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GRAPHICAL CORRELATION OF PRESSURE DROP IN ADIABATIC

FLOW OF COMPRESSIBLE FLUIDS IN PIPES

BY

GUY E. STRAUB

A THESIS SUBMITTED TO THE FACULTY OF THE DEPARTMENT OF CHEMICAL ENGINEERING OF NEWARK COLLEGE OF ENGINEERING

>

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ABSTRACT

Analytic equations for the flow of compressible fluids in conduits with friction are awkward to solve, since the variables involved usually necessitate trial and error solution. A study of the situation indicated that a series of graphs for different values of the specific heat ratio would eliminate the trial and error solution.

The purpose of this paper is to present in graphical form a solution of the equations relating the ratios of the pressure and temperatures in adiabatic flow of compressible fluids to the friction and velocity.

The equations for adiabatic flow of compressible fluids in horizontal ducts of constant cross section have been employed to calculate a series of values of the pressure ratio and the friction at each of several Mach numbers for each of three specific heat ratios. The temperature ratio has also been calculated for each of several mach numbers for each of the three specific heat ratios.

The graphs make it possible to calculate the pressure drop and temperature drop of compressible fluids, in adiabatic flow in horizontal ducts of constant cross section with friction, without resorting to trial and error solution.

The results are presented graphically on pages 9 to 11.

INTRODUCTION

Where the flow is isothermal, general methods of calculation, both algebraic and graphical, have been developed. (2, 4, 5, 6, 7, 8) Equations have been developed for conditions of adiabatic flow, (1, 2, 8) but these are so complex that for most design purposes their use involves lengthy and tedious graphical solutions. There is also available a general method of calculation based on the Thermodynamic properties of the fluid which involves the Fanno lines (3). This method also involves lengthy calculations before any solution can be obtained.

Mechanical Engineering texts have plots relating the pressure ration in adiabatic flow of compressible fluids for several initial Mach numbers - to the friction and velocity (4). The plots appearing in Advance Fluid Dynamics and Fluid Machinery by Binder are representative, but only 4 Mach numbers have been plotted for only one specific heat ration. These graphs cannot be read with any degree of accuracy.

It is the purpose of this paper to present in graphical form a solution of the equations relating the ratios of the pressures and temperatures in adiabatic flow of compressible fluids to the friction and velocity and to offer a simplified method of determining the performance of horizontal ducts handling compressible fluids at high pressure drops. Lapple has studied adiabatic flow of compressible

fluids and has devised a graphical method for finding the rate of discharge from a storage tank through a pipe line. Lapple's graphs are not well adapted to the calculation of the pressure drop between two sections in a line. It is possible to calculate the pressure drop between two sections of a line by means of Lapple's graphs, but a trial and error solution must be used, whereas the graphical method developed here requires no trial.

DISCUSSION

Binder (4) has plotted graphs relating the pressure ratio to the friction and velocity in horizontal ducts of constant cross section for both isothermal and adiabatic flow for several initial Mach numbers for a specific heat ratio of 1.4. The only Mach numbers plotted were M = 0.05, M = 0.10, M = 0.125 and M = 0.20. These graphs were plotted for adiabatic flow using the following equations: $\beta = (\kappa - i) M_i^2 \left[\lambda + (\kappa - i) M_i^2 \right]$ (1) $\frac{f'L}{D} = \frac{\kappa - i}{\kappa B} \left[1 - \left(\frac{R}{r_i} \right)^2 - \frac{R^2}{P_i} \sqrt{\left(\frac{R}{r_i} \right)^2 + B} + \sqrt{1 + B} \right] + \frac{\kappa + i}{\kappa} \log \left[\frac{R}{r_i} + \sqrt{\frac{R}{R} + B} \right]$ (2)

Binder's graph (4) relating the pressure ratio to friction and velocity for adiabatic flow of compressible fluids in horizontal ducts for several Mach numbers indicates that it is possible to construct a series of graphs for different values for the specific heat ratio which would be extremely useful in solving problems of design for compressible fluids. When Binder has related the pressure ratio to the friction factor and velocity, the friction factor (f^1) is the Mechanical Engineering friction factor which is four times the friction factor commonly used by Chemical Engineers. No attempt has been made by Binder to include the temperature ratio on the graph relating the pressure ratio to the friction and velocity.

It was felt that it would be advantageous to include the temperature ratio on a graph where the pressure ratio is related to the friction and velocity. For this reason, the following equations have been chosen to construct graphs for specific heat ratios of K = 1.1, K = 1.4 and K = 1.67. For cases met in practice the specific heat ratio would fail in the chosen range.

If the initial acoustic velocity $C_1 = \sqrt{kg_{c}RT_{i}/M}$ Then $\frac{fL}{O} = -2.30 \frac{(k+1)}{k} \log_{10} \frac{\sqrt{2}}{\sqrt{i}} + \frac{1}{k} \left[\frac{C_{i}^{A}}{\sqrt{i}} + \frac{(k-1)}{2} \right] \left(1 - \frac{\sqrt{2}}{\sqrt{2}}\right) (3)$ $\frac{R_{1}}{R_{i}} = \frac{\sqrt{i}}{\sqrt{2}} \left[1 + \frac{(k-1)V_{i}^{A}}{2C_{i}^{A}} \left(1 - \frac{\sqrt{2}}{\sqrt{2}} \right) \right]$ (4) $\frac{T_{i}}{T_{i}} = \left(\frac{R_{i}}{T_{i}}\right) \left(\frac{\sqrt{2}}{\sqrt{2}}\right)$

Where $\frac{\sqrt{2}}{C_{1}} = \frac{M}{M_{1}}$ and $R_{11} = \frac{M}{4}$ for circular cross section

These formulas relating the pressure ratio to the friction and velocity for adiabatic flow of compressible fluids are shown in the Chemical Engineers' Handbook (2).

Sufficient data were calculated to relate both the pressure and temperature ratios to the friction and velocity. Since V_2 and V_1 always appear as a ratio $(\underline{V_2})$ or its reciprocal, except where $(\overline{V_1})$

the initial velocity (V_1) appears as a ratio with the initial acoustic velocity (C_1) it becomes simple to establish the temperature ratio $(\underline{T_2})$, since the temperature ratio is equal to the pressure $(\underline{T_1})$ ratio multiplied by the velocity ratio.

A curved line for each initial Mach number relating the pressure ratio to the friction can be plotted by solving Equations 3 and 4 simultaneously for a series of values of the velocity ratio from 1.25 to 8.00. Curved lines can be constructed for each temperature ratio by carrying the calculation one step further and multiplying the pressure ratios by velocity ratios for each Mach number in accordance with Equation 5.

SAMPLE CALCULATIONS

The graphs for specific heat ratios of K = 1.1, K = 1.4 and K = 1.67 were constructed by keeping the specific heat ratio constant and solving Equations 3 and 4 simultaneously for a series of values of the velocity ratio from 1.25 to 8.00 for each Mach number. The temperature ratio was then calculated by multiplying the pressure ratio by the velocity ratio for each Mach number:

When k = 1.4 and $\overline{R}_{H} = \frac{D}{4}$ for ducts of circular cross section Assume $M_{1} = \frac{V_{1}}{C_{1}} = 0.150$ and $\frac{V_{2}}{V_{1}} = 2.00$ substituting in Equation 3. $\frac{44fL}{D} = -\frac{2.150}{1.4} \frac{(1.4+1)}{1.4} \log_{10}(2.00) + \frac{1}{1.4} \left[\frac{1}{(15)} + \frac{(1.4-1)}{2} \right] (1-\frac{1}{(27)})$ $\frac{44fL}{D} = 22.729$ $\frac{1}{D} = 5.682$ substituting in Equation 1

substituting in Equation 4.

$$\frac{R}{R} = \frac{1}{2} \left[1 + \frac{(1.4 - 1)(a.5)^2}{2} (1 - 2^2) \right]$$

$$\frac{R}{R} = 0.49325$$

Substituting the assumed value of the velocity ratio and the calculated value of the pressure ratio in Equation 5, the temperature ration can be calculated

$$\frac{I_2}{T_1} = 0,49325 \times 2,00 = 0,9865$$

RESULTS

The results are presented by graphs I, II and III on the next three pages where both the pressure ratio and the temperature ratio are related to the friction and velocity.

The limiting conditions of maximum flow are calculated from

 $\frac{P_2}{P_1} = M^2 \left\{ \left(\frac{k-i}{k+i} \right) \left[1 + \frac{2}{(k-i)M^2} \right] \right\}^2$







SAMPLE PROBLEM

For purposes of comparison, a sample problem has been solved by several methods, including the method developed by the writer.

Air enters a steel pipe 6 inches in diameter at a rate of 2000 ft.³/ minute at 70° F. and 18 psia. Determine the discharge conditions at a point 500 feet from the entrance of the pipe.

I. Method involving trial-and-error solution by means of simultaneous equations.

Using $\binom{P}{2} = .0916$ and $u = 000672 \text{ X} .018 \frac{16. \text{ mass}}{\text{ft. sec.}}$ the Reynolds number at the entrance will be $\mathcal{R}_{\mathcal{C}} = \frac{DVC}{M} = \frac{2000}{60} \frac{16}{2\pi} \frac{.0916}{.006672 \times 018} = 6.4 \times 10^{\circ}$ from the Chemical Engineers Handbook (2), f = .0045, K = 1.4, $C_{I} = \sqrt{Kg_{C}RT_{I}/M} = \sqrt{\frac{1.4 \times 32.17 \times 546 \times 530}{29}} = 1128.13$ The inlet velocity is $V_{I} = \frac{2000}{60}/\frac{\pi}{16} = 169.7$ The initial Mach number is $\mathcal{M}_{I} = \frac{V_{I}}{C_{I}} = \frac{169.7}{1128.13} = 0.150$ then $\frac{fL}{D} = \frac{0045 \times 500}{25} = 4.5$ Using the formulas given in the Chemical Engineers Handbook (2)

to solve the sample problem

$$\frac{FL}{R_{W}} = -\frac{2}{K} \frac{3(K+1)}{K} \log_{10} \frac{V_{L}}{V_{1}} + \frac{1}{K} \left[\frac{c_{1}^{2}}{V_{1}^{2}} + \frac{(K-1)}{L} \right] \left(1 - \frac{V_{1}^{2}}{V_{L}^{2}} \right) (3)$$

$$\frac{R_{1}}{R} = \frac{V_{1}}{V_{2}} \left[1 + \frac{(K-1)V_{1}^{2}}{2c_{1}^{2}} \left(1 - \frac{V_{L}^{2}}{V_{1}^{2}} \right) \right] \qquad (4)$$

Assume $\frac{V_2}{V_1}$ 1.500 and substitute in Equation 3 $\frac{f_L}{R_N} = -\frac{2.3(1.4+1)}{1.4} \log_{10} 1.5 + \frac{1}{1.4} \left[\frac{1}{(1.5)^2} + \frac{(1.4-1)}{2}\right] \left(1 - \frac{1}{(1.5)^4}\right)$ $\frac{f_L}{R_N} = 17.024 \qquad \frac{f_L}{D} = 4.256$ Adjusting $\frac{V_2}{V_1} = 1.564$

From the equation, we obtain:

.

$$\frac{fL}{R_{\rm H}} = 18.04 \qquad \frac{fL}{D} = 4.51$$

This checks the value 4.5 given above.

Then substituting $\underline{V_2} = 1.564$ in Equation 4

$$\frac{R_{i}}{R_{i}} = \frac{1}{1.564} \left[1 + \frac{(1.4-1)(1.5)^{2}}{2} \left(1 - (1.564)^{2} \right)^{2} \right]$$

$$\frac{R_{i}}{R_{i}} = 0.635$$

$$R_{i} = 0.635 R_{i} = 0.635 \times 18 = 11.43 P_{5} n_{2}$$

$$\frac{T_{i}}{T_{i}} = \frac{R_{i}}{R_{i}} \left(\frac{V_{i}}{V_{i}} \right) = .635 \times 4564 = 0.993$$

$$T_{i} = 0.993 \times T_{i} = 0.993 \times 530 = 526.3^{\circ}R$$

II. Writer's Method

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By means of the graphs presented in this paper, the solution is obtained much more simply.

From the previous calculation, we know that f = .0045 $M_1 = 0.150$ K = 1.4and $\frac{fL}{D} = 4.5$

Using the graph, Figure 2, for a specific heat ratio of K = 1.4, we can read directly the pressure ratio and temperature ratio corresponding to M1 = 0.150 and $\frac{fL}{D}$ = 4.5

Reading directly from the graph, we obtain the following values: $\frac{P_2}{P_1} = 0.635, \frac{T_2}{T_1} = 0.993$ Therefore - P_2 = .635 X 18 = 11.43 psia $T_2 = .993 X 530 = 526.3^{\circ}R$

III. Lapple's Method

Working the sample problem presented previously by Lapple's graphical method (5) we must assume that the air enters the pipe from a chamber through a frictionless nozzle. This means of solving the problem must be done by trial-and-error solution

From previous calculations we know that $V_1 = 169.7 ft/sec.$

then

$$G_1 = V_1 C = 169.7 \times .0916 = 15.54 \ 16/ft^2 \ s=c,$$

 $P_1 = 18 \times 144 = 2592 \ 16/ft^2$
 $N = \frac{fL}{R_H} = \frac{4fL}{D} = 4.5 \times 4 = 18.0$

From the Chemical Engineers Handbook (2)

$$G_{Ci} = P_0 \sqrt{\frac{3c}{2.718}} = P_0 \sqrt{\frac{32.17\times17}{2.718\times1545}}$$

reading from Lapple's graph for K = 1.4 for adiabatic flow, assume

$$\frac{P_{i}}{P_{o}} = 0.96 , \quad N = 0$$
We find $\frac{G_{i}}{G_{ci}} = 0.40 , \quad \frac{T_{i}}{T_{o}} = .9905$
therefore $G_{ci} = \frac{15.54}{.4} = 38.85 , \quad T_{o} = \frac{530}{.9905} = 535 \text{ R}$

Substituting these values in the formula

$$\begin{aligned}
B &= \frac{G c_{i}}{\sqrt{\frac{3c}{N}}} = \frac{38.85}{\sqrt{\frac{32.17\times29}{2.718\times1545\times535}}} \\
P_{o} &= 1913 \quad 16/5t^{2} \\
\text{then } \frac{P_{i}}{P_{o}} = \frac{2592}{1913} = 1.35 \\
\text{This does not check the assumed value of } \frac{P_{i}}{P_{o}} = 0.96
\end{aligned}$$

Adjusting
$$\frac{P_{i}}{P_{0}} = 0.975$$
, $N = 0$
Therefore $\frac{G_{i}}{G_{ci}} = 0.288$, $\frac{T_{i}}{T_{0}} = 0.9935$
 $G_{ci} = \frac{15.54}{0.288} = 53.95$, $T_{0} = \frac{530}{.9935} = 533$

Substituting in the formula

$$P_{o} = \frac{Gci}{\sqrt{\frac{g_{o} M}{2.718 R T_{o}}}}$$

We obtain Po=2657

then
$$\frac{P_1}{R_0} = \frac{2592}{2657} = 0.975$$

This checks the assumed value for $\frac{P}{R} = 0.975$

for
$$N = 18^{\circ}$$
, $\frac{P_1}{P_0} = 0.62^{\circ}$, $\frac{T_1}{T_0} = 0.988^{\circ}$
then $P_2 = \frac{\frac{P_1}{P_0}}{\frac{P_1}{P_0}} \times P_1 = \frac{0.62^{\circ}}{0.975^{\circ}} \times 18 = 11.45^{\circ} \text{ ps/a}.$

and
$$T_{2} = \frac{T_{2}}{T_{0}} \times T_{1} = \frac{0.988}{0.7935} \times 530 = 527^{\circ}R$$

IV. Method Based on Fanno Tables

$$\underbrace{ACTUAL}_{Duct} \xrightarrow{KE}[ct][c] = S _ Duct \xrightarrow{K}[ct][c] = Duct \xrightarrow{K}[ct][c] = S _ Duct \xrightarrow{K}[ct][c] = S _$$

solving the sample problem by means of the Fanno Tables (3):

The friction length is

$$f'\frac{(x_1 - x_1)}{D} = 0.018 \times \frac{500}{1/2} = 18$$

The inlet velocity is $V_{i} = \frac{2000}{60} / \frac{\pi}{16} = 169.7$

at the inlet

$$M_{i} = \frac{V}{2} = \frac{169.7}{49.03\sqrt{530}} = 0.150$$

The Fanno Tables have a column headed $\frac{fL}{D}$ from which we read the

value for
$$f'(x^{*-x_{i}})$$
 at M=0,150 of 27.932

By reference to the figure above we know that

$$f'(\frac{x^{*}-x_{1}}{0} - f'(\frac{x_{2}-x_{1}}{0}) = f'(\frac{x^{*}-x_{2}}{0})$$

$$27,932 - 18 = 9.932$$

or

For the value of $f(x^*-x_1)=9.932$ we find from the Fanno Tables that $M_2 = .2347$.

The Fanno Tables have columns headed \underline{P} and \underline{T} corresponding to \underline{P}^* \underline{T}^*

various values of M. Therefore, for M1 and M2 we can read

$$\frac{P_{FF}}{FF} = 4.6441 \qquad \frac{P_{FF}}{FF} = 7.2866 \\ \frac{T_{FF}}{FF} = 1.1869 \qquad \frac{T_{FF}}{FF} = 1.1946$$

The exit conditions are then found:

$$\begin{array}{l}
P_{1} = P_{1} \times \frac{P_{1}}{P_{\pi}} \times \frac{P_{\pi}}{P_{i}} = \frac{18 \times 4.6441}{7.2866} = 11.47 \, \text{ps}_{1a} \\
T_{k} = T_{i} \times \frac{T_{k}}{T^{*}} \times \frac{T^{*}}{T_{i}} = \frac{530 \times 1.1869}{1.1946} = 526.5 \, ^{\circ}R \\
\end{array}$$
This method does not involve trial-and-error, but requires the

use of a special set of tables.

CONCLUSIONS

There should be little doubt that certain design problems concerning fluid flow of compressible fluids can be solved more simply through the use of the three graphs presented in this paper relating the pressure ratio and temperature ratio to the friction and velocity. While it is possible to interpolate between graphs for specific heat ratios other than those presented in this paper (K = 1.1, K = 1.4 and K = 1.67), valuable additions could be made to this problem by constructing graphs for specific heat ratios of 1.2, 1.3, 1.5 and 1.6.

With the graphs presented in this paper the maximum velocity limitation is about 340 feet per second for air at 70°F. At greater velocities the Mach number would exceed 0.300. A further contribution could be made by constructing graphs for Mach numbers in excess of 0.300. Graphs with Mach numbers higher than 0.300 would increase the possible maximum velocity range.

D		diameter, ft.
f	1	friction factor
fl		4 £
gc		32.1740 (lb. mass) (ft.)/ (lb. force) (sec.2)
G	1	C=V/~ mass velocity, lb. mass/(sec.) (sq. ft.)
Gci	8	190Po/2718No = Po 190 M/2718 RTO
K		Cp/c, ratio of specific heat at constant pressure to that a constant volume
L	8	Length, ft.
М	Ξ	Molecular weight, lb. mass/lb. mole
	=	viscosity, lb. mass/ (ft.) (sec.)
N	11	$\frac{fL}{R_{H}}$ = frictional resistance in velocity heads
P ₁ ,	P ₂	= pressure at inlet and outlet, respectively. <u>lb. force</u> sq. inch
R	=	gas constant 1546 ftlb. force/ (lbmole) (^O R)
T1,	T2	= absolute temperature at inlet and outlet, respectively, ^O R
V _l ,	V 2	= linear velocity at inlet and outlet, respectively, ft./sec.
Ml	Ξ	$\frac{V_{l}}{C_{l}}$ initial Mach number
M2		outlet Mach number
Cl	Ξ	initial acoustic velocity
P	1	density of fluid, 1b. mass/cu. ft.
Po	*	pressure in theoretical chamber, lb./sq. ft.
To		temperature in theoretical chamber, ^{OR}

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APPENDIX

CALCULATED DATA

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		al de aler a car		
V2 V1	ŝ	<u>fl</u> D	P2 P1	T2 T1
1.10		15.734	0.90907	0.9999
1.25		32.625	0.79994	0.9999
1.50	¥	50.318	0.66656	0.9998
1.75		60.965	0.57128	0.9998
2.00		67.861	0.49981	0.9996
2.25		72.574	0.44421	0.9995
2.50		75.937	0.39973	0.9992
2.75		78.416	0.36333	0.9991
3.00		80.294	0.33300	0.9990
3.25		81.751	0.30732	0.9987
3.50		82.901	0.28531	0.9986
3.75		83.825	0.26623	0.9983
4.00		84.578	0.24953	0.9981
4.25		85.197	0.23479	0.9979
4.50		85.964	0.22168	
4.75		86.148	0,20995	
5.00		86.517	0.19940	
5.25		86.831	0.18984	
5.50		87.103	0.18115	
5.75		87.337	0.17321	
6.00		87.541	0.16593	
8.00		88.508	0.12402	
			MVK = 0.052	

K = 1.1

.

M = 0.05

.

K = 1.1 M = 0.075

$\frac{\overline{v}_2}{\overline{v}_1}$	$\frac{\mathrm{fL}}{\mathrm{D}}$	P2 P1	$rac{\mathrm{T}_2}{\mathrm{T}_1}$
1.10	6.969	0.90904	0.9999
1.25	14.443	0.79987	0,9999
1.50	22.260	0.66643	0.9996
1.75	26.952	0.57109	0.9993
2.00	29.981	0.49957	0.9991
2.25	32.045	0.44393	0.9988
2.50	33.512	0,39940	0,9985
2.75	34.589	0.36296	0,9983
3.00	35.401	0.33258	0.9977
3.25	36.027	0,30686	0.9974
3,50	36.519	0,28481	0,9968
3.75	36.911	0,26568	0,9964
4.00	37.229	0.24894	0,9958
4.25	37.488	0.23416	0,9954
4.50	37.703	0,22101	0.9945
4.75	37.881	0,20924	0.9937
5.00	38,032	0.19865	0.9933
5.25	38,159	0,18905	0,9928
5.50	38,267	0.18032	0.9917
5.75	38.359	0.17234	0.9907
6.00	38,439	0.16502	0.9901
8.00	38.793	0.12279	0.9823
		MVK = 0.0786	6

K = 1.1

M = 0.100

$\frac{v_2}{v_1}$	<u>fL</u> D	P2 P1	$\frac{T_2}{T_1}$
1.10	3.651	0.90899	0.9999
1.25	8.080	0.79977	0.9998
1.50	12.439	0.66625	0.9995
1.75	15.047	0.57083	0.9989
2.00	16.724	0.49925	0.9985
2.25	17.860	0.44354	0.9979
2.50	18.664	0.39895	0.9975
2.75	19.250	0.36244	0.9966
3.00	19.688	0.33200	0.9960
3.25	20.024	0,30622	0.9952
3.50	20.285	0.28410	0.9944
3.75	20,492	0.26492	0.9934
4.00	20,657	0.24812	0,9925
4.25	20.790	0,23328	0.9914
4,50	20.899	0,22008	0,9904
4.75	20,988	0.20825	0,9892
5.00	21.062	0,19760	0.9880
5,25	21.123	0.18794	0.9867
5,50	21.174	0.17915	0,9853
5.75	21.217	0.17112	0.9839
6.00	21.253	0.16375	0,9825
8.00	21.392	0.12106	0.9685
		$M\sqrt{K} = 0.1048$	8

K = 1.1

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M = 0.125

V2 V1	<u>fL</u> D	P2 P1	$\frac{T_2}{T_1}$
1.10	2.481	0.90894	0.9998
1.25	5.134	0.79964	0.9995
1.50	7.894	0.66601	0,9990
1.75	9•537	0.57050	0.9984
2.00	10.587	0.49882	0,9976
2,25	11.295	0.44303	0,9968
2.50	11.791	0.39835	0.9958
2,75	12.150	0.36177	0,9949
3,00	12.417	0,33125	0.9937
3.25	12.617	0,30539	0,9925
3,50	12,771	0.28320	0.9912
3.75	12.892	0.26394	0.9898
4.00	12,986	0.24707	0.9883
4.25	13.061	0.23215	0.9866
4.50	13.121	0.21888	0.9850
4.75	13.169	0.20697	0.9831
5.00	13.207	0.19625	0.9813
5.25	13.238	0.18652	0.9792
5.50	13.263	0.17766	
5.75	13.283	0.16955	
6.00	13.298	0.16210	
7.00	13.332	0.13750	
		MVK = 0.1311	

	K = 1.1	M = 0,150	
$\frac{v_2}{v_1}$	$\frac{fL}{D}$	$\frac{P_2}{P_1}$	$\frac{T_2}{T_1}$
1.10	1.710	0,90886	0,9998
1.25	3.534	0 .7 9949	0.9994
1.50	5.425	0.66572	0,9986
1.75	6.544	0.57010	0,9977
2.00	7.254	0,49831	0,9966
2.25	7.728	0.44241	0,9954
2.50	8.058	0.39763	0,9940
2.75	8,293	0,36095	0 .99 26
3.00	8.465	0,33033	0,9910
3.25	8。593	0.30438	0,9892
3,50	8,690	0,28209	0,9873
3.75	8,763	0,26274	0,9853
4.00	8,845	0,24578	0,9831
4.25	8,863	0.23077	0,9808
4.50	8.896	0,21740	0,9783
4.75	8,921	0.20541	
5.00	8.941	0.19460	
5.25	8.955	0.18478	
5,50	8.965	0.17583	
5.75	8.975	0.16763	
6.00	8.977	0.16010	
		M√K ■ 0.15732	

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N= 0.175

V2 Vl	fL D	P2 F1	T <u>2</u> T1
1.10	1.245	0,90000	0.9957
1.25	2,569	0 . 79931	0,9991
1.50	3. 936	0.66539	0,9931
1.75	4.729	0.56962	0,7963
2.00	5.244	0.49770	0,9954
2.25	5.578	C.44167	0,9938
2,50	5.807	0,39678	0.9920
2.75	5.970	0.35998	0,9900
3.00	6.083	0.32925	0,9378
3.25	6,142	0.30318	0,9854
3.50	6.229	0.28079	0,9828
3.75	6.274	0.26133	0.9798
4.00	6.307	0.24425	
4.25	6.331	0.22914	
4.50	6.348	0.21567	
4.75	6,260	0.20357	
5 .0 0	6,363	0.19265	
		M √K = 0.18354	

K = 1.1 M = 0.200 $\frac{v_2}{v_1}$ $\frac{\text{fL}}{\text{D}}$ $\frac{F_2}{P_1}$ $\frac{T_2}{T_1}$ 1,10 0.943 0.90871 0,9996 1.25 1.943 0.79910 0,9988 1.50 2,970 0.66500 0.9975 1.75 3.567 0.56907 0.9959 2.00 3.939 0.49700 0.9940 2.25 4.182 0.44083 0.9918 2.50 4.345 0,39580 0,9895 2.75 4.458 0.35886 0,9870 3,00 4.537 0,32800 0,9840 3,25 4.592 0,30180 0,9809 3.50 4,631 0,27928 0.9776 3.75 4**.6**58 0,25970 4.00 4.676 0.24250 4.25 4.688 0.22726 4.50 4.695 0.21366

 $M\sqrt{K} = 0.20976$

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v ₂ v ₁	<u>fl</u> D	$\frac{P_2}{P_1}$	$\frac{T_2}{T_1}$
1.10	0.736	0.90801	0.999 <u>)</u>
1.25	1.514	0.79886	0.9986
1.50	2,307	0.66455	0.9969
1.75	2.764	0.56844	0.9947
2.00	3.045	0.49620	0.9924
2.25	3,225	0.43987	0,9898
2.50	3.344	0.39468	0.9868
2.75	3.423	0,35759	0,9834
3.00	3.477	0.32658	0,9798
3,25	3.513	0.30024	
3,50	3.536	0.27757	
3.75	3.551	0,25784	
4.00	3.559	0,24050	

MVK = 0.23598

	K ≅ 1.1	M = 0.250	
$\frac{v_2}{v_1}$	<u>fL</u> D	$\frac{P_2}{P_1}$	$\frac{T_2}{T_1}$
1.10	0.588	0.90849	0.9994
1.25	1.207	0.79859	0,9983
1.50	1.833	0.66406	0.9962
1.75	2.190	0.56776	0.9937
2,00	2.405	0.49531	0.9906
2.25	2.541	0,43880	0.9873
2,50	2.627	0.39343	0,9835
2.75	2.683	0.35617	0,9796
3,00	2.719	0,32500	
3,25	2.740	0,29849	
3,50	2.753	0.27566	
		MVK 0.26220	
	K = 1.1	MVR 0.26220 M = 0.275	
1.10	K = 1.1 0.476	MVR 0.26220 M ≈ 0.275 0.90838	0.9992
1.10 1.25	K = 1.1 0.476 0.980	MVR 0.26220 M = 0.275 0.90838 0.79829	0.9992 0.9979
1.10 1.25 1.50	K = 1.1 0.476 0.980 1.483	MVR 0.26220 M = 0.275 0.90838 0.79829 0.66351	0.9992 0.9979 0.9953
1.10 1.25 1.50 1.75	K = 1.1 0.476 0.980 1.483 1.765	MVK 0.26220 M ≈ 0.275 0.90838 0.79829 0.66351 0.56697	0.9992 0.9979 0.9953 0.9923
1.10 1.25 1.50 1.75 2.00	K = 1.1 0.476 0.980 1.483 1.765 1.932	MVK 0.26220 M = 0.275 0.90838 0.79829 0.66351 0.56697 0.49432	0.99992 0.9979 0.9953 0.9923 0.9886
1.10 1.25 1.50 1.75 2.00 2.25	K = 1.1 0.476 0.980 1.483 1.765 1.932 2.034	MVR 0.26220 M = 0.275 0.90838 0.79829 0.66351 0.56697 0.49432 0.43761	0.99992 0.9979 0.9953 0.9923 0.9886 0.9846
1.10 1.25 1.50 1.75 2.00 2.25 2.50	K = 1.1 0.476 0.980 1.483 1.765 1.932 2.034 2.097	MVK 0.26220 M ≈ 0.275 0.90838 0.79829 0.66351 0.56697 0.49432 0.43761 0.39205	0.9992 0.9979 0.9953 0.9923 0.9886 0.9846 0.9803
1.10 1.25 1.50 1.75 2.00 2.25 2.50 2.75	K = 1.1 0.476 0.980 1.483 1.765 1.932 2.034 2.097 2.135	MVK 0.26220 M = 0.275 0.90838 0.79829 0.66351 0.56697 0.49432 0.49432 0.43761 0.39205 0.35461	0.99992 0.9979 0.9953 0.9923 0.9886 0.9846 0.9803 0.9752
1.10 1.25 1.50 1.75 2.00 2.25 2.50 2.75 3.00	K = 1.1 0.476 0.980 1.483 1.765 1.932 2.034 2.097 2.135 2.158	MVK 0.26220 M ≈ 0.275 0.90838 0.79829 0.66351 0.56697 0.49432 0.43761 0.39205 0.35461 0.32325	0.9992 0.9979 0.9953 0.9923 0.9886 0.9846 0.9803 0.9752
1.10 1.25 1.50 1.75 2.00 2.25 2.50 2.75 3.00 3.25	K = 1.1 0.476 0.980 1.483 1.765 1.932 2.034 2.097 2.135 2.158 2.194	MVK 0.26220 M ≈ 0.275 0.90838 0.79829 0.66351 0.66351 0.56697 0.49432 0.49432 0.43761 0.39205 0.35461 0.32325 0.29656	0.9992 0.9979 0.9953 0.9923 0.9886 0.9846 0.9803 0.9752

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	K = 1.1	М	= 0.300
$\frac{v_2}{v_1}$	<u>fL</u> D	P2 F1	T2 T1
1.10	0.395	0.90823	0.9990
1.25	0.807	0.79797	0.9975
1.50	1.216	0.66291	0.9944
1.75	1.442	0.56612	0,9907
2.00	1.572	0.49325	0.9865
2.25	1.649	0.43631	0.9817
2,50	1.694	0.39055	0,9765
2,75	1.719	0.35289	
3.00	1.731	0,32133	

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MVK _ 0.31464

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	К = 1.4	M = 0.050	
$\frac{V_2}{V_1}$	$\frac{fL}{D}$	P2 P1	T ₂ T1
1.10	12,362	0,90899	0,9999
1.25	25.636	0.79977	0 .9 998
1,50	39 . 535	0,66625	0,9995
1.75	47.396	0.57083	0,9989
2,00	53.310	0.49925	0,9985
2.25	57.009	0,44354	0.9979
2,50	59.647	0,39895	0.9975
2.75	61,591	0.36244	0,9966
3.00	63,063	0,33200	0,9960
3.25	64.0204	0,30622	0.9952
3.50	65,104	0,28410	0.9944
3/75	65.827	0.26492	0,9934
4.00	66.415	0,24812	0,9925
4.25	66.399	0.23328	0,9915
4.50	67.302	0,22008	0,9905
4.75	67/640	0,20325	0,9894
5.00	67.923	0,19760	0,9880
5.25	68.172	0.18794	0,9865
5.50	68.383	0.17915	0,9856
5.75	68,565	0.17112	0 ,9 838
6.00	68 ,723	0.16375	0.9825
8.00	69,457	0.12063	0,9648
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 $M \sqrt{K} = 0.05916$

K = 1.4

M = 0.075

V ₂ V1	$\frac{\text{fL}}{D}$	P2 P1	T ₂ T1
1.10	5.475	0,90337	0,9998
1.25	11.348	0.79949	0,9994
1.50	17.486	0,66572	0,9986
1.75	21,168	0,57010	0.9977
2,00	23.543	0.49831	0,9966
2,25	25,161	0,44241	0.9954
2,50	26.309	0.39763	0.9940
2.75	27.150	0.36095	0,9928
3.00	27.734	0.33033	0.9910
3.25	28,273	0.30438	0,9893
3.50	28,655	0,23209	0,9874
3,75	28,960	0,26274	0,9851
4,00	29.207	0,24578	0,9831
4.25	29.407	0.23077	0,9809
4.50	29.573	0,21740	0,9783
4.75	29.711	0,20541	
5,00	29.826	0,19460	
5,25	29.923	0,18478	
5,50	30,006	0,17583	
5.75	30.077	0.16763	
6,00	30.137	0.16010	
ප . 00	30.395	0.11614	
		M √ K = 0,08874	, +

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	K = 1.4	M = C.100	
<u>v₂ v1</u>	$\frac{\texttt{fL}}{\texttt{D}}$	P2 P1	T ₂ T1
1.10	3.065	0,90870	0.9996
1.25	6.347	0.79910	0 .9 989
1.50	9.768	0,66500	0.9975
1.75	11.812	0,56907	0.9959
2.00	1 3.125	0.49700	0.9940
2.25	14.013	0,44083	0.9918
2,50	14.0640	0,39580	0.9895
2.75	15.096	0.35896	0.9870
3.00	15.437	0,32800	0,9840
3.25	15.697	0,30180	0.9809
3,50	15.898	0,27928	0,9776
3.75	16.057	0,25970	0,9739
4.00	16.184	0,24250	0,9700
4.25	16 . 285	0,22726	0,9658
4.50	16,368	0,21366	0,9615
4.75	16.435	0,20144	0,9568
5,00	16.491	0.19040	
5.25	16.536	0.18035	
5.50	16.574	0,17118	

16,606

16.632

16.695

5.75

6,00

7.00

0,16276

0.15500

0.12915

M √ K = 0,11832

K = 1.4

M = 0.125

v ₂ v ₁	<u>fL</u> D	P ₂ P1	$\frac{T_2}{T_1}$
1.10	1.949	0.90849	0.9993
1.25	4.032	0.79859	0.9983
1.50	6.196	0.66406	0.9962
1.75	7.482	0.56774	0.9935
2.00	8.303	0.49531	0,9906
2.25	8.854	0.43880	0.9873
2.50	9.239	0.39343	0.9835
2.75	9.517	0.35617	0.9796
3.00	9.720	0.32500	0.9750
3.25	9.876	0.29849	0.9701
3.50	9.994	0.27566	0.9648
3.75	10.085	0.25578	0.9592
4.00	10.156	0.23828	
4.25	10.212	0.22274	
4.50	10.256	0.20885	
4.75	10.291	0.19634	
5.00	10.319	0.18500	
5.25	10.340	0.17466	
5.50	10.357	0.16519	
5.75	10.371	0.15648	
6.00	10.381	0.14843	
		MVK = 0.14790	

	K 🖫 1.4	M = 0.150	
$\frac{v_2}{v_1}$	$rac{\mathrm{fL}}{\mathrm{D}}$	<u>P2</u> P1	$\frac{\mathbb{T}_2}{\mathbb{T}_1}$
1.10	1.343	0,90823	0.9991
1.25	2.775	0.79797	0.9975
1.50	4.256	0.66291	0.9944
1.75	5.130	0,56612	0.9907
2,00	5.682	0.49325	0.9865
2.25	6.051	0.43631	0.9817
2.50	6.305	0.39055	0.9765
2.75	6.486	0.35289	0.9704
3.00	6.617	0,32133	0.9640
3.25	6.714	0.29445	0.9569
3.50	6.786	0.27125	
3.75	6.841	0.25099	
4.00	6.882	0.23312	
4.25	6.913	0.21722	
4.50	6,936	0.20297	
4.75	6,953	0.19009	
5.00	6,966	0.17840	
		MVK = 0.17748	

K = 1.4

M = 0.175

fL D	$\frac{P_2}{P_1}$	$\frac{T_2}{T_1}$
0.977	0.90792	0.9987
2.017	0.79724	0,9965
3.086	0.66156	0.9924
3.711	0.56420	0.9874
4.104	0.49081	0.9816
4.361	0.43338	0,9752
4.536	0.38713	0.9678
4.659	0.34901	0,9598
4.745	0.31700	
4.807	0.28967	
4.852	0,26602	
4.884	0.24533	
4.907	0.22703	
4.923	0.21070	
	$\frac{fL}{D}$ 0.977 2.017 3.086 3.711 4.104 4.361 4.361 4.536 4.659 4.745 4.807 4.852 4.884 4.907 4.923	$\frac{fT}{D}$ $\frac{P_2}{P_1}$ 0.9770.907922.0170.797243.0860.661563.7110.564204.1040.490814.3610.433384.5360.387134.6590.349014.7450.317004.8070.289674.8520.266024.8840.245334.9070.227034.9230.21070

MVR : 0.20706

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11 = 0,200

$\frac{v_2}{v_1}$	fL D	P ₂ P ₁	Т <u>2</u> Т <u>1</u>
1.10	0.740	0.90756	0,9983
1.25	1.525	0.79640	0.9955
1.50	2.327	0.66000	0,9900
1.75	2.791	0.56200	0,9835
2.00	3.078	0,48800	0,9760
2.25	3.266	0.43000	0,9675
2,50	3.388	0.38320	0,9530
2.75	3.473	0.34454	
3.00	3.530	0.31200	
3.25	3.570	0.28415	
3.50	3.597	0.26000	
3.75	3.615	0,23880	

M√K = 0.23664

	K = 1.4.	M = 0.225	
$\frac{v_2}{v_1}$	fL D	$\frac{P_2}{P_1}$	$\frac{T_2}{T_1}$
1. 10	0.578	0.90715	0,9979
1.25	1.187	0.79544	0,9943
1.50	1.806	0,65822	0.9873
1.75	2.160	0.55949	0.9791
2,00	2.376	0 .48431	
2.25	2,513	0,42616	
2,50	2.601	0.37873	
2.75	2.659	0.33947	
3.00	2.697	0.30633	
3,25	2,722	0.27790	
		M√K = 0,26622	an a
	K = 1.4	M = 0.250	
1.10	0.461	0,90670	0.9974
1.25	0.946	0.79437	0,9930
1.50	1.434	0.65625	0,9845
1.75	1.709	0.55669	0.9742
2,00	1.873	0.48125	0.9625
2.25	1.975	0.42187	0.9492
2.50	2,038	0.37375	
2.75	2,078	0.33380	
3.00	2,101	0,30000	
		p	

 $M\sqrt{K} = 0.29580$

	R = Sal	Mr 0.275	
₹72 ¹⁷ 1	f L D	F2 P1	
1.10	0,375	0,90620	0,9963
1.25	0,768	0.79319	0.9915
1.50	1.153	0.65406	0.9810
1.75	1,375	0,55360	0,9683
2.00	1,501	0.47731	
2.25	1,577	0.41713	
2,50	1.622	0,36823	
2.75	1.647	0.32754	
		$M\sqrt{K} = 0.32538$	
LALIER MERINALIS CONSIGNATION OF MICH.	an a	***************************************	n na han san san san san san san san san san s
K	= 1.4	M = 0,300	
r 1.02	.⇔1.4	M ≈ 0,300 0,97968	0 .9993
k 1.02 1.05	i≂1.4	M ≈ 0,300 0.97968 0.95062	0 . 9993 0. 9982
k 1.02 1.05 1.10	0,366	M ≈ 0,300 0,97968 0,95062 0,90565	0 .9993 0.9982 0.9962
K 1.02 1.05 1.10 1.25	0,366 0,632	M ≈ 0,300 0.97968 0.95062 0.90565 0.79190	0.9993 0.9982 0.9962 0.9399
K 1.02 1.05 1.10 1.25 1.50	0,366 0,632 0,949	M ≈ 0.300 0.97968 0.95062 0.90565 0.79190 0.65166	0.9993 0.9962 0.9962 0.9399 0.9776
k 1.02 1.05 1.10 1.25 1.50	0.366 0.632 0.949 1.121	M ≈ 0.300 0.97968 0.95062 0.90565 0.79190 0.65166 0.55021	0.9993 0.9982 0.9962 0.9399 0.9776 0.9639
K 1.02 1.05 1.10 1.25 1.50 1.75 2.00	0,366 0,632 0,949 1,121 1,218	M ≈ 0,300 0.97968 0.95062 0.90565 0.79190 0.65166 0.55021 0,47300	0.9993 0.9962 0.9962 0.9399 0.9776 0.9629 0.9460
K 1.02 1.05 1.10 1.25 1.50 1.75 2.00 2.25	0.366 0.632 0.949 1.121 1.218 1.218 1.274	M = 0.300 0.97968 0.95062 0.90565 0.79190 0.65166 0.55021 0.47300 0.41194	0.9993 0.9962 0.9962 0.9399 0.9776 0.9629 0.9460
K 1.02 1.05 1.10 1.25 1.50 1.75 2.00 2.25 2.50	0.366 0.632 0.949 1.121 1.218 1.274 1.305	M = 0.300 0.97968 0.95062 0.90565 0.79190 0.65166 0.55021 0.47300 0.41194 0.36220	0.9993 0.9962 0.9399 0.9776 0.9639 0.9460

	K = 1.67	M ∞ 0.050	
<u>V2</u> V1	$\frac{fL}{D}$	<u>P2</u> P1	<u>T2</u> T1
1.10	10.363	0 。90 893	0 .999 8
1.25	21.486	0.79962	0.9995
1.50	33,132	0.66592	0,9989
1.75	40.137	0.57044	0.9982
2,00	44.671	0,49874	0,9975
2.25	47.768	0.44293	0.9965
2,50	50.026	0.39824	0,9955
2.75	51.601	0.36163	0.9944
3.00	52.882	0.33110	0,9933
3.25	53.785	0,30522	0.9919
3.50	54.537	0.28302	0.9905
3.75	55.140	0.26374	0,9888
4.00	55.631	0.24685	0.9874
4.25	56.034	0.23193	0,9856
4.50	56.370	0.21863	0.9837
4.75	56.651	0.20672	0,9818
5.00	56.890	0.19598	0 . 9799
5.25	57.093	0.18623	0.9775
5.50	57.268	0.17736	0.9757
5.75	57.419	0.16924	
6.00	57.550	0.16178	
8.00	58.164	0.11841	
		$M\sqrt{K} = 0.06462$	

K =	1.67	
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M = 0.075

$\frac{v_2}{v_1}$	<u>fl</u> D	P2 P1	$\frac{T_2}{T_1}$
1.10	4.589	0.90873	0.9996
1.25	9.510	0.79915	0,9989
1.50	14.651	0.66509	0.9976
1.75	1 7 .733	0.56920	0.9961
2.00	19.721	0.49717	0.9943
2.25	21.073	0.44104	0.9923
2.50	22.031	0,39604	0.9901
2.75	22.734	0.35913	0.9876
3.00	23.262	0.32830	0.9849
3.25	23.668	0.30214	0.9819
3.50	23.987	0.27965	0.9788
3.75	24.239	0.26010	0.9754
4.00	24.444	0.24293	0.9717
4.25	24.609	0.22772	0.9678
4.50	24.746	0.21416	0.9637
4.75	24.859	0.20197	0.9594
5.00	24.954	0.19095	
5.25	25.034	0.18094	
5:50	25.101	0.17179	
5.75	25.159 ^{°;}	0,16340	
6.00	25.207	0.15567	
8.00	25.417	0.11016	
		MVX = 0.09692	

V2 V1	<u>fl</u> D	$\frac{P_2}{P_1}$	<u>T2</u> T1
1.10	2.569	0.90845	0.9993
1.25	5.318	0.79849	0.9981
1.50	8.183	0,66387	0.9958
1.75	9.892	0,56748	0.9931
2.00	10.988	0,49497	0.9899
2.25	11.729	0.43839	0.9864
2.50	12.251	0,39296	0.9824
2.75	12.630	0.35564	-0.9780
3.00	12.912	0.32440	0.9732
3.25	13.127	0.29783	0.9679
3.50	13.294	0,27494	0.9623
3.75	13.424	0.25499	0.9562
4.00	13.528	0.23743	
4.25	13.611	0.22184	
4.50	13.678	0.20789	
4.75	13.732	0.19531	
5.00	13.777	0.18392	
5.25	13.813	0.17352	
5.50	13.843	0.16400	
5.75	13.867	0,15523	
6.00	13.888	0,14712	
		NA = 0.12923	l

K = 1.67

M = 0.100

• •	K = 1.0%	M 2 0.	(21
$\frac{v_2}{v_1}$	<u>fl</u> D	P2 P1	$\frac{\mathbf{T}_2}{\mathbf{T}_1}$
1.10	1.633	0.90809	0.9989
1.25	·3.378	0.79764	0.9970
1.50	5.189	0.66230	0.993 5
1.75	6.263	0.56525	0.9892
2.00	6.946	0,49214	0.9843
2.25	7.405	0,43499	`_0.9787
2.50	7.724	0,38900	0.9725
2.75	7 •9 53	0.35114	0.9656
3.00	8.122	0,31937	0.9581
3.25	8.249	0.29229	
3.50	8.345	0,26888	
3.75	8.418	0.24843	
4.00	8.476	0.23037	
4.25	8.520	0.21427	
4.50	8.555	0,19983	
4.75	8.581	0.18676	
5.00	8.603	0.17487	
5.25	8.619	0.16399	
		NVE = 0.161	54

K = 1.67

M = 0.125

	K = 1.67	K = 0.15	50
v ₂ v ₁	<u>fl</u> D	P2 P1	T2 T1
1.05		0.95164	0.9992
1.10	1,125	0.90764	0.9984
1.25	2.324	0.79660	0.9958
1.50	3.562	0.66038	0.9906
1.75	4.291	0.56254	0.9844
2.00	4.751	0.48869	0.9774
2.25	5.056	0.43083	0.9694
2.50	5.265	0.38417	0.9604
2.75	5.413	0.34564	0.9505
3.00	5.520	0.31323	
3.25	5.398	0.28551	
3.50	5.656	0.26148	
3.75	5.699	0.24041	
4.00	5.731	0.22173	
4.25	5.755	0.20503	

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NVE = 0.19385

	K = 1.67	М = О.	175
<u>v</u> 2 v1	tl D	P2 P1	T2 T1
1.05		0.95138	0.9989
1.10	0.819	0.90714	0.9979
1.25	1.689	0.79538	0.9942
1.50	2.582	0.65811	0.9872
1.75	3.102	0,55933	0.9788
2.00	3.427	0.48461	0.9692
2.25	3.639	0.42592	0.9583
2.50	3.782	0.37845	
2.75	3.881	0.33915	
3.00	3.951	0.30597	
3.25	4.000	0.27750	,
3.50	4.035	0.25273	
3.75	4.099	0.23092	

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NVK : 0.22615

	K = 1.67	M = 0.2	200
v ₂ v ₁	<u>fl</u> D	P2 P1	$\frac{T_2}{T_1}$
1.10	0.620	0.90654	0.9972
1.25	1.276	0.79397	0.9925
1.50	1.945	0.65550	0.9833
1.75	2.331	0.55563	0.9724
2.00	2.568	0,47990	0.9598
2.25	2.720	0,42025	
2.50	2.820	0.37186	
2.75	2.887	0.33165	
3.00	2.933	0.29760	
3.25	2.963	0.26826	
		MVE = 0.25846	
an a	K = 1.67	M = 0.4	225
1.10	0.484	0.90585	0.9965
1.25	0.993	0.79236	0.9905
1.50	1.509	0.65253	0.9788
1.75	1.802	0.55144	
2.00	1.979	0.47456	
2.25	2.089	0.41382	
2.50	2.160	0.36438	
2.75	2.206	0.32316	
	· · · · ·	MVK = 0.29077	
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		·	

	K = 1.67	M = 0.250	
<u>v</u> 2 v1		P2 P1	$\frac{\mathbf{r}_2}{\mathbf{r}_1}$
1.05		0.95033	0.9978
1.10	0.386	0.90510	0.9956
1.25	1.791	0,79057	0.9882
1.50	1.197	0,64921	0.9738
1.75	1.423	0.54675	0,9568
2.00	1.557	0,46859	
2.25	1.639	0,40664	
2.50	1.688	0,35603	
		MVK = 0.32308	
(* <u>Understandigenschunder</u>	K = 1,67	M = 0.275	
1.10	0.304	0.90425	0 .99 47
1.25	0,642	0,78859	0.9858
1.50	0,966	0.64555	0.968 4
1.75	1,163	0.541.57	
2.00	1,246	0.46199	
2.25	1.305	0.39870	
		MVX = 0.35538	

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	K = 1.6 7	M = 0.300	
<u>v₂</u> ⊽1		P2 P1	$\frac{\mathtt{T}_2}{\mathtt{T}_1}$
1.01		0,98950	0.9994
1.02		0,97919	0.9988
1.10	0.259	0.90334	°°0,9937
1.25	0,528	0.78643	0.9830
1.50	0.790	0,64154	0.9623
1.75	0,931	0,53589	0.9378
2.00	1.008	0,45477	
2.25	1.051	0,39000	
		MVX = 0.38769	

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