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AN ECONOMIC EVALUATION
OF A PRESSURE SULFURIC ACID
MANUFACTURING PLANT

BY

DANIEL A. MARLIN

A THESIS

PRESENTED IN PARTIAL FULFILLMENT OF
THE REQUIREMENTS FOR THE DEGREE
OF
MASTERS OF SCIENCE IN CHEMICAL ENGINEERING
AT
NEWARK COLLEGE OF ENGINEERING

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NEWARK, NEW JERSEY

1970

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FOR

DEPARTMENT OF CHEMICAL ENGINEERING
NEWARK COLLEGE OF ENGINEERING

BY

FACULTY COMMITTEE

APPROVED: _____

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1970

The author wishes to thank Dr. Deran Hanesian for his guidance in carrying this work to completion. Also, thanks to Miss M. Foernsler for the preparation of the manuscript.

Dedicated with all my love to Mary Frances and a little girl named Kimberly Ann.

ABSTRACT

The production of Sulfuric Acid at elevated pressures is presented with the economics for operating at one, three, five, seven, nine and thirteen atmospheres. The conventional and pressure plants are compared operating on the same basis.

The basis for a pressure plant design is obtained from an I.B.M. 1130 computer program developed for One Thousand Net Tons per Day of Sulfuric Acid. Recognized operating conditions for a conventional acid plant are incorporated in the pressure plant design program.

Sulfuric Acid manufacture with operating pressures higher than atmospheric, but less than five atmospheres, results in maximum rate of return on investment and operating economy.

A compressor and gas expander arrangement can significantly reduce operating costs in a pressure plant without sacrifice to the steam generation capabilities associated with sulfuric acid production.

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AN ECONOMIC EVALUATION
OF A PRESSURE SULFURIC ACID
MANUFACTURING PLANT
INTRODUCTION

The chamber and contact process are the two main procedures used to commercially manufacture sulfuric acid.⁽⁵⁾ In each process, the operating pressure is atmospheric.

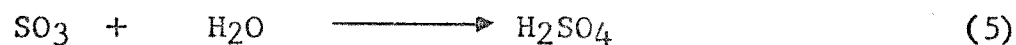
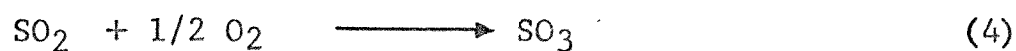
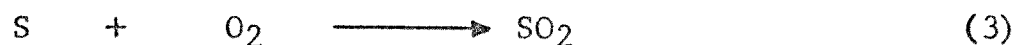
Laboratory demonstrations have shown the manufacture of sulfuric acid benefits from higher operating pressures. Zeisberg ⁽¹⁵⁾ and Busching ⁽¹⁾ have shown that both higher yields in sulfur dioxide conversion and effectively improved sulfur trioxide absorption are observed at elevated pressures.

Relatively few articles have been written describing the detail economics associated with a pressure sulfuric acid plant. Richterova ⁽¹³⁾ cites a reduction in both plant investment and operating costs using a gas expander. Cathala ⁽⁴⁾ suggests a minimum conversion pressure should be three atmospheres.

A comparison study is undertaken to define the economics, based on return on investment, for a pressure and conventional sulfuric acid plant.

DESIGN OF PROCESSIntroduction

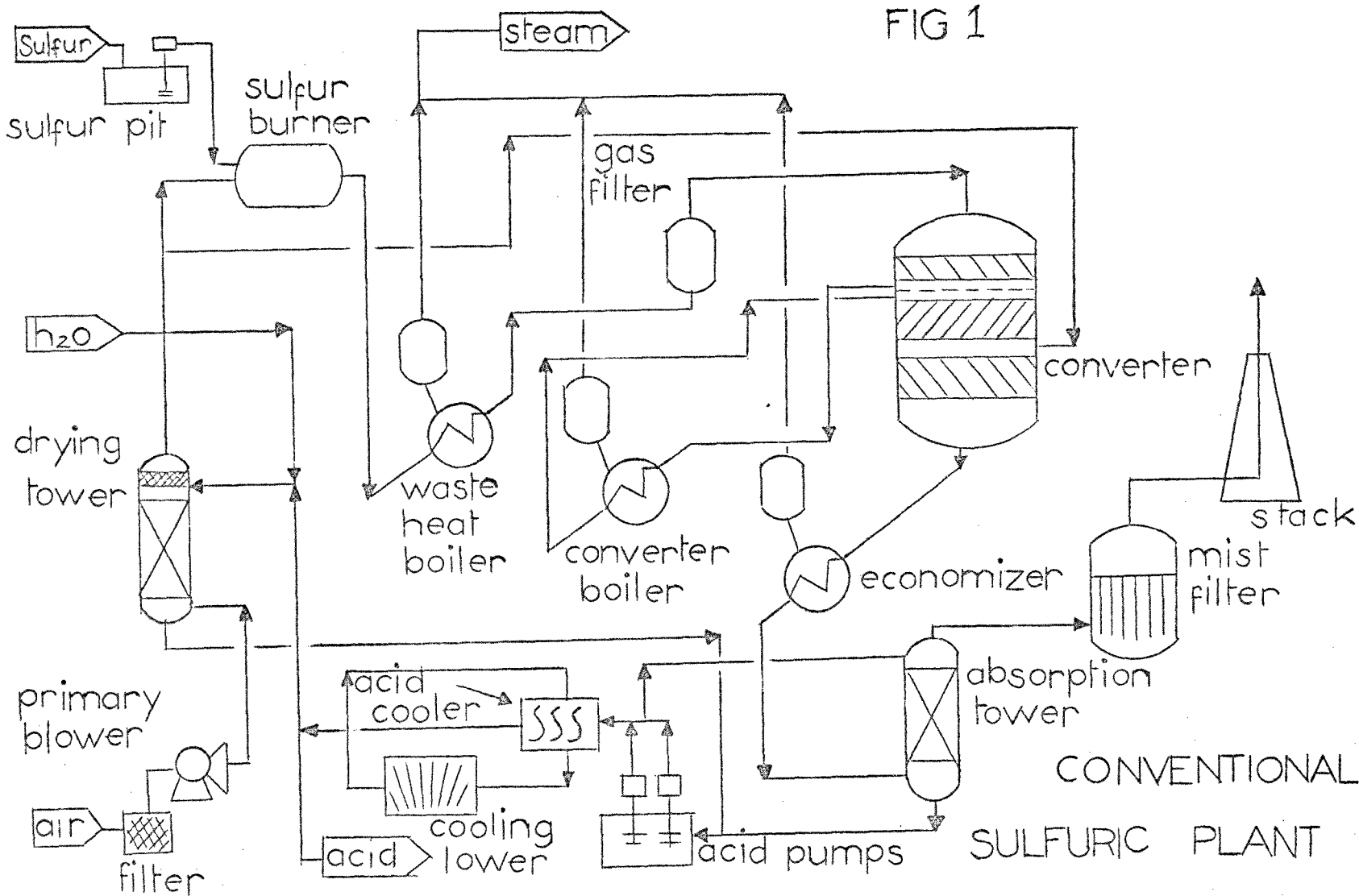
Sulfuric acid manufacture involves three exothermic reactions.



The conventional acid plant using the contact process and sulfur burning as the source of sulfur dioxide is schematically shown in Figure 1.

The burning of sulfur, described with equation (3), is accomplished in the presence of excess air which has been dried with concentrated acid. The reaction products leave the sulfur burner at the adiabatic flame temperature and are then cooled and filtered before entering the sulfur dioxide converter.

The oxidation of sulfur dioxide, reaction (4), is carried out in an adiabatic fixed bed catalytic reactor. The equilibrium characteristics and high degree of conversion required in sulfuric acid manufacture is



obtained in several stages or beds with cooling between stages. The reaction gases are cooled indirectly after the first bed and directly after subsequent beds.

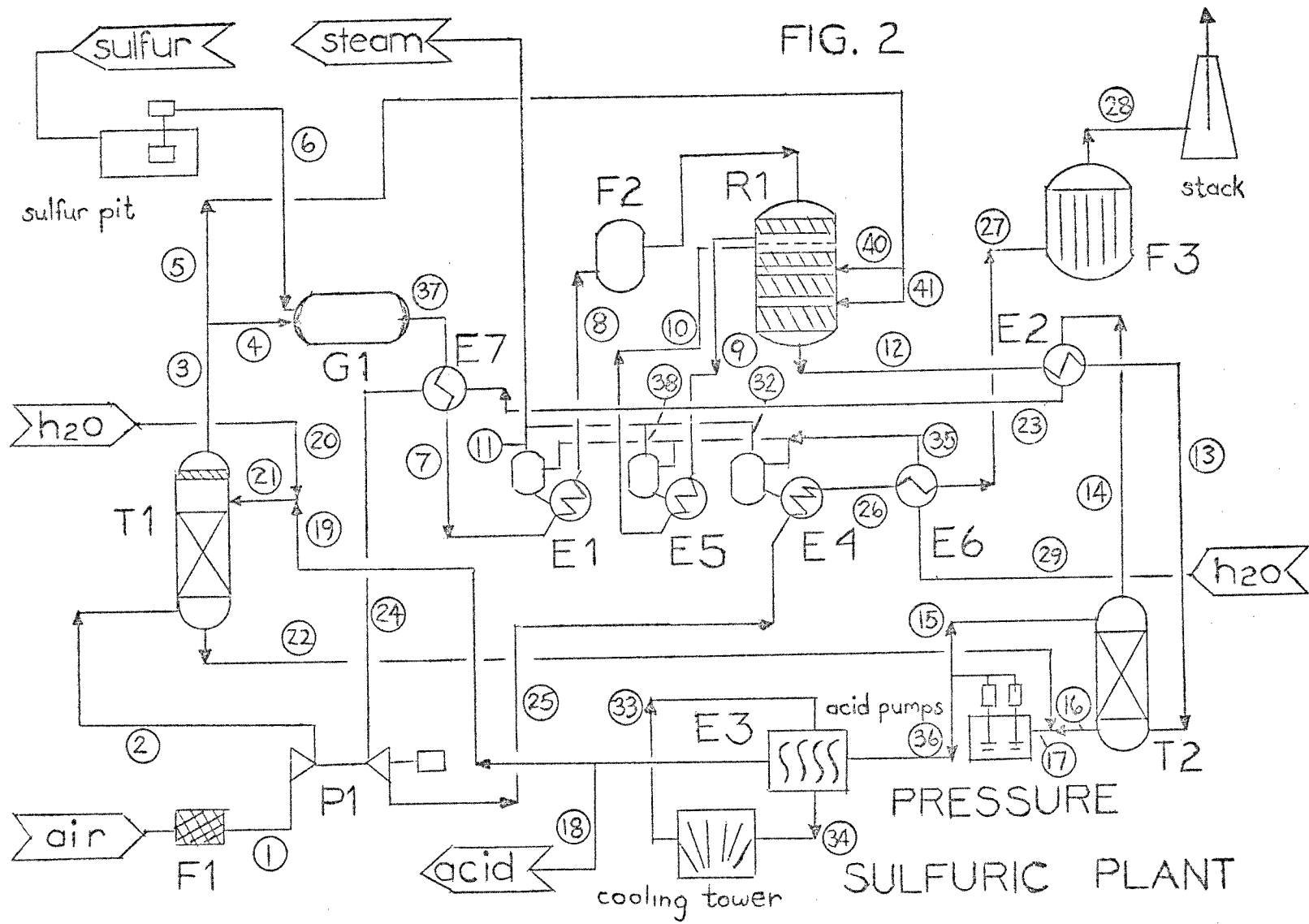
Prior to entering the absorption column, the converter gases are cooled in a waste heat boiler.

Energy released in cooling gases leaving the sulfur burner, and first and last bed of the converter is recovered as steam.

Finally, the absorption of sulfur trioxide in concentrated acid is accomplished in a ceramic ring packed tower. The gases leaving the absorption tower are essentially free of sulfur trioxide. The emission however, does contain unconverted sulfur dioxide and acid mist. Acid mist is removed in the mist filter and unconverted sulfur dioxide is released to the atmosphere.

The primary blower, usually operating at a discharge pressure less than five psig, delivers the process gases throughout the plant.

A proposed pressure plant, shown in Figure 2, closely resembles the conventional plant except an air



compressor-expander replaces the primary blower and heat exchangers are re-arranged to provide energy recovery. The heat exchanger arrangement generates steam and provides a high temperature gas feed to the expander required for operating cost economy. The equipment numbers, introduced in Figure 2, are defined in Tables III, IV and V.

For the purposes of this study the following operating conditions and rates are used to illustrate the effect of pressure on plant economics.

Rate = 1000 N.T. Sulfuric per Day

Converter Feed Concentration = 8 percent SO_2

Efficiency = 95 percent (based on sulfur)

Overall Conversion = 99 percent

Approach to Equilibrium = 95 percent of equil. conver.

The above conditions could be varied by modifications in the input data of the computer program.

Conditions that are considered fixed, based on operating guidelines for conventional plants, are listed in Table I. The stream numbers are identified in Figure 2.

TABLE I
OPERATING GUIDELINES FOR A
SULFURIC ACID PLANT

Stream Number	Assumptions (5)
(1)	Relative humidity equals 50 percent
(3)	Water content at 1 mg per cu. ft. and at a temperature 1 degree less than (21)
(6)	Temperature at 127 deg. F
(8)	Temperature at 450 deg. C
(11) & (32) & (38)	Saturated steam at 250 lbs. per sq. in.
(13)	Temperature at 450 deg. F
(14)	Temperature at 1 deg. > than (15)
(15) & (21)	Circulation rate at 2.5 gal per daily ton of acid, gravity at 15.2 lbs. per gal.
(10) & (40) & (41)	Converter feed temperature at 400 deg. C
(18)	Acid temperature at 100 deg. F and concentration at 98 percent
(16) & (22)	Heats of dilution based on graphical data D.&W. (5) p. 440
(24)	Temperature at 1000 deg. F
(26)	Temperature at 450 deg. F
(27)	Temperature at 250 deg. F
(37)	Adiabatic flame temperature at 890 deg. C based on 8 percent SO ₂ from burner

I. Heat and Material Balance

The calculations for the heat and material balance follow the conventional mole balance starting with an overall material balance. Assumed thermal conditions are matched by successive iterations through the plant balance.

Enthalpy coefficients are derived for the process gases in the plant using the definition of heat capacity as a function of temperature. (14)

$$C_{pi}^* = \alpha_i + \beta_i T + \gamma_i T^2 \quad (6)$$

$$C_{PMIXTURE}^* = y_A C_{pA} + y_B C_{pB} + y_C C_{pC} + \dots \quad (7)$$

$$\Delta H_i = m_i \times 1.8 \int_{T_0}^T C_{pi}^* dT = \sum_i m_i \times COFi \quad (8)$$

The total change in enthalpy resulting from heating and cooling gases is developed for plant components.

$$\Delta H = \sum_i \Delta H_i \equiv Btu \quad (9)$$

Using the reference temperature, T_0 , equal to 298°K, the component coefficients are calculated.

$$COF_{N_2} : m_{N_2} [11.743 * T + 1.125 * 10^{-3} * T^2 - 3599.] \quad (10)$$

$$COF_{O_2} : m_{O_2} [11.066 * T + 2.792 * 10^{-3} * T^2 - 3546.] \quad (11)$$

$$\text{COF SO}_2 : \eta_{\text{SO}_2} \left[12.809 * T + 8.561 \times 10^{-3} * T^2 - 4578. \right] \quad (12)$$

$$\text{COF SO}_3 : \eta_{\text{SO}_3} \left[10.939 * T + 21.182 \times 10^{-3} * T^2 - 5141. \right] \quad (13)$$

Physical properties used in the balance are used from average operating conditions for the equipment. (See Appendix)

The logic and sequence of calculations are summarized in a computer block diagrams shown in Figure 3, 4 and 5.

II. Equipment Design

The design calculations for the various equipment in the pressure plant are located in Subprograms N89 and N90. Design equations and assumptions are described in this section.

(1) Converter design. The conversion of sulfur dioxide to sulfur trioxide is accomplished in a fixed bed catalytic reactor operating under adiabatic conditions. The high degree of conversion required in sulfuric acid manufacture is achieved by removing the large quantities of heat evolved by the heat of reaction.

The path of the reaction is illustrated in Figure 6.

Feed enters the converter, at point g, with essentially zero conversion and at a temperature greater than the ignition temperature of the catalyst. On contact

FIG 3
HEAT AND MATERIAL BALANCE
SUBPROGRAMS M88, N88
BLOCK DIAGRAM

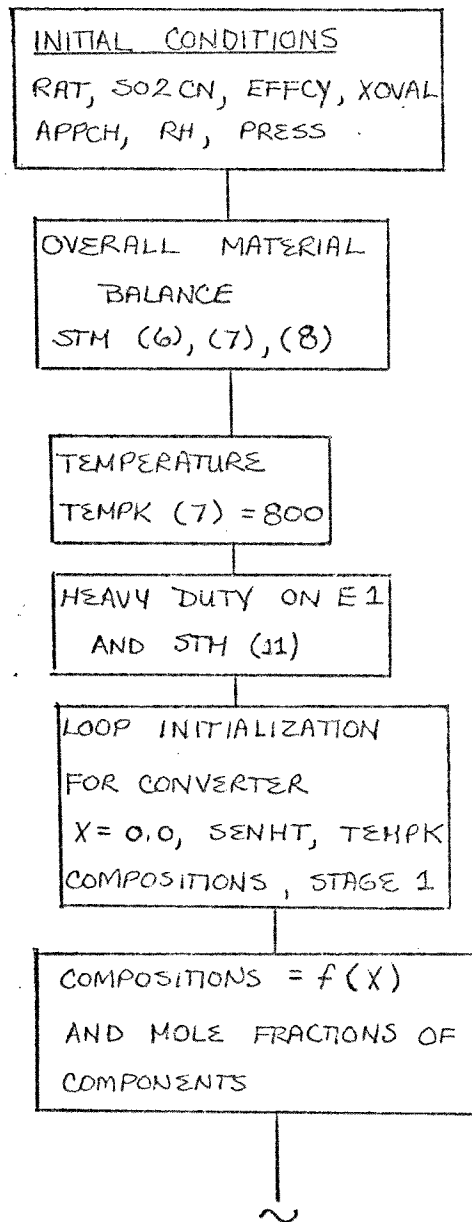


FIG4
HEAT AND MATERIAL BALANCE
M88 BLOCK DIAGRAM
(continued)

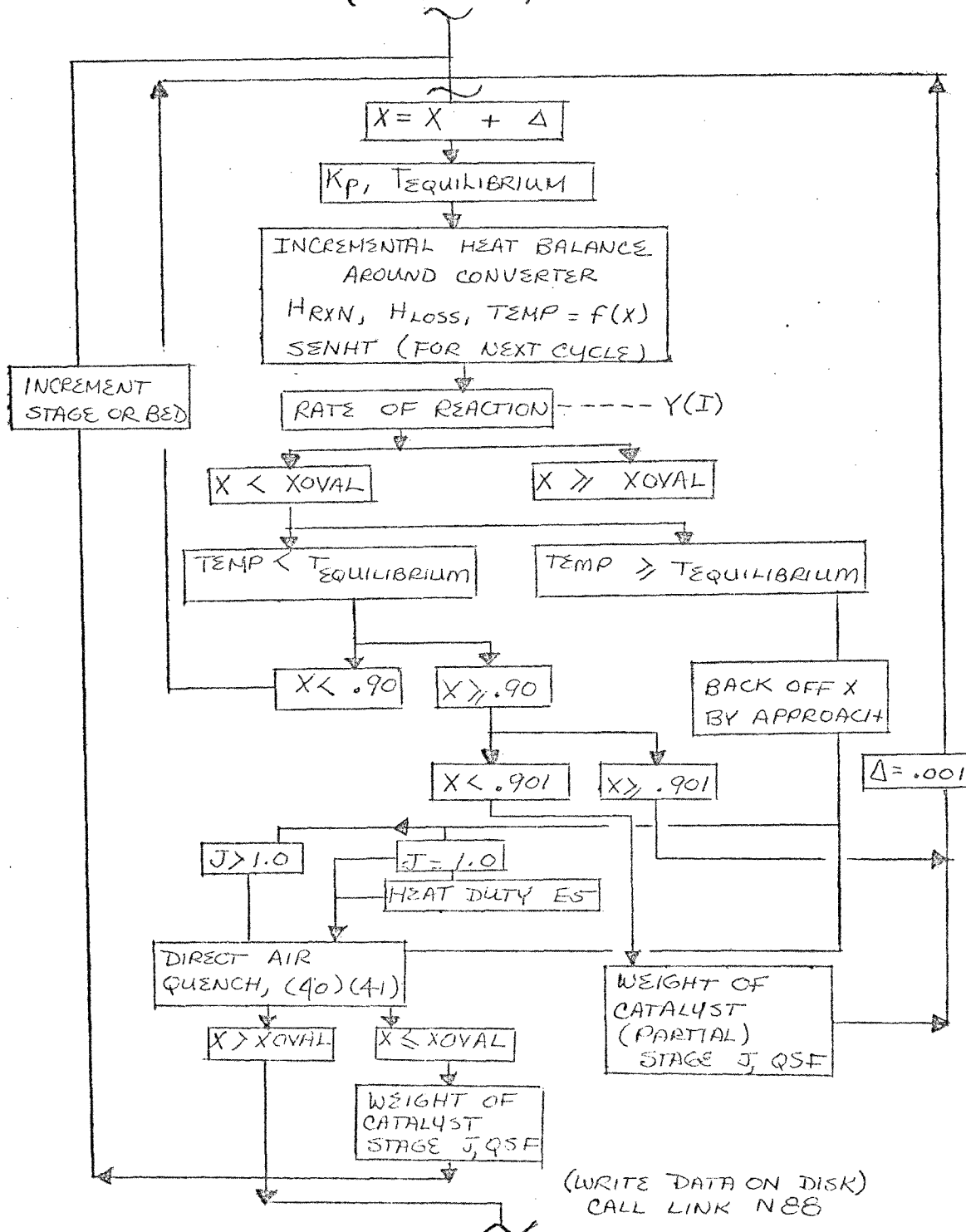


FIG 5
HEAT AND MATERIAL BALANCE
N88 BLOCK DIAGRAM

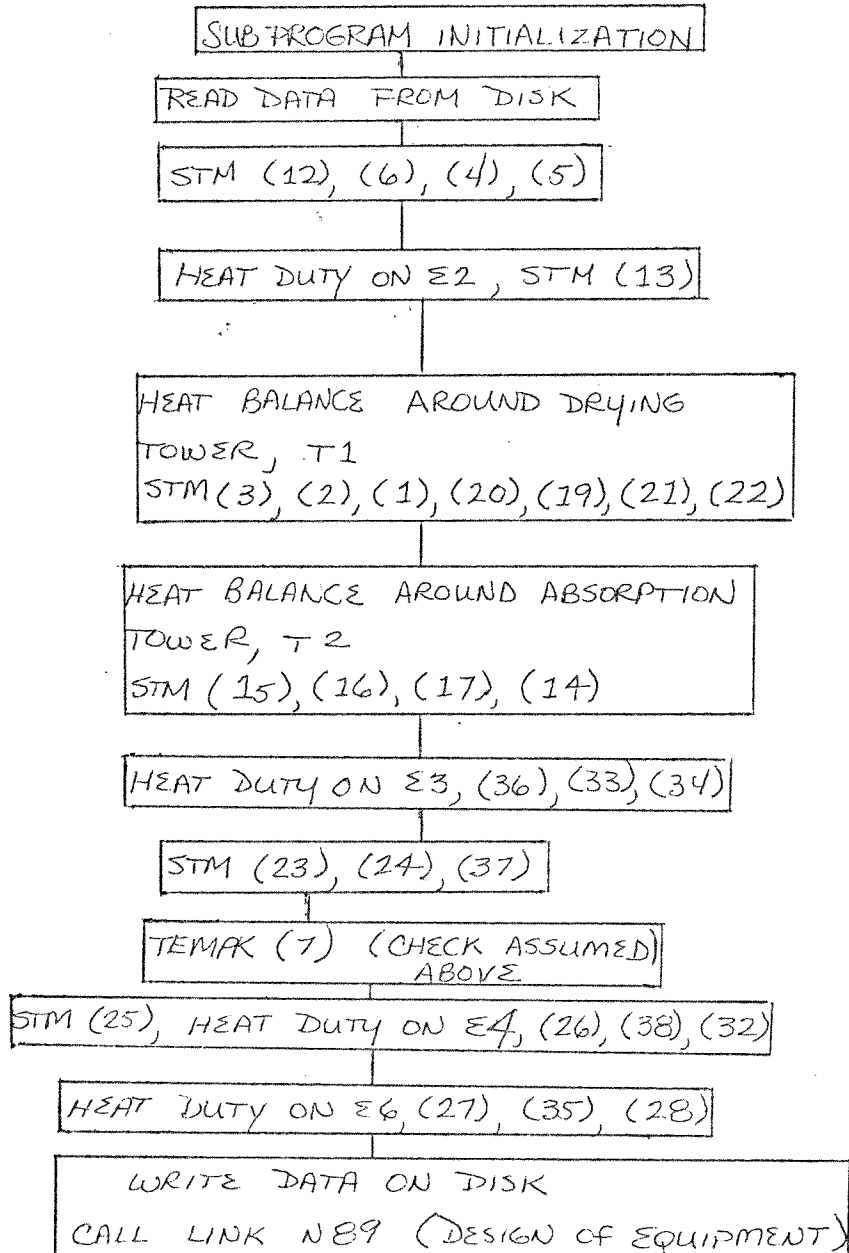
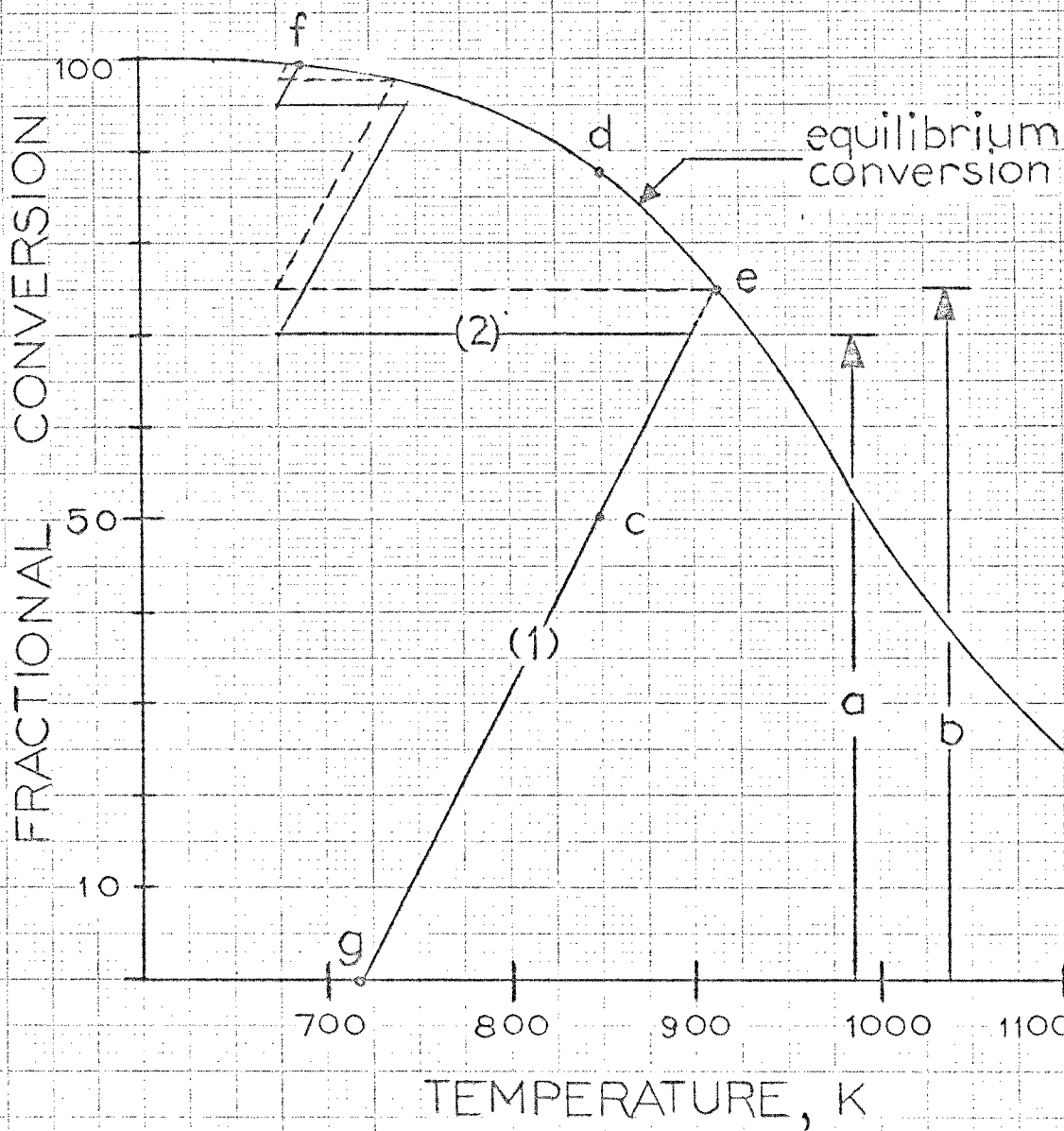


FIG. 6
CONVERSION
VS
TEMPERATURE



with vanadium pentoxide catalyst, the converter gas temperature increases along path (1) until the equilibrium conversion line is reached.

(a) Converter heat balance. An incremental heat balance equation is used to determine the resulting gas temperature for each unit conversion.⁽⁵⁾

$$\left(\begin{array}{c} \text{SENSIBLE HEAT} \\ \text{OF} \\ \text{REACTION PRODUCTS} \end{array} \right) = \left(\begin{array}{c} \text{SENSIBLE HEAT} \\ \text{OF} \\ \text{REACTANTS} \end{array} \right) - (\Delta H_{\text{RXN}} * X) \quad (14)$$

Incorporating the enthalpy coefficients for the reaction gases developed in the previous heat and material balance section, the incremental heat balance is expressed as a second order linear equation.

$$COFA * T^2 + COFB * T + COFC = 0 \quad (15)$$

The coefficients of T squared in the gas enthalpy equation are expressed as a function of the moles of the various components.

$$\begin{aligned} & [m_{N_2} * 1.125 * 10^{-3}] + [m_{O_2} * 2.792 * 10^{-3}] + \\ & [m_{SO_2} * 8.561 * 10^{-3}] + [m_{SO_3} * 21.182 * 10^{-3}] \end{aligned} \quad (16)$$

Likewise COFB is expressed in similar terms.

$$[m_{N_2} * 11.743] + [m_{O_2} * 11.066] + [m_{SO_2} * 12.809] + [m_{SO_3} * 10.939] \quad (17)$$

The remaining elements in the heat balance are collected in COFC.

$$COFC = - \left(\begin{array}{c} \text{SENSIBLE} \\ \text{HEAT} \\ \text{OF REACTANTS} \end{array} \right) + (\Delta H_{RXN} * X) - H_{LOSS} + \text{CONSTANT} \quad (18)$$

The constant term collects residual terms in the gas enthalpy equation not covered with COFA or COFB.

$$\text{CONSTANT} = -3599 * m_{N_2} - 3546 * m_{O_2} - 4578 * m_{SO_2} + 5141 * m_{SO_3} \quad (19)$$

The resulting gas temperature for increments of conversion is obtained using the standard square root formula.

$$T = \frac{-COFB \pm (COFB^2 - 4COFA * COFC)^{1/2}}{2 * COFA} \quad (20)$$

The heat of reaction is expressed as a function of temperature using the heat capacity coefficients of the reaction gases. (5)

$$\Delta H_{RXN} = \Delta H_{RXN}^{\circ} + \Delta \alpha T + \frac{\Delta \beta}{2} T^2 + \frac{\Delta \gamma}{3} T^3 \quad (21)$$

where

$$\Delta H_{RXN}^{\circ} = -42,340 \frac{\text{Btu}}{\# \text{moles } SO_2} \text{ AT } 298^{\circ}K$$

In the final form:

$$\Delta H_{RXN} = -41,631 - 4,113T + 6.236 \times 10^{-3} T^2 - 1.246 \times 10^{-6} T^3 \quad (22)$$

(b) Converter equilibrium calculation. The true equilibrium constant for the reversible reaction is a product of three terms.

$$K = K_Y * K_\phi * K_P \quad (23)$$

K_Y is difficult to calculate and usually is assumed equal to 1.0. ⁽¹⁴⁾ Gas mixtures produced in sulfuric acid manufacture behave as ideal gases. A deviation < 2% has been calculated at pressures of 200 psig. For ideal gas behavior K_ϕ equals 1.0. Therefore, for the reaction $SO_2 + 1/2 O_2 \longrightarrow SO_3$:

$$K = K_P = K_Y \pi^{c_{SO_3} - c_{SO_2} - c_{O_2}} \quad (24)$$

$$K = K_P = \left(\frac{y_{SO_3}}{y_{SO_2} y_{O_2}^{1/2}} \right) * \pi^{-1/2} \quad (25)$$

Using an empirical equation for the equilibrium constant K_P , ⁽⁵⁾

$$\log_{10} K_P = 4956/T - 4.678 \equiv \text{ATM}^{-1/2} \quad (26)$$

A relation is obtained to calculate the equilibrium temperature for a gas mixture and absolute pressure.

$$T_{\text{EQUILIBRIUM}} = \frac{4956}{\text{Log}_{10} K_p + 4.678} \quad (27)$$

Equation (27) is used to determine when the equilibrium conversion line has been reached, along path (1). A comparison is made between the reaction gas temperature, calculated by the incremental heat balance, and the equilibrium temperature. This comparison is illustrated in Figure 6 using points C and D. A comparison continues until the maximum conversion for the first converter bed is obtained at point E. No further conversion is possible until the reaction gases are cooled along path (2).

Alternate adiabatic heating by the reaction and cooling between beds continues until the desired final conversion is obtained (point F.).

The cooling of reaction gases is accomplished by direct injection of air after the second and subsequent stages.

(c) Converter catalyst requirement. The quantity of catalyst required to obtain a specified conversion is dependent on the rate of reaction as described in the following equation.⁽²⁾

$$W = F \int_{x_1}^{x_2} \frac{1}{r} dx \quad (28)$$

At points close to the equilibrium conversion line, the weight of catalyst required becomes infinite due to the small concentration driving force. For the purposes of this study a practical approach to the equilibrium conversion point is set. Using the ordinate as a reference, the approach to equilibrium is fixed at 95 percent as illustrated in Figure 6.

$$\frac{\text{Approach to Equilibrium}}{\text{Equilibrium}} = \frac{a}{b} = 95 \text{ percent} \quad (29)$$

Design conversion is obtained by calculating the equilibrium conversion and backing off 5 percent to the design approach.

Several forms of the rate of reaction have been developed from experimental investigators. Calderbank's ⁽²⁾ equation for an operating pressure of 1

atmosphere is presented in equation (30)

$$r = (8.5 \times 10^6) \left\{ \left(e^{-31,000/RT + 12.07} \times \frac{P_{O_2} P_{SO_2}}{P_{SO_2}^{1/2}} \right) + \right. \\ \left. - \left(e^{-53,600/RT + 22.75} \times \frac{P_{SO_3} P_{O_2}^{1/2}}{P_{SO_2}^{1/2}} \right) \right\} \quad (30)$$

The total pressure contributes in the rate expression by using the definition of partial pressure of the gas in terms of mole fractions and total pressure.

$$P_i = y_i \pi \quad (31)$$

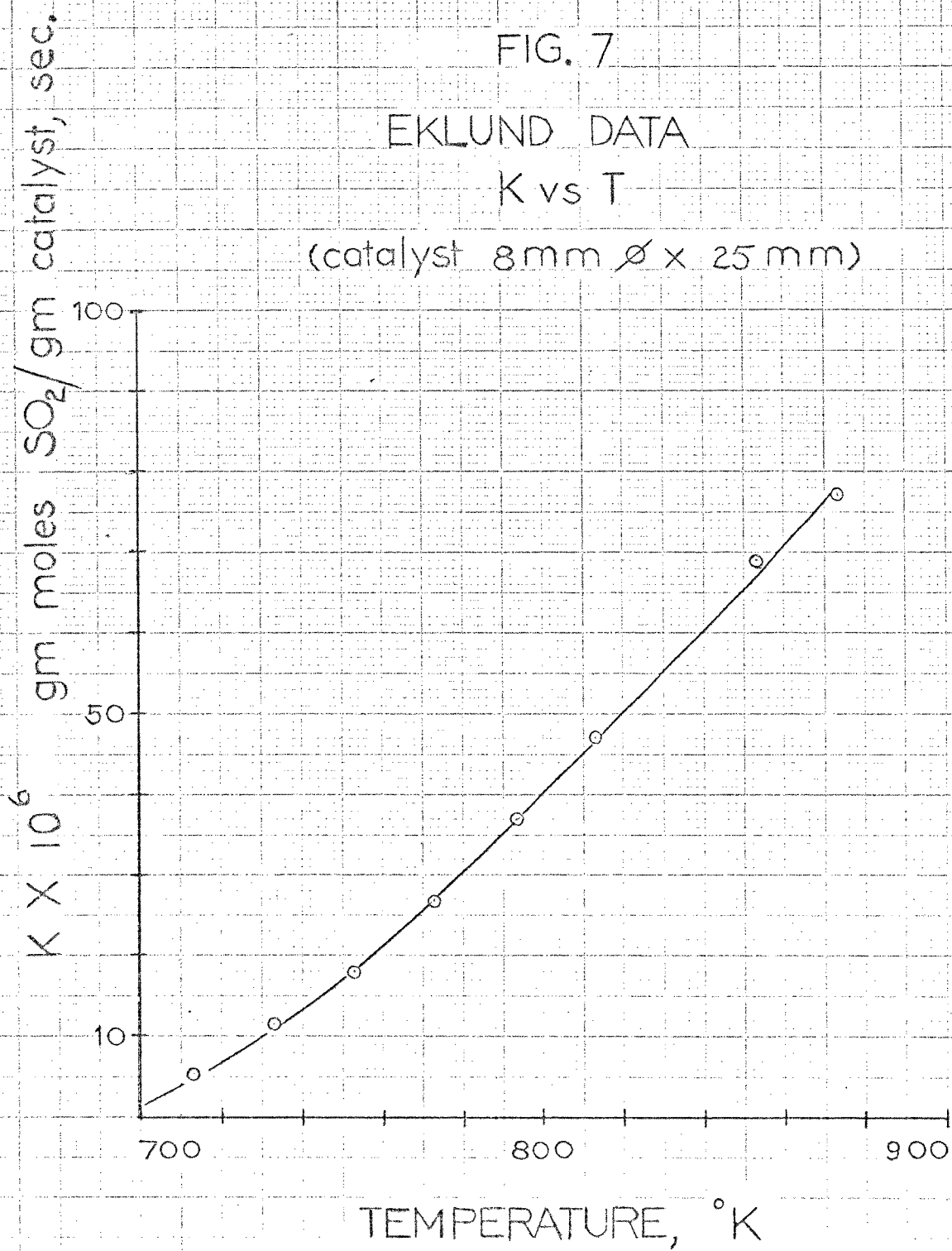
$$\text{ie } \left(\frac{P_{O_2} P_{SO_2}}{P_{SO_2}^{1/2}} \right) = \left(\frac{y_{O_2} y_{SO_2}}{y_{SO_2}^{1/2}} \right) \pi^{1.5} \quad (32)$$

$$\text{and } \left(\frac{P_{SO_3} P_{O_2}^{1/2}}{P_{SO_2}^{1/2}} \right) = \left(\frac{y_{SO_3} y_{O_2}}{y_{SO_2}^{1/2}} \right) \pi \quad (33)$$

A form of the rate expression which has been found to more closely characterize commercial converters is defined by Eklund. (5)

$$r = (8.5 \times 10^6) k^* \left(\frac{P_{SO_2}}{P_{SO_3}} \right)^{1/2} \left[P_{O_2} - \left(\frac{P_{SO_3}}{P_{SO_2} K_{eq}} \right)^2 \right] \quad (33A)$$

Eklund presents empirical data for k^* , the velocity constant, using 8 MM diameter by 25 MM catalyst. The data is graphically shown in Figure 7. The value of k^* can be expressed as a polynomial as a function of temperature. The results of a regression computer



program gives an expression with an average difference of 1.22 percent from empirical data.

$$k^* = \eta + \mu' T + \xi T^2 + \eta T^4 \quad (34)$$

where

$$\begin{aligned} \eta &= .004248 \\ \mu' &= - .00001471 \\ \xi &= .1411 \times 10^{-7} \\ \eta &= - .3580 \times 10^{-14} \end{aligned}$$

The final Eklund rate equation, including the effect of total operating pressure, is expressed in equation (35).

$$r = 8.5 \times 10^6 * k^* * \left(\frac{y_{SO_2}}{y_{SO_3}} \right)^{1/2} * \left(\left(y_{O_2} \pi - \frac{y_{SO_3}}{y_{SO_2} K_p} \right)^2 \right) \quad (35)$$

Laboratory studies (2) show the rate determining step in the oxidation of SO_2 is the rate of oxidation of chemisorbed SO_2 by oxygen in the gas phase. These results indicate the velocity constant, k^* , is dependent only on temperature and equation (35) should be valid at higher pressures.

The Scientific Subprogram, QSF, used to integrate quantities in equation 28, requires equal increments of conversion. On this basis, increments of conversion at 0.01 are used for conversion up to and including 90 percent. Further conversions are incremented by units equal to 0.001.

The size of the converter is established from the calculated catalyst per stage obtained in program M88. An additional 15 percent catalyst is added to each bed to compensate for gradual loss in catalyst activity with age.

A block diagram of the computer program used to calculate the quantity of catalyst and stages is shown in Figure 4.

(d) Converter diameter and height calculation. The diameter of the vessel is obtained on the basis of superficial gas velocity in the converter. Ideal gas behavior is used for the gas volume calculations. Using a design velocity of conventional plants at 100 cu. ft. per min. per square feet (S.C.), the allowable velocity is adjusted for pressure to maintain an approximate equivalent pressure drop per foot of catalyst. The

allowable superficial velocity is found to be inversely proportional to the square root of the total pressure. The converter cross sectional area and diameter are obtained from the ratio of volumetric gas flow per unit time and allowable superficial velocity. The converter height is obtained from the total catalyst per stage, in pounds, and a catalyst density at 40 pounds per cu. ft.

$$\text{Height of Catalyst} = \frac{\text{Volume of Catalyst}}{\text{Cross Sectional Area}} \quad (36)$$

An additional height of 15 percent times the diameter is allowed between stages for access and air sparge connections. Also an additional 15 percent is provided for the height of vessel above and below the first and last bed.

(e) Converter pressure drop calculation. The pressure drop across the converter is calculated from a correlation for flow through static beds. ⁽¹¹⁾

$$\Delta P = \frac{2f_m G^2 L}{144 D_p g_c f} \quad (37)$$

where $f_m = 79.5 / Re^{.375}$

and $Re = \frac{D_p G}{\mu}$

The computer program to calculate the converter design is shown in Figure 4.

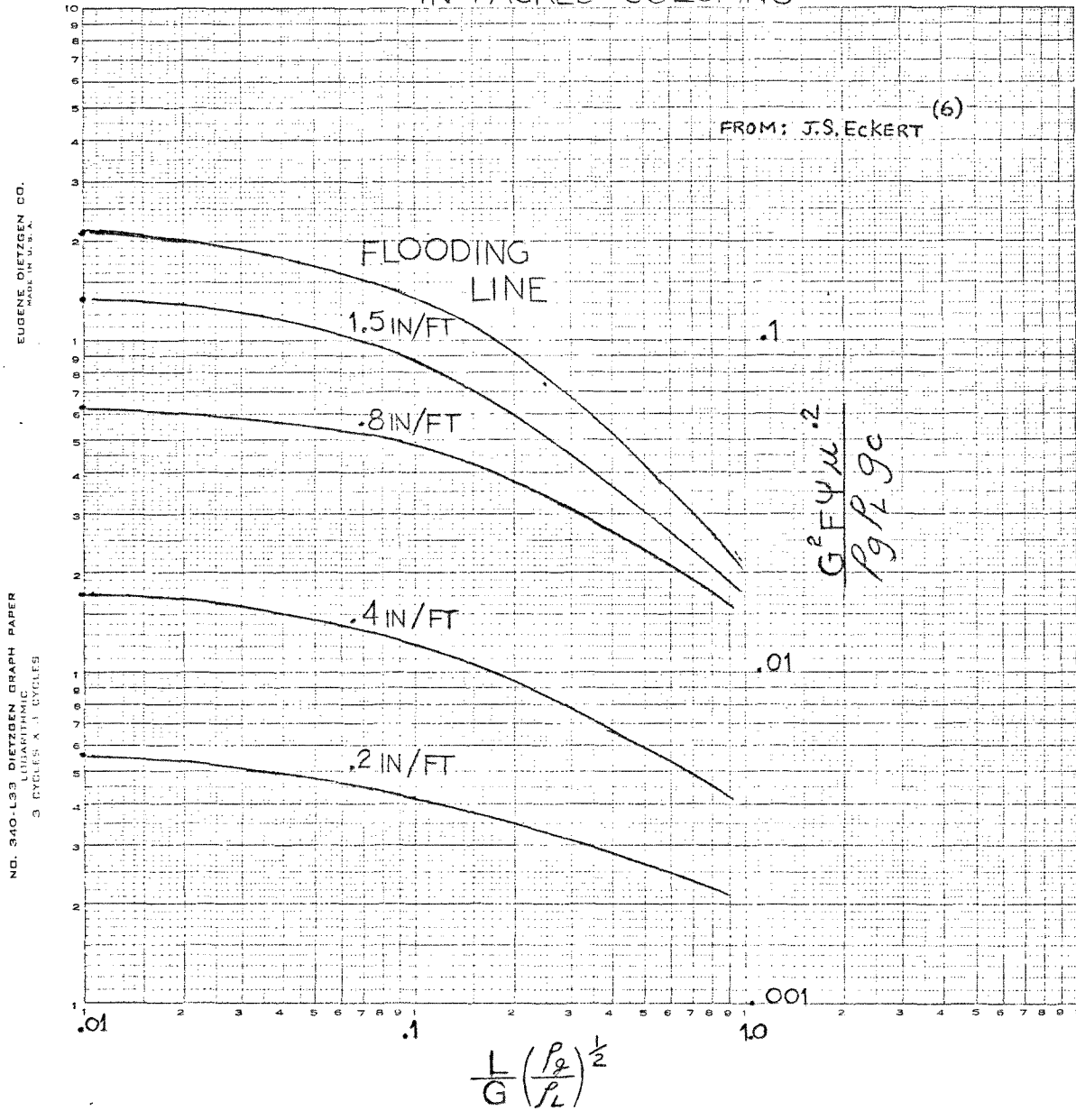
(2) Absorption and drying tower. The absorption of sulfur trioxide in concentrated sulfuric acid and the drying of air with concentrated acid requires intimate contact for both heat and mass transfer. Towers packed with ceramic raschig rings provide the surface necessary for both modes of transfer.

The relations used in a packed tower design are derived from empirical correlations developed by Eckert.⁽⁶⁾ A generalized correlation is shown in Figure 8.⁽⁶⁾

The general curve establishes the maximum allowable gas velocity in a packed column as a function of gas and liquid flow rates, physical properties and packing characteristics.

The flooding line on Figure 8 identifies the point at which the gas velocity prevents the "regular" down flow of liquid. For Raschig Rings, flooding generally occurs at pressure drops equal to 1.5 inches of water per foot of packing. The normal design basis is to select a

FIG. 8
GENERALIZED PRESSURE DROP
IN PACKED COLUMNS



column diameter which gives an approximate 50 percent of the flooding point. (ie .75 inches of water per foot of packing) A 50% loading factor is used in the computer program.

(a) Calculation of tower diameters. A graphical procedure is not convenient for use on a computer. An arithmetical procedure has been developed by Prah1 (12) based on the generalized curve. In this procedure, the design correlation is numerically characterized by four numbers, particular to the type of packing and packing size.

The arithmetical equations defined by Prah1 to calculate the cross sectional area are listed below.

$$a = l/g \left(\rho_g / \rho_L \right)^{1/2} \quad (37)$$

$$m' = pa + q \quad (38)$$

$$n' = ra + s \quad (39)$$

$$0 = \Delta P' / (m' \Delta P' + n') \quad (40)$$

$$A_x = g / \rho_L \left(\frac{u^2}{\rho_g \times 0} \right) \quad (41)$$

The specific parameters that characterize 2 inch Rasching Rings used in this study are listed.

$$p = 3.77 \times 10^3$$

$$q = .754 \times 10^3$$

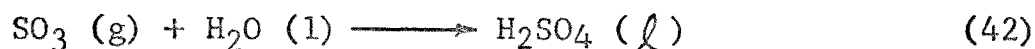
$$r = 2.09 \times 10^3$$

$$s = .880 \times 10^3$$

$$\Delta p' = .75 \text{ inch of H}_2\text{O per foot (based on superficial flooding velocity)}$$

The design diameter of the tower is obtained from the cross sectional area. Volumetric gas flow are based on ideal gas behavior.

(b) Calculation of tower height. The absorption of SO_3 (g) involves the release of large quantities of energy.



$$\Delta H_{\text{RXN}} = 697 \frac{\text{Btu}}{\# \text{SO}_3 \text{ (g)}}$$

It is assumed the surface required to transfer heat from the gas to the acid is the controlling factor in the packed tower design and controls over the area required for the corresponding mass transfer rate.

The "j" factor analysis (10) is used to calculate the heat transfer coefficient.

$$j = \frac{h}{C_p G_o} \left[\frac{C_p \mu}{k} \right]_f^{2/3} = 1.06 \left[\frac{D_p G_o}{\mu} \right]^{-.41} \quad (43)$$

The temperature driving force is calculated using a point to point analysis up the column. The surface area of packing required for heat transfer is calculated using:

$$\int_{T_o}^{T_1'} \frac{dT'}{T' - t} = \int_0^A \frac{h}{w C_p} = \frac{hA}{w C_p} \quad (44)$$

The relationship between the packing surface area and packed volume is used to calculate the height of packing.

For 2" raschig rings

$$\text{Height of Packing} = \frac{\text{AREA (HEAT TRANSFER)}}{28 \frac{\text{sq.ft.}}{\text{cu. ft.}} \times A_x} \quad (45)$$

For operating conditions which are not controlled by heat transfer considerations, the geometric relationships of conventional plants are maintained.

The total column height is made up of the packing sections, and an additional 30 percent of the column diameter to allow for top and bottom sections of the column and regions required liquid distribution mechanisms.

No single bed height is allowed to exceed three times the diameter.

Physical properties used in design are based on average operating conditions in the towers. (See Appendix)

(3) Heat exchange design. Heat transfer equipment is essential in maintaining energy economy in a sulfuric acid plant. A very important by product of acid manufacture is the energy released from heat of reaction in the gas burner and converter and heats of formation in the absorber and drying tower. The design arrangement selected for the pressure plant involves several types of heat transfer design conditions.

- (a) Gas/Gas Transfer
- (b) Gas/Steam Transfer
- (c) Gas/Liquid Transfer
- (d) Liquid/Liquid Transfer

(a) Design of gas/gas heat exchangers. The transfer of heat from one gas stream to another gas stream requires the greatest capital investment due to the low heat transfer coefficients. The basis for this equipment design is selected from McAdams,⁽¹⁰⁾ using the section

covering optimum operating conditions for shell and tube exchangers.

The following assumptions are used in the heat exchanger calculations.

- a) Physical properties are estimated from average operating conditions. (See Appendix)
- b) Annual fixed charges estimated at 20 percent of installed cost.
- c) Operating factor is 95 percent or 8300 hrs. per year.
- d) Power cost to the fluid equals \$.02 per kw - hr. for both shell and tubes.
- e) Exchanger tubes @ nominal 2 inch diameter and square pitch @ 1.5 Do
- f) Turbulent flow in tubes and shell.
- g) Safety factor is 1.25 or 25 percent over design from theoretical.
- h) Shell side is unbaffled.
- i) Length of tubes equal 20 feet

(a1) Calculation of optimum velocities. The optimum mass velocities for both the shell and tubes are obtained from the physical properties of the process streams and

mechanical features of the heat exchanger. The basic design equations used include the following dimensionless terms,

$$* B_i = \frac{1 + NTP K_1 (a_3/a_i)}{\phi_i F_s} \quad (46)$$

$$* B_i = 1.0 \quad (\text{unbaffled exchanger}) \quad (47)$$

where

$$\phi_i = \frac{(t_2 - t_1)_i}{\Delta t_i} \left[\frac{c_p \mu}{k} \right]_i^{2/3} \quad (48)$$

and dimensional terms:

$$K_i = \frac{B_i a_i \mu_i^{m_i}}{2g_c f_i^2 D_i^{m_i}} \quad (49)$$

* Subscript i, o refer to conditions in tubes or in shell respectively.

$$K_o = \frac{2B_o a_o y_o \mu_o^{m_o}}{\pi g_c f_o^2 D_i D_o^{m_o}} \quad (50)$$

The optimum mass velocities for the tubes and shell, G_{io} and G_{oo} respectively are calculated from unit cost C_{Ai} , C_{ei} and C_{eo} .

$$G_{io} = \left[\frac{(C_{Ai}/C_{ei}) K_i}{2.5 + 2.76 \left(\frac{\Delta t_o}{\Delta t_i} \right)} \right] \quad (51)$$

$$G_{oo} = \left[\frac{(C_{Ai}/C_{eo}) K_o}{3.75 + 3.39 (\Delta t_i/\Delta t_o)} \right]^{.351} \quad (52)$$

(a2) Calculation of heat transfer coefficients. The knowledge of mass velocities is used to calculate individual heat transfer coefficients.

$$\left(\frac{h}{c_p G_o} \right)_i \left(\frac{c_p \mu}{k} \right)_i^{2/3} = \frac{a_3 / F_s}{(DG/\mu)_i^{1-n_i}} \quad (53)$$

$$\left(\frac{h}{c_p G_o} \right)_o \left(\frac{c_p \mu}{k} \right)_o^{2/3} = \frac{a_4 / F_s}{(DG/\mu)_o^{1-n_o}} \quad (54)$$

(a3) Calculation of heat transfer area. The design procedures used on heat exchangers involves assuming an overall heat transfer coefficient, U , and calculating a trial surface area.

$$A = Q / (U \Delta T_{em}) \quad (55)$$

The surface area is used to determine the hourly fixed charges on equipment, C_{Ai} , which is a function of area.

Individual film temperature drops Δt_i and Δt_o are assumed for the shell and tube side of the exchanger.

The assumed thermal resistances are checked from the mass velocities and individual heat transfer coefficients.

$$\frac{\Delta t_i}{\Delta t_o} = \frac{h_o}{h_i} \frac{D_o}{D_i} = 1.14 \frac{h_o}{h_i} \quad (56)$$

Using the correct thermal resistances, the assumed surface area is compared with calculated area, from individual heat coefficients.

$$U = \frac{1}{h_i} + .876 / h_o \quad (57)$$

(a4) Calculation of heat exchanger pressure drop.

Pressure drops in the heat exchanger ⁽¹⁰⁾ are calculated from physical conditions.

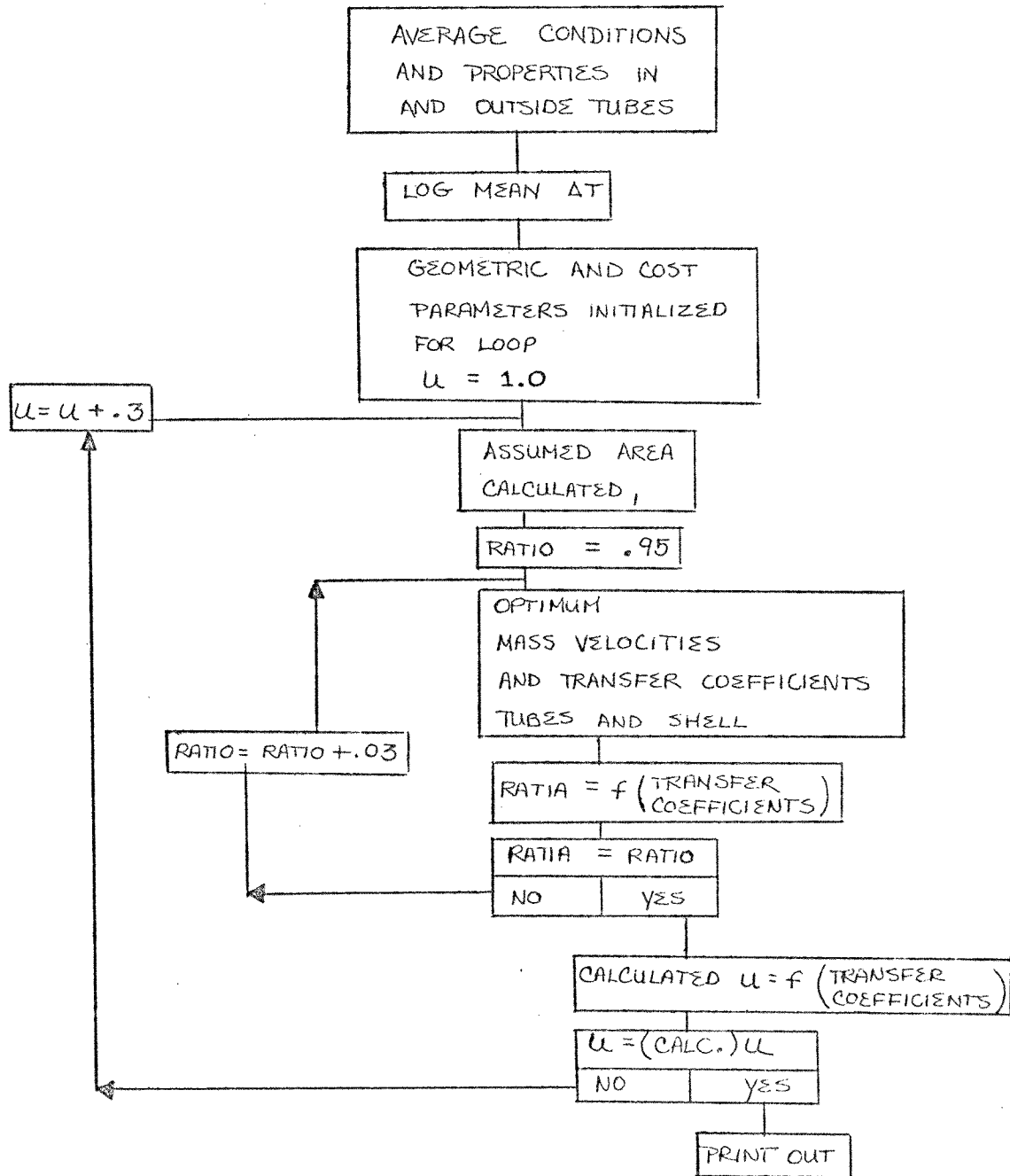
$$\Delta P_i = \frac{B_i 4 f_i L_H G_{i0}^2}{2 g_c \rho_i D_i} \quad (58)$$

where $f_i = a_i \left(\mu / D G \right)_i^{m_i}$

and $\Delta P_o = \frac{4 f_o B_o N G_{o0}^2}{2 g_c \rho_o} \quad (59)$

The block diagram shown in Figure 9 illustrates the computer logic involved in heat exchanger design.

FIG 9
HEAT EXCHANGER
BLOCK DIAGRAM



b. Design of gas/liquid heat exchangers. Waste heat boilers involve heat transfer from the gas stream to water by generating steam.

The design equations used are again from McAdams⁽¹⁰⁾ for the situation in which the shell side power is immaterial. The design equations, assumption, logic are essentially identical to the gas/gas design outlined above. For turbulent flow and low shell power costs the following equation is used.

$$G_{io} = \left[\frac{C_{Ai} C_{Ei} K_i}{3.5 \frac{\Delta t_m}{\Delta t_i} - 1} \right]^{.357} \quad (60)$$

The design equations for the transfer of heat from a gas to a liquid involves the same design equation for optimum velocity on the tube side described above, however a heat transfer coefficient on the shell side is assumed. An individual shell side coefficient of 150 Btu/(Hr) (Sq. ft.) (deg F) is used.⁽¹⁰⁾

(c) Design of liquid/liquid heat exchangers. The liquid to liquid heat transfer for acid coolers is not contained in the computer design program since it is not strictly dependent on pressure. For this exchanger an

overall coefficient equal to 100 Btu/(Hr) (Sq ft) ($^{\circ}$ F) is used. (10)

(4) Compressor / expander. A compressor and gas expander combination has been accepted as an effective device to maintain thermal economy in systems with exothermic reactions and favoring higher operating pressures. The modern nitric acid and ammonia plants are examples of this arrangement.

The characteristics of sulfuric acid manufacture are very similar to both nitric acid and ammonia, however, the stoichiometry is less favorable. It would appear a compressor - expander combination could also contribute in making a pressure sulfuric acid plant more attractive by reducing investment and operating costs. To date, no commercial pressure plants are in operation. (5)

The compression of large volumes of air used in the oxidation of sulfur requires considerable power. The conventional sulfuric acid plant is designed to give a minimum pressure drop across process equipment and thereby reducing total power consumption. A typical power requirement for a conventional plant, rated at

1000 NT per day Sulfuric Acid and 40 inch water S.P., is 700 H.P. This primary blower is usually turbine driven.

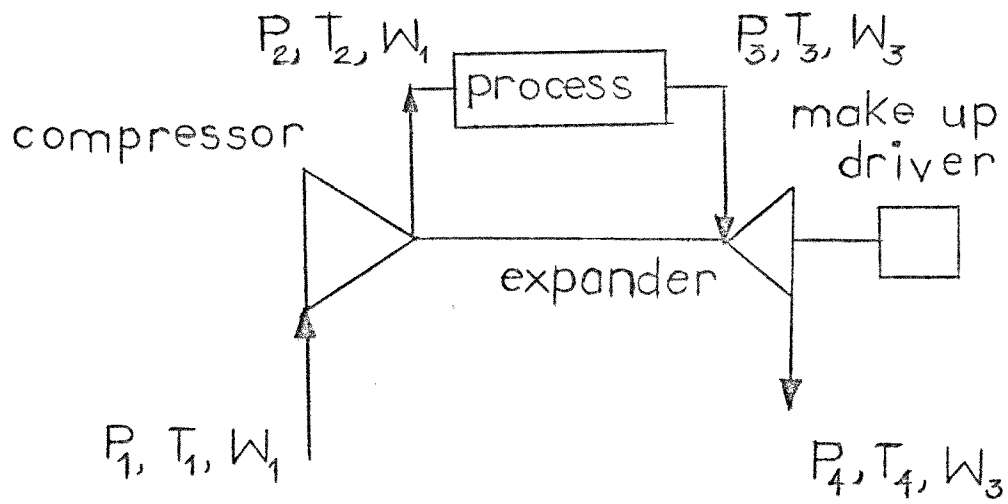
The power needed for a pressure sulfuric acid plant is much higher than the conventional plant. To justify the pressure operation, an energy recovery system is a necessity.

A compressor - expander shown in Figure 10, illustrates a typical scheme for power recovery. Air, at atmospheric pressure, is compressed in a centrifugal compressor and used to advantage in the process. Gases are released to a gas expander at a slightly reduced absolute pressure due to pressure loss across the plant. The exothermic reactions allow for maximum utilization of power recovery through the expander.

The compressor and expander are linked together through a common shaft. Additional power, not recovered by the expander, is supplied by an electric motor.

The basic relations for power recovery have been developed by Canova.⁽³⁾ For uncooled compression the

FIG 10
COMPRESSOR / EXPANDER SYSTEM



make up horsepower, E , (3) is calculated from operating conditions and physical properties.

$$\frac{E}{\sqrt{Q_1} \frac{P}{Z_1}} = \left(50.3 \times \frac{\omega_1 \sqrt{Q_1}}{\frac{P}{Z_1}} \right) * \left[\frac{\left(\frac{P_2}{P_1} \right)^{\frac{k'-1}{k'}} - 1 - C \left(1 - \frac{P_3 P_2}{P_2 P_1} \right)^{\frac{k'-1}{k'}}}{\eta_c (k'-1) / k'} \right] \quad (61)$$

The compressor - expander will sustain itself if the temperature to the expander is sufficiently high. The basic relation for the self-sustaining temperature is given by the following equation. (3)

$$\frac{T_3}{T_1} = \frac{R_1/R_3 \left[\left(\frac{P_2}{P_1} \right)^{-(k'-1)/k'} - 1 \right]}{\left(\omega_3/\omega_1 \right) \eta_c \eta_e \left[1 - \left(\frac{P_3 P_2}{P_2 P_1} \right)^{-(k'-1)/k'} \right]} \quad (62)$$

The assumptions (3, 13) used to calculate power usage for the expander arrangement are listed.

$$k' = 1.401$$

$$\eta_c = .65$$

$$\eta_e = .85$$

$$P_3/P_1 = .90$$

$$T_3(\text{MAX}) = 1000^\circ\text{F}$$

$$\left(\frac{P_2}{P_1} \right)_{\text{MAX}} = 13$$

$$\omega_1 = \text{based on 1000 NT acid/day}$$

The sizing of the centrifugal compressor, for cost purposes, is obtained from a series of charts and graphs presented by Hallock.⁽⁸⁾

(5) Miscellaneous equipment. Four pieces of equipment are not dependent on pressure due to gas volume. These items include: 1) Acid Cooler, E3, 2) Sulfur Pit 3) Acid Pumps and 4) Cooling Tower.

The acid cooler is designed from heat and material balance considerations using an overall heat transfer coefficient equal to $100 \text{ Btu/Hr/Ft}^2/\text{°F}$.⁽¹⁰⁾

The Sulfur Pit is designed from conventional arrangements outlined in Duecker & West⁽⁵⁾.

Acid pumps are designed from material balance considerations.

The cooling tower is designed from the heat balances and a range of cooling equal to 20 degrees F on the recirculated cooling water.

BASIS FOR ECONOMIC CONSIDERATIONS

Return on Investment is used as a measure of the economic worth of a project by incorporating capital investment, operating cost, and burdens. Rate of return can also serve as a basis for comparing and selecting the most attractive operating condition for a pressure sulfuric acid plant.

Cost data, developed for a conventional sulfuric acid plant, (9) are adjusted to reflect economic differences resulting from operating pressure.

The fixed capital investment is obtained from estimating factors developed to calculate installed cost, on a preliminary basis.

Operating costs and burdens are assumed equivalent for equal capacity plants, however, power costs are adjusted to incorporate the use of the compressor expander.

Equations used to compare the effect of operating pressure include rate of return on investment and payout period.

$$\begin{array}{l} \text{Return on} \\ \text{Investment} \\ (\%) \end{array} = \frac{\text{Profit after Taxes}}{\text{Fixed Capital} + \text{Working Capital}} \quad (1)$$

$$\begin{array}{l} \text{Payout} \\ \text{(years)} \end{array} = \frac{\text{Fixed Capital}}{\text{Profit after Taxes} + \text{Depreciation}} \quad (2)$$

DETERMINATION OF TOTAL COST

A module technique used to make preliminary capital cost estimates is presented by Guthrie (7). The method consists of applying multipliers on the F.O.B. equipment cost to estimate the various components that make up the total installed cost. A variety of specific chemical plant equipment is correlated and allows a comparative analysis of the effect of pressure on capital investment.

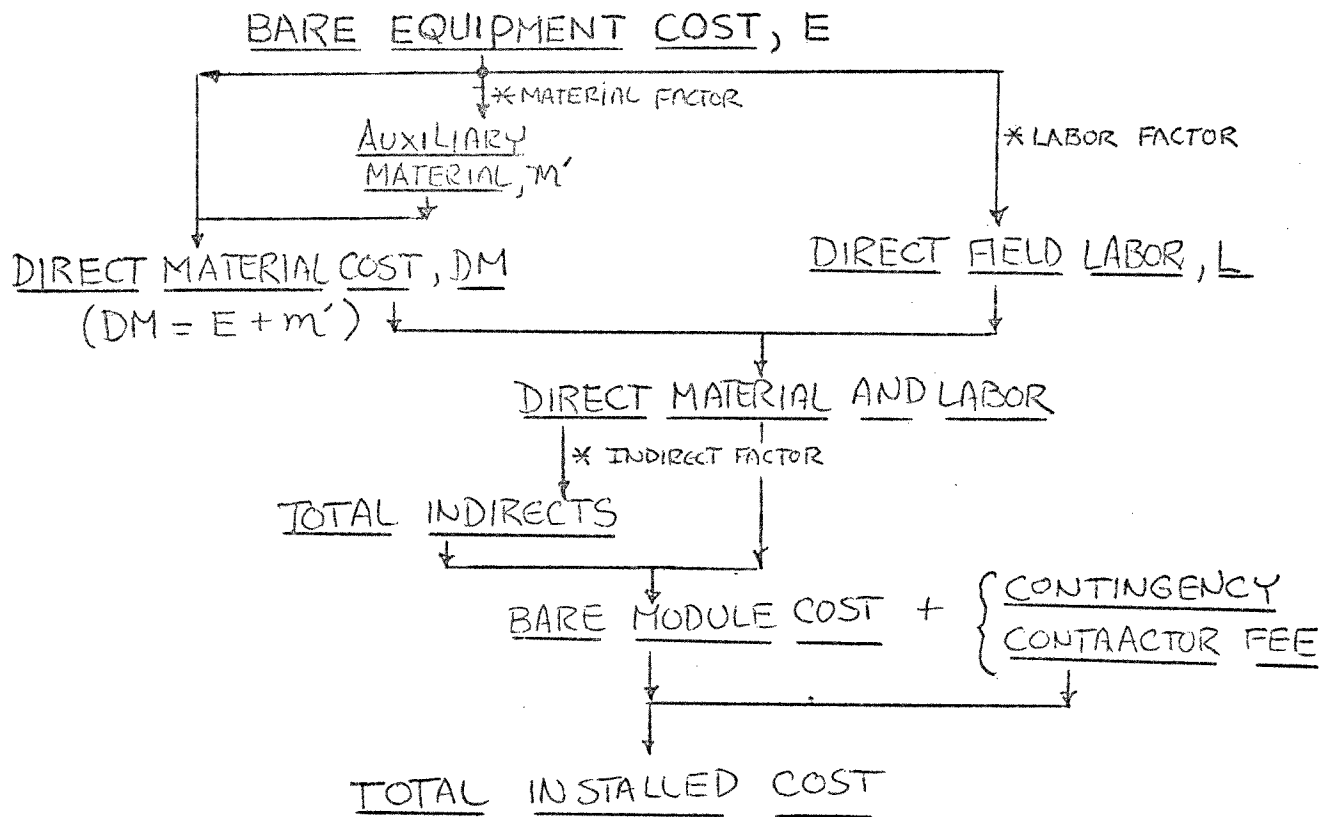
An schematic outline of factors that make up the total installed cost is shown in Figure 11.

The material factor, M.F., is the multiplier applied on F.O.B. cost which gives the total direct material cost including auxiliary material, such as piping, concrete, equipment steel, instruments, electrical, insulation and paint.

A labor factor, LF, when applied on the F.O.B. cost gives the cost of material erection and equipment setting.

Indirect costs are also obtained by applying a multiplier on F.O.B. cost. Indirects include elements of cost which are a percentage of direct material cost

FIG. 11
TOTAL INSTALLED COST BY
MODULE ANALYSIS



such as sales, taxes, freight, insurance and duties. Also included in the indirect cost are items which are a percentage of direct field labor. Fringe benefits, labor burdens, supervision, temporary facilities, construction equipment, and construction overhead make up this category. Finally, the cost for project engineering process engineering, design and drafting, procurement, office overhead and contractor engineering cost, all of which being a percentage of direct material cost, make up the balance of indirect cost.

The sum of direct material and labor and indirect cost is defined as the bare module cost.

The addition of contingency and contractor's fees to the bare module cost gives the total installed cost. A 20 percent adjustment to the bare module cost is used to cover the contractor's fee and contingency.

The F.O.B. cost, based on mid-1968 and c - steel construction, is presented by Guthrie (7) in graphical form related to equipment size. The base cost is adjusted for materials of construction, pressure rating, and design arrangement.

The following equations are used to adjust the Base Equipment Cost.⁽⁷⁾

$$\text{EXCHANGER COST, \$} = (\text{BASE COST } (F_d + F_p) * F_M) * \text{INDEX} \quad (63)$$

$$\text{PROCESS VESSEL COST, \$} = (\text{BASE COST } * F_M * F_p) * \text{INDEX} \quad (64)$$

$$\text{CENTRIFUGAL PUMP COST, \$} = (\text{BASE COST } * F_M) * \text{INDEX} \quad (65)$$

$$\text{COMPRESSOR COST, \$} = (\text{BASE COST } * F_d) * \text{INDEX} \quad (66)$$

The factors that make up the total installed cost for various equipment are presented in Table II from Guthrie.⁽⁷⁾

The components and their design characteristics for the pressure plant are summarized in Tables III, IV, and V.

TABLE II
FIELD MODULE COSTS (7)

	SHELL AND TUBE HEAT EXCHANGER	VERTICAL TANK FABRICATION	HORIZONTAL TANK FABRICATION	CENTRIFUGAL PUMPS	COMPRESSOR
EQUIPMENT COST, F.O.B., E	100.0	100.0	100.0	100.0	100.0
PIPING	45.6	60.0	41.1	30.2	20.6
CONCRETE	5.1	10.0	6.2	4.0	12.3
STEEL	3.1	8.0	--	--	--
INSTRUMENTATION	10.2	11.5	6.2	3.0	8.2
ELECTRICAL	2.0	5.0	5.2	31.0	15.4
INSULATION	4.9	8.0	5.2	2.5	2.6
PAINT	0.6	1.3	.5	.8	.5
MATERIAL FACTOR, m	71.4	103.8	64.5	71.5	59.6
DIRECT MATERIAL AND LABOR	171.4	203.8	164.5	171.5	159.6
MATERIAL ERECTION AND EQUIP. SETTING	55.4 7.6	84. 15.2	52.2 9.3	60.0 9.7	49.8 11.6
DIRECT FIELD LABOR, L	63.0	99.2	61.5	69.7	61.4
DIRECT MATERIAL AND LABOR	234.4	303.	226.0	241.2	221.0
FREIGHT, INSURANCE, TAXES	8.0	8.0	8.0	8.0	8.0
INDIRECTS	86.7	112.0	83.6	89.2	81.8
BARE MODULE FACTOR	329.1	423.0	317.6	338.4	310.8

TABLE III
EQUIPMENT DESIGN

EQUIPMENT	ARRANGEMENT	MATERIALS OF CONSTRUCTION	F _M (RANGE)	F _p (RANGE)	NOTES ⁽⁵⁾
TOWERS T1, T2	VERT. TANK, PACKING & SUPPORTS INTERN. ACID DISTRIBUTERS	C.S. SHELL; 4" TK. ACID BRICK (LINED); C.I. INTERNALS	1.0	1.0/1.15	PACKING - 2" RASCHING RINGS @\$3.5/CU.FT.; ACID BRICK @ \$5.50/SQ. FT. (1968 PRICES M.&LABOR)
SULFUR BURNER, G1	HORIZ. TANK WITH BURNER NOZZLE	C.S. SHELL; 4" TK. REFRACT. & 9" TK. FIRE- BRICK	1.0	1.0/1.15	4" REFRACT. @ \$10.52 PER SQ. FT; 9" FIREBRICK AT \$10.79 PER SQ. FT. TOTAL \$21.31/SQ. FT.
HOT GAS FILTER, F2	VERT. TANK FILLED WITH FINELY CRUSHED FIREBRICK	C.S. SHELL	1.0	1.0/1.15	CRUSHED REFRACT. AT \$3.0/CU. FT.
WASTE HEAT BOILERS, E1, E4, E5	BOILER: WATER TUBE TYPE (SPECIAL S./T. EXCHANGER)	C.S. S./T. REFRACT. LINED HEADS	1.2/ 2.0	0.0/.05	DESIGN FACTOR, F _d = 1.20 * TUBES ALONIZED
MIST ELIMINATOR F3	VERT. TANK WITH SPECIAL TUBE FIL- TERS	C.S. SHELL ACID BRICK LINED INTERN. 304 S.S.	1.15	1.0/1.15	MIST FILTERS AT \$2000/ELEM. EACH ELEM. @ 1500 ACFM

TABLE IV
EQUIPMENT DESIGN

EQUIPMENT	ARRANGEMENT	MATERIALS OF CONSTRUCTION	F _M (RANGE)	F _p (RANGE)	NOTES (5)
CONVERTER R1	VERT. TANK WITH CATALYST AND SUPPORTS	C.S. SHELL EXCEPT TOP SECTION ALONIZED.	1.07	1.0/1.15	VANADIUM PENTOXIDE CATALYST @ \$1400 PER TON OR \$1.30/LITER
RECUP. EXCHANGERS, E2 and E7	FIXED TUBE, S./T. EXCHANGER	C.S. SHELL AND ALONIZED TUBES	1.2/2.0	0.0/.05	DESIGN FACTOR, F _d EQUALS .8
ACID COOLER E3	FIXED TUBE, S./T. EXCHANGER	C.S. SHELL KARBATE TUBES	-	-	U ⁽¹⁰⁾ = 100 BTU/HR. FT ² °F COST = \$6 PER SQ. FT.
CONDENSATE HEATER, E6	FIXED TUBE, S./T. EXCHANGER	C.S. SHELL AND TUBES	1.0	1.0/1.15	DESIGN FACTOR, F _d EQUALS .8

TABLE V
EQUIPMENT DESIGN

EQUIPMENT	ARRANGEMENT	MATERIALS OF CONSTRUCTION	FM	Fp	NOTES
AIR COMPRESSOR, Pl	MULTIPLE STAGE CENTRIFUGAL WITH EXPANDER	C - STEEL	-	-	DESIGN AND COST HALLOCK (8) TEMP. IN 67°F Cp/Cv = 1.401 UNCOOLED COMPRESSION Fd = 1.15 (GUTHRIE (7))
SULFUR PIT	SUBMERGED CON- CRETE PIT WITH PUMP AND HEAT PLATES	CONCRETE AND CAST IRON	-	-	DESIGN BASED ON DEUCKER AND WEST (5)
ACID PUMPS	CENTRIFUGAL PUMPS	CAST IRON	1.0	1.0	DESIGNED AT 5500 gpm AND 100 psig
COOLING TOWER	COOLING TOWER, CONCRETE BASIN, PUMPS, DRIVES	-	-	-	CAPACITY EQUALS 10,000 GPM AT COOLING RANGE = 20°F GUTHRIE (7)

ECONOMIC COMPARISONS

Operating pressure is reflected in the investment and income factors for a one thousand ton per day plant to obtain a comparison of earnings at pressures equal to one, three, five, seven, nine and thirteen atmospheres.

A summary of the material unit prices and various components that make up the total manufacturing cost is presented in Table VI for a Conventional Sulfuric Acid Plant as presented by Krnonseder.⁽⁹⁾ The total manufacturing cost is adjusted for operating pressure using Krnonseder's values as a base.

A summary of typical earnings is presented for the conventional plant ⁽⁹⁾ in Table VII. Operating pressure effect both capital investment and manufacturing cost in this tabulation. Working capital, sales income, and other expenses are assumed equal for equivalent capacity plants.

TABLE VI
TOTAL MANUFACTURING COST
CONVENTIONAL H₂SO₄ PLANT
(1000 NT/DAY)

DOLLARS/TON ACID

I	RAW MATERIALS @ \$40.00/TON SULFUR	<u>13.48</u>
II	UTILITIES AND BY PRODUCT CREDIT	
	STEAM CREDIT @ \$.50/M lbs	(1.10)
	WATER @ \$.02/M gal	.14
	POWER @ \$.01/KWH	.53
	FUEL @ \$.05/M gal	<u>.16</u>
		(.27)
III	OPERATING LABOR AND SUPERVISION	<u>.23</u>
IV	CAPITAL CHARGES, TAXES AND INSURANCE	<u>.39</u>
V	MAINTENANCE, REPAIRS, ETC.	<u>.46</u>
VI	OVERHEADS	<u>.91</u>
		<u>=====</u>
	TOTAL	15.20

TABLE VIIEARNINGS FOR CONVENTIONALH₂SO₄ PLANT - 1000 NT/DAYI. INVESTMENT

FIXED CAPITAL		2,990,000
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WORKING CAPITAL

ACCOUNT RECEIVABLE	106,000	
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CASH	460,000	
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IN PROCESS INVENTORY	94,000	
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RAW MATERIAL INVENT.	409,000	
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FINISH PRODUCT INVENT.	<u>106,000</u>	
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		<u>1,175,000</u>
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A) TOTAL INVESTMENT		4,075,000
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II. INCOME

B) SALES 330,000 TPY @ \$ 18.03/Ton		5,954,000
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LESS MANUFACTURING COST @ \$ 15.20/Ton		<u>5,016,000</u>
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C) GROSS INCOME		938,000
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LESS SALE EXPENSE		
-------------------	--	--

RESEARCH EXPENSE		
------------------	--	--

ADMINISTRATION EXPENSE		
------------------------	--	--

ALL OTHER EXPENSE		
-------------------	--	--

		<u>500,000</u>
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PROFIT BEFORE TAXES		438,000
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D) PROFIT AFTER TAX @ 50%		219,000
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RESULTS OF STUDY

The results of the computer study are presented in a series of five tables and one graph.

Table VIII outlines the effect of pressure on equipment design for components which are pressure dependent.

Table IX illustrates how effectively the gas expander reduces operating costs compared to a motor driven compressor by the lower power requirement. The Table also lists the operating costs for various pressures of a compressor expander.

Table X summarizes equipment cost as a function of operating pressure. All major pieces of equipment, except the compressor expander P1, are reduced in cost. A significant reduction in catalyst requirements is obtained at elevated pressures.

Table XI is used to compare capital investment of the conventional sulfuric acid plant with the proposed scheme developed for a pressure plant. For this design study,

the additional fixed capital cost required at one atmosphere is due to the added recuperative heat exchangers needed in the pressure plant arrangement.

The economics evaluation of a pressure plant is summarized in Table XII. Since the scheme presented for one atmosphere is unattractive, the listing at pressure equals one is for the conventional sulfuric acid plant adjusted for the cost index operating basis.⁽⁹⁾ The Table shows that a maximum return on investment occurs at a pressure of about 3 atmospheres. The return on investment, the basis objective function for this study, is shown graphically in Figure 12 as a function of pressure. These results are in agreement with experimental investigations by Cathala ⁽⁴⁾ and recent studies by Richterova ⁽¹³⁾ applying the modern gas expander in sulfuric acid manufacturing plants.

TABLE VIII
EQUIPMENT DESIGN

Equipment	Pressure, ATM					
	(1)	(3)	(5)	(7)	(9)	(13)
F1 (Thousand ACFM)	79	26	14	10	7	5
*T1 (Diam. x Height)	16.5/21.5	13.1/17	11.4/14.8	10.7/13.9	10.2/13.3	9.6/12.5
G1 (Diam. x Height)	22.2/33.4	15.4/23.1	13. /19.5	11.6/17.4	10.7/16.	9.4/14.2
E1 (Sq. Ft.)	7,000	3,650	2,670	2,170	1,920	1,920
F2 (Diam. x Height)	10.8/10.8	6.4/6.4	5. /5.	4.3/4.3	3.8/3.8	3.2/3.2
R1 (Diam. x Height)	29.2/52.8	22.1/31.3	19.5/22.6	17.9/19.7	16.8/17.9	15.3/15.4
E5 (Sq. Ft.)	1,810	710	2,970	2,450	2,250	2,280
E2 (Sq. Ft.)	41,480	19,400	11,860	9,540	8,210	6,730
T2 (Diam. x Height)	18. /23.	14. /18.	12. /16.	11. /15.	11. /14.	10. /13.
E4 (Sq. Ft.)	15,840	7,690	5,120	4,180	3,620	2,890
E6 (Sq. Ft.)	1,820	950	660	560	560	560
E7 (Sq. Ft.)	33,120	15,300	8,000	6,430	5,600	4,590
P1 Make Up Horse- power (Stages)	**	500 (3)	3,100 (5)	5,000 (6)	7,200 (8)	12,000 (10)

* Linear Dimensions equal feet

** Primary Blower @ 700 hp.

TABLE IX
COMPRESSOR/EXPANDER
CHARACTERISTICS

PRESSURE (ATM)	HORSEPOWER (COMPRESSOR)	MAKE UP HORSEPOWER (COMP/EXPAND.)	OPERATING* COST/YEAR (COMP/EXPAND.)	OPERATING* COST/TON OF ACID
1	-	700**	42,000	.12
3	9,000	500	30,000	.086
5	15,000	3,100	186,000	.53
7	18,500	5,000	300,000	.86
9	22,000	7,200	432,000	1.23
13	29,000	12,000	726,000	2.08

* in Dollars

** Primary Blower @ 700 h.p.

TABLE X
EQUIPMENT COST,
 (\$1000 UNITS)

EQUIPMENT	PRESSURE, ATM					
	(1)	(3)	(5)	(7)	(9)	(13)
T1	147	77	57	52	49	42
G1	191	102	73	55	46	35
E1	107	72	57	50	44	43
F2	46	15	10	9	8	7
R1	365	186	131	109	92	75
E5	43	25	64	53	52	51
E2	290	164	118	115	102	92
T2	157	83	61	56	53	45
E4	152	106	78	75	71	61
E6	39	28	22	20	20	20
E7	264	147	99	92	82	72
P1	52*	645	875	985	1,150	1,340
Sulfur Pit	50	50	50	50	50	50
Acid Pumps	90	90	90	90	90	90
Acid Cooler	138	138	138	138	138	138
Cooling Tower	500	500	500	500	500	500
Catalyst	660	195	116	83	46	44

* Priced as a conventional
 Primary Blower
 @ 700 hp

TABLE XI
FIXED CAPITAL VS PRESSURE

(\$1MM UNITS)

PRESSURE (ATM)	INVESTMENT (CATALYST FREE BASIS)	CATALYST COST	TOTAL INSTALLED COST **	<u>F.C. PRESSURE *</u> <u>F.C. CONVEN-</u> <u>TIONAL</u>
1	2.630	.660	3.421 *	1.18
3	2.427	.190	2.738	.95
5	2.427	.115	2.663	.92
7	2.447	.082	2.652	.915
9	2.546	.050	2.723	.94
13	2.663	.042	2.841	.98

** Based on Battery Limits
including 5 percent
contractor fee

* Conventional plant @
\$2.9 MM

TABLE XII
ECONOMIC EVALUATION OF A

PRESSURE PLANT (1000 NT/DAY H₂SO₄)

PRESSURE (ATM)	TOTAL ** INVESTMENT (\$ MM)	SALES (\$ MM)	OPERATING COST (DOLLARS/TON)	GROSS INCOME (\$ M)	PROFIT AFTER TAXES (\$ M)	RETURN ON INVEST- MENT (PERCENT)	PAYOUT (YEARS) F.C./ (PROFIT + DEP.)
1 ***	4.075	5.954	15.20	938	219	5.38	5.7
3	3.913	5.954	15.066	994	247	6.32	5.25
5	3.838	5.954	15.61	804	152	3.96	6.37
7	3.827	5.954	15.94	604	52	1.36	8.4
9	3.898	5.954	16.31	554	27	.7	9.1

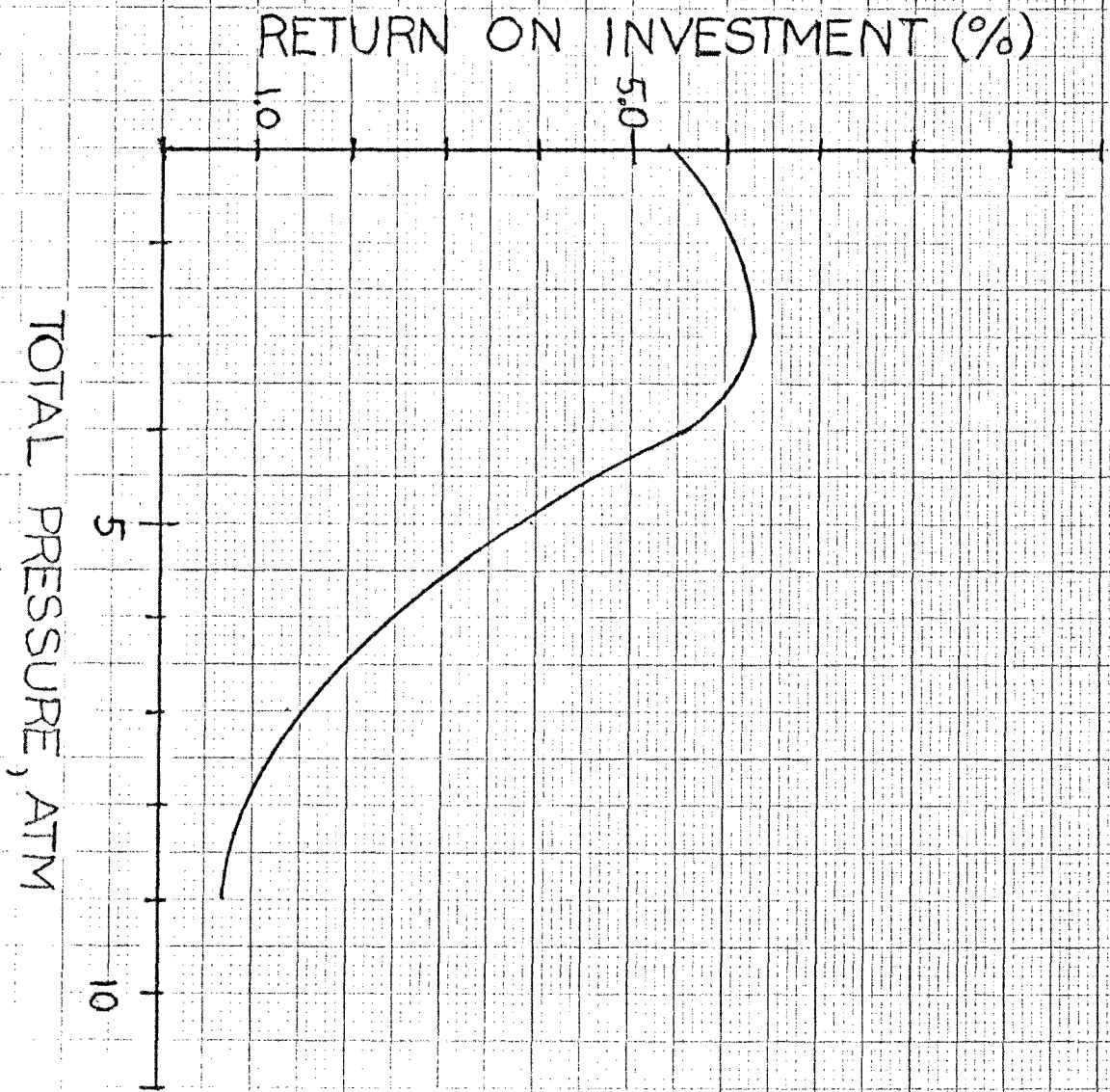
*** Economics Based on Conventional Plant
for 1 ATM.

* Gross Income Reduced by \$500,000 Expenses
& then taxed at 50 percent.

** Includes \$1,175,000 working capital

FIG 12

RETURN ON INVESTMENT
VS
PRESSURE



CONCLUSIONS

1. The manufacture of sulfuric acid is economically attractive at pressures above one atmosphere but less than five atmospheres.

Operating pressures exceeding five atmospheres results in only slight improvements in fixed capital cost compared to rapidly rising operating cost.

2. The proposed pressure plant scheme is economically unattractive at one atmosphere compared with conventional plant economics developed by Kronseder ⁽⁹⁾.

3. A significant reduction in catalyst requirements is obtained at elevated pressures.

4. The compressor expander arrangement significantly reduces equipment cost, without prohibitive operating cost, for pressure ratios less than five. The gas expander reduces the required horsepower as much as twenty two times from that required by a motor driven centrifugal compressor.

RECOMMENDATIONS

Based on the results of this investigation, the following recommendations for further study are made:

1. An extension of this study that might be of interest would involve pressure plant designs at higher conversions and reduced SO₂ pollution.
2. Improvements in the equilibrium conversion characteristics, with pressure, allows operating at higher sulfur dioxide concentrations. The effect of gas concentration on plant economics could also be studied for pressure plants.
3. Other pressure plant equipment arrangements could be studied to determine optimum economy.

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NOMENCLATURE

A	= Heat exchanger surface area, sq. ft.
Appch	= Approach to equilibrium, y axis of of equil. conversion graph
A_x	= Cross sectional Area, sq. ft.
a	= Correlation factor, equat. (37), deminsionless
a_o	= Dimensionless constant, a function of physical properties, .31
a_3, a_4	= Dimensionless constant equal .023 and .33 respectively (turbulent)
a_i	= Dimensionless constant equal .055 a function of tube arrangement
B_i	= Dimensionless term defined by equation (46) for tubes
B_o	= Dimensionless term defined by equation (47) for shell
C	= Coefficient equal $(W_3/W_1) (n_c) (n_e) (R_3 T_3 / R_1 T_1)$
C_{Ai}	= Hourly fixed charges on equipment, \$/hr. ft. ² (inside area)
C_e	= Cost of mechanical energy supplied to fluid, \$/ft. - lb.
C_p	= Mean specific heat, Btu/lb - deg. F
C_p^x	= Mean specific heat, Btu/lb - deg. F for component i
COFA	= Coefficient of T^2 , Btu/(deg F) ²
COFB	= Coefficient of T, Btu/(deg F)

$C_{SO_3}, C_{SO_2}, C_{O_2}$	=	Concentration of reaction components, moles
D	=	Tube diameter, Ft
D_p	=	Average particle size, Ft
E	=	Make-up horsepower, hp
e	=	Log constant = 2.718 - - - -
Eff	=	Efficiency of operating plant based on sulfur
F_s	=	Safety factor, deminsionless
F	=	Feed, equivalent tons H_2SO_4 /day
F_d, F_p, F_M	=	Correction factors to adjust capital cost as a result of design, pressure or material considerations, dimensionless
f_M	=	Friction factor define by equation (37)
G_{io}, G_{oo}	=	Optimum mass velocities tubes and shell respectively, (lb/hr. - sq. ft.)
G_o	=	Superficial mass velocity, #/Hr. - sq. ft.
g	=	Gas flow rate, #/sec.
G	=	Superficial mass velocity based on empty shell, lb/sec. - sq. ft.
g_c	=	Gravitational constant, 4.17×10^8 (lb fluid) (ft)/(hr) (hr) (lb _f)
H_{LOSS}	=	Heat loss to atmosphere by radiation, Btu/min
h	=	Individual heat transfer coefficient Btu/hr. - sq. ft.
Index	=	Cost index to update equipment cost, dimensionless

J	=	Stage or bed in converter
k'	=	Ratio of specific heats, C_p/C_v , dimensionless
K_i	=	Dimensionless term defined by equation (49)
K_o	=	Dimensionless term defined by equation (50)
k	=	Thermal conductivity, Btu/(hr) (sq. ft.) (deg F per ft.)
K_1	=	.55 for single pass and turbulent flow
$K_{Eq.}$	=	Equilibrium constant, dimensionless
k^*	=	Velocity constant define for Eklund equation
K_γ, K_ϕ, K_p	=	Equilibrium constants developed for overall conversion
L	=	Depth of bed, feet
L_H	=	Tube length, feet
l	=	Liquid flow rate, #/sec.
m_o, m'_i	=	Exponents dimensionless defined in equation (49) & (50) equal to .15 and .20 respectively
m'	=	Numerical value defined in equation (38)
m_i	=	Moles of component i
η_e	=	Efficiency of expander
η_c	=	Efficiency of compressor
n'	=	Numerical value defined in equation (38)

N	=	Number of tubes
n_1, n_0	=	Exponents defined in equations (53) & (54)
N_{TP}	=	Number of tube passes
O	=	Numerical value defined for equation (38)
$P_{O_2}, P_{SO_2}, P_{SO_3}$	=	Partial pressures of O ₂ , SO ₂ , SO ₃ respect, psia
P, Press	=	Pressure, psia
Q	=	Heat transfer, Btu/hr
QSF	=	IBM scientific subprogram for intergration
RAT	=	Acid production, Tons Acid/Day
Re	=	Reynold's number, dimensionless
R	=	Gas constant, various dimensions
RH	=	Relative humidity
r	=	Reaction rate, Tons H ₂ SO ₄ produced/ton catalyst - day
SENHT	=	Enthalpy of stream, Btu
SO2CN	=	SO ₂ fractional concentration
T, Temp	=	Temperature, °K
TEMPK	=	Temperature, °K
t	=	Average liquid temperature for interval, °F
T'	=	Average gas temperature for interval, °F
T _{EQUIL}	=	Equilibrium temperature, °K
t_2, t_1	=	Bulk temperature, deg F

U	= Overall heat transfer coefficient, Btu/(hr) (sq. ft.) $^{\circ}\text{F}$
$W_{A,B}$	= Tons of catalyst
w	= Mass flow, #/sec.
X	= Fractional conversion
X'	= Unit fractional conversion
X_{OVAL}	= Overall fractional conversion
y_i	= Mole fraction of component i
$y_{\text{SO}_3}, y_{\text{SO}_2}, y_{\text{O}_2}$	= Mole fraction of respective components
y_0	= Clearance between outer tube surface in a bundle, feet
α, β, γ	= Coefficients of heat capacity equation defined in equation (6)
Δ	= Incremental conversion
Δt	= Mean value of temperature drop across individual thermal resistance, $^{\circ}\text{F}$
Δt_m	= Mean overall temperature drop from hot to cold fluid, deg. F.
ΔP^*	= Pressure drop, inch $\text{H}_2\text{O}/\text{ft}$.
ρ	= Density of fluid, lbs per cu. ft.
ξ	= Normalized Pressure, $P/14.7$
θ	= Normalized Temperature, $T/520$
ΔP	= Pressure drop, Psi
$\Delta\alpha, \Delta\beta, \Delta\gamma$	= Heat capacity coefficients used in H_{RXN} adjusting for stiochemistry.

- ΔH_{RXN} = Heat of reaction, Btu/# mole
- ΔT_{lm} = Log mean temperature difference
- Π = Total pressure, ATM
- η, μ, ξ, η = Equation to define velocity constant
eq. (34)
- ϕ_i = Dimensionless term, a function of
temperature
- μ = Fluid viscosity, lbs/(hr) (ft)

APPENDIX A - SAMPLE PROBLEMS

APPENDIX A - SAMPLE PROBLEMS1. Equipment Estimates

Preliminary cost estimate for reactor, R1

Operating pressure 1 atm

Design parameters 29.2' diam. x 52.8'
 473 tons of catalyst
 3 beds in converter

Using Figure 5 in Guthrie, (7) the base cost for a pressure vessel is read. (Vertical arrangement).

a) Base cost = \$83,000

The fraction of the converter vessel with an aluminum alloy is calculated from a ratio of catalyst in the top bed to total catalyst. F_m , the material correction factor is calculated:

$$\frac{18 \text{ Tons}}{473} = 5.3\%$$

$$F_m = 1.0 \text{ (c - steel)}$$

$$F_m = 2.33 \text{ (Aluminum Alloy)}$$

$$F_m = 94.7 \times 1 + 5.3 \times 2.33$$

$$F_m = 1.07$$

$$F_p = 1.0 \text{ for pressure to 50 psig}$$

Index = 1.08 to correct cost from mid-1968 to mid-1970.

- b) Expected cost mid 70 = (Fm * Fp * Index)
 * Base = 1.16 * Base = \$93,000, F.O.B. Cost

Using the bare modual factor developed by Guthrie for vertical tank under \$200,000.

- c) Bare Modual Cost = Base * Bare Mod. Factor
 = 4.24 * 83,000 = \$352,000.

- d) Due to design changes to base equipment an additional \$(93,000 - 83,000) = \$10,000 is added.

Total Bare Modual Cost = 352,000 + 10,000
 = \$362,000

Other equipment are estimated on a similar basis using graph and factors developed by Guthrie.

2. Make up Horsepower for Compressor - Expander

Using equation developed by Canova (3) the make up horsepower is calculated from material flows and physical properties.

For a pressure ratios equal to 3.0:

$$\frac{E}{\sqrt{Q_1} \frac{P}{Q_1}} = \left[\frac{50.3 (w_1 \sqrt{Q_1} / \frac{P}{Q_1})}{\eta_c} \right] * \left[\ln \frac{P_2}{P_1} - C \left(\frac{K'}{K'-1} \right) \left(1 - \frac{P_3 P_2}{P_2 P_1} \right)^{\frac{-K'-1}{K'}} \right]$$

$$Q_1 = \frac{T_1}{520} = 1.0 @ T_1 = 60^\circ F$$

$$\frac{P_1}{14.7} = 1.0 @ P_1 = 14.7$$

$$\omega_1 = \frac{79,300}{359} * \frac{29.9}{60} \quad 110 \text{ \#/sec}$$

$$K' = C_p/C_v = 1.401$$

$$\left(\frac{P_2}{P_1}\right)^{(K'-1/K')} = 1.365$$

$$\omega_3 = .90 \omega_1$$

$$C = \frac{\omega_3}{\omega_1} \eta_c \eta_e \frac{R_3 T_3}{R_1 T_1} = (.90)(.65)(.85) \left(\frac{1460}{520}\right)$$

$$= 1.390$$

$$P_3 = .90 P_2 \text{ (ASSUMED)}$$

$$\left(\frac{P_3}{P_1}\right)^{-\frac{K'-1}{K'}} = .75$$

$$E = (50.3)(110) \left\{ \frac{1.365 - 1.0 - (1.390)(.25)}{(.65)(.285)} \right\}$$

$$E = 510 \text{ hp}$$

3. Income Statement - Calculations - P = 3 ATMInvestment

Fixed Capital (From preliminary cost estimates for 3 ATM) -----	\$ 2,738,000
Working Capital (Assumed identical for the conventional plant operating at same basis) -----	\$ 1,175,000
(A) Total Investment ----- (F. C. + W. C.)	\$ 3,913,000

Income

(B) Sales 330,000 T.P.Y. @ $\frac{\$18.03}{\text{Ton}}$ ---	\$ 5,954,000
Operating Cost @ \$15.066 ----- (Adjusted for power cost due to expander)	\$ 4,960,000
(C) Gross Income	
Sales - Operating Cost -----	\$ 994,000
All Other Expenses ----- (Assumed equivalent to the conventional plant)	\$ 500,000
Profit before Taxes	
Gross Income - Expenses =	\$ 494,000
(D) Profit after Taxes @ 50% =	\$ 247,000
Capital Turnover (B/A) -----	1.52
Gross Return on Sales (C/B) --	16.7%

(D) continued

Return on Investment (D/A) ----- 6.32%

Payout = $\frac{\text{F. C.}}{(\text{D} + \text{Depreciation})}$ = 5.25 years

APPENDIX B - COMPUTER PROGRAM

M87, M88, N88, N89, N90

N87

```
COMMONIC  
IC = 1  
CALL LINK(M88)  
END  
*STORE
```

```
N87 0010  
N87 0020  
N87 0030  
N87 0040  
N87 0050
```

This subprogram is used for initialization of iteration sequence in subprograms M88 and N88.

SUB PROGRAM M88 - HEAT AND MATERIAL BALANCE

<u>STATEMENT NOS.</u>	<u>DESCRIPTION</u>
10 / 290	Initialization of Balance
300 / 410	Overall Material Balance
420 / 690	Heat Duty on E 1
700 / 750	Loop Initialization for Converter
780 / 890	Compositions of Components
900 / 1230	Incremental Heat Balance Around Converter
1240 / 1300	Rate Expression
1310 / 2550	Catalyst Requirements
2560 / 2730	Call Link to N 88

```

M88 PAGE 1
DEFINEFILE8(10,40,U,18),9(1,40,U,19) M88 0010
DIMENSIONACIDM(11),H2OM(11),TOTM(11),SO2M(11),ZN2M(11),O2M(11)M88 0020
1),TEMPK(11),X(7),SO3M(11),TXSO2(90),Y(90),Z(90),WTCAT(90),M88 0030
2VAT(8) M88 0040
COMMONIC M88 0050
GOTO(5,30),IC M88 0060

```

```

5 DO I=1,11 M88 0070
  ACIDM(I) = 0.0 M88 0080
  H2OM(I) = 0.0 M88 0090
  TOTM(I) = 0.0 M88 0100
  SO2M(I) = 0.0 M88 0110
  ZN2M(I) = 0.0 M88 0120
  O2M(I) = 0.0 M88 0130
  TEMPK(I) = 0.0 M88 0140
  SO3M(I) = 0.0 M88 0150
10 CONTINUE M88 0160
  WRITE(3,15) M88 0170
15 FORMAT(1X,'PRESSURE SULFURIC ACID PLANT ') M88 0180
  WRITE(3,20) M88 0190
20 FORMAT(/// ' HEAT + MATERIAL BALANCE') M88 0200
  WRITE(3,25) M88 0210
25 FORMAT(///// ' REACTOR DESIGN CONDITIONS') M88 0220
  RAT = 1000. M88 0230
  SO2CN = .08 M88 0240
  EFFCY = .95 M88 0250
  APPCH = .95 M88 0260
  RH = .50 M88 0270
  XOVAL = .99 M88 0280
  PRESS = - 1.0 M88 0290
  DO 245 IJ=1,6 M88 0300
  PRESS = PRESS + 2.0 M88 0310
  ACIDP = RAT * 1.3888 / EFFCY M88 0320
  H2OP = ACIDP * .02041 M88 0330
  ACIDM(6) = ACIDP / 98. M88 0340
  H2OM(6) = H2OP / 18. M88 0350
  TOTM(6) = ACIDM(6) + H2OM(6) M88 0360
  TEMPK(6) = 311. M88 0370
  TOTM(7) = ACIDM(6) / SO2CN M88 0380
  SO2M(7) = TOTM(7) * SO2CN M88 0390
  ZN2M(7) = TOTM(7) * .79 M88 0400
  O2M(7) = TOTM(7) * (.21 - SO2CN) M88 0410
  TEMPK(8) = 723. M88 0420
  TEMPK(7) = 800. M88 0430
30 TEMPK(7) = TEMPK(7) + 20. M88 0440
  QN2 = ZN2M(7) * (11.743 * TEMPK(7) + .1125E - 02 * TEMPK(7) * * 2. M88 0450
  1 - 3599.) M88 0460
  QO2 = O2M(7) * (11.066 * TEMPK(7) + .2792E - 02 * TEMPK(7) * * 2. M88 0470
  1. - 4578.) M88 0480
  QSO2 = SO2M(7) * (12.809 * TEMPK(7) + .8561E - 02 * TEMPK(7) * * 2 M88 0490
  1. - 4578.) M88 0500
  QTOT = QN2 + QO2 + QSO2 M88 0510
  QN2 = ZN2M(7) * (11.743 * TEMPK(8) + .1125E - 02 * TEMPK(8) * * 2. M88 0520
  1 - 3599.) M88 0530
  QO2 = O2M(7) * (11.066 * TEMPK(8) + .2792E - 02 * TEMPK(8) * * 2. M88 0540
  1. - 4578.) M88 0550
  QSO2 = SO2M(7) * (12.809 * TEMPK(8) + .8561E - 02 * TEMPK(8) * * 2 M88 0560
  1. - 4578.) M88 0570
  QTOTB = QN2 + QO2 + QSO2 M88 0580
  Q = QTOT - QTOTB M88 0590
  TOTP = Q / 825.1 M88 0590
  TEMPK(11) = 478. M88 0590

```

```

M88 PAGE 2
O2M(11) = 0.0
TOTM(11) = TOTP / 18.
H2OM(11) = TOTM(11)
TOTM(8) = TOTM(7)
SO2M(8) = SO2M(7)
O2M(8) = O2M(7)
ZN2M(8) = ZN2M(7)
NJ = 1
X(1) = 0.0
SENHT = QTOTH
ZN2 = ZN2M(8)
TO2 = O2M(8)
TSO2 = SO2M(8)
XSO2 = -.01
TEMP = TEMPK(8)
ADD = .01
DO 210 J=1,10
DO 105 I=1,99
XSO2 = XSO2 + ADD
TXSO2(1) = XSO2
SO3 = TSO2 * XSO2
SO2 = TSO2 - SO3
O2 = TO2 - .5 * SO3
ZN2 = ZN2
TOT = SO3 + SO2 + O2 + ZN2
SO2MF = SO2 / TOT
SO3MF = SO3 / TOT
O2MF = O2 / TOT
ZN2MF = ZN2 / TOT
SO2X = TSO2 * ADD
TKP = SO3MF / (SO2MF * O2MF * * .5 * PRESS * * .5)
IF (TKP - .001) 35,35.40
35 TMEQ = 3000.
GOTO 45
40 TLGKP = ALOG(TKP)
DIV = ALOG(10.)
TLGKP = TLGKP / DIV
TMEQ = 4956. / (TLGKP + 4.678)
45 IF (I - 1) 50,50.55
50 SO2X = 0.0
ROOT = TEMP
GOTO 75
55 HRXN = SO2X * (- 43049. - 4.113 * TEMP + .6236E - 02 * TEMP * *
1. - .1246E - 05 * TEMP * * 3.)
COFA = ZN2 * .1125E - 02 + O2 * .2792E - 02 + SO2 * .8561E - 02 +
1SO3 * .2118E - 01
COFB = ZN2 * 11.743 + O2 * 11.066 + SO2 * 12.809 + SO3 * 10.939
HLOSS = 0.0
CONST = - 3599. * ZN2 - 3546. * O2 - 4578. * SO2 - 5141. * SO3
COFC = HRXN - SENHT - HLOSS + CONST
DISCR = SQRT(COFC * * 2. - 4. * COFA * COFC)
ROOT = (- COFB + DISCR) / (2. * COFA)
IF (ROOT) 60,60.65
60 ROOT = (- COFB - DISCR) / (2. * COFA)
IF (ROOT) 70,70.65
70 IF (ROOT - 2000.) 75,75.70
75 TEMP = ROOT
QO2 = ZN2 * (11.743 * TEMP + .1125E - 02 * TEMP * * 2. - 3599.)
O2 = O2 * (11.066 * TEMP + .2792E - 02 * TEMP * * 2. - 3546.)
M88 0600
M88 0610
M88 0620
M88 0630
M88 0640
M88 0650
M88 0660
M88 0670
M88 0680
M88 0690
M88 0700
M88 0710
M88 0720
M88 0730
M88 0740
M88 0750
M88 0760
M88 0770
M88 0780
M88 0790
M88 0800
M88 0810
M88 0820
M88 0830
M88 0840
M88 0850
M88 0860
M88 0870
M88 0880
M88 0890
M88 0900
M88 0910
M88 0920
M88 0930
M88 0940
M88 0950
M88 0960
M88 0970
M88 0980
M88 0990
M88 1000
M88 1010
M88 1020
M88 1030
M88 1040
M88 1050
M88 1060
M88 1070
M88 1080
M88 1090
M88 1100
M88 1110
M88 1120
M88 1130
M88 1140
M88 1150
M88 1160
M88 1170
M88 1180
M88 1190
M88 1200

```

M88 PAGE 3

QSO2 = SO2 * (12.809 * TEMP + .8561E - 02 * TEMP ** 2. - 4578.) M88 1210
 QSO3 = SO3 * (10.939 * TEMP + .2118E - 01 * TEMP ** 2. - 5141.) M88 1220
 SENHT = QN2 + QO2 + QSO2 + QSO3 M88 1230
 EXPO1 = (- 15600. / TEMP) + 12.07 M88 1240
 EXPO2 = (- 26970. / TEMP) + 22.75 M88 1250
 AB = EXP(1.) M88 1260
 RATE = 8.5E + 06 * ((AB ** EXPO1 * O2MF * SO2MF / SO2MF ** .5M88 1270
 1) * PRESS ** 1.5) - (AB ** EXPO2 * SO3MF * O2MF ** .5 / SO2MF M88 1280
 2 * * .5) * PRESS)) M88 1290
 Y(I) = 1. / RATE M88 1300
 IF (ABS(XSO2 - XOVAL) - .0001) 110,110,80 M88 1310
 80 IF (TMEQ - TEMP) 110,85,85 M88 1320
 85 CONTINUE M88 1330
 IF (ABS(XSO2 - .90) - .0001) 90,105,105 M88 1340
 90 IF (ABS(XSO2 - .901) - .0001) 105,95,95 M88 1350
 95 CONTINUE M88 1360
 NDIM = I M88 1370
 H = ADD M88 1380
 CALL QSF(H,Y,Z,NDIM) M88 1390
 DO 100 K=1,NDIM M88 1400
 WTCAT(K) = Z(K) * RAT M88 1410
 100 CONTINUE M88 1420
 CAT = WTCAT(NDIM) M88 1430
 ADD = .001 M88 1440
 105 CONTINUE M88 1450
 110 I = I - 1 M88 1460
 IF (ABS(XSO2 - XOVAL) - .0001) 170,170,115 M88 1470
 115 WRITE(3,120)J M88 1480
 170 FORMAT(/ / 22H MAX. CONVERSION STAGE 14) M88 1490
 WRITE(3,125)XSO2 M88 1500
 125 FORMAT(/17H MAX. CONVERSION=FR.5) M88 1510
 RANGE = XSO2 - X(J) M88 1520
 DESG = APPCH * RANGE M88 1530
 XSO2 = DESG + X(J) M88 1540
 N = I M88 1550
 DO 130 L=1,N M88 1560
 IF (XSO2 - TXSO2(L)) 135,130,130 M88 1570
 130 CONTINUE M88 1580
 135 I = I - 1 M88 1590
 XSO2 = TXSO2(I) M88 1600
 WRITE(3,140)XSO2 M88 1610
 140 FORMAT(/19H DESIGN CONVERSION=FR.5) M88 1620
 JJ = J + 1 M88 1630
 X(JJ) = XSO2 M88 1640
 XRANG = (1.0 - APPCH) * RANGE * TS02 M88 1650
 HRXN = XRANG * (- 41697. - 4.113 * TEMP + .6236E - 02 * TEMP ** M88 1660
 12. - .1246E - 05 * TEMP ** 3.) M88 1670
 WS03 = SO3 - XRANG M88 1680
 WS02 = SO2 + (SO3 - WS03) M88 1690
 W02 = O2 + .5 * (SO3 - WS03) M88 1700
 WZN2 = ZN2 M88 1710
 COFA = WZN2 * .1125E - 02 + W02 * .2792E - 02 + WS02 * .8561E - 02M88 1720
 1 + WS03 * .2118E - 01 M88 1730
 COFB = WZN2 * 11.743 + W02 * 11.066 + WS02 * 12.809 + WS03 * 10.93M88 1740
 19 M88 1750
 CONST = - 3599. * WZN2 - 3546. * W02 - 4578. * WS02 - 5141. * WS03M88 1760
 HLOSS = 0.0 M88 1770
 COFC = - HRXN - SENHT - HLOSS + CONST M88 1780
 DISCR = SQRT(COFB ** 2. - 4 * COFA * COFC) M88 1790
 ROOT = (- COFB + DISCR) / (2. * COFA) M88 1800
 IF (ROOT) 145,145,150 M88 1810

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M88 page 4
145 ROOT = ( - COFR - DISCR) / (2. * COFA) M88 1820
    IF (ROOT) 155,155,150 M88 1830
150 IF (ROOT - 2000.) 160,160,155 M88 1840
155 CONTINUE M88 1850
C ERROR 2 M88 1860
160 TEMP = ROOT M88 1870
    WRITE(3,165)TEMP M88 1880
165 FORMAT(/26H DESIGN CONV. TEMPERATURE=F6.0) M88 1890
    GOTO 175 M88 1900
170 I = I - 1 M88 1910
175 NDIM = I M88 1920
    H = ADD M88 1930
    CALL QSF(H,Y,Z,NDIM) M88 1940
    DO 180 K=1,NDIM M88 1950
        WTCAT(K) = Z(K) * RAT M88 1960
180 CONTINUE M88 1970
    WTCAT(NDIM) = WTCAT(NDIM) + CAT M88 1980
    VAT(J) = WTCAT(NDIM) M88 1990
    WRITE(3,185)J,WTCAT(NDIM),XS02 M88 2000
185 FORMAT( /7H STAGE 13,2X,9HCATALYST=F8.4,4X,5HXS02=F8.5) M88 2010
    CAT = 0.0 M88 2020
    IF (ABS(XS02 - XOVAL) - .0001) 215,215,190 M88 2030
190 TTEMP = TEMP M88 2040
    QBN2 = WZN2 * (11.743 * TEMP + .1125E - 02 * TEMP * * 2. - 3599.) M88 2050
    QBO2 = WO2 * (11.066 * TEMP + .2792E - 02 * TEMP * * 2. - 3546.) M88 2060
    QBSO2 = WSO2 * (12.809 * TEMP + .8561E - 02 * TEMP * * 2. - 4578.) M88 2070
    QBSO3 = WSO3 * (10.939 * TEMP + .2118E - 01 * TEMP * * 2. - 5141.) M88 2080
    TOTQ = QBN2 + QBO2 + QBSO2 + QBSO3. M88 2090
    TEMP = 673. M88 2100
    QBN2 = WZN2 * (11.743 * TEMP + .1125E - 02 * TEMP * * 2. - 3599.) M88 2110
    QBO2 = WO2 * (11.066 * TEMP + .2792E - 02 * TEMP * * 2. - 3546.) M88 2120
    QBSO2 = WSO2 * (12.809 * TEMP + .8561E - 02 * TEMP * * 2. - 4578.) M88 2130
    QBSO3 = WSO3 * (10.939 * TEMP + .2118E - 01 * TEMP * * 2. - 5141.) M88 2140
    TOTQA = QBN2 + QBO2 + QBSO2 + QBSO3 M88 2150
    A = TOTQ - TOTQA M88 2160
    IF (J - 1) 195,195,200 M88 2170
195 ZN2M(9) = WZN2 M88 2180
    O2M(9) = WO2 M88 2190
    SO3M(9) = WSO3 M88 2200
    SO2M(9) = WSO2 M88 2210
    TOTM(9) = ZN2M(9) + O2M(9) + SO3M(9) + SO2M(9) M88 2220
    TEMPK(9) = TTEMP M88 2230
    ZN2M(10) = ZN2M(9) M88 2240
    O2M(10) = O2M(9) M88 2250
    SO3M(10) = SO3M(9) M88 2260
    SO2M(10) = SO2M(9) M88 2270
    TOTM(10) = TOTM(9) M88 2280
    TEMPK(10) = 673. M88 2290
    SENHT = TOTQA M88 2300
    GOTO 205 M88 2310
200 R = A / 5972. M88 2320
    ZN2M(NJ) = R M88 2330
    O2M(NJ) = R * .266 M88 2340
    TOTM(NJ) = ZN2M(NJ) + O2M(NJ) M88 2350
    NJ = NJ + 1 M88 2360
    ZN2 = ZN2 + R M88 2370
    TO2 = WO2 + .266 * R M88 2380
    QBN2 = ZN2 * (11.743 * TEMP + .1125E - 02 * TEMP * * 2. - 3599.) M88 2390
    QBO2 = TO2 * (11.066 * TEMP + .2792E - 02 * TEMP * * 2. - 3546.) M88 2400
    QSO2 = WSO2 * (12.809 * TEMP + .8561E - 02 * TEMP * * 2. - 4578.) M88 2410
    QSO3 = WSO3 * (10.939 * TEMP + .2118E - 01 * TEMP * * 2. - 5141.) M88 2420

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M88 PAGE 5
SENHT = QBN2 + QB02 + QS02 + QS03
205 XSO2 = XSO2 - ADD
210 CONTINUE
215 WRITE(3,220)J
220 FORMAT(/ /24H FINAL CONVERSION STAGE 13)
WRITE(3,225)TEMP
225 FORMAT(/ 13H FINAL TEMP.=F7.1)
WRITE(3,230)XSO2
230 FORMAT(/15H FINAL CONVER.=F8.5)
NJ = NJ + 38
TT = J
WRITE(3,235)
235 FORMAT(///// ' POLUTION EMISSION')
PPM = (SO2 * 64. / (SO2 * 64. + O2 * 32. + ZN2 * 28. + SO3 * 80.))
1* 1000000.
WRITE(3,240)PPM
240 FORMAT(1X, 9HEXIT SO2=F8.2)
245 CONTINUE
WRITE(9'1) SO3,SO2,O2,ZN2,TOT,TEMP,RH,RAT,Q,A,PRESS,NJ,TT
WRITE(8'1) TEMPK
WRITE(8'2) H2OM
WRITE(8'3) TOTM
WRITE(8'4) SO2M
WRITE(8'5) ZN2M
WRITE(8'6) O2M
WRITE(8'7) ACIDM
WRITE(8'8) SO3M
WRITE(8'9) VAT
CALL LINK(N88)
END
*STORE
M88 2430
M88 2440
M88 2450
M88 2460
M88 2470
M88 2480
M88 2490
M88 2500
M88 2510
M88 2520
M88 2530
M88 2540
M88 2550
M88 2560
M88 2570
M88 2580
M88 2590
M88 2600
M88 2610
M88 2620
M88 2630
M88 2640
M88 2650
M88 2660
M88 2670
M88 2680
M88 2690
M88 2700
M88 2710
M88 2720
M88 2730

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M88 PAGE 6

Eklund RATE EQUATION

$$ELOGK = 4956. / TEMP - 4.678$$

$$TKPE = 10. ** ELOGK$$

$$C1 = 0.4248762 E-02$$

$$C2 = -0.1471751 E-04$$

$$C3 = 0.1411377 E-07$$

$$C4 = -0.3580165 E-14$$

$$TK = C1 + C2 * TEMP + C3 * TEMP**2$$

$$+ C4 * TEMP**4$$

$$RATE = 8.5 E+06 * TK * (SO2MF/SO3MF)$$

$$** .5 * (O2MF * PRESS -$$

$$(SO3MF / (SO2MF * TKPE)) ** 2)$$

SUBPROGRAM N88 - HEAT AND MATERIAL BALANCE

<u>STATEMENT NOS.</u>	<u>DESCRIPTION</u>
10 / 250	Initialization
260 / 690	Balance Around E 1, E 2
700 / 1570	Balance Around T 1
1580 / 1740	Balance Around T 2
1750 / 2740	Balance all Streams
2750 / 2960	Call Link to N 89

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N 88 PAGE 1
DEFINE FILE 8(10,40,U,18) ,9(1,40,U,19) ,7(10,90,U,17) N88 0010
DIMENSION SM(45) ,ACIDM(45) ,H2OM(45) ,TOTM(45) ,SO2M(45) ,ZN2M(45) N88 0020
1,O2M(45) ,TEMPK(45) ,SO3M(45) ,VAT(8) N88 0030
DO 5 I=1,45 N88 0040
SM(I) = 0.0 N88 0050
ACIDM(I) = 0.0 N88 0060
H2OM(I) = 0.0 N88 0070
TOTM(I) = 0.0 N88 0080
SO2M(I) = 0.0 N88 0090
ZN2M(I) = 0.0 N88 0100
O2M(I) = 0.0 N88 0110
TEMPK(I) = 0.0 N88 0120
SO3M(I) = 0.0 N88 0130
5 CONTINUE N88 0140
READ(9'1) SO3,SO2,O2,ZN2,TOT,TEMP,RH,RAT,Q,A,PRESS,NJ,TT N88 0150
READ(8'6) (O2M(I) ,I = 40,44) ,O2M(18) , (O2M(I) ,I = 7,11) N88 0160
READ(8'9) (VAT(I) ,I = 1,8) N88 0170
READ(8'1) (TEMPK(I) ,I = 40,44) ,TEMPK(18) , (TEMPK(I) ,I = 7,11) N88 0180
READ(8'7) (ACIDM(I) ,I = 40,44) ,ACIDM(18) , (ACIDM(I) ,I = 7,11) N88 0190
READ(8'2) (H2OM(I) ,I = 40,44) ,H2OM(18) , (H2OM(I) ,I = 7,11) N88 0200
READ(8'3) (TOTM(I) ,I = 40,44) ,TOTM(18) , (TOTM(I) ,I = 7,11) N88 0210
READ(8'8) (SO3M(I) ,I = 40,44) ,SO3M(18) , (SO3M(I) ,I = 7,11) N88 0220
READ(8'5) (ZN2M(I) ,I = 40,44) ,ZN2M(18) , (ZN2M(I) ,I = 7,11) N88 0230
READ(8'4) (SO2M(I) ,I = 40,44) ,SO2M(18) , (SO2M(I) ,I = 7,11) N88 0240
WRITE(3,10) N88 0250
10 FORMAT(///' HEAT EXCHANGER DUTY') N88 0260
WRITE(3,15)Q N88 0270
15 FORMAT(//18H HEAT DUTY ON E-1=E12.4) N88 0280
SO3M(12) = SO3 N88 0290
SO2M(12) = SO2 N88 0300
O2M(12) = O2 N88 0310
ZN2M(12) = ZN2 N88 0320
TOTM(12) = TOT N88 0330
TEMPK(12) = TEMP N88 0340
SM(6) = ACIDM(18) N88 0350
TOTM(6) = SM(6) N88 0360
TEMPK(6) = 400. N88 0370
O2M(4) = TOTM(7) * .21 N88 0380
ZN2M(4) = TOTM(7) * .79 N88 0390
H2OM(4) = TOTM(7) * 359. * 1. / (1000. * 454. * 18.) N88 0400
TOTM(4) = O2M(4) + ZN2M(4) + H2OM(4) N88 0410
NJJ = NJ N88 0420
NY = NJ - 39 N88 0430
NJ = 39 N88 0440
DO 20 KK=1,NY N88 0450
NJ = NJ + 1 N88 0460
ZN2M(5) = ZN2M(NJ) + ZN2M(5) N88 0470
O2M(5) = O2M(NJ) + O2M(5) N88 0480
20 CONTINUE N88 0490
TOTM(5) = ZN2M(5) + O2M(5) N88 0500
H2OM(5) = H2OM(4) * (TOTM(5) / TOTM(4)) N88 0510
TOTM(5) = TOTM(5) + H2OM(5) N88 0520
QCN2 = ZN2M(12) * (11.743 * (TEMPK(12) - 505.) + .1125E - 02 * (TEN88 0530
1MPK(12) * * 2. - 505. * * 2)) N88 0540
QCO2 = O2M(12) * (11.066 * (TEMPK(12) - 505.) + .2792E - 02 * (TEMN88 0550
1PK(12) * * 2. - 505. * * 2)) N88 0560
QCSO2 = SO2M(12) * (12.839 * (TEMPK(12) - 505.) + .8561E - 02 * (TN88 0570
1EMPK(12) * * 2. - 505. * * 2)) N88 0580
QCSO3 = SO3M(12) * (10.939 * (TEMPK(12) - 505.) + .2118E - 01 * (TN88 0590
1EMPK(12) * * 2. - 505. * * 2)) N88 0600

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NBB PAGE 2

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B = QCN2 + QCO2 + QCSO2 + QCSO3
WRITE(3,25)B
25 FORMAT(//18H HEAT DUTY ON E-2=E12.4)
ZN2M(13) = ZN2M(12)
O2M(13) = O2M(12)
SO3M(13) = SO3M(12)
SO2M(13) = SO2M(12)
TOTM(13) = TOTM(12)
TEMPK(13) = 505.
O2M(3) = O2M(4) + O2M(5)
ZN2M(3) = ZN2M(4) + ZN2M(5)
H2OM(3) = H2OM(4) + H2OM(5)
TOTM(3) = O2M(3) + ZN2M(3) + H2OM(3)
O2M(2) = O2M(3)
ZN2M(2) = ZN2M(3)
TOTC = O2M(2) + ZN2M(2)
H2OM(2) = RH * (23.756 / 760.) * TOTC
TOTM(2) = O2M(2) + ZN2M(2) + H2OM(2)
TEMPK(2) = 298.
O2M(1) = O2M(2)
ZN2M(1) = ZN2M(2)
H2OM(1) = H2OM(2)
TOTM(1) = TOTM(2)
TEMPK(1) = TEMPK(2)
H2OM(20) = ACIDM(18) + (ACIDM(18) * 98. * (.02 / .98)) / 18. - H2ON
1M(2)
O2M(20) = 0.0
ZN2M(20) = 0.0
SO3M(20) = 0.0
TOTM(20) = H2OM(20)
TEMPK(20) = 298.
ACIDM(19) = (RAT * 2.5 * 15.374 * .98) / 98.
H2OM(19) = (ACIDM(19) * 98. * (.02 / .98)) / 18.
TEMPK(19) = 311.
TOTM(19) = ACIDM(19) + H2OM(19)
H2OM(21) = H2OM(20) + H2OM(19)
ACIDM(21) = ACIDM(20) + ACIDM(19)
TOTM(21) = H2OM(21) + ACIDM(21)
PCT = ACIDM(21) * 98. / (ACIDM(21) * 98. + H2OM(21) * 18.)
HEAT = ( (1.0 - PCT) * 800. - 16.) * ACIDM(21) * 98.
DELT = HEAT / ( (ACIDM(21) * 98. + H2OM(21) * 18.) * .35)
T = 100. + DELT
TEMPK(21) = (T - 32.) * (5. / 9.) + 273.
NJ = NJJ
NY = NJ - 39
NJ = 39
TEMPK(3) = TEMPK(21) - 1.
DO 30 KK=1,NY
NJ = NJ + 1
TEMPK(NJ) = TEMPK(3)
30 CONTINUE
TEMPK(5) = TEMPK(3)
TEMPK(4) = TEMPK(3)
QDN2 = ZN2M(2) * (11.743 * (TEMPK(3) - 298.) + .1125E - 02 * (TEMPN
1K(3) * * 2. - 298. * * 2))
QDO2 = O2M(2) * (11.066 * (TEMPK(3) - 298.) + .2792E - 02 * (TEMPKN
1(3) * * 2. - 298. * * 2))
QDTOT = QDN2 + QDO2
RFF = (1.0 - PCT) * 800.
AH2O = H2OM(2) * 18. - H2OM(3) * 18.
PCT = (ACIDM(21) * 98.) / (ACIDM(21) * 98. + (H2OM(21) + H2OM(2)))

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N88 0610
N88 0620
N88 0630
N88 0640
N88 0650
N88 0660
N88 0670
N88 0680
N88 0690
N88 0700
N88 0710
N88 0720
N88 0730
N88 0740
N88 0750
N88 0760
N88 0770
N88 0780
N88 0790
N88 0800
N88 0810
N88 0820
N88 0830
N88 0840
N88 0850
N88 0860
N88 0870
N88 0880
N88 0890
N88 0900
N88 0910
N88 0920
N88 0930
N88 0940
N88 0950
N88 0960
N88 0970
N88 0980
N88 0990
N88 1000
N88 1010
N88 1020
N88 1030
N88 1040
N88 1050
N88 1060
N88 1070
N88 1080
N88 1090
N88 1100
N88 1110
N88 1120
N88 1130
N88 1140
N88 1150
N88 1160
N88 1170
N88 1180
N88 1190
N88 1200
N88 1210

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1* 18.)
HEAT = ( (1.0 - PCT) * 800. - REF) * ACIDM(21) * 98. N88 1220
QAA = HEAT + QOTOT N88 1230
DELT = (QAA / ( (ACIDM(21) * 98. + H2OM(21) * 18. + H2OM(2) * 18.) N88 1240
1* .35)) / 1.8 N88 1250
TEMPK(22) = TEMPK(21) + DELT N88 1260
ACIDM(22) = ACIDM(21) N88 1270
H2OM(22) = H2OM(21) + H2OM(2) N88 1280
TOTM(22) = ACIDM(22) + H2OM(22) N88 1290
QRN = SO3M(13) * 80. * 697. N88 1300
TOTP = RAT * 2.5 * 15.2 N88 1310
ACIDM(15) = (.98 * TOTP) / 98. N88 1320
H2OM(15) = (.02 * TOTP) / 18. N88 1330
TOTM(15) = ACIDM(15) + H2OM(15) N88 1340
ACIDM(16) = SO3M(13) + ACIDM(15) N88 1350
H2OM(16) = H2OM(15) - SO3M(13) N88 1360
TOTM(16) = ACIDM(16) + H2OM(16) N88 1370
ACIDM(17) = ACIDM(16) + ACIDM(22) N88 1380
H2OM(17) = H2OM(16) + H2OM(22) N88 1390
TOTM(17) = H2OM(17) + ACIDM(17) N88 1400
PCT = ACIDM(16) * 98. / (ACIDM(16) * 98. + H2OM(16) * 18.) N88 1410
HEAT = ( (1.0 - PCT) * 800. - 16.) * ACIDM(16) * 98. N88 1420
T1 = 100. N88 1430
DO 35 MN=1,300 N88 1440
T1 = T1 + 1.0 N88 1450
T22 = 1.8 * (TEMPK(22) - 273.) + 32. N88 1460
T2 = (T1 + T22) / 2. N88 1470
QN2 = ZN2M(13) * (11.743 * (505. - T2) + .1125E - 02 * (505. * * ZN88 1490
1 - T2 * * 2)) N88 1500
QO2 = O2M(13) * (11.066 * (505. - T2) + .2792E - 02 * (505. * * 2 N88 1510
QSO2 = SO2M(13) * (12.809 * (505. - T2) + .8561E - 02 * (505. * * N88 1520
12 - T2 * * 2)) N88 1530
QE = QN2 + QO2 + QSO2 N88 1540
QTOTB = QRN + HEAT + QE N88 1550
QTOTA = (ACIDM(16) * 98. + H2OM(16) * 18.) * .40 * (T1 - T2) N88 1560
SENTV = TOTP * .40 * 1.0 * 2.0 N88 1570
IF (ABS(QTOTA - QTOTB) - SENTV) 40,40,35 N88 1580
35 CONTINUE N88 1590
40 TEMPK(16) = .555 * (T1 - 32.) + 273. N88 1600
TEMPK(15) = .555 * (T2 - 32.) + 273. N88 1610
TEMPK(14) = TEMPK(15) + 1.0 N88 1620
O2M(14) = O2M(13) N88 1630
ZN2M(14) = ZN2M(13) N88 1640
SO2M(14) = SO2M(13) N88 1650
SO3M(14) = 0.0 N88 1660
TOTM(14) = O2M(14) + ZN2M(14) + SO2M(14) N88 1670
TEMPK(17) = TEMPK(15) N88 1680
TEMPK(36) = TEMPK(17) N88 1690
ACIDM(36) = ACIDM(17) - ACIDM(15) N88 1700
H2OM(36) = H2OM(17) - H2OM(15) N88 1710
TOTM(36) = ACIDM(36) + H2OM(36) N88 1720
TEM = (TEMPK(36) - 273.) * 1.8 + 32. N88 1730
QBB = (ACIDM(36) * 98. + H2OM(36) * 18.) * .4 * (TEM - 100.) N88 1740
WRITE(3,45)QBB N88 1750
45 FORMAT(//18H HEAT DUTY ON E-3=E12.4) N88 1760
TOTM(33) = (QBB / (1. * 63.)) / 18. N88 1770
H2OM(33) = TOTM(33) N88 1780
TOTM(34) = TOTM(33) N88 1790
H2OM(34) = H2OM(33) N88 1800
TEMPK(33) = 298. N88 1810
TEMPK(34) = 333. N88 1820

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E = COFA * (TEMPK(25) * * 2. - TEMPK(26) * * 2.) + COFB * (TEMPK(2N88 2440
15) - TEMPK(26)) N88 2450
WRITE(3,110)E N88 2460
110 FORMAT(//18H HEAT DUTY ON E-4=E12.4) N88 2470
WRITE(3,115)A N88 2480
115 FORMAT(//18H HEAT DUTY ON E-5=E12.4) N88 2490
H2OM(32) = (E / 825.1) / 18. N88 2500
TOTM(32) = H2OM(32) N88 2510
H2OM(29) = H2OM(32) + H2OM(11) + H2OM(38) N88 2520
TOTM(29) = TOTM(32) + TOTM(11) + TOTM(38) N88 2530
TEMPK(27) = 393. N88 2540
F = COFA * (TEMPK(26) * * 2. - TEMPK(27) * * 2.) + COFB * (TEMPK(2N88 2550
16) - TEMPK(27)) N88 2560
WRITE(3,120)F N88 2570
120 FORMAT(//18H HEAT DUTY ON E-6=E12.4) N88 2580
WRITE(3,125)C N88 2590
125 FORMAT(// ' HEAT DUTY ON E7 = 'E12.4) N88 2600
TEMPK(29) = 298. N88 2610
H2OM(35) = H2OM(29) N88 2620
TOTM(35) = H2OM(35) N88 2630
DTT = (F / (H2OM(29) * 18. * 1.1)) / 1.8 N88 2640
TEMPK(35) = TEMPK(29) + DTT N88 2650
DO 130 L=26,28 N88 2660
SO3M(L) = SO3M(14) N88 2670
ZN2M(L) = ZN2M(14) N88 2680
O2M(L) = O2M(14) N88 2690
SO2M(L) = SO2M(14) N88 2700
TOTM(L) = TOTM(14) N88 2710
130 CONTINUE N88 2720
TEMPK(28) = TEMPK(27) N88 2730
WRITE(3,135) N88 2740
135 FORMAT(///// ' COMPOSITIONS + TEMPERATURES') N88 2750
WRITE(3,140) N88 2760
140 FORMAT(/1X, 6HSTREAM,2X,5HTEMPK,5X,4HZN2M,9X,3HO2M,10X,4HH2OM,10X,N88 2770
14HSO2M,7X,4HSO3M,8X,5HH2SO4,8X,2HSM,8X,4HTOTM) N88 2780
DO 150 KK=1,NJJ N88 2790
WRITE(3,145)KK,TEMPK(KK),ZN2M(KK),O2M(KK),H2OM(KK),SO2M(KK),SO3M(KK),N88 2800
1K),ACIDM(KK),SM(KK),TOTM(KK) N88 2810
145 FORMAT(4X,I3,3X,F5.0,2X,E10.4,7(3X,E10.4)) N88 2820
150 CONTINUE N88 2830
WRITE(9'1) SO3,SO2,O2,ZN2,TOT,TEMP,RH,RAT,Q,A,PRESS,NJ,TT,QE,E,B,FN88 2840
1,C N88 2850
WRITE(7'1) TEMPK N88 2860
WRITE(7'2) H2OM N88 2870
WRITE(7'3) TOTM N88 2880
WRITE(7'4) SO2M N88 2890
WRITE(7'5) ZN2M N88 2900
WRITE(7'6) O2M N88 2910
WRITE(7'7) ACIDM N88 2920
WRITE(7'8) SO3M N88 2930
CALL LINK(N89) N88 2940
END N88 2950
*STORE N88 2960

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SUBPROGRAM - N 89 - EQUIPMENT DESIGN

<u>STATEMENT NOS.</u>	<u>DESCRIPTION</u>
150 / 220	F 1 Design
210 / 950	T 1 Design
960 / 1060	G 1 Design
1070 / 1440	E 1 Design
1450 / 1560	F 2 Design
1570 / 1920	R 1 Design
1930 / 2200	E 5 Design
2230 / 2290	Call Link N 90


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N&A PAQ I
DEFINE FILE8(10,40,U,I8) ,9(1,40,U,I9) ,7(10,90,U,I7) N89 0010
DIMENSION ACIDM(45) ,H2OM(45) ,TOTM(45) ,SO2M(45) ,ZN2M(45) ,O2M(45) N89 0020
1) ,TEMPK(45) ,SO3M(45) ,VAT(8) ,HIGH(8) ,PRES(20) ,Y(90) ,Z(90) N89 0030
READ(9'1) SO3,SO2,O2,ZN2,TOT,TEMP,RH,RAT,Q,A,PRESS,NJ,IT,UE,E,B,F,N89 0040
1C N89 0050
READ(7'1) (TEMPK(I) ,I = 1,45) N89 0060
READ(7'2) (H2OM(I) ,I = 1,45) N89 0070
READ(7'3) (TOTM(I) ,I = 1,45) N89 0080
READ(7'4) (SO2M(I) ,I = 1,45) N89 0090
READ(7'5) (ZN2M(I) ,I = 1,45) N89 0100
READ(7'6) (O2M(I) ,I = 1,45) N89 0110
READ(7'7) (ACIDM(I) ,I = 1,45) N89 0120
READ(7'8) (SO3M(I) ,I = 1,45) N89 0130
READ(8'9) (VAT(I) ,I = 1,3) N89 0140
C DESIGN OF AIR FILTER F1 N89 0150
VOLM = TOTM(1) * 359. / PRESS N89 0160
WRITE(3,5) N89 0170
5 FORMAT(//'/ ' **** DESIGN PARAMETERS FOR F1' ) N89 0180
WRITE(3,10) VOLM N89 0190
10 FORMAT(//4X, ' FILTER CAPACITY(ACFM) = 'F7.0) N89 0200
WRITE(3,65) PRESS N89 0210
C DESIGN OF DRYING TOWER COLUMN T1 N89 0220
DO 15 M=1,20 N89 0230
PRES(M) = PRESS N89 0240
15 CONTINUE N89 0250
G = (ZN2M(2) * 28. + O2M(2) * 32. + H2OM(2) * 18.) / 60. N89 0260
TL = (RAT * 2.5 * 15.2) / 60. N89 0270
DENLL = 15.2 / .1337 N89 0280
DENVV = 22. * PRES(1) / ((TEMPK(2) + TEMPK(3)) / 2.) N89 0290
VISCL = 3.8 N89 0300
COA = (TL / G) * (DENVV / DENLL) * * .5 N89 0310
TM = .377E + 04 * COA + .754E + 03 N89 0320
TN = .209E + 04 * COA + .880E + 03 N89 0330
TLOAD = .50 N89 0340
DELTP = 1.5 * TLOAD N89 0350
TO = DELTP / (TM * DELTP + TN) N89 0360
AREA = (G / DENLL) * (VISCL * * .2 / (DENVV * TO)) * * .5 N89 0370
GO = G * 3600. / AREA N89 0380
COFJ = 1.06 * (.146 * GO / .046) * * (-.41) N89 0390
PEN0 = (.24 * .046 / .0157) * * .666 N89 0400
HH = COFJ * .24 * GO / PEN0 N89 0410
TF2 = (9. / 5.) * (TEMPK(2) - 273.) + 32. N89 0420
TF3 = (9. / 5.) * (TEMPK(3) - 273.) + 32. N89 0430
TF22 = (9. / 5.) * (TEMPK(22) - 273.) + 32. N89 0440
TF21 = (9. / 5.) * (TEMPK(21) - 273.) + 32. N89 0450
TEMRO = (TF3 - TF2) / (TF22 - TF21) N89 0460
TINCR = (TF3 - TF2) / 50. N89 0470
TEMG = TF2 N89 0480
TEML = TF22 N89 0490
DO 20 LM=1,50 N89 0500
TEMG1 = TEMG N89 0510
TEMG = TEMG + TINCR N89 0520
TAVG = (TEMG1 + TEMG) / 2. N89 0530
TEML1 = TEML N89 0540
TEML = TEML - TINCR * (1. / TEMRO) N89 0550
TAVL = (TEML1 + TEML) / 2. N89 0560
DIFF = TAVL - TAVG N89 0570
RECP = 1. / DIFF N89 0580
Y(LM) = RECP N89 0590
20 CONTINUE N89 0600

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N89 PA9C 2
NDIM = 50
H = TINCR
CALL QSF(H,Y,Z,NDIM)
NDIM = 50
TINTR = Z(NDIM)
TM2 = (O2M(2) * 32. + ZN2M(2) * 28. + H2OM(2) * 18.) * 60.
TM3 = (O2M(3) * 32. + ZN2M(3) * 28. + H2OM(3) * 18.) * 60.
TMAVG = (TM2 + TM3) / 2.
CP = .24
SAREA = (TINTR * TMAVG * CP) / HH
THIGH = SAREA / (28. * AREA)
THIGH = 1.333 * THIGH
DIAM = SQRT(AREA / .785)
IF (THIGH - 10.) 25,25,30
25 THIGH = DIAM
30 PAC = AREA * THIGH
BEDS = THIGH / (3. * DIAM)
NBEDS = BEDS + 1
PRED = (.75 * THIGH + .30 * (.75 * THIGH)) / 12. * (14.7 / 34.)
THIGH = THIGH + .30 * DIAM
WRITE(3,35)
35 FORMAT(///// ' **** DESIGN PARAMETERS FOR T1')
WRITE(3,40)DIAM
40 FORMAT(/4X,' DIAMETER(FT) =F6.1)
WRITE(3,45)THIGH
45 FORMAT(1X,3X,' HEIGHT OR LENGHT(FT) =F6.1)
WRITE(3,50)PAC
50 FORMAT(1X,3X,' PACKING (CU FT) =F6.0)
WRITE(3,55)NBEDS
55 FORMAT(1X,3X,' NUMBER OF BEDS =I3)
WRITE(3,60)PRED
60 FORMAT(1X,3X,' PRESSURE DROP INSIDE (PSI) =F5.1)
WRITE(3,65)PRESS
65 FORMAT(1X,3X,' PRESSURE(ATM) =F5.0)
C
DESIGN OF SULFUR BURNER G1
VOLM = 40. * RAT * (32. / 98.) * (1. / PRES(2))
DIAM = (VOLM / 1.19) * * .3333
HT = 1.5 * DIAM
PD = .5
WRITE(3,70)
70 FORMAT(///// ' **** DESIGN PARAMETERS FOR G1')
WRITE(3,40)DIAM
WRITE(3,45)HT
WRITE(3,60)PD
WRITE(3,65)PRESS
C
DESIGN OF BURNER COOLER E-1
AVGTI = (TEMPK(7) + TEMPK(8)) / 2.
DENSI = 22. * PRES(3) / AVGTI
PHI = ( (TEMPK(7) - TEMPK(8)) / (AVGTI - 478.)) * .782
BI = 1.0 + .00184 / PHI
CI = .570E - 10 * BI / DENSI * * 2
CEI = .750E - 08
QQ = Q * 60.
DT = (AVGTI - TEMPK(11)) * 1.8
U = 1.0
ENR = 800.
DO 75 MM=1,100
U = U + .3
AREA = QQ / (U * DT)
CA = (16000. * (AREA / 1000.) * * .88) / AREA * (ENR / 1000.)
CAI = CA * .20 / 8300.
N89 0610
N89 0620
N89 0630
N89 0640
N89 0650
N89 0660
N89 0670
N89 0680
N89 0690
N89 0700
N89 0710
N89 0720
N89 0730
N89 0740
N89 0750
N89 0760
N89 0770
N89 0780
N89 0790
N89 0800
N89 0810
N89 0820
N89 0830
N89 0840
N89 0850
N89 0860
N89 0870
N89 0880
N89 0890
N89 0900
N89 0910
N89 0920
N89 0930
N89 0940
N89 0950
N89 0960
N89 0970
N89 0980
N89 0990
N89 1000
N89 1010
N89 1020
N89 1030
N89 1040
N89 1050
N89 1060
N89 1070
N89 1080
N89 1090
N89 1100
N89 1110
N89 1120
N89 1130
N89 1140
N89 1150
N89 1160
N89 1170
N89 1180
N89 1190
N89 1200
N89 1210

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N89 PAGE 3
RATIO = 1.0
GIO = ( (CA1 / (CE1 * C1)) / (3.5 * RATIO - 1.)) * * .357
HI = .555E - 02 * GIO * * .8
REPV = 1. / HI
V = 1. / REPV
IF (ABS(U - V) - .5) 80,80,75
75 CONTINUE
80 FI = .055 * (.094 / ( (2.067 / 12.) * GIO)) * * .2
PDP1 = (1.00248 * 4. * FI * 20. * GIO * * 2) / (2. * 4.17E + 08 *
1DENS1 * 2.067)
WRITE(3,85)
85 FORMAT(////// '**** DESIGN PARAMETERS FOR E1')
WRITE(3,90)QO
90 FORMAT(//4X, ' HEAT EXCHANGE DUTY(BTU/HR) = 'E12.4)
WRITE(3,95)U
95 FORMAT(1X,3X, ' HEAT TRANSFER COFF(BTU/SQ FT/DEG F)='F6.1)
WRITE(3,100)DT
100 FORMAT(1X,3X, ' TEMPERATURE DIFFERENCE(DEG F) = 'F6.0)
WRITE(3,105)AREA
105 FORMAT(1X,3X, ' SURFACE AREA(SQ FT) = 'F7.0)
WRITE(3,60)PDP1
WRITE(3,65)PRESS
C DESIGN OF HOT GAS FILTER F2
VOLM = TOTM(7) * (1.0 / PRES(4)) * 359.
TUBES = VOLM / 2000.
DIAM = 1.37 * TUBES * * .475 * 1.5
HT = DIAM
PD = .3
WRITE(3,110)
110 FORMAT(////// '**** DESIGN PARAMETERS FOR F2')
WRITE(3,40)DIAM
WRITE(3,45)HT
WRITE(3,60)PD
WRITE(3,65)PRESS
C DESIGN OF CONVERTER
L = TT
TOTCT = 0.0
DO 115 LM=1,L
VAT(LM) = (VAT(LM) + VAT(LM) * .15) * 2000.
TOTCT = VAT(LM) + TOTCT
115 CONTINUE
TOTCT = TOTCT / 2000.
VOLF = TOTM(8) * 359. / PRES(5)
VOLF = VOLF * PRES(5) * * .5
XAREA = VOLF / 100.
DIAM = (XAREA / .785) * * .5
HT = 0.0
DO 120 LN=1,L
HIGH(LN) = (VAT(LN) / 40.) / XAREA
HT = HIGH(LN) + .15 * DIAM + HT
120 CONTINUE
TOTHT = HT + .15 * DIAM
G = ( (ZN2M(8) * 28. + O2M(8) * 32. + SO2M(8) * 64.) / 60.) / XARE
1A
DP = .0525
VISC = .286E - 04
RENO = (DP * G) / VISC
F = 79.5 / RENO * * .375
DEN = 26.6 * (PRES(5) / TEMPK(9))
DROP = (2. * F * G * * 2 * TOTHT) / (144. * DP * 32.2 * DEN)
DROP = DROP + .20 * DROP

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N89 Page 4
PDR = DROP * (VAT(1) / (VAT(1) + VAT(2) + VAT(3) + VAT(4)))
WRITE(3,125)
125 FORMAT(///// '**** DESIGN PARAMETERS FOR R1')
WRITE(3,40)DIAM
WRITE(3,45)TOTHT
WRITE(3,130)TOTCT
130 FORMAT(1X,3X,' CATALYST (TONS)                =F6.0)
WRITE(3,55)L
WRITE(3,60)DROP
WRITE(3,65)PRESS
C DESIGN OF BURNER COOLER E-5
AVGTI = (TEMPK(9) + TEMPK(10)) / 2.
DENSI = 22. * PRES(5) / AVGTI
PHI = ( (TEMPK(9) - TEMPK(10)) / (AVGTI - 478.)) * .782
BI = 1.0 + .00184 / PHI
CI = .570E - 10 * BI / DENSI * * 2
CEI = .750E - 08
QA = A * 60.
DT = (AVGTI - TEMPK(38)) * 1.8
U = 1.0
ENR = 800.
DO 135 MM=1,100
U = U + .3
AREA = QA / (U * DT)
CA = (16000. * (AREA / 1000.) * * .88) / AREA * (ENR / 1000.)
CAI = CA * .20 / 8300.
RATIO = 1.0
GIO = ( (CAI / (CEI * CI)) / (3.5 * RATIO - 1.)) * * .357
HI = .555E - 02 * GIO * * .8
REPV = 1. / HI
V = 1. / REPV
IF (ABS(U - V) - .5) 140,140,135
135 CONTINUE
140 FI = .055 * (.094 / ( (2.067 / 12.) * GIO)) * * .2
PDP1 = (1.00248 * 4. * FI * 20. * GIO * * 2) / (2. * 4.17E + 08 *
IDENSI * 2.067)
WRITE(3,145)
145 FORMAT(///// '**** DESIGN PARAMETERS FOR E5')
WRITE(3,90)JA
WRITE(3,95)U
WRITE(3,100)DT
WRITE(3,105)AREA
WRITE(3,60)PDP1
WRITE(3,65)PRESS
CALL LINK(N90)
END
*STORE

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N89 1830
N89 1840
N89 1850
N89 1860
N89 1870
N89 1880
N89 1890
N89 1900
N89 1910
N89 1920
N89 1930
N89 1940
N89 1950
N89 1960
N89 1970
N89 1980
N89 1990
N89 2000
N89 2010
N89 2020
N89 2030
N89 2040
N89 2050
N89 2060
N89 2070
N89 2080
N89 2090
N89 2100
N89 2110
N89 2120
N89 2130
N89 2140
N89 2150
N89 2160
N89 2170
N89 2180
N89 2190
N89 2200
N89 2210
N89 2220
N89 2230
N89 2240
N89 2250
N89 2260
N89 2270
N89 2280
N89 2290

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SUBPROGRAM N 90 - EQUIPMENT DESIGN

<u>STATEMENT NOS.</u>	<u>DESCRIPTION</u>
160 / 800	E 2 Design
810 / 1500	T 2 Design
1510 / 1860	E 4 Design
1870 / 2240	E 6 Design
2250 / 2790	E 7 Design

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N90 PAGE1
DEFINEFILE8(10,40,U,I8) ,9(1,40,U,I9) ,7(10,90,U,I7) N90 0010
DIMENSIONACIDM(45) ,H2OM(45) ,TOTM(45) ,SO2M(45) ,ZN2M(45) ,O2M(45)N90 0020
1) ,TEMP(45) ,SO3M(45) ,VAT(8) ,PRES(20) ,Y(90) ,Z(90) N90 0030
READ(9'1) SO3,SO2,O2,ZN2,TOT,TEMP,RH,RAT,U,A,PRESS,NJ,TT,UE,E,B,F,N90 0040
1C N90 0050
READ(7'1) (TEMPK(I) ,I = 1,45) N90 0060
READ(7'2) (H2OM(I) ,I = 1,45) N90 0070
READ(7'3) (TOTM(I) ,I = 1,45) N90 0080
READ(7'4) (SO2M(I) ,I = 1,45) N90 0090
READ(7'5) (ZN2M(I) ,I = 1,45) N90 0100
READ(7'6) (O2M(I) ,I = 1,45) N90 0110
READ(7'7) (ACIDM(I) ,I = 1,45) N90 0120
READ(7'8) (SO3M(I) ,I = 1,45) N90 0130
READ(8'9) (VAT(I) ,I = 1,8) N90 0140
C DESIGN OF REACTOR COOLER E-2 N90 0150
DO 5 M=1,20 N90 0160
PRES(M) = PRESS N90 0170
5 CONTINUE N90 0180
AVGTO = (TEMPK(14) + TEMPK(23)) / 2. N90 0190
AVGTI = (TEMPK(12) + TEMPK(13)) / 2. N90 0200
DENSO = 22. * PRES(9) / AVGTO N90 0210
DENSI = 22. * PRES(6) / AVGTI N90 0220
DELT1 = TEMPK(12) - TEMPK(14) N90 0230
DELT2 = TEMPK(23) - TEMPK(13) N90 0240
TLMDL = ( (DELT2 - DELT1) / ALOG(DELT2 / DELT1)) * 1.8 N90 0250
PHI = ( (TEMPK(13) - TEMPK(12)) / (AVGTO - AVGTI)) * .731 N90 0260
BI = 1.0 + .00184 / PHI N90 0270
CI = .554E - 10 * BI / DENSI * * 2 N90 0280
CO = .953E - 09 / DENSO * * 2 N90 0290
CEI = .750E - 08 N90 0300
CEO = CEI N90 0310
QB = B * 60. N90 0320
FNR = 800. N90 0330
U = 1.0 N90 0340
DO 20 MJ=1,100 N90 0350
U = U + .3 N90 0360
AREA = QB / (U * TLMDL) N90 0370
CA = 7.0 * (AREA / 1000.) * * ( - .6) * (ENR / 400.) N90 0380
CAI = CA * .20 / 8300. N90 0390
RATIO = .95 N90 0400
DO 10 MJ=1,50 N90 0410
RATIO = RATIO + .03 N90 0420
GIO = ( (CAI / (CEI * CI)) / (2.5 + 2.76 * (1. / RATIO))) * * .357N90 0430
GOO = ( (CAI / (CEO * CO)) / (3.75 + 3.39 * RATIO)) * * .351 N90 0440
HI = .529E - 02 * GIO * * .8 N90 0450
HO = .72E - 01 * GOO * * .6 N90 0460
RATIA = 1.14 * (HO / HI) N90 0470
IF (ABS(RATIA - RATIO) - .02) 15,15,10 N90 0480
10 CONTINUE N90 0490
15 REPV = 1. / HI + .876 / HO N90 0500
V = 1. / REPV N90 0510
IF (ABS(U - V) - .5) 25,25,20 N90 0520
20 CONTINUE N90 0530
25 FI = .055 * (.0772 / ( (2.067 / 12.) * GIO)) * * .2 N90 0540
PDP1 = (BI * 4. * FI * 20. * GIO * * 2) / (2. * 4.17E + 08 * DENSI)N90 0550
1 * 2.^67) N90 0560
FO = .31 * (.382 / ( (2.375 / 12.) * GOO)) * * .15 N90 0570
FO = .31 * (.382 / ( (2.375 / 12.) * GOO)) * * .15 N90 0580
TMASS = TOTM(14) * 29. * 50. N90 0590
TUBE = TMASS / (GIO * .0234) N90 0600

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N90 page 2

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PDP2 = (4. * FO * 1. * TUBE * GOO * * 2) / (2. * 4.17E + 08 * DENS
10)
PDP2 = PDP1 * 1.75
WRITE(3,30)
30 FORMAT(///// '**** DESIGN PARAMETERS FOR E2')
WRITE(3,35)QB
35 FORMAT(//4X, ' HEAT EXCHANGE DUTY(BTU/HR) =E12.4)
WRITE(3,40)U
40 FORMAT(1X,3X, ' HEAT TRANSFER COFF(BTU/SQ FT/DEG F)=F6.1)
WRITE(3,45)TLMDL
45 FORMAT(1X,3X, ' TEMPERATURE DIFFERENCE(DEG F) =F6.0)
WRITE(3,50)AREA
50 FORMAT(1X,3X, ' SURFACE AREA(SQ FT) =F7.0)
WRITE(3,55)PDP1
55 FORMAT(1X,3X, ' PRESSURE DROP INSIDE (PSI) =F5.1)
WRITE(3,60)PDP2
60 FORMAT(1X,3X, ' PRESSURE DROP OUTSIDE(PSI) =F5.1)
WRITE(3,65)PRESS
65 FORMAT(1X,3X, ' PRESSURE(ATM) =F5.0)
C
DESIGN CALCULATIONS OF ABSORPTION COLUMN T2
G = (SO3M(13) * 80. + SO2M(13) * 64. + ZN2M(13) * 28. + O2M(13) *
132.) / 60.
TL = (RAT * 2.5 * 15.2) / 60.
DENLL = 15.2 / .1337
DENVV = 22. * PRES(7) / ((TEMPK(13) + TEMPK(14)) / 2.)
VISCL = 3.8
COA = (TL / G) * (DENVV / DENLL) * * .5
TM = .377E + 04 * COA + .754E + 03
TN = .209E + 04 * COA + .880E + 03
TLOAD = .50
DELTP = 1.5 * TLOAD
TO = DELTP / (TM * DELTP + TN)
AREA = (G / DENLL) * (VISCL * * .2 / (DENVV * TO)) * * .5
GO = G * 3600. / AREA
COFJ = 1.06 * (.146 * GO / .0675) * * (-.41)
PEN0 = (.25 * .0675 / .027) * * .666
MH = COFJ * .25 * GO / PEN0
TF13 = 1.8 * (TEMPK(13) - 273.) + 32.
TF14 = 1.8 * (TEMPK(14) - 273.) + 32.
TF16 = 1.8 * (TEMPK(16) - 273.) + 32.
TF15 = 1.8 * (TEMPK(15) - 273.) + 32.
TEMRO = (TF13 - TF14) / (TF16 - TF15)
TINCR = (TF13 - TF14) / 50.
TEMG = TF13
TEML = TF16
DO 70 LM=1,50
TEMG1 = TEMG
TEMG = TEMG - TINCR
TAVG = (TEMG1 + TEMG) / 2.
TEML1 = TEML
TEML = TEML - TINCR * (1. / TEMRO)
TAVL = (TEML1 + TEML) / 2.
DIFF = TAVG - TAVL
RECP = 1. / DIFF
Y(LM) = RECP
70 CONTINUE
NDIM = 50
H = TINCR
CALL QSF(H,Y,Z,NDIM)
NDIM = 50
TINTR = Z(NDIM)

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N90 PAGE 3
TM13 = (SO3M(13) * 80. + SO2M(13) * 64. + O2M(13) * 32. + ZN2M(13)N90 1220
1* 28.) * 60. N90 1230
TM14 = (SO3M(14) * 80. + SO2M(14) * 64. + O2M(14) * 32. + ZN2M(14)N90 1240
1* 28.) * 60. N90 1250
TMAVG = (TM13 + TM14) / 2. N90 1260
CP = .25 N90 1270
SAREA = (TINTR * TMAVG * CP) / HH N90 1280
THIGH = SAREA / (28. * AREA) N90 1290
THIGH = 1.333 * THIGH N90 1300
DIAM = (AREA / .785) ** .5 N90 1310
IF (THIGH - 10.) 75,75,80 N90 1320
75 THIGH = DIAM N90 1330
80 PAC = AREA * THIGH N90 1340
BEDS = THIGH / (3. * DIAM) N90 1350
NBEDS = BEDS + 1 N90 1360
PRED = (.75 * THIGH + .30 * (.75 * THIGH)) / 12. * (14.7 / 34.) N90 1370
THIGH = THIGH + .30 * DIAM N90 1380
WRITE(3,95) N90 1390
85 FORMAT(///// ' **** DESIGN PARAMETERS FOR T2') N90 1400
WRITE(3,90)DIAM N90 1410
90 FORMAT(//4X,' DIAMETER(FT) =F5.0) N90 1420
WRITE(3,95)THIGH N90 1430
95 FORMAT(1X,3X,' HEIGHT OR LENGHT(FT) =F5.0) N90 1440
WRITE(3,100)PAC N90 1450
100 FORMAT(1X,3X,' PACKING (CU FT) =F6.0) N90 1460
WRITE(3,105)NBEDS N90 1470
105 FORMAT(1X,3X,' NUMBER OF BEDS =I3) N90 1480
WRITE(3,55)PRED N90 1490
WRITE(3,65)PRESS N90 1500
C DESIGN OF WASTE HEAT BOILER E-4 N90 1510
AVGTI = (TEMPK(25) + TEMPK(26)) / 2. N90 1520
DENSI = 22. * PRES(12) / AVGTI N90 1530
PHI = ( (TEMPK(25) - TEMPK(26)) / (AVGTI - 478.)) * .782 N90 1540
BI = 1.0 + .00184 / PHI N90 1550
CI = .580E - 10 * BI / DENSI * * 2 N90 1560
CEI = .75E - 08 N90 1570
QE = E * 60. N90 1580
ENR = 800. N90 1590
DTT = (AVGTI - TEMPK(32)) * 1.8 N90 1600
U = 1.0 N90 1610
DO 110 ML=1,200 N90 1620
U = U + .3 N90 1630
AREA = QE / (U * DTT) N90 1640
CA = (16000. * (AREA / 1000.) * * .88) / AREA * (ENR / 1000.) N90 1650
CAI = CA * .20 / 3300. N90 1660
RATIO = 1.0 N90 1670
GIO = ( (CAI / (CEI * CI)) / (3.5 * RATIO - 1.)) * * .357 N90 1680
HI = .530E - 02 * GIO * * .8 N90 1690
REPV = 1. / HI N90 1700
V = 1. / REPV N90 1710
ARE = QE / (V * DTT) N90 1720
IF (ABS(U - V) - .5) 115,115,110 N90 1730
110 CONTINUE N90 1740
115 FI = .055 * (.094 / ( (2.067 / 12.) * GIO)) * * .2 N90 1750
PDP1 = (BI * 4. * FI * 20. * GIO * * 2) / (2. * 4.17E + 08 * DENSI)N90 1760
1 * 2.067) N90 1770
WRITE(3,120) N90 1780
120 FORMAT(///// ' **** DESIGN PARAMETERS FOR E4') N90 1790
WRITE(3,35)QE N90 1800
WRITE(3,40)U N90 1810
WRITE(3,45)DTT N90 1820

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N90 PAGE 4
WRITE(3,50)AREA
WRITE(3,55)PDP1
WRITE(3,65)PRESS
C DESIGN OF FEED WATER HEATER E6
AVGTI = (TEMPK(26) + TEMPK(27)) / 2.
DENSI = 22. * PRES(13) / AVGTI
PHI = (TEMPK(26) - TEMPK(27)) / 230. * 1.8 * .780
BI = 1.0 + .00184 / PHI
CI = 7.1E - 11 / DENSI * * 2
CEI = .75E - 08
FQ = F * 60.
DT = 840.
U = 1.0
ENR = 800.
DO 125 NN=1,100
U = U + .3
AREA = FQ / (U * DT)
CA = (16000. * (AREA / 1000.) * * .88) / AREA * (ENR / 1000.)
CAI = CA * .20 / 8300.
RATIO = 1.0
GIO = ( (CAI / (CEI * CI)) / (3.5 * RATIO - 1.0)) * * .357
HI = .55E - 02 * GIO * * .8
HO = 150.
REPV = 1. / HI + .876 / HO
V = 1. / REPV
ARE = FQ / (V * 940.)
IF (ABS(U - V) - .5) 130,130,125
125 CONTINUE
130 FI = .055 * (.0635 / ( (1.033 / 12.) * GIO)) * * .2
PDP1 = (BI * 4. * FI * 20. * GIO * * 2) / (2. * 4.17E + 08 * DENSI
1 * 1.033)
PDP2 = 3.0
WRITE(3,135)
135 FORMAT(///// ' **** DESIGN PARAMETERS FOR E6')
WRITE(3,35)FQ
WRITE(3,40)U
WRITE(3,45)DT
WRITE(3,50)AREA
WRITE(3,55)PDP1
WRITE(3,60)PDP2
WRITE(3,65)PRESS
C DESIGN OF ECONOMIZER E7
AVGTO = (TEMPK(23) + TEMPK(24)) / 2.
AVGTI = (TEMPK(37) + TEMPK(7)) / 2.
DENSO = 22. * PRES(9) / AVGTO
DENSI = 22. * PRES(3) / AVGTI
DELT1 = TEMPK(7) - TEMPK(24)
DELT2 = TEMPK(37) - TEMPK(23)
TLMDL = ( (DELT2 - DELT1) / ALOG(DELT2 / DELT1)) * 1.8
PHI = ( (TEMPK(37) - TEMPK(7)) / (AVGTI - AVGTO)) * .782
BI = 1.0 + .00184 / PHI
CI = .570E - 10 * BI / DENSI * * 2
CO = .100E - 08 / DENSO * * 2
CEI = .750E - 08
CEO = CEI
QC = C * 60.
ENR = 800.
U = 1.0
DO 150 JM=1,100
U = U + .3
AREA = QC / (U * TLMDL)

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N90 1830
N90 1840
N90 1850
N90 1860
N90 1870
N90 1880
N90 1890
N90 1900
N90 1910
N90 1920
N90 1930
N90 1940
N90 1950
N90 1960
N90 1970
N90 1980
N90 1990
N90 2000
N90 2010
N90 2020
N90 2130
N90 2040
N90 2050
N90 2060
N90 2070
N90 2080
N90 2090
N90 2100
N90 2110
N90 2120
N90 2130
N90 2140
N90 2150
N90 2160
N90 2170
N90 2180
N90 2190
N90 2200
N90 2210
N90 2220
N90 2230
N90 2240
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N90 2270
N90 2280
N90 2290
N90 2300
N90 2310
N90 2320
N90 2330
N90 2340
N90 2350
N90 2360
N90 2370
N90 2380
N90 2390
N90 2400
N90 2410
N90 2420
N90 2430

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APPENDIX C - MISCELLANEOUS INFORMATION

- 1) "The Effect of Pressure on Equilibrium Conversion"
- 2) Total catalyst requirement vs pressure

FIG. (1)
EQUILIBRIUM
CONVERSION
VS
TEMPERATURE

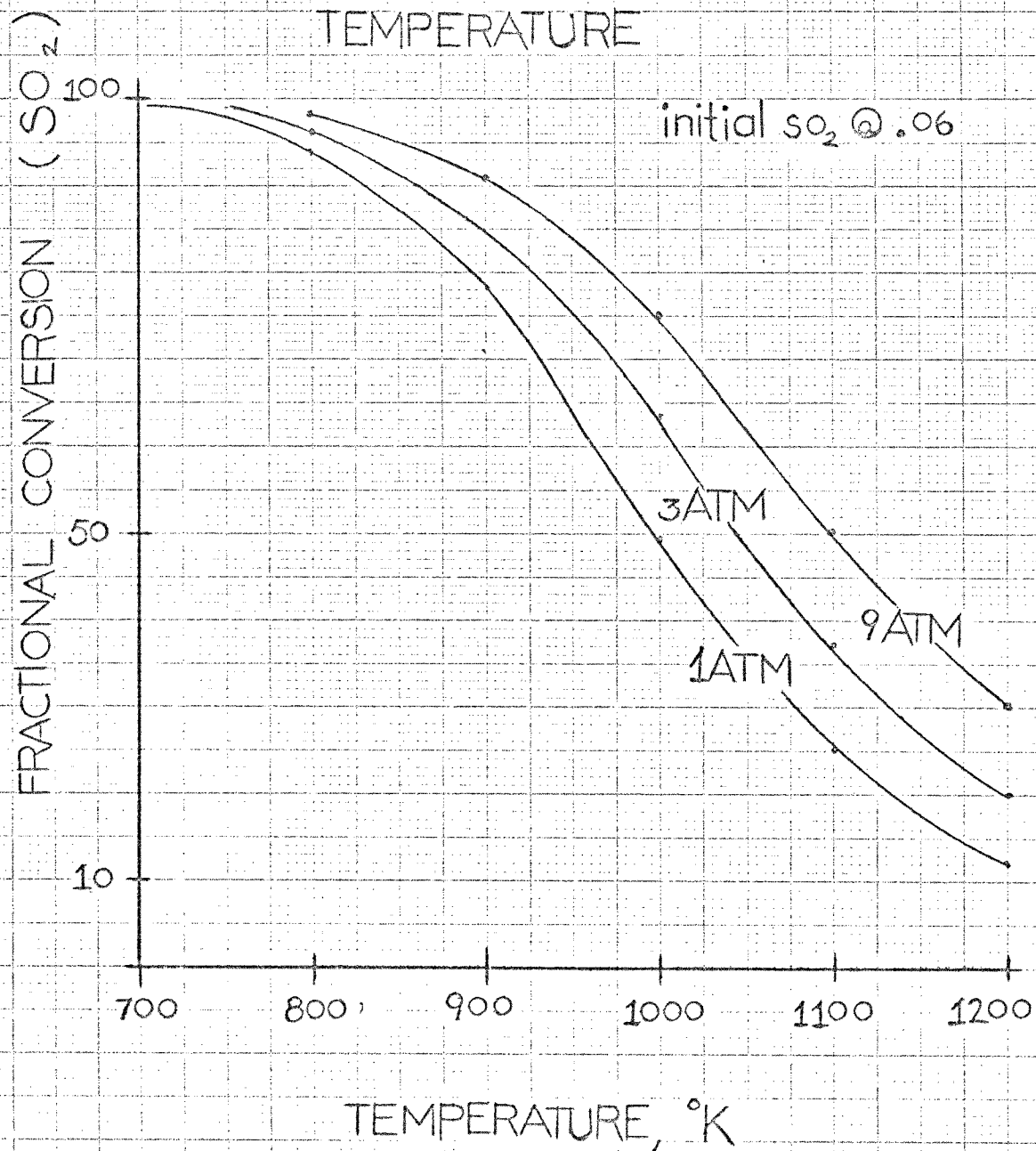
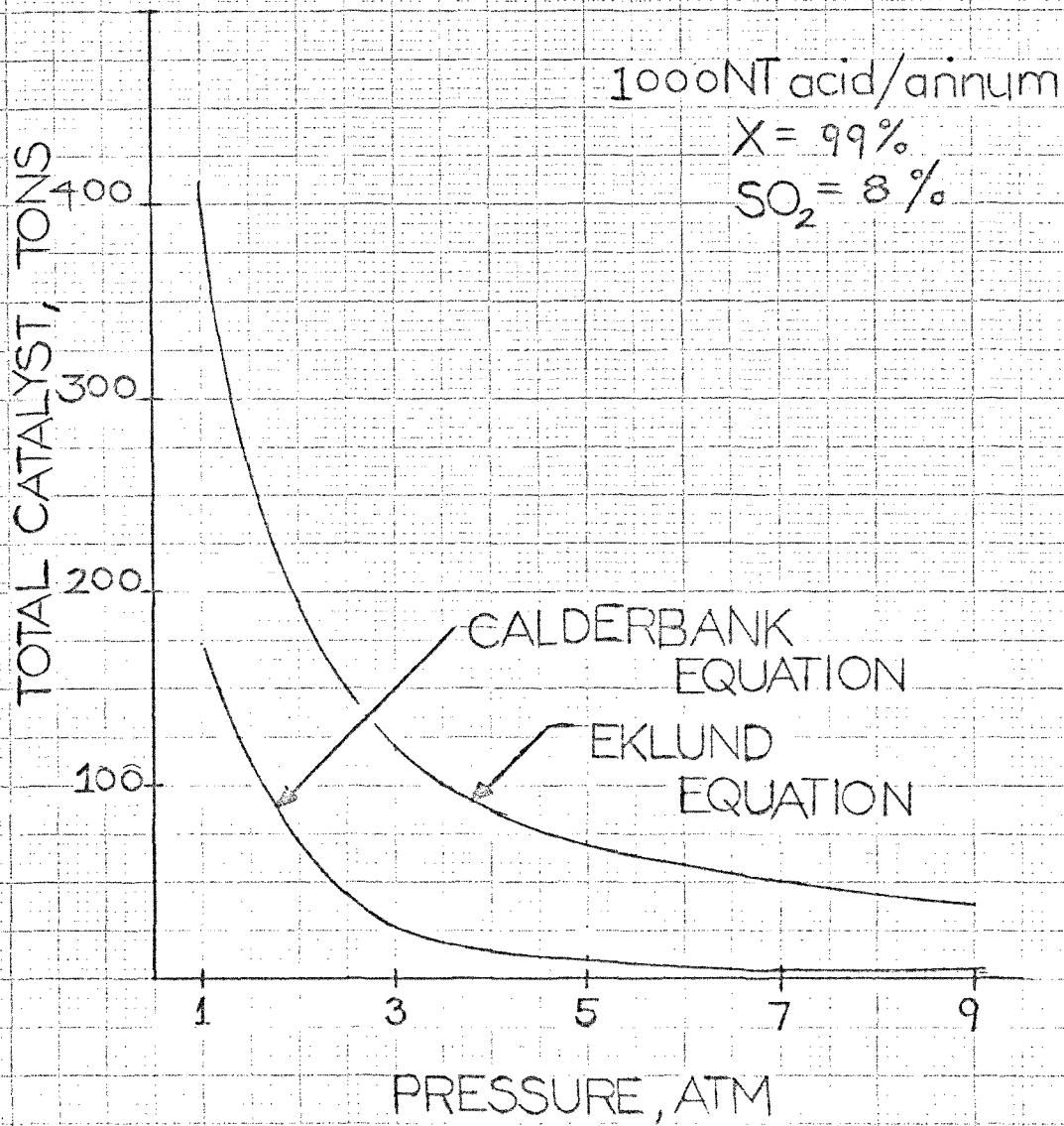


FIG (2)
TOTAL CATALYST
VS
PRESSURE



APPENDIX D - COMPUTER OUTPUT

**** DESIGN PARAMETERS FOR F1

FILTER CAPACITY (ACFM)	= 79310.
PRESSURE (ATM)	= 1.

**** DESIGN PARAMETERS FOR T1

DIAMETER (FT)	= 16.5
HEIGHT OR LENGHT (FT)	= 21.4
PACKING (CU FT)	= 3538.
NUMBER OF BEDS	= 1
PRESSURE DROP INSIDE (PSI)	= 0.5
PRESSURE (ATM)	= 1.

**** DESIGN PARAMETERS FOR G1

DIAMETER (FT)	= 22.2
HEIGHT OR LENGHT (FT)	= 33.4
PRESSURE DROP INSIDE (PSI)	= 0.5
PRESSURE (ATM)	= 1.

**** DESIGN PARAMETERS FOR E1

HEAT EXCHANGE DUTY (BTU/HR)	= 0.3856E 08
HEAT TRANSFER COFF (BTU/SQ FT/DEG F)	= 8.4
TEMPERATURE DIFFERENCE (DEG F)	= 648.
SURFACE AREA (SQ FT)	= 7001.
PRESSURE DROP INSIDE (PSI)	= 1.3
PRESSURE (ATM)	= 1.

 **** DESIGN PARAMETERS FOR F2

DIAMETER(FT)	=	10.8
HEIGHT OR LENGHT(FT)	=	10.8
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	1.

 **** DESIGN PARAMETERS FOR R1

DIAMETER(FT)	=	29.2
HEIGHT OR LENGHT(FT)	=	52.8
CATALYST (TONS)	=	473.
NUMBER OF BEDS	=	3
PRESSURE DROP INSIDE (PSI)	=	3.5
PRESSURE(ATM)	=	1.

 **** DESIGN PARAMETERS FOR E5

HEAT EXCHANGE DUTY(BTU/HR)	=	0.8789E 07
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	9.0
TEMPERATURE DIFFERENCE(DEG F)	=	533.
SURFACE AREA(SQ FT)	=	1811.
PRESSURE DROP INSIDE (PSI)	=	1.5
PRESSURE(ATM)	=	1.

 **** DESIGN PARAMETERS FOR E2

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3091E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	2.5
TEMPERATURE DIFFERENCE(DEG F)	=	298.
SURFACE AREA(SQ FT)	=	41479.
PRESSURE DROP INSIDE (PSI)	=	0.2
PRESSURE DROP OUTSIDE(PSI)	=	0.5
PRESSURE(ATM)	=	1.

 **** DESIGN PARAMETERS FOR I2

DIAMETER(FT)	= 18.
HEIGHT OR LENGHT(FT)	= 23.
PACKING (CU FT)	= 4580.
NUMBER OF BEDS	= 1
PRESSURE DROP INSIDE (PSI)	= 0.6
PRESSURE(ATM)	= 1.

**** DESIGN PARAMETERS FOR E4

HEAT EXCHANGE DUTY(BTU/HR)	= 0.4603E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	= 8.7
TEMPERATURE DIFFERENCE(DEG F)	= 330.
SURFACE AREA(SQ FT)	= 15838.
PRESSURE DROP INSIDE (PSI)	= 1.3
PRESSURE(ATM)	= 1.

**** DESIGN PARAMETERS FOR E6

HEAT EXCHANGE DUTY(BTU/HR)	= 0.1666E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	= 10.8
TEMPERATURE DIFFERENCE(DEG F)	= 840.
SURFACE AREA(SQ FT)	= 1819.
PRESSURE DROP INSIDE (PSI)	= 3.3
PRESSURE DROP OUTSIDE(PSI)	= 3.0
PRESSURE(ATM)	= 1.

**** DESIGN PARAMETERS FOR E7

HEAT EXCHANGE DUTY(BTU/HR)	= 0.3524E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	= 1.9
TEMPERATURE DIFFERENCE(DEG F)	= 560.
SURFACE AREA(SQ FT)	= 33120.
PRESSURE DROP INSIDE (PSI)	= 0.2
PRESSURE DROP OUTSIDE(PSI)	= 0.5
PRESSURE(ATM)	= 1.

**** DESIGN PARAMETERS FOR F1

FILTER CAPACITY (ACFM)	= 25669.
PRESSURE (ATM)	= 3.

**** DESIGN PARAMETERS FOR T1

DIAMETER (FT)	= 13.1
HEIGHT OR LENGTH (FT)	= 17.0
PACKING (CU FT)	= 1772.
NUMBER OF BEDS	= 1
PRESSURE DROP INSIDE (PSI)	= 0.4
PRESSURE (ATM)	= 3.

**** DESIGN PARAMETERS FOR G1

DIAMETER (FT)	= 15.4
HEIGHT OR LENGTH (FT)	= 23.1
PRESSURE DROP INSIDE (PSI)	= 0.5
PRESSURE (ATM)	= 3.

**** DESIGN PARAMETERS FOR E1

HEAT EXCHANGE DUTY (BTU/HR)	= 0.3856E 08
HEAT TRANSFER COEFF (BTU/SQ FT/DEG F)	= 16.2
TEMPERATURE DIFFERENCE (DEG F)	= 648.
SURFACE AREA (SQ FT)	= 3650.
PRESSURE DROP INSIDE (PSI)	= 1.9
PRESSURE (ATM)	= 3.

**** DESIGN PARAMETERS FOR F2

DIAMETER(FT) = 6.4
 HEIGHT OR LENGHT(FT) = 6.4
 PRESSURE DROP INSIDE (PSI) = 0.3
 PRESSURE(ATM) = 3.

**** DESIGN PARAMETERS FOR R1

DIAMETER(FT) = 22.1
 HEIGHT OR LENGHT(FT) = 31.3
 CATALYST (TONS) = 139.
 NUMBER OF BEDS = 3
 PRESSURE DROP INSIDE (PSI) = 1.7
 PRESSURE(ATM) = 3.

**** DESIGN PARAMETERS FOR E5

HEAT EXCHANGE DUTY(BTU/HR) = 0.7003E 07
 HEAT TRANSFER COFF(BTU/SQ FT/DEG F)= 18.0
 TEMPERATURE DIFFERENCE(DEG F) = 543.
 SURFACE AREA(SQ FT) = 712.
 PRESSURE DROP INSIDE (PSI) = 2.2
 PRESSURE(ATM) = 3.

**** DESIGN PARAMETERS FOR E2

HEAT EXCHANGE DUTY(BTU/HR) = 0.2964E 08
 HEAT TRANSFER COFF(BTU/SQ FT/DEG F)= 5.1
 TEMPERATURE DIFFERENCE(DEG F) = 293.
 SURFACE AREA(SQ FT) = 19399.
 PRESSURE DROP INSIDE (PSI) = 0.5
 PRESSURE DROP OUTSIDE(PSI) = 0.8
 PRESSURE(ATM) = 3.

**** DESIGN PARAMETERS FOR T2

DIAMETER(FT)	=	14.
HEIGHT OR LENGHT(FT)	=	18.
PACKING (CU FT)	=	2247.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.4
PRESSURE(ATM)	=	3.

**** DESIGN PARAMETERS FOR E4

HEAT EXCHANGE DUTY(BTU/HR)	=	0.4444E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	17.4
TEMPERATURE DIFFERENCE(DEG F)	=	330.
SURFACE AREA(SQ FT)	=	7689.
PRESSURE DROP INSIDE (PSI)	=	1.8
PRESSURE(ATM)	=	3.

**** DESIGN PARAMETERS FOR E6

HEAT EXCHANGE DUTY(BTU/HR)	=	0.1609E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	20.1
TEMPERATURE DIFFERENCE(DEG F)	=	840.
SURFACE AREA(SQ FT)	=	948.
PRESSURE DROP INSIDE (PSI)	=	4.7
PRESSURE DROP OUTSIDE(PSI)	=	3.0
PRESSURE(ATM)	=	3.

**** DESIGN PARAMETERS FOR E7

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3435E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	3.9
TEMPERATURE DIFFERENCE(DEG F)	=	561.
SURFACE AREA(SQ FT)	=	15297.
PRESSURE DROP INSIDE (PSI)	=	0.4
PRESSURE DROP OUTSIDE(PSI)	=	0.9
PRESSURE(ATM)	=	3.

**** DESIGN PARAMETERS FOR F1

FILTER CAPACITY (ACFM) = 13597.
 PRESSURE (ATM) = 5.

**** DESIGN PARAMETERS FOR T1

DIAMETER (FT) = 11.4
 HEIGHT OR LENGTH (FT) = 14.8
 PACKING (CU FT) = 1164.
 NUMBER OF BEDS = 1
 PRESSURE DROP INSIDE (PSI) = 0.4
 PRESSURE (ATM) = 5.

**** DESIGN PARAMETERS FOR G1

DIAMETER (FT) = 13.0
 HEIGHT OR LENGTH (FT) = 19.5
 PRESSURE DROP INSIDE (PSI) = 0.5
 PRESSURE (ATM) = 5.

**** DESIGN PARAMETERS FOR E1

HEAT EXCHANGE DUTY (BTU/HR) = 0.3856E 08
 HEAT TRANSFER COEFF (BTU/SQ FT/DEG F) = 22.2
 TEMPERATURE DIFFERENCE (DEG F) = 648.
 SURFACE AREA (SQ FT) = 2668.
 PRESSURE DROP INSIDE (PSI) = 2.3
 PRESSURE (ATM) = 5.

**** DESIGN PARAMETERS FOR F2

DIAMETER(FT)	=	5.0
HEIGHT OR LENGHT(FT)	=	5.0
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	5.

**** DESIGN PARAMETERS FOR R1

DIAMETER(FT)	=	19.5
HEIGHT OR LENGHT(FT)	=	22.6
CATALYST (TONS)	=	83.
NUMBER OF BEDS	=	2
PRESSURE DROP INSIDE (PSI)	=	1.1
PRESSURE(ATM)	=	5.

**** DESIGN PARAMETERS FOR E5

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3782E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	23.1
TEMPERATURE DIFFERENCE(DEG F)	=	548.
SURFACE AREA(SQ FT)	=	2973.
PRESSURE DROP INSIDE (PSI)	=	2.3
PRESSURE(ATM)	=	5.

**** DESIGN PARAMETERS FOR E2

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3356E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	7.2
TEMPERATURE DIFFERENCE(DEG F)	=	387.
SURFACE AREA(SQ FT)	=	11861.
PRESSURE DROP INSIDE (PSI)	=	0.6
PRESSURE DROP OUTSIDE(PSI)	=	1.2
PRESSURE(ATM)	=	5.

**** DESIGN PARAMETERS FOR T2

DIAMETER (FT)	=	12.
HEIGHT OR LENGHT (FT)	=	16.
PACKING (CU FT)	=	1528.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.4
PRESSURE (ATM)	=	5.

**** DESIGN PARAMETERS FOR E4

HEAT EXCHANGE DUTY (BTU/HR)	=	0.4027E 08
HEAT TRANSFER COFF (BTU/SQ FT/DEG F)	=	23.7
TEMPERATURE DIFFERENCE (DEG F)	=	330.
SURFACE AREA (SQ FT)	=	5123.
PRESSURE DROP INSIDE (PSI)	=	2.2
PRESSURE (ATM)	=	5.

**** DESIGN PARAMETERS FOR E6

HEAT EXCHANGE DUTY (BTU/HR)	=	0.1456E 08
HEAT TRANSFER COFF (BTU/SQ FT/DEG F)	=	26.1
TEMPERATURE DIFFERENCE (DEG F)	=	840.
SURFACE AREA (SQ FT)	=	661.
PRESSURE DROP INSIDE (PSI)	=	5.6
PRESSURE DROP OUTSIDE (PSI)	=	3.0
PRESSURE (ATM)	=	5.

**** DESIGN PARAMETERS FOR E7

HEAT EXCHANGE DUTY (BTU/HR)	=	0.2469E 08
HEAT TRANSFER COFF (BTU/SQ FT/DEG F)	=	5.7
TEMPERATURE DIFFERENCE (DEG F)	=	532.
SURFACE AREA (SQ FT)	=	8002.
PRESSURE DROP INSIDE (PSI)	=	0.6
PRESSURE DROP OUTSIDE (PSI)	=	1.4
PRESSURE (ATM)	=	5.

**** DESIGN PARAMETERS FOR E1

FILTER CAPACITY (ACFM)	=	9712.
PRESSURE (ATM)	=	7.

**** DESIGN PARAMETERS FOR T1

DIAMETER (FT)	=	10.7
HEIGHT OR LENGHT (FT)	=	13.9
PACKING (CU FT)	=	974.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE (ATM)	=	7.

**** DESIGN PARAMETERS FOR G1

DIAMETER (FT)	=	11.6
HEIGHT OR LENGHT (FT)	=	17.4
PRESSURE DROP INSIDE (PSI)	=	0.5
PRESSURE (ATM)	=	7.

**** DESIGN PARAMETERS FOR E1

HEAT EXCHANGE DUTY (BTU/HR)	=	0.3856E 08
HEAT TRANSFER COEFF. (BTU/SQ FT/DEG F)	=	27.3
TEMPERATURE DIFFERENCE (DEG F)	=	648.
SURFACE AREA (SQ FT)	=	2171.
PRESSURE DROP INSIDE (PSI)	=	2.5
PRESSURE (ATM)	=	7.

**** DESIGN PARAMETERS FOR F2

DIAMETER(FT)	=	4.3
HEIGHT OR LENGHT(FT)	=	4.3
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR R1

DIAMETER(FT)	=	17.9
HEIGHT OR LENGHT(FT)	=	19.7
CATALYST (TONS)	=	59.
NUMBER OF BEDS	=	2
PRESSURE DROP INSIDE (PSI)	=	0.9
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR E5

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3817E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	28.2
TEMPERATURE DIFFERENCE(DEG F)	=	549.
SURFACE AREA(SQ FT)	=	2452.
PRESSURE DROP INSIDE (PSI)	=	2.6
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR E2

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3319E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	9.0
TEMPERATURE DIFFERENCE(DEG F)	=	382.
SURFACE AREA(SQ FT)	=	9536.
PRESSURE DROP INSIDE (PSI)	=	0.8
PRESSURE DROP OUTSIDE(PSI)	=	1.4
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR T2

DIAMETER(FT)	=	11.
HEIGHT OR LENGHT(FT)	=	15.
PACKING (CU FT)	=	1269.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.4
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR E4

HEAT EXCHANGE DUTY(BTU/HR)	=	0.4027E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	29.1
TEMPERATURE DIFFERENCE(DEG F)	=	330.
SURFACE AREA(SQ FT)	=	4176.
PRESSURE DROP INSIDE (PSI)	=	2.5
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR E6

HEAT EXCHANGE DUTY(BTU/HR)	=	0.1456E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	30.9
TEMPERATURE DIFFERENCE(DEG F)	=	840.
SURFACE AREA(SQ FT)	=	559.
PRESSURE DROP INSIDE (PSI)	=	6.3
PRESSURE DROP OUTSIDE(PSI)	=	3.0
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR E7

HEAT EXCHANGE DUTY(BTU/HR)	=	0.2507E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	7.2
TEMPERATURE DIFFERENCE(DEG F)	=	533.
SURFACE AREA(SQ FT)	=	6434.
PRESSURE DROP INSIDE (PSI)	=	0.8
PRESSURE DROP OUTSIDE(PSI)	=	1.7
PRESSURE(ATM)	=	7.

**** DESIGN PARAMETERS FOR F1

FILTER CAPACITY(ACFM)	=	7554.
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR T1

DIAMETER(FT)	=	10.2
HEIGHT OR LENGHT(FT)	=	13.3
PACKING (CU FT)	=	856.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR G1

DIAMETER(FT)	=	10.7
HEIGHT OR LENGHT(FT)	=	16.0
PRESSURE DROP INSIDE (PSI)	=	0.5
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR E1

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3856E 08
HEAT TRANSFER COEF(BTU/SQ FT./DEG F)	=	30.9
TEMPERATURE DIFFERENCE(DEG F)	=	648.
SURFACE AREA(SQ FT)	=	1919.
PRESSURE DROP INSIDE (PSI)	=	2.7
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR F2

DIAMETER(FT)	=	3.8
HEIGHT OR LENGHT(FT)	=	3.8
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR R1

DIAMETER(FT)	=	16.8
HEIGHT OR LENGHT(FT)	=	17.9
CATALYST (TONS)	=	46.
NUMBER OF BEDS	=	2
PRESSURE DROP INSIDE (PSI)	=	0.8
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR E5

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3852E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	30.9
TEMPERATURE DIFFERENCE(DEG F)	=	551.
SURFACE AREA(SQ FT)	=	2252.
PRESSURE DROP INSIDE (PSI)	=	2.8
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR E2

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3281E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	10.5
TEMPERATURE DIFFERENCE(DEG F)	=	377.
SURFACE AREA(SQ FT)	=	8206.
PRESSURE DROP INSIDE (PSI)	=	0.9
PRESSURE DROP OUTSIDE(PSI)	=	1.5
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR T2

DIAMETER(FT)	=	11.
HEIGHT OR LENGHT(FT)	=	14.
PACKING (CU FT)	=	1109.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR E4

HEAT EXCHANGE DUTY(BTU/HR)	=	0.4027E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	33.6
TEMPERATURE DIFFERENCE(DEG F)	=	330.
SURFACE AREA(SQ FT)	=	3618.
PRESSURE DROP INSIDE (PSI)	=	2.7
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR E6

HEAT EXCHANGE DUTY(BTU/HR)	=	0.1456E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	30.9
TEMPERATURE DIFFERENCE(DEG F)	=	840.
SURFACE AREA(SQ FT)	=	559.
PRESSURE DROP INSIDE (PSI)	=	6.7
PRESSURE DROP OUTSIDE(PSI)	=	3.0
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR E7

HEAT EXCHANGE DUTY(BTU/HR)	=	0.2544E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	8.4
TEMPERATURE DIFFERENCE(DEG F)	=	535.
SURFACE AREA(SQ FT)	=	5590.
PRESSURE DROP INSIDE (PSI)	=	0.9
PRESSURE DROP OUTSIDE(PSI)	=	1.9
PRESSURE(ATM)	=	9.

**** DESIGN PARAMETERS FOR F1

FILTER CAPACITY (ACFM)	=	5229.
PRESSURE (ATM)	=	13.

**** DESIGN PARAMETERS FOR T1

DIAMETER (FT)	=	9.6
HEIGHT OR LENGTH (FT)	=	12.5
PACKING (CU FT)	=	712.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE (ATM)	=	13.

**** DESIGN PARAMETERS FOR G1

DIAMETER (FT)	=	9.4
HEIGHT OR LENGTH (FT)	=	14.2
PRESSURE DROP INSIDE (PSI)	=	0.5
PRESSURE (ATM)	=	13.

**** DESIGN PARAMETERS FOR E1

HEAT EXCHANGE DUTY (BTU/HR)	=	0.3856E 08
HEAT TRANSFER COEFF (BTU/SQ FT/DEG F)	=	30.9
TEMPERATURE DIFFERENCE (DEG F)	=	648.
SURFACE AREA (SQ FT)	=	1919.
PRESSURE DROP INSIDE (PSI)	=	3.1
PRESSURE (ATM)	=	13.

**** DESIGN PARAMETERS FOR F2

DIAMETER(FT)	=	3.2
HEIGHT OR LENGHT(FT)	=	3.2
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR R1

DIAMETER(FT)	=	15.3
HEIGHT OR LENGHT(FT)	=	15.4
CATALYST (TONS)	=	31.
NUMBER OF BEDS	=	2
PRESSURE DROP INSIDE (PSI)	=	0.6
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR E5

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3922E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	30.9
TEMPERATURE DIFFERENCE(DEG F)	=	554.
SURFACE AREA(SQ FT)	=	2280.
PRESSURE DROP INSIDE (PSI)	=	3.1
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR E2

HEAT EXCHANGE DUTY(BTU/HR)	=	0.3206E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	12.9
TEMPERATURE DIFFERENCE(DEG F)	=	366.
SURFACE AREA(SQ FT)	=	6726.
PRESSURE DROP INSIDE (PSI)	=	1.4
PRESSURE DROP OUTSIDE(PSI)	=	2.4
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR T2

DIAMETER(FT)	=	10.
HEIGHT OR LENGHT(FT)	=	13.
PACKING (CU FT)	=	915.
NUMBER OF BEDS	=	1
PRESSURE DROP INSIDE (PSI)	=	0.3
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR E4

HEAT EXCHANGE DUTY(BTU/HR)	=	0.4027E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	42.0
TEMPERATURE DIFFERENCE(DEG F)	=	330.
SURFACE AREA(SQ FT)	=	2896.
PRESSURE DROP INSIDE (PSI)	=	3.1
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR E6

HEAT EXCHANGE DUTY(BTU/HR)	=	0.1456E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	30.9
TEMPERATURE DIEFFERENCE(DEG F)	=	840.
SURFACE AREA(SQ FT)	=	559.
PRESSURE DROP INSIDE (PSI)	=	7.5
PRESSURE DROP OUTSIDE(PSI)	=	3.0
PRESSURE(ATM)	=	13.

**** DESIGN PARAMETERS FOR E7

HEAT EXCHANGE DUTY(BTU/HR)	=	0.2619E 08
HEAT TRANSFER COFF(BTU/SQ FT/DEG F)	=	10.5
TEMPERATURE DIFFERENCE(DEG F)	=	538.
SURFACE AREA(SQ FT)	=	4585.
PRESSURE DROP INSIDE (PSI)	=	1.0
PRESSURE DROP OUTSIDE(PSI)	=	2.2
PRESSURE(ATM)	=	13.

APPENDIX E - PHYSICAL PROPERTIES

PHYSICAL PROPERTIES

<u>Temperature</u> °F	<u>Cp, Btu/# F</u>				<u>μ, #/ft-hr.</u>			<u>K, Btu</u> <u>hr. Ft² (°F/Ft.)</u>	
	Air	SO ₃	SO ₂	98% H ₂ SO ₄	Air	SO ₂	98% Acid	Air	SO ₂
100				.35			26.4	.0181	.0069
200	.2412	.161	.141	.37	.0521	.00039	8.2		
300				.38					
400	.2446	.174	.161	.40	.0630	.00049		.0225	.0075
500									
600	.2502	.185	.168		.0726	.00059		.0266	.0080
800	.2566	.194	.172		.0813	.00070		.0303	.0085
1000	.2630	.201	.176		.0892	.00081		.0337	.0088
1200	.2687	.204	.18		.0967	.00090		.0369	.0090