04 | 2   |
| 40 | 0.56982E-04 | 0.83468E-04 | 2   |
| 41 | 0.55755E-04 | 0.81464E-04 | 2   |
| 42 | 0.54545E-04 | 0.79507E-04 | 2   |
| 43 | 0.53352E-04 | 0.77599E-04 | 2   |
| 44 | 0.52176E-04 | 0.75740E-04 | 2   |
| 45 | 0.51019E-04 | 0.73931E-04 | 2   |
| 46 | 0.49882E-04 | 0.72172E-04 | 2   |
| 47 | 0.48764E-04 | 0.70463E-04 | 2   |
| 48 | 0.47667E-04 | 0.68803E-04 | 2   |
| 49 | 0.46592E-04 | 0.67194E-04 | 2   |
| 50 | 0.45538E-04 | 0.65634E-04 | 2   |
| 51 | 0.44506E-04 | 0.64123E-04 | 2   |
| 52 | 0.43497E-04 | 0.62660E-04 | 2   |
| 53 | 0.42510E-04 | 0.61246E-04 | 2   |
| 54 | 0.41547E-04 | 0.59879E-04 | 2   |
| 55 | 0.40607E-04 | 0.58559E-04 | 2   |
| 56 | 0.39690E-04 | 0.57284E-04 | 2   |
| 57 | 0.38797E-04 | 0.56054E-04 | 2   |
| 58 | 0.37927E-04 | 0.54867E-04 | 2   |
| 59 | 0.37081E-04 | 0.53724E-04 | 2   |
| 60 | 0.36256E-04 | 0.52622E-04 | 2   |
| 61 | 0.35455E-04 | 0.51561E-04 | 2   |
| 62 | 0.34682E-04 | 0.50540E-04 | 2   |
| 63 | 0.33929E-04 | 0.49558E-04 | 2   |
| 64 | 0.33198E-04 | 0.48614E-04 | 2   |
| 65 | 0.32489E-04 | 0.47706E-04 | 2   |
| 66 | 0.31903E-04 | 0.46833E-04 | 2   |

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Wellington Business Forms, LTD.

H 17 IND 1 TT 60,077

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.22114E-04 | 0.27137E-04 | 1   |
| 2  | 0.22114E-04 | 0.27137E-04 | 1   |
| 3  | 0.22114E-04 | 0.27137E-04 | 1   |
| 4  | 0.22114E-04 | 0.27137E-04 | 1   |
| 5  | 0.22115E-04 | 0.27138E-04 | 1   |
| 6  | 0.22115E-04 | 0.27139E-04 | 1   |
| 7  | 0.22116E-04 | 0.27142E-04 | 1   |
| 8  | 0.22117E-04 | 0.27146E-04 | 1   |
| 9  | 0.22121E-04 | 0.27154E-04 | 1   |
| 10 | 0.22126E-04 | 0.27167E-04 | 1   |
| 11 | 0.22133E-04 | 0.27186E-04 | 1   |
| 12 | 0.22145E-04 | 0.27213E-04 | 2   |
| 13 | 0.22162E-04 | 0.27251E-04 | 2   |
| 14 | 0.22186E-04 | 0.27302E-04 | 2   |
| 15 | 0.22219E-04 | 0.27370E-04 | 2   |
| 16 | 0.22263E-04 | 0.27458E-04 | 2   |
| 17 | 0.22321E-04 | 0.27570E-04 | 2   |
| 18 | 0.22395E-04 | 0.27709E-04 | 2   |
| 19 | 0.22488E-04 | 0.27880E-04 | 2   |
| 20 | 0.22603E-04 | 0.28084E-04 | 3   |
| 21 | 0.22744E-04 | 0.28328E-04 | 3   |
| 22 | 0.22912E-04 | 0.28614E-04 | 3   |
| 23 | 0.23111E-04 | 0.28945E-04 | 3   |
| 24 | 0.23343E-04 | 0.29324E-04 | 3   |
| 25 | 0.23611E-04 | 0.29752E-04 | 3   |
| 26 | 0.23917E-04 | 0.30232E-04 | 3   |
| 27 | 0.24260E-04 | 0.30765E-04 | 3   |
| 28 | 0.24646E-04 | 0.31351E-04 | 3   |
| 29 | 0.25073E-04 | 0.31992E-04 | 3   |
| 30 | 0.25544E-04 | 0.32688E-04 | 3   |
| 31 | 0.26058E-04 | 0.33437E-04 | 3   |
| 32 | 0.26615E-04 | 0.34239E-04 | 3   |
| 33 | 0.27215E-04 | 0.35092E-04 | 3   |
| 34 | 0.27858E-04 | 0.35997E-04 | 3   |
| 35 | 0.28543E-04 | 0.36950E-04 | 3   |
| 36 | 0.29268E-04 | 0.37950E-04 | 3   |
| 37 | 0.30033E-04 | 0.38995E-04 | 3   |
| 38 | 0.30836E-04 | 0.40084E-04 | 3   |
| 39 | 0.31675E-04 | 0.41213E-04 | 2   |
| 40 | 0.32550E-04 | 0.42381E-04 | 2   |
| 41 | 0.33458E-04 | 0.43586E-04 | 2   |
| 42 | 0.34395E-04 | 0.44827E-04 | 2   |
| 43 | 0.35371E-04 | 0.46102E-04 | 2   |
| 44 | 0.36373E-04 | 0.47410E-04 | 2   |
| 45 | 0.37402E-04 | 0.48749E-04 | 2   |
| 46 | 0.38460E-04 | 0.50119E-04 | 2   |
| 47 | 0.39543E-04 | 0.51519E-04 | 2   |
| 48 | 0.40651E-04 | 0.52947E-04 | 2   |
| 49 | 0.41785E-04 | 0.54404E-04 | 2   |
| 50 | 0.42942E-04 | 0.55888E-04 | 2   |
| 51 | 0.44122E-04 | 0.57400E-04 | 2   |
| 52 | 0.45326E-04 | 0.58940E-04 | 2   |
| 53 | 0.46552E-04 | 0.60507E-04 | 2   |
| 54 | 0.47802E-04 | 0.62102E-04 | 2   |
| 55 | 0.49073E-04 | 0.63725E-04 | 2   |
| 56 | 0.50368E-04 | 0.65375E-04 | 2   |
| 57 | 0.51685E-04 | 0.67054E-04 | 2   |
| 58 | 0.53025E-04 | 0.68760E-04 | 2   |
| 59 | 0.54388E-04 | 0.70496E-04 | 2   |
| 60 | 0.55774E-04 | 0.72260E-04 | 2   |
| 61 | 0.57184E-04 | 0.74053E-04 | 2   |
| 62 | 0.58616E-04 | 0.75876E-04 | 2   |
| 63 | 0.60073E-04 | 0.77727E-04 | 2   |
| 64 | 0.61553E-04 | 0.79608E-04 | 2   |
| 65 | 0.63056E-04 | 0.81517E-04 | 2   |
| 66 | 0.64583E-04 | 0.83455E-04 | 2   |

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Wellington Business Forms LTD

M 17 IND 2 TT 60.851

| I  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.99378E-04 | 0.17885E-03 | 1   |
| 2  | 0.98957E-04 | 0.17666E-03 | 2   |
| 3  | 0.98464E-04 | 0.17447E-03 | 2   |
| 4  | 0.97905E-04 | 0.17220E-03 | 2   |
| 5  | 0.97282E-04 | 0.16985E-03 | 2   |
| 6  | 0.96598E-04 | 0.16744E-03 | 2   |
| 7  | 0.95856E-04 | 0.16498E-03 | 2   |
| 8  | 0.95058E-04 | 0.16246E-03 | 2   |
| 9  | 0.94207E-04 | 0.15991E-03 | 2   |
| 10 | 0.93307E-04 | 0.15731E-03 | 2   |
| 11 | 0.92360E-04 | 0.15469E-03 | 2   |
| 12 | 0.91369E-04 | 0.15203E-03 | 2   |
| 13 | 0.90337E-04 | 0.14936E-03 | 2   |
| 14 | 0.89267E-04 | 0.14667E-03 | 2   |
| 15 | 0.88152E-04 | 0.14397E-03 | 2   |
| 16 | 0.87025E-04 | 0.14127E-03 | 2   |
| 17 | 0.85859E-04 | 0.13856E-03 | 2   |
| 18 | 0.84667E-04 | 0.13585E-03 | 2   |
| 19 | 0.83450E-04 | 0.13315E-03 | 2   |
| 20 | 0.82213E-04 | 0.13045E-03 | 2   |
| 21 | 0.80956E-04 | 0.12777E-03 | 2   |
| 22 | 0.79683E-04 | 0.12511E-03 | 1   |
| 23 | 0.78396E-04 | 0.12246E-03 | 1   |
| 24 | 0.77098E-04 | 0.11933E-03 | 1   |
| 25 | 0.75790E-04 | 0.11722E-03 | 1   |
| 26 | 0.74475E-04 | 0.11464E-03 | 1   |
| 27 | 0.73155E-04 | 0.11208E-03 | 1   |
| 28 | 0.71831E-04 | 0.10956E-03 | 1   |
| 29 | 0.70505E-04 | 0.10706E-03 | 1   |
| 30 | 0.69180E-04 | 0.10460E-03 | 1   |
| 31 | 0.67857E-04 | 0.10217E-03 | 1   |
| 32 | 0.66537E-04 | 0.99783E-04 | 1   |
| 33 | 0.65223E-04 | 0.97431E-04 | 1   |
| 34 | 0.63915E-04 | 0.95119E-04 | 1   |
| 35 | 0.62615E-04 | 0.92848E-04 | 1   |
| 36 | 0.61324E-04 | 0.90619E-04 | 1   |
| 37 | 0.60044E-04 | 0.88433E-04 | 1   |
| 38 | 0.58775E-04 | 0.86292E-04 | 2   |
| 39 | 0.57520E-04 | 0.84197E-04 | 2   |
| 40 | 0.56279E-04 | 0.82148E-04 | 2   |
| 41 | 0.55052E-04 | 0.80147E-04 | 2   |
| 42 | 0.53842E-04 | 0.78194E-04 | 2   |
| 43 | 0.52649E-04 | 0.76290E-04 | 2   |
| 44 | 0.51474E-04 | 0.74435E-04 | 2   |
| 45 | 0.50318E-04 | 0.72631E-04 | 2   |
| 46 | 0.49181E-04 | 0.70876E-04 | 2   |
| 47 | 0.48064E-04 | 0.69171E-04 | 2   |
| 48 | 0.46968E-04 | 0.67516E-04 | 2   |
| 49 | 0.45894E-04 | 0.65911E-04 | 2   |
| 50 | 0.44841E-04 | 0.64356E-04 | 2   |
| 51 | 0.43811E-04 | 0.62850E-04 | 2   |
| 52 | 0.42803E-04 | 0.61392E-04 | 2   |
| 53 | 0.41818E-04 | 0.59982E-04 | 2   |
| 54 | 0.40857E-04 | 0.58620E-04 | 2   |
| 55 | 0.39919E-04 | 0.57304E-04 | 2   |
| 56 | 0.39004E-04 | 0.56033E-04 | 2   |
| 57 | 0.38112E-04 | 0.54808E-04 | 2   |
| 58 | 0.37245E-04 | 0.53626E-04 | 2   |
| 59 | 0.36400E-04 | 0.52487E-04 | 2   |
| 60 | 0.35580E-04 | 0.51390E-04 | 2   |
| 61 | 0.34782E-04 | 0.50333E-04 | 2   |
| 62 | 0.34008E-04 | 0.49316E-04 | 2   |
| 63 | 0.33256E-04 | 0.48338E-04 | 2   |
| 64 | 0.32527E-04 | 0.47397E-04 | 2   |
| 65 | 0.31821E-04 | 0.46493E-04 | 2   |
| 66 | 0.31136E-04 | 0.45625E-04 | 2   |

YYAV  
0.21913E-04

Wellington Business Forms LTD

H  
20

IND  
1

TT  
60,077

| 1  | YY          | XX          | NTR |
|----|-------------|-------------|-----|
| 1  | 0.21721E-04 | 0.26654E-04 | 1   |
| 2  | 0.21721E-04 | 0.26654E-04 | 1   |
| 3  | 0.21721E-04 | 0.26654E-04 | 1   |
| 4  | 0.21721E-04 | 0.26655E-04 | 1   |
| 5  | 0.21721E-04 | 0.26655E-04 | 1   |
| 6  | 0.21722E-04 | 0.26657E-04 | 1   |
| 7  | 0.21723E-04 | 0.26659E-04 | 1   |
| 8  | 0.21724E-04 | 0.26664E-04 | 1   |
| 9  | 0.21727E-04 | 0.26672E-04 | 1   |
| 10 | 0.21732E-04 | 0.26684E-04 | 1   |
| 11 | 0.21740E-04 | 0.26703E-04 | 1   |
| 12 | 0.21751E-04 | 0.26730E-04 | 2   |
| 13 | 0.21766E-04 | 0.26767E-04 | 2   |
| 14 | 0.21772E-04 | 0.26810E-04 | 2   |
| 15 | 0.21825E-04 | 0.26886E-04 | 2   |
| 16 | 0.21868E-04 | 0.26973E-04 | 2   |
| 17 | 0.21928E-04 | 0.27084E-04 | 2   |
| 18 | 0.21995E-04 | 0.27221E-04 | 2   |
| 19 | 0.22091E-04 | 0.27390E-04 | 2   |
| 20 | 0.22206E-04 | 0.27593E-04 | 3   |
| 21 | 0.22345E-04 | 0.27835E-04 | 3   |
| 22 | 0.22512E-04 | 0.28119E-04 | 3   |
| 23 | 0.22709E-04 | 0.28447E-04 | 3   |
| 24 | 0.22939E-04 | 0.28822E-04 | 3   |
| 25 | 0.23204E-04 | 0.29247E-04 | 3   |
| 26 | 0.23507E-04 | 0.29722E-04 | 3   |
| 27 | 0.23848E-04 | 0.30251E-04 | 3   |
| 28 | 0.24231E-04 | 0.30833E-04 | 3   |
| 29 | 0.24655E-04 | 0.31469E-04 | 3   |
| 30 | 0.25122E-04 | 0.32159E-04 | 3   |
| 31 | 0.25631E-04 | 0.32902E-04 | 3   |
| 32 | 0.26184E-04 | 0.33699E-04 | 3   |
| 33 | 0.26780E-04 | 0.34546E-04 | 3   |
| 34 | 0.27418E-04 | 0.35444E-04 | 3   |
| 35 | 0.28098E-04 | 0.36391E-04 | 3   |
| 36 | 0.28819E-04 | 0.37385E-04 | 3   |
| 37 | 0.29579E-04 | 0.38423E-04 | 3   |
| 38 | 0.30377E-04 | 0.39505E-04 | 3   |
| 39 | 0.31210E-04 | 0.40628E-04 | 2   |
| 40 | 0.32080E-04 | 0.41789E-04 | 2   |
| 41 | 0.32983E-04 | 0.42987E-04 | 2   |
| 42 | 0.33918E-04 | 0.44221E-04 | 2   |
| 43 | 0.34884E-04 | 0.45490E-04 | 2   |
| 44 | 0.35891E-04 | 0.46792E-04 | 2   |
| 45 | 0.36905E-04 | 0.48125E-04 | 2   |
| 46 | 0.37957E-04 | 0.49488E-04 | 2   |
| 47 | 0.39035E-04 | 0.50882E-04 | 2   |
| 48 | 0.40139E-04 | 0.52305E-04 | 2   |
| 49 | 0.41267E-04 | 0.53756E-04 | 2   |
| 50 | 0.42419E-04 | 0.55235E-04 | 2   |
| 51 | 0.43595E-04 | 0.56742E-04 | 2   |
| 52 | 0.44795E-04 | 0.58276E-04 | 2   |
| 53 | 0.46017E-04 | 0.59839E-04 | 2   |
| 54 | 0.47262E-04 | 0.61429E-04 | 2   |
| 55 | 0.48530E-04 | 0.63047E-04 | 2   |
| 56 | 0.49821E-04 | 0.64694E-04 | 2   |
| 57 | 0.51134E-04 | 0.66369E-04 | 2   |
| 58 | 0.52471E-04 | 0.68072E-04 | 2   |
| 59 | 0.53831E-04 | 0.69804E-04 | 2   |
| 60 | 0.55214E-04 | 0.71565E-04 | 2   |
| 61 | 0.56621E-04 | 0.73356E-04 | 2   |
| 62 | 0.58051E-04 | 0.75175E-04 | 2   |
| 63 | 0.59505E-04 | 0.77024E-04 | 2   |
| 64 | 0.60983E-04 | 0.78903E-04 | 2   |
| 65 | 0.62484E-04 | 0.80810E-04 | 2   |
| 66 | 0.64010E-04 | 0.82747E-04 | 2   |

YYAV  
0.97917E-04

Wellington Business Forms, Ltd.

Wellington Business Forms, Ltd.

Table No: 24

N      NNT      NCASE

10      600      1

H      FEHP

82.0600 - 298.0000

H      TIME      VOID      A      ERR      YU      DENS

100.0000    15.0000    0.4200    7.9173    0.0010    0.0100    0.7300

AK      250.0000000000

G      PH      PL

10.0000    4.0816    1.3605

PURGE RATIO

2.2000

DH      DL

0.16691E-03      0.55635E-04

DP      VISC      DIFF      ALPHA      BETA      ANAV

0.10000    0.01190    4.20000    1.25000    -0.51000    4.40000

OZ1      OZ2      NNT1      VO1      VO2

0.55421E-01      0.25191E-01      270      0.38073E-01      0.66160E-01

CPH= 0.16691E-02      CPL= 0.12240E-02      CKGH= 0.13001E-00      CKGL= 0.33515E-00

END VALUES OF LOW PRESSURE COLUMN WHEN N= 1

END VALUES    1      NW(1)      YY(1)

1      0.14090272E-03      0.99999979E-02

21      0.41145575E-03      0.29420532E-01

41      0.41342736E-03      0.29723972E-01

61      0.41345530E-03      0.29725978E-01

81      0.41348324E-03      0.29727984E-01

101      0.41351118E-03      0.29729990E-01



|     |                  |                   |
|-----|------------------|-------------------|
| 141 | 0.41352492E-03   | 0.29731117E-01    |
| 161 | 0.41352492E-03   | 0.29731117E-01    |
| 181 | 0.41352492E-03   | 0.29731117E-01    |
| 201 | 0.41352492E-03   | 0.29731117E-01    |
| 221 | 0.41352492E-03   | 0.29731117E-01    |
| 241 | 0.41352492E-03   | 0.29731117E-01    |
| 261 | 0.41352492E-03   | 0.29731117E-01    |
| 281 | 0.41352492E-03   | 0.29731117E-01    |
| 301 | 0.41352492E-03   | 0.29731117E-01    |
| 321 | 0.41352492E-03   | 0.29731117E-01    |
| 341 | 0.41352492E-03   | 0.29731117E-01    |
| 361 | 0.41352492E-03   | 0.29731117E-01    |
| 381 | 0.41352492E-03   | 0.29731117E-01    |
| 401 | 0.41352492E-03   | 0.29731117E-01    |
| 421 | 0.41352492E-03   | 0.29731117E-01    |
| 441 | 0.41352492E-03   | 0.29731117E-01    |
| 461 | 0.41352492E-03   | 0.29731117E-01    |
| 481 | 0.41352492E-03   | 0.29731117E-01    |
| 501 | 0.41352492E-03   | 0.29731117E-01    |
| 521 | 0.41352492E-03   | 0.29731117E-01    |
| 541 | 0.41352492E-03   | 0.29731117E-01    |
| 561 | 0.41352492E-03   | 0.29731117E-01    |
| 581 | 0.41352492E-03   | 0.29731117E-01    |
| 601 | 0.41352492E-03   | 0.29731117E-01    |
| M=1 | YYAV=0.29677E-01 | YYAV1=0.29731E-01 |

END VALUES OF HIGH-PRESSURE COLUMN WHEN M=2

| END VALUES | I  | WH(I)          | YY(I)          |
|------------|----|----------------|----------------|
|            | 1  | 0.41672844E-03 | 0.9999979E-02  |
|            | 21 | 0.41356171E-03 | 0.99110119E-02 |
|            | 41 | 0.41356171E-03 | 0.99110119E-02 |
|            | 61 | 0.41356171E-03 | 0.99110119E-02 |

|      |                   |                    |
|------|-------------------|--------------------|
| 141  | 0.41356171E-03    | 0.99110119E-02     |
| 161  | 0.41356171E-03    | 0.99110119E-02     |
| 181  | 0.41356171E-03    | 0.99110119E-02     |
| 201  | 0.41356171E-03    | 0.99110119E-02     |
| 221  | 0.41356171E-03    | 0.99110119E-02     |
| 241  | 0.41356171E-03    | 0.99110119E-02     |
| 261  | 0.41356171E-03    | 0.99110119E-02     |
| 281  | 0.41356171E-03    | 0.99110119E-02     |
| 301  | 0.41356171E-03    | 0.99110119E-02     |
| 321  | 0.41356171E-03    | 0.99110119E-02     |
| 341  | 0.41356171E-03    | 0.99110119E-02     |
| 361  | 0.41356171E-03    | 0.99110119E-02     |
| 381  | 0.41356171E-03    | 0.99110119E-02     |
| 401  | 0.41356171E-03    | 0.99110119E-02     |
| 421  | 0.42173592E-03    | 0.10106854E-01     |
| 441  | 0.42176075E-03    | 0.10107640E-01     |
| 461  | 0.42177906E-03    | 0.10107934E-01     |
| 481  | 0.42177900E-03    | 0.10107934E-01     |
| 501  | 0.42176596E-03    | 0.10107670E-01     |
| 521  | 0.42173825E-03    | 0.10106999E-01     |
| 541  | 0.42171055E-03    | 0.10106340E-01     |
| 561  | 0.42168284E-03    | 0.10105677E-01     |
| 581  | 0.42082253E-03    | 0.10093942E-01     |
| 601  | 0.18430728E-03    | 0.50751120E-02     |
| N= 2 | YYAV= 0.44062E-02 | YYAV1= 0.50751E-02 |

END VALUES OF LOW PRESSURE COLUMN WHEN N= 2

| END VALUES | 1   | WN(I)          | YY(I)          |
|------------|-----|----------------|----------------|
|            | 1   | 0.69009053E-04 | 0.50721057E-02 |
|            | 21  | 0.41084317E-03 | 0.29327113E-01 |
|            | 41  | 0.41342713E-03 | 0.29723957E-01 |
|            | 61  | 0.41345507E-03 | 0.29725958E-01 |
|            | 81  | 0.41348501E-03 | 0.29727977E-01 |
|            | 101 | 0.41351504E-03 | 0.29729977E-01 |

161 0.41352492E-03 0.29731125E-01

181 0.41352492E-03 0.29731125E-01

201 0.41352492E-03 0.29731125E-01

221 0.41352492E-03 0.29731125E-01

241 0.41352492E-03 0.29731125E-01

261 0.41352492E-03 0.29731125E-01

281 0.41352492E-03 0.29731125E-01

301 0.41352492E-03 0.29731125E-01

321 0.41352492E-03 0.29731125E-01

341 0.41352492E-03 0.29731125E-01

361 0.41352492E-03 0.29731125E-01

381 0.41352492E-03 0.29731125E-01

401 0.41352492E-03 0.29731125E-01

421 0.41352492E-03 0.29731125E-01

441 0.41352492E-03 0.29731125E-01

461 0.41352492E-03 0.29731125E-01

481 0.41352492E-03 0.29731125E-01

501 0.41352492E-03 0.29731125E-01

521 0.41352492E-03 0.29731125E-01

541 0.41352492E-03 0.29731125E-01

561 0.41352492E-03 0.29731125E-01

581 0.41352492E-03 0.29731117E-01

601 0.41352492E-03 0.29731117E-01

N= 2 YYAV= 0.29677E-01 YYAV1= 0.29731E-01

END VALUES OF HIGH PRESSURE COLUMN WHEN N= 6

| END VALUES | I  | NW(I)          | YY(I)          |
|------------|----|----------------|----------------|
|            | 1  | 0.41579013E-03 | 0.99999979E-02 |
|            | 21 | 0.40641380E-03 | 0.97396635E-02 |
|            | 41 | 0.40644174E-03 | 0.97403340E-02 |
|            | 61 | 0.40648129E-03 | 0.97406060E-02 |

101 0.40645129E-03 0.97406060E-02

121 0.40645129E-03 0.97406060E-02

141 0.40645129E-03 0.97406060E-02

161 0.40645129E-03 0.97406060E-02

181 0.40645129E-03 0.97406060E-02

201 0.40645129E-03 0.97406060E-02

221 0.40645129E-03 0.97406060E-02

241 0.40645106E-03 0.97406022E-02

261 0.40645106E-03 0.97406022E-02

281 0.40645106E-03 0.97406022E-02

301 0.40645106E-03 0.97406022E-02

321 0.40645106E-03 0.97406022E-02

341 0.40645106E-03 0.97406022E-02

361 0.40645106E-03 0.97406022E-02

381 0.40663220E-03 0.97430348E-02

401 0.41740481E-03 0.99813603E-02

421 0.43005287E-03 0.10306165E-01

441 0.43008546E-03 0.10306951E-01

461 0.43007542E-03 0.10306805E-01

481 0.43005706E-03 0.10306355E-01

501 0.43003797E-03 0.10305904E-01

521 0.43001864E-03 0.10305427E-01

541 0.43041515E-03 0.10314733E-01

561 0.42922865E-03 0.10294028E-01

581 0.44765317E-03 0.87158494E-02

601 0.79939925E-04 0.95096882E-03

M = 6 YYAV = 0.69375E-03 YYAVI = 0.95097E-03

END VALUES OF LHM PRESSURE COLUMN WHEN M = 6

END VALUES I MW(I) YY(I)  
1 0.11904299E-04 0.94966567E-03

21 0.28005507E-03 0.17918322E-01

GLT (array) reading in file

|      |                   |                    |
|------|-------------------|--------------------|
| 11   | 0.42157900E-03    | 0.30330010E-01     |
| 11   | 0.42205094E-03    | 0.30363976E-01     |
| 161  | 0.42207609E-03    | 0.30365809E-01     |
| 121  | 0.42209541E-03    | 0.30367190E-01     |
| 141  | 0.42211437E-03    | 0.30368562E-01     |
| 161  | 0.42213313E-03    | 0.30369914E-01     |
| 181  | 0.42215199E-03    | 0.30371266E-01     |
| 201  | 0.42021926E-03    | 0.30270241E-01     |
| 221  | 0.40677423E-03    | 0.29266614E-01     |
| 241  | 0.40634256E-03    | 0.29214732E-01     |
| 261  | 0.40634256E-03    | 0.29214729E-01     |
| 281  | 0.40634256E-03    | 0.29214729E-01     |
| 301  | 0.40634256E-03    | 0.29214729E-01     |
| 321  | 0.40634256E-03    | 0.29214736E-01     |
| 341  | 0.40634256E-03    | 0.29214736E-01     |
| 361  | 0.40634256E-03    | 0.29214729E-01     |
| 381  | 0.40634279E-03    | 0.29214755E-01     |
| 401  | 0.40634279E-03    | 0.29214747E-01     |
| 421  | 0.40634279E-03    | 0.29214751E-01     |
| 441  | 0.40634279E-03    | 0.29214747E-01     |
| 461  | 0.40634279E-03    | 0.29214751E-01     |
| 481  | 0.40634279E-03    | 0.29214751E-01     |
| 501  | 0.40634279E-03    | 0.29214751E-01     |
| 521  | 0.40634279E-03    | 0.29214751E-01     |
| 541  | 0.40634279E-03    | 0.29214755E-01     |
| 561  | 0.40634256E-03    | 0.29214736E-01     |
| 581  | 0.40632370E-03    | 0.29213689E-01     |
| 601  | 0.40691276E-03    | 0.29231593E-01     |
| H= 6 | YYAV= 0.29267E-01 | YYAV1= 0.29232E-01 |

END VALUES OF HIGH-PRESSURE COLUMN WHEN H=10

| END VALUES | I | HW(I)          | YY(I)          |
|------------|---|----------------|----------------|
|            | 1 | 0.41492586E-03 | 0.99999979E-02 |

81 0.39973636E-03 0.95797151E-02  
101 0.39974018E-03 0.95797140E-02  
121 0.39974018E-03 0.95797140E-02  
141 0.39973995E-03 0.95797103E-02

161 0.39973948E-03 0.95797628E-02

181 0.39973971E-03 0.95797628E-02

201 0.39973971E-03 0.95797628E-02

221 0.39973948E-03 0.95797628E-02

241 0.39973925E-03 0.95797591E-02

261 0.39973925E-03 0.95797591E-02

281 0.39973925E-03 0.95797591E-02

301 0.39973948E-03 0.95797591E-02

321 0.39973925E-03 0.95797591E-02

341 0.39973925E-03 0.95797591E-02

361 0.39974344E-03 0.95798112E-02

381 0.40399469E-03 0.96634477E-02

401 0.42454922E-03 0.10140067E-01

421 0.43754606E-03 0.10485742E-01

441 0.4375909E-03 0.10486156E-01

461 0.43752720E-03 0.10485996E-01

481 0.43748249E-03 0.10484353E-01

501 0.43739076E-03 0.10482296E-01

521 0.43725618E-03 0.10478806E-01

541 0.43762964E-03 0.10491809E-01

561 0.40963129E-03 0.99498630E-02

581 0.41869749E-03 0.49883313E-02

601 0.69931612E-05 0.23496668E-03

H=10 YYAV=0.15970E-03 YYAV1=0.23497E-03

END VALUES OF LOW PRESSURE COLUMN WHEN H=10

END VALUES (HW(1)) (YY(1))

1 0.28551750E-05 0.20301177E-05

101 0.429659381-03 0.368889081-01

121 0.429700371-03 0.368899681-01

141 0.429842391-03 0.369000281-01

161 0.429875921-03 0.369000781-01

181 0.429904321-03 0.369000781-01

201 0.427010991-03 0.368210991-01

221 0.405571421-03 0.292535091-01

241 0.399611091-03 0.287354071-01

261 0.399563001-03 0.287273121-01

281 0.399563001-03 0.287273081-01

301 0.399563001-03 0.287273121-01

321 0.399563001-03 0.287273151-01

341 0.399563001-03 0.287273121-01

361 0.399563001-03 0.287273081-01

381 0.399563461-03 0.287273231-01

401 0.399563461-03 0.287273341-01

421 0.399563231-03 0.287273381-01

441 0.399563231-03 0.287273231-01

461 0.399563231-03 0.287273361-01

481 0.399563691-03 0.287273341-01

501 0.399563461-03 0.287273411-01

521 0.399563461-03 0.287273271-01

541 0.399553111-03 0.287269171-01

561 0.399527371-03 0.287249161-01

581 0.399490591-03 0.287224431-01

601 0.400814461-03 0.287676791-01

N=10 YYAV=0.288931-01 YYAV1=0.287688-01

| N | <FIP>       | <YBP>       | YPT/YU      | YBP/YU      |
|---|-------------|-------------|-------------|-------------|
| 1 | 0.100001-01 | 0.296771-01 | 0.100001-01 | 0.296771-01 |
| 2 | 0.440621-02 | 0.296771-01 | 0.440621-00 | 0.296771-01 |
| 3 | 0.279021-02 | 0.294651-01 | 0.279021-00 | 0.294651-01 |
| 4 | 0.164321-02 | 0.294651-01 | 0.164321-00 | 0.294651-01 |
| 5 | 0.108741-02 | 0.292671-01 | 0.108741-00 | 0.292671-01 |
| 6 | 0.693751-03 | 0.292671-01 | 0.693751-01 | 0.292671-01 |
| 7 | 0.476531-03 | 0.296771-01 | 0.476531-01 | 0.296771-01 |

|   |             |             |             |             |
|---|-------------|-------------|-------------|-------------|
| 6 | 0.321691-03 | 0.290771-01 | 0.320691-01 | 0.290771-01 |
| 8 | 0.229171-03 | 0.288931-01 | 0.229171-01 | 0.288931-01 |